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Supporting information for article:

Testing the tools for revealing and characterizing the iodine–iodine halogen bond in crystals

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Table S1 Interatomic distances (\AA) after optimization of atomic positions in crystals and calculated values at bond critical points (a. u.) for I–I and I...I interactions

Refcode in CSD	Name of compound	Sort of bonding	$d_{\text{I,I}}$	$\rho(\mathbf{r}_{\text{bcp}})$	$\nabla^2\rho(\mathbf{r}_{\text{bcp}})$	$P(\mathbf{r}_{\text{bcp}})$	$\eta(\mathbf{r}_{\text{bcp}})$
DOWMAF P $\bar{1}$	$\text{C}_{12}\text{H}_{11}\text{INS}^+\cdot\text{I}_3^-$ 1-(Iodomethyl)-1,2-dihydro[1,3]thiazolo[3,2-a]quinolin-10-i um tri-iodide (Bartashevich <i>et.al.</i> , 2014)	[I–I–I] $^-$	2.973	0.044	0.039	-0.218	0.491
			3.029	0.040	0.041	-0.259	0.436
			4.149	0.006	0.016	-0.623	0.036
		I...I	4.798	0.002	0.006	-0.605	0.013
			4.389	0.005	0.012	-0.612	0.027
DOWMEJ P $\bar{1}$	$(\text{C}_{12}\text{H}_{11}\text{INS}^+\cdot\text{I}_3^-)\cdot\text{I}_2$ 1-(Iodomethyl)-1,2-dihydro[1,3]thiazolo[3,2-a]quinolin-10-i um tri-iodide diiodine (Bartashevich <i>et.al.</i> , 2014)	I–I	2.856	0.055	0.026	-0.115	0.621
		[I–I–I] $^-$	2.936	0.046	0.038	-0.207	0.505
			3.016	0.039	0.043	-0.270	0.421
		I...I	3.441	0.019	0.037	-0.477	0.157
			3.741	0.012	0.027	-0.589	0.069
			4.241	0.005	0.013	-0.650	0.026
			4.628	0.003	0.008	-0.626	0.016
			4.954	0.003	0.006	-0.624	0.013
			4.020	0.008	0.022	-0.637	0.045
DULZOZ01	$\text{C}_4\text{H}_{12}\text{N}^+\cdot\text{I}_5^-$	I–I	2.861	0.054	0.029	-0.134	0.601
C2/c	Tetramethylammonium pentaiodide (Filguieras <i>et.al.</i> , 2001)	I...I	3.221	0.027	0.044	-0.407	0.250
EJUPOQ	$\text{C}_{12}\text{H}_8\text{ClINO}^+\cdot\text{I}_3^-$						
P2 $_1$ /n	(3E)-8-chloro-3-iodomethylidene-2,3-dihydro-1,4-oxazino[2,3,4-ij]quinolin-4-i um triiodide (Bartashevich <i>et.al.</i> , 2016)	I–I	2.923	0.049	0.033	-0.170	0.554
HAFLAC	$\text{C}_7\text{H}_{10}\text{N}_2\text{OS}$ I ₂	I–I	2.901	0.050	0.031	-0.155	0.573
P2 $_1$ /c	(6-Propyl-thiouracil) diiodine (Antoniadis <i>et.al.</i> , 2003)	I...I	5.096	0.002	0.004	-0.581	0.009
			3.969	0.007	0.019	-0.626	0.041
I ₂	I ₂	I–I	2.796	0.061	0.021	-0.085	0.655
Cmca	(Bertolotti <i>et.al.</i> , 2014)	I...I	3.470	0.018	0.037	-0.510	0.135
			4.357	0.005	0.012	-0.667	0.021
IVOVOG	$\text{C}_{22}\text{H}_{17}\text{I}_2\text{N}_2\text{S}_2^+\cdot\text{I}_3^-$						
C2/c	(E)-8-((2,3-diido-4-(quinolin-8-ylthio)but-2-en-1-yl)thio)quinolin-1-i um triiodide (Bartashevich <i>et.al.</i> , 2016)	I–I	2.995	0.042	0.040	-0.238	0.465
		I...I	3.722	0.011	0.027	-0.606	0.066

