

Bis(4*H*-1,2,4-triazol-3-yl)disulfane

Dongsheng Liu,* Yaping Xu, Xinfu Li, Shaoming Ying and Wentong Chen

Department of Chemistry, JingGangShan University, Ji'an, Jiangxi 343009, People's Republic of China

Correspondence e-mail: liudongsheng@jgsu.edu.cn

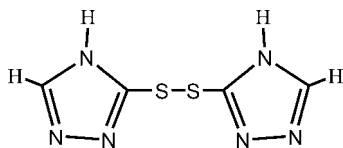
Received 30 October 2007; accepted 4 December 2007

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{N}-\text{C}) = 0.002$ Å; R factor = 0.029; wR factor = 0.080; data-to-parameter ratio = 13.6.

The title compound, $\text{C}_4\text{H}_4\text{N}_6\text{S}_2$, was synthesized by the reaction of 3-mercapto-1*H*-1,2,4-triazole with sodium hydroxide in ethanol. The molecule possesses a crystallographically imposed twofold axis. Intermolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds link the molecules into chains along the c axis.

Related literature

For related literature, see: De Luca (2006); Di Santo, Tafi, Costi, Botta, Artico, Corelli, Forte, Caporuscio, Angiolella & Palamara (2005); Fringuelli *et al.* (2005); Menozzi *et al.* (2004).



Experimental

Crystal data

$\text{C}_4\text{H}_4\text{N}_6\text{S}_2$

$M_r = 200.25$

Monoclinic, $C2/c$

$a = 14.052$ (3) Å

$b = 6.4044$ (13) Å

$c = 9.928$ (2) Å

$\beta = 122.18$ (3)°

$V = 756.2$ (4) Å³

$Z = 4$

Mo $K\alpha$ radiation

$\mu = 0.65$ mm⁻¹

$T = 293$ (2) K

$0.12 \times 0.09 \times 0.06$ mm

Data collection

Rigaku R-Axis RAPID IP

diffractometer

Absorption correction: multi-scan

(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.932$, $T_{\max} = 0.962$

3518 measured reflections

859 independent reflections

742 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.029$

$wR(F^2) = 0.081$

$S = 1.09$

859 reflections

63 parameters

All H-atom parameters refined

$\Delta\rho_{\max} = 0.25$ e Å⁻³

$\Delta\rho_{\min} = -0.22$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{N3}^i$	0.89 (2)	1.97 (2)	2.8617 (19)	174.9 (19)

Symmetry code: (i) $x, -y + 1, z + \frac{1}{2}$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO* (Rigaku, 1998); data reduction: *RAPID-AUTO* (Rigaku, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL/PC* (Sheldrick, 1993); software used to prepare material for publication: *SHELXL97* (Sheldrick, 1997).

This work was supported financially by the Natural Science Project of Jinggangshan University (JZ0731).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RZ2177).

References

- De Luca, L. (2006). *Curr. Med. Chem.* **13**, 1–23.
- Di Santo, R., Tafi, A., Costi, R., Botta, M., Artico, M., Corelli, F., Forte, M., Caporuscio, F., Angiolella, L. & Palamara, A. T. (2005). *J. Med. Chem.* **48**, 5140–5153.
- Fringuelli, R., Milanese, L. & Schiaffella, F. (2005). *Mini-Rev. Med. Chem.* **5**, 1061–1073.
- Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
- Menozzi, G., Merello, L., Fossa, P., Schenone, S., Ranise, A., Mosti, L., Bondavalli, F., Loddo, R., Murgioni, C., Mascia, V., La Colla, P. & Tamburini, E. (2004). *Bioorg. Med. Chem.* **12**, 5465–5483.
- Rigaku (1998). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.
- Sheldrick, G. M. (1993). *SHELXTL/PC*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.

supporting information

Acta Cryst. (2008). E64, o247 [https://doi.org/10.1107/S1600536807065452]

Bis(4*H*-1,2,4-triazol-3-yl)disulfane

Dongsheng Liu, Yaping Xu, Xinfu Li, Shaoming Ying and Wentong Chen

S1. Comment

It is well known that derivatives of pyrazole, imidazole, triazole, tetrazole and indole exhibit extensive biological activities (De Luca, 2006; Fringuelli *et al.*, 2005; Di Santo *et al.*, 2005; Menozzi *et al.*, 2004). In a search for more efficient antibacterial medicines, we have synthesized a new azole derivative and its crystal structure is reported here.

In the molecule of the title compound (Fig. 1), which possesses a crystallographically imposed twofold axis, the torsion angles C1—S1—S1ⁱ—C1ⁱ and S1ⁱ—S1—C1—N3 are 83.69 (8) and -93.69 (13)°, respectively [symmetry code: (i) -*x*, *y*, -0.5 - *z*]. The dihedral angle formed by the triazole rings is 21.80 (7)°. In the crystal structure (Fig. 2 and 3), molecules are linked by N—H···N hydrogen bonding interactions (Table 1) to form stepped chains running parallel to the *c* axis.

