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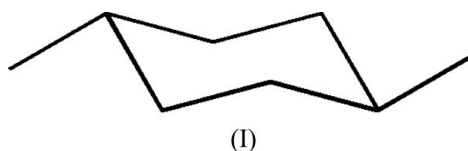
Key indicators

Single-crystal X-ray study
 $T = 150$ K
Mean $\sigma(\text{C}-\text{C}) = 0.001$ Å
 R factor = 0.044
 wR factor = 0.091
Data-to-parameter ratio = 66.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.*trans*-1,4-Dimethylcyclohexane

trans-1,4-Dimethylcyclohexane, C_8H_{16} , was studied as part of a project to develop a computer-controlled low-temperature crystal-growing device. The liquid crystallizes with the molecule lying on a centre of symmetry, leading to $Z' = \frac{1}{2}$.

Comment

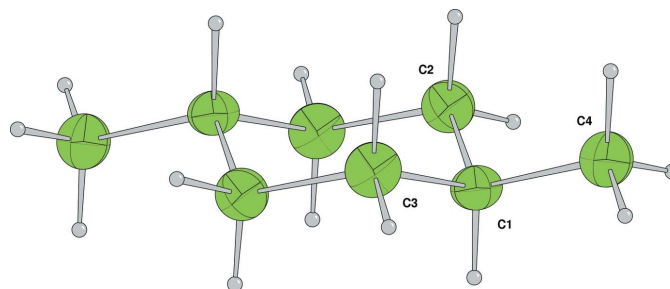
trans-1,4-Dimethylcyclohexane, (I) (Fig. 1), was one of eight alkylcyclohexanes whose thermodynamic properties were published in 1949 (Huffman *et al.*, 1949). That work reported a melting point of 236.217 K and showed no evidence for phase changes in the range down to liquid-nitrogen temperatures.



The sample used for the present study was one of several sealed in 0.2 mm Lindeman tubes for preliminary work carried out in 1979. Data had been collected at that time on a Stoe Weissenberg diffractometer and the structure solved, but it was not of a publishable quality (Courseille *et al.*, 1979).

The sample solidified spontaneously to a polycrystalline mass on flash cooling to 150 K. The temperature was then raised to 230 K and the sample zone-refined into a single-crystal using tandem computer-controlled heating elements. The temperature was then slowly reduced to 150 K for data collection.

The structure of (I) consists of molecules lying on centres of symmetry. This leads to the packing consisting of columns of molecules lying along the b axis (Fig. 2), with the mean plane of the molecule inclined at 145° to that axis (Fig. 3).

**Figure 1**

The structure of (I), with displacement ellipsoids drawn at the 50% probability level and H atoms shown as spheres of arbitrary radii. Unlabelled atoms are related to labelled atoms by a centre of symmetry.

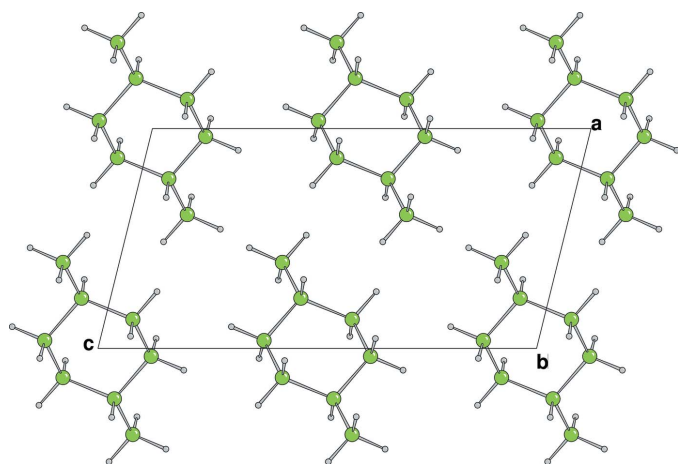


Figure 2
A projection of (I) along the *b* axis. The low specific gravity (0.98 Mg m^{-3}) is explained by the open texture of the structure.

Experimental

The material was used as supplied by Aldrich Chemical Company Inc. in 1979.