NULBUR	C ₈ H ₆ N ₂ I ₂						
Pbca	Quinoxaline diiodine (Bailey <i>et.al.</i> , 1997)	I-I	2.808	0.061	0.016	-0.066	0.676
NUTSOL	C ₁₂ H ₃₀ N ₂ ²⁺ 2I ⁻ I ₂						
C2/m	Hexane-1,6-bis(trimethylammonium) iodine diiodide (Abate <i>et. al.</i> , 2010)	I-I	2.881	0.053	0.028	-0.132	0.606
NUTSUR	C ₁₂ H ₃₀ N ₂ ²⁺ 2I ₅ ⁻	I-I	2.880	0.052	0.031	-0.146	0.583
C2/m	Hexane-1,6-bis(trimethylammonium) bis(pentaiodide) (Abate <i>et. al.</i> , 2010)	I-I	2.862	0.054	0.029	-0.136	0.594
			3.155	0.031	0.044	-0.357	0.306
			3.143	0.032	0.044	-0.343	0.323
			4.487	0.004	0.010	-0.643	0.019
		I...I	4.592	0.003	0.009	-0.639	0.016
			4.571	0.003	0.009	-0.625	0.018
			4.518	0.003	0.008	-0.632	0.017
			4.528	0.004	0.010	-0.611	0.022
QIRZEY	C ₁₃ H ₁₃ INS ⁺ I ⁻						
P2 ₁ /c	3-(Iodomethyl)-3-methyl-2,3-dihydro[1,4]thiazo[3,2-a]quinolin-4-i-um iodide (Batalov <i>et.al.</i> , 2013)	I...I	3.700	0.012	0.027	-0.568	0.079
			4.207	0.006	0.015	-0.624	0.034
			4.458	0.004	0.011	-0.655	0.021
SUFLUC	C ₁₂ H ₁₁ NS ⁺ I ⁻						
P2 ₁ /n	1-(Iodomethyl)-1,2-dihydro[1,3]thiazolo[3,2-a]quinolin-10-ium iodide (Slepukhin, 2015)	I...I	3.670	0.013	0.029	-0.556	0.086
TIJLUU	C ₁₇ H ₁₂ N ₂ O ₄ S ₅ I ₂						
P ¹	4,5-bis(2-nitrobenzylsulfanyl)-1,3-dithiole-2-diiodothione (Skabara <i>et.al.</i> , 2007)	I-I	2.853	0.055	0.028	-0.128	0.602
YUSYUH	C ₅ H ₇ N ₂ ⁺ 0.5I ₂ I ⁻	I-I	2.879	0.053	0.029	-0.138	0.594
P ¹	1-Methylpyrazin-1-ium iodide diiodine (Nelyubina <i>et.al.</i> , 2010)	I...I	3.386	0.022	0.038	-0.442	0.190
ZOJPAR	C ₁₂ H ₉ INO ⁺ I ₃ ⁻ 0.5I ₂	I-I	2.830	0.058	0.022	-0.097	0.644
P ¹	(E)-3-(iodomethylene)-2,3-dihydro-[1,4]oxazino[2,3,4-ij]quinolin-4-ium triiodide diiodine	[I-I-I] ⁻	2.940	0.047	0.038	-0.204	0.508
			3.027	0.040	0.043	-0.267	0.425
		I...I	3.504	0.017	0.034	-0.497	0.135