S2. Experimental

3-Mercapto-1*H*-1,2,4-triazole (0.025 mol, 5.05 g) and sodium hydroxide (0.025 mol, 1.01 g) were dissolved in ethanol (15 ml). The mixture was refluxed at 353 K for five hours, cooled to room temperature, acidified with HCl (12 *M*) and filtered. Colourless crystal of the title compound were obtained on slow evaporation of the solvent after several days at room temperature.

S3. Refinement

All H atoms were located in a difference Fourier map and refined isotropically.

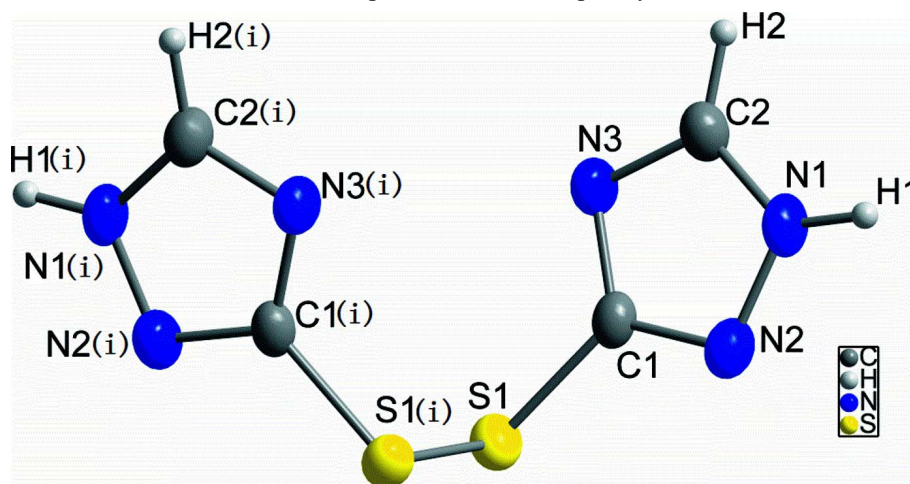


Figure 1

The molecular structure of the title compound showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. [Symmetry code: (i) -*x*, *y*, -*z* - 1/2]

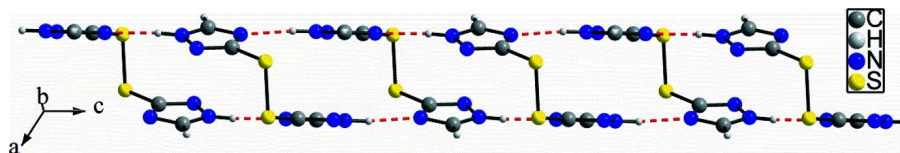


Figure 2

The chain of hydrogen-bonded molecules running along the *c* axis. Hydrogen bonding interactions are shown as red dashed lines.

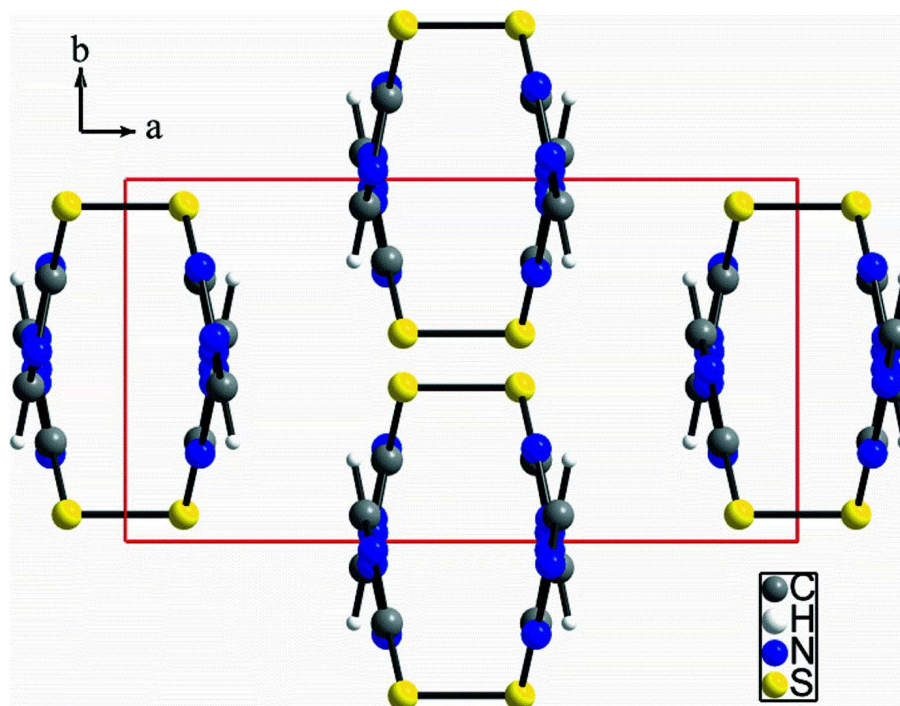


Figure 3

Packing diagram of the title compound viewed along the *c* axis.