Crystal data

C_8H_{16}	$D_x = 0.978 \text{ Mg m}^{-3}$
$M_r = 112.22$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 882 reflections
$a = 6.0843 (2) \text{ \AA}$	$\theta = 1-27^\circ$
$b = 5.4818 (2) \text{ \AA}$	$\mu = 0.05 \text{ mm}^{-1}$
$c = 11.7629 (5) \text{ \AA}$	$T = 150 \text{ K}$
$\beta = 103.8918 (18)^\circ$	Cylinder, colourless
$V = 380.85 (2) \text{ \AA}^3$	1.00 (length) \times 0.20 mm (diameter)
$Z = 2$	

Data collection

Nonius KappaCCD area-detector diffractometer	862 independent reflections
ω scans	861 reflections with $I > 3\sigma(I)$
Absorption correction: multi-scan (<i>DENZO/SCALEPACK</i> ; Otwinowski & Minor, 1997)	$R_{\text{int}} = 0.027$
$T_{\text{min}} = 0.758$, $T_{\text{max}} = 1.000$	$\theta_{\text{max}} = 27.6^\circ$
4141 measured reflections	$h = -7 \rightarrow 7$
	$k = -7 \rightarrow 6$
	$l = -15 \rightarrow 15$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F^2) + (0.03P)^2 + 0.07P]$
$R[F^2 > 2\sigma(F^2)] = 0.044$	where $P = (\max(F_o^2, 0) + 2F_c^2)/3$
$wR(F^2) = 0.091$	$(\Delta\sigma)_{\text{max}} < 0.001$
$S = 0.99$	$\Delta\rho_{\text{max}} = 0.19 \text{ e \AA}^{-3}$
861 reflections	$\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$
13 parameters	
H-atom parameters constrained	

The H atoms were all located in a difference map and then repositioned geometrically. The H atoms were initially refined with soft restraints on the bond lengths and angles to regularize their

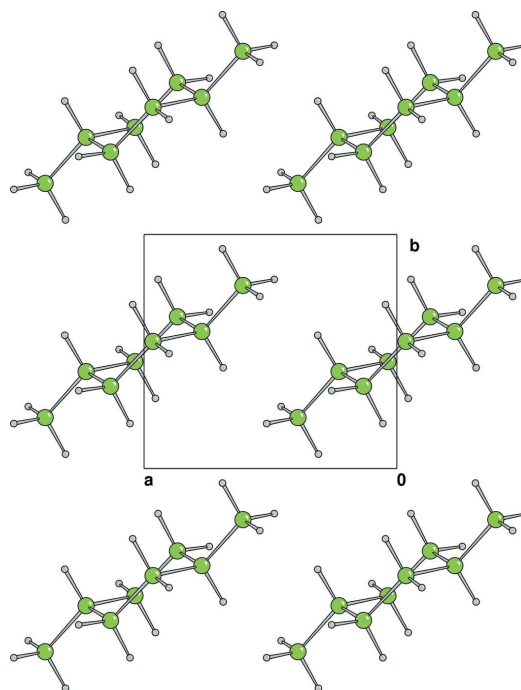


Figure 3
A projection of (I) along the *c* axis, showing the molecular stacks parallel to the *b* axis.

geometry, with C–H distances in the range $0.93\text{--}0.98 \text{ \AA}$, and on the displacement parameters, with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5$ times U_{eq} of the parent atom, after which their positions were refined with riding constraints.

Data collection: *COLLECT* (Nonius, 2001); cell refinement: *DENZO/SCALEPACK*; data reduction: *DENZO/SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1994); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *CAMERON* (Watkin *et al.*, 1996); software used to prepare material for publication: *CRYSTALS*.

References

- Altomare, A., Cascarano, G., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). *J. Appl. Cryst.* **27**, 435.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, C. K. & Watkin, D. J. (2003). *J. Appl. Cryst.* **36**, 1487.
- Courseille, D., Hospital, M., Leroy, F. & Watkin, D. (1979). 5th European Crystallographic Meeting, Copenhagen, Denmark, p. 285.
- Huffman, H. M., Todd, S. S. & Oliver, G. D. (1949). *J. Am. Chem. Soc.* **71**, 584–592.
- Nonius (2001). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Watkin, D. J., Prout, C. K. & Pearce, L. J. (1996). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.

supporting information

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trans-1,4-Dimethylcyclohexane