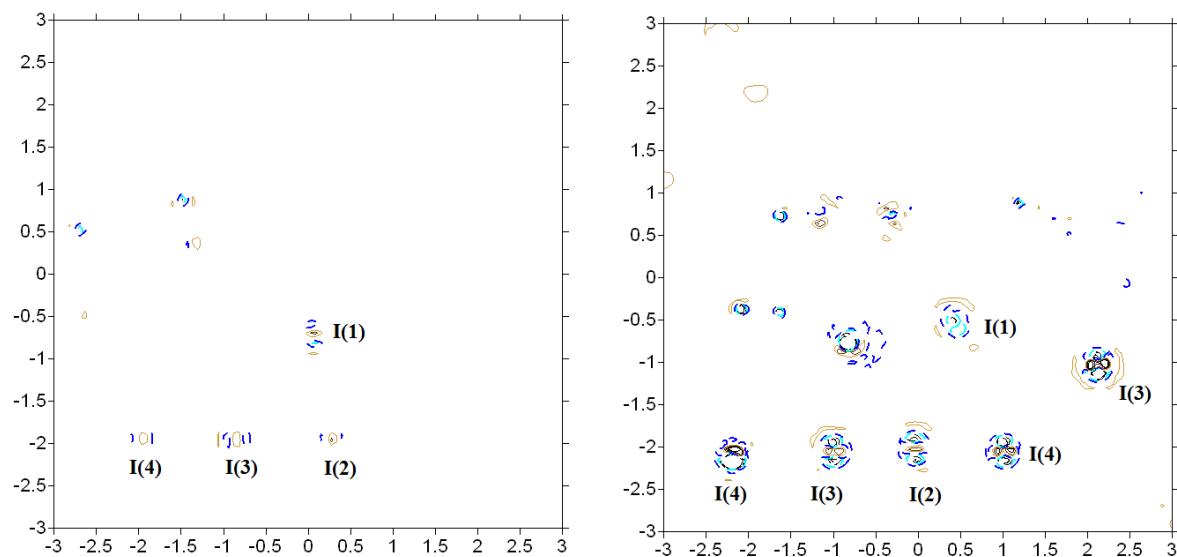
(Batalov <i>et.al.</i> , 2014)	3.604	0.013	0.032	-0.601	0.080
	3.991	0.008	0.021	-0.629	0.046
	4.048	0.007	0.019	-0.645	0.037
	4.150	0.007	0.017	-0.642	0.034
	4.277	0.005	0.014	-0.669	0.025
	4.526	0.004	0.009	-0.650	0.017
	4.782	0.003	0.006	-0.600	0.014
	4.488	0.004	0.011	-0.620	0.022
	4.489	0.004	0.010	-0.624	0.021

Table S2 Experimental and calculated Raman data for structures with bound diiodine molecule and triiodide ions

Refcode in CSD	Calculated wave number, $\nu(I_2)$, cm^{-1}	Experimental wave number $\nu(I_2)$, cm^{-1}	References of Raman data
DOWMAF	113 $\nu_{\text{sym}}(I_3^-)$	116 $\nu_{\text{sym}}(I_3^-)$	(Yushina <i>et.al.</i> , 2015)
DOWMEJ	114 $\nu_{\text{sym}}(I_3^-)$	108 $\nu_{\text{sym}}(I_3^-)$	(Yushina <i>et.al.</i> , 2016)
EJUPOQ	165	165	
IVOVOG	148	149	(Bartashevich <i>et.al.</i> , 2016)
ZOJPAR	115 $\nu_{\text{sym}}(I_3^-)$	113 $\nu_{\text{sym}}(I_3^-)$	(Yushina <i>et.al.</i> , 2015)
NUTSOL	168	172	
DULZOZ01	158	161	(Abate <i>et. al.</i> , 2010)
HAFLAC	165	155	(Nour <i>et.al.</i> , 1986)
TIJLUU	141	151	(Antoniadis <i>et. al.</i> , 2003)
NULBUR	162	160	(Skabara <i>et.al.</i> , 2007)
I ₂	182	182	(Bailey <i>et.al.</i> , 1997)
	179	180	(Congeduti <i>et.al.</i> , 2000)

Table S3 The disagreement R-factors for the multipolar refinements

DOWMAF	DOWMEJ	EJUPOQ	IVOVOG	ZOJPAR	DULZOZ01	QIRZEY	SUFLUC
0.0011	0.0009	0.0020	0.0013	0.0010	0.0009	0.0017	0.0015

Figure S1 Typical residual density maps corresponding to the structures on the Fig. 3, yellow-brown lines are positive values, blue are negative ones, the step is 0.05 e/Å³, zero level is omitted: a) polyiodide fragment in crystal DOWMEJ; b) polyiodide fragment in crystal EJUPOQ**Table S4** The minima (top line) and maxima of the one-electron potential P(**r**) (bottom line), Å, represented using relativistic analytical wave functions