Bis(4*H*-1,2,4-triazol-3-yl)disulfane

Crystal data

$C_4H_4N_6S_2$
 $M_r = 200.25$
 Monoclinic, $C2/c$
 Hall symbol: $-C\ 2yc$
 $a = 14.052\ (3)\ \text{\AA}$
 $b = 6.4044\ (13)\ \text{\AA}$
 $c = 9.928\ (2)\ \text{\AA}$
 $\beta = 122.18\ (3)^\circ$
 $V = 756.2\ (4)\ \text{\AA}^3$
 $Z = 4$

$F(000) = 408$
 $D_x = 1.759\ \text{Mg m}^{-3}$
 Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$
 Cell parameters from 25 reflections
 $\theta = 12\text{--}18^\circ$
 $\mu = 0.65\ \text{mm}^{-1}$
 $T = 293\ \text{K}$
 Block, colourless
 $0.12 \times 0.09 \times 0.06\ \text{mm}$

Data collection

Rigaku R-Axis RAPID IP
 diffractometer
 Radiation source: fine-focus sealed tube

Graphite monochromator
 Oscillation scans

Absorption correction: multi-scan

(ABSCOR; Higashi, 1995)

 $T_{\min} = 0.932$, $T_{\max} = 0.962$

3518 measured reflections

859 independent reflections

742 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.035$ $\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 3.4^\circ$ $h = -18 \rightarrow 18$ $k = -8 \rightarrow 7$ $l = -12 \rightarrow 12$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.029$ $wR(F^2) = 0.081$ $S = 1.09$

859 reflections

63 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

All H-atom parameters refined

 $w = 1/[\sigma^2(F_o^2) + (0.0491P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.22 \text{ e } \text{\AA}^{-3}$ *Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.08684 (3)	0.07452 (6)	-0.17828 (4)	0.03333 (18)
N1	0.13246 (12)	0.4379 (2)	0.15290 (16)	0.0330 (3)
N2	0.11117 (12)	0.2436 (2)	0.08916 (15)	0.0350 (3)
N3	0.13151 (11)	0.4757 (2)	-0.06502 (15)	0.0315 (3)
C1	0.11097 (11)	0.2743 (2)	-0.04273 (16)	0.0275 (3)
C2	0.14361 (13)	0.5723 (3)	0.06089 (18)	0.0331 (4)
H1	0.1365 (17)	0.462 (4)	0.244 (3)	0.053 (6)*
H2	0.1600 (16)	0.725 (3)	0.083 (2)	0.042 (5)*

Atomic displacement parameters ($\text{\AA}^2$)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0383 (3)	0.0327 (3)	0.0299 (3)	0.00536 (14)	0.0188 (2)	-0.00149 (13)
N1	0.0378 (7)	0.0423 (8)	0.0238 (7)	0.0035 (5)	0.0197 (6)	-0.0008 (5)
N2	0.0434 (7)	0.0387 (8)	0.0284 (7)	0.0037 (5)	0.0228 (6)	0.0035 (5)
N3	0.0383 (7)	0.0365 (7)	0.0258 (7)	-0.0025 (5)	0.0212 (6)	-0.0017 (5)
C1	0.0280 (7)	0.0346 (8)	0.0215 (7)	0.0031 (5)	0.0143 (6)	0.0022 (5)
C2	0.0357 (8)	0.0387 (9)	0.0270 (8)	-0.0029 (6)	0.0181 (7)	-0.0034 (6)

Geometric parameters (Å, °)

S1—C1	1.7541 (15)	N2—C1	1.3225 (19)
S1—S1 ⁱ	2.0693 (11)	N3—C2	1.322 (2)
N1—C2	1.324 (2)	N3—C1	1.3653 (19)
N1—N2	1.3549 (18)	C2—H2	1.004 (19)
N1—H1	0.89 (2)		
C1—S1—S1 ⁱ	101.72 (5)	N2—C1—N3	114.30 (13)
C2—N1—N2	110.63 (13)	N2—C1—S1	123.30 (12)
C2—N1—H1	128.6 (15)	N3—C1—S1	122.40 (11)
N2—N1—H1	120.8 (15)	N3—C2—N1	110.21 (15)
C1—N2—N1	102.11 (13)	N3—C2—H2	124.4 (12)
C2—N3—C1	102.74 (13)	N1—C2—H2	125.4 (12)

Symmetry code: (i) $-x, y, -z-1/2$.*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N1—H1 $\cdots$ N3 ⁱⁱ	0.89 (2)	1.97 (2)	2.8617 (19)	174.9 (19)

Symmetry code: (ii) $x, -y+1, z+1/2$.