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Trans-1,4 dimethyl cyclohexane*Crystal data*

C_8H_{16}

$M_r = 112.22$

Monoclinic, $P2_1/c$

$a = 6.0843$ (2) Å

$b = 5.4818$ (2) Å

$c = 11.7629$ (5) Å

$\beta = 103.8918$ (18)°

$V = 380.85$ (2) Å³

$Z = 2$

$F(000) = 128$

$D_x = 0.978$ Mg m⁻³

Melting point: 236.217 K

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 882 reflections

$\theta = 1\text{--}27^\circ$

$\mu = 0.05$ mm⁻¹

$T = 150$ K

Cylinder, colourless

1.00 × 0.20 (radius) mm

Data collection

Nonius KappaCCD area-detector
diffractometer

Graphite monochromator

ω scans

Absorption correction: multi-scan

(DENZO/SCALEPACK; Otwinowski & Minor,
1997)

$T_{\min} = 0.758$, $T_{\max} = 1.000$

4141 measured reflections

862 independent reflections

861 reflections with $I > 3\sigma(I)$

$R_{\text{int}} = 0.027$

$\theta_{\max} = 27.6^\circ$, $\theta_{\min} = 3.5^\circ$

$h = -7 \rightarrow 7$

$k = -7 \rightarrow 6$

$l = -15 \rightarrow 15$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.044$

$wR(F^2) = 0.091$

$S = 0.99$

861 reflections

13 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F^2) + (0.03P)^2 + 0.07P]$

where $P = (\max(F_o^2, 0) + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.000356$

$\Delta\rho_{\max} = 0.19$ e Å⁻³

$\Delta\rho_{\min} = -0.17$ e Å⁻³

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.22877 (14)	0.08561 (16)	0.56580 (7)	0.0289
C2	0.13233 (14)	0.14820 (17)	0.43681 (7)	0.0323
C3	0.03582 (15)	0.04284 (17)	0.62583 (8)	0.0331
C4	0.39112 (15)	0.28192 (18)	0.62800 (9)	0.0389

H11	0.3132	-0.0696	0.5701	0.0332*
H21	0.0554	0.3079	0.4325	0.0369*
H22	0.2582	0.1675	0.3975	0.0380*
H31	-0.0450	0.2001	0.6270	0.0402*
H32	0.0987	-0.0087	0.7079	0.0404*
H41	0.4578	0.2351	0.7116	0.0558*
H42	0.3105	0.4379	0.6274	0.0552*
H43	0.5149	0.3077	0.5892	0.0549*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0289	0.0268	0.0303	0.0042	0.0054	0.0008
C2	0.0333	0.0345	0.0302	0.0002	0.0098	0.0043
C3	0.0370	0.0383	0.0243	0.0006	0.0078	-0.0002
C4	0.0356	0.0360	0.0418	-0.0014	0.0028	-0.0019

Geometric parameters (Å, °)

C1—C2	1.5286 (12)	C2—H22	0.991
C1—C3	1.5264 (12)	C3—H31	0.994
C1—C4	1.5238 (12)	C3—H32	0.990
C1—H11	0.989	C4—H41	1.003
C2—C3 ⁱ	1.5246 (12)	C4—H42	0.985
C2—H21	0.988	C4—H43	0.980
C2—C1—C3	109.82 (7)	C1—C3—C2 ⁱ	112.54 (7)
C2—C1—C4	111.60 (7)	C1—C3—H31	108.1
C3—C1—C4	111.61 (7)	C2 ⁱ —C3—H31	109.0
C2—C1—H11	108.4	C1—C3—H32	109.5
C3—C1—H11	107.0	C2 ⁱ —C3—H32	109.4
C4—C1—H11	108.3	H31—C3—H32	108.1
C1—C2—C3 ⁱ	112.25 (7)	C1—C4—H41	110.7
C1—C2—H21	108.3	C1—C4—H42	110.2
C3 ⁱ —C2—H21	109.2	H41—C4—H42	108.2
C1—C2—H22	109.4	C1—C4—H43	111.1
C3 ⁱ —C2—H22	110.6	H41—C4—H43	108.6
H21—C2—H22	106.9	H42—C4—H43	108.0

Symmetry code: (i) $-x, -y, -z+1$.