DOWMEJ	Halogen-bond donor	0.039	0.129	0.361	1.099
I ₃ ⁻ ...I-I		0.022	0.077	0.236	0.687
3.431 Å	Halogen-bond acceptor	0.039	0.129	0.361	1.117
		0.022	0.077	0.236	0.679
ZOJPAR	Halogen-bond donor	0.040	0.128	0.363	1.110
I ₃ ⁻ ...I-I		0.018	0.075	0.239	0.686
3.530 Å	Halogen-bond acceptor	0.040	0.128	0.363	1.119
		0.018	0.075	0.239	0.677
SUFLUC	Halogen-bond donor	0.041	0.127	0.363	1.130
I ⁻ ...I-C		0.018	0.078	0.239	0.685
3.670 Å	Halogen-bond acceptor	0.041	0.127	0.363	1.121
		0.018	0.078	0.239	0.680

QIRZEY	Halogen-bond donor	0.037	0.129	0.361	1.135
I ⁻ ...I-C		0.019	0.079	0.236	0.685
3.700 Å	Halogen-bond acceptor	0.037	0.129	0.361	1.121
		0.019	0.079	0.236	0.681
EJUPOQ	Halogen-bond donor	0.038	0.129	0.359	1.125
I ₃ ⁻ ...I-C		0.020	0.077	0.239	0.690
3.826 Å	Halogen-bond acceptor	0.038	0.129	0.359	1.140
		0.019	0.077	0.239	0.685
EJUPOQ	I(1) Type I interaction	0.041	0.127	0.360	1.137
I ₃ ⁻ ...I-C		0.020	0.076	0.239	0.680
4.054 Å	I(3) Type I interaction	0.041	0.127	0.360	1.147
		0.020	0.076	0.239	0.685
					2.045

Table S5 The minima (top line) and maxima of the one-electron potential P(**r**) (bottom line), Å, represented using Clementi and Roetti wave functions

DOWMEJ	Halogen-bond donor	0.039	0.133	0.365	1.117
I ₃ ⁻ ...I-I		0.022	0.082	0.241	0.700
3.431 Å	Halogen-bond acceptor	0.039	0.133	0.365	1.121
		0.022	0.082	0.241	0.691
ZOJPAR	Halogen-bond donor	0.040	0.133	0.367	1.128
I ₃ ⁻ ...I-I		0.022	0.080	0.239	0.699
3.530 Å	Halogen-bond acceptor	0.040	0.133	0.367	1.123
		0.022	0.080	0.239	0.690
SUFLUC	Halogen-bond donor	0.041	0.133	0.367	1.144
I ⁻ ...I-C		0.018	0.078	0.239	0.703
3.670 Å	Halogen-bond acceptor	0.041	0.133	0.367	1.125
		0.018	0.078	0.239	0.694
QIRZEY	Halogen-bond donor	0.042	0.134	0.366	1.149
I ⁻ ...I-C		0.019	0.079	0.241	0.704
3.700 Å	Halogen-bond acceptor	0.042	0.134	0.366	1.130
		0.019	0.079	0.241	0.695
EJUPOQ	Halogen-bond donor	0.038	0.134	0.364	1.140
I ₃ ⁻ ...I-C		0.019	0.081	0.239	0.704
3.826 Å	Halogen-bond acceptor	0.038	0.134	0.364	1.145
		0.019	0.081	0.239	0.694
EJUPOQ	I(1) Type I interaction	0.041	0.132	0.365	1.142
I ₃ ⁻ ...I-C		0.020	0.081	0.244	0.695
4.054 Å	I(3) Type I interaction	0.041	0.132	0.365	1.147
		0.020	0.081	0.239	0.690
					2.045