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Hexaaquacobalt(II) bis[4-(2-hydroxybenzylideneamino)benzenesulfonate]

Xi-Shi Tai,* Yi-Min Feng and Fan-Yuan Kong

Department of Chemistry and Chemical Engineering, Weifang University, Weifang 261061, People's Republic of China
Correspondence e-mail: taixishi@lzu.edu.cn

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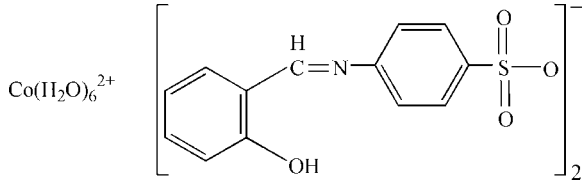
Key indicators: single-crystal X-ray study; T = 298 K; mean σ(C–C) = 0.009 Å; R factor = 0.071; wR factor = 0.141; data-to-parameter ratio = 13.4.

In the cation of the title compound, $[Co(H_2O)_6](C_{13}H_{10}NO_4S)_2$, the Co atom lies on a centre of symmetry and its coordination geometry is octahedral. The crystal structure is stabilized by water–anion O—H···O hydrogen bonds. An intramolecular O—H···N hydrogen bond occurs in the anion.

Related literature

For related literature, see: Allen *et al.* (1987); Tai & Feng (2008); Tai *et al.* (2003); Tai *et al.* (2008); Tai, Yin & Feng (2007); Tai, Yin & Hao (2007); Tai, Yin, Feng & Kong (2007); Wang *et al.* (2007).



Experimental

Crystal data

$[Co(H_2O)_6](C_{13}H_{10}NO_4S)_2$
M_r = 719.59
Monoclinic, $P2_1/n$
a = 6.3216 (13) Å
b = 35.211 (3) Å
c = 6.9924 (15) Å
β = 90.186 (2)°
V = 1556.4 (5) Å^3
Z = 2
Mo $K\alpha$ radiation
μ = 0.76 mm^{-1}
T = 298 (2) K
0.40 × 0.35 × 0.15 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2000)
T_{min} = 0.752, T_{max} = 0.895
7328 measured reflections
2749 independent reflections
2171 reflections with $I > 2\sigma(I)$
R_{int} = 0.040

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.070
$wR(F^2)$ = 0.141
S = 1.17
2749 reflections
205 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.34 e Å^{-3}
$\Delta\rho_{min}$ = −0.69 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O4—H4···N1	0.82	1.88	2.588 (7)	143
O5—H5A···O2^{i}	0.85	1.96	2.736 (6)	151
O5—H5B···O1	0.85	1.97	2.744 (6)	151
O6—H6A···O1^{ii}	0.85	1.99	2.757 (5)	150
O6—H6B···O3	0.85	2.03	2.768 (5)	144
O7—H7A···O3^{i}	0.85	1.96	2.759 (5)	157
O7—H7B···O2^{iii}	0.85	1.98	2.761 (5)	152
C6—H6···O3	0.93	2.56	2.918 (7)	104

Symmetry codes: (i) $x, y, z-1$; (ii) $x+1, y, z$; (iii) $x+1, y, z-1$.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: AT2558).

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supporting information

Acta Cryst. (2008). E**64**, m645 [doi:10.1107/S1600536808009422]

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S1. Comment

As part of our ongoing studies of the coordination chemistry of Schiff base ligands (Xi-Shi & Yi-Min, 2008; Tai, Feng & Zhang, 2008; Tai, Yin & Feng, 2007; Tai, Yin, Feng & Kong, 2007; Tai, Yin & Hao, 2007; Wang *et al.*, 2007; Tai *et al.*, 2003), we now report the synthesis and structure of the title compound, (I), (Fig. 1).

In the molecule of (I), the Co (II) centre is six-coordinate with six O donors of the water molecules. The C7—N1 bond length of 1.281 (8) Å is close to double-bond. Otherwise, the geometrical parameters for (I) are in normal range (Allen *et al.*, 1987). The dihedral angle between the two benzene ring is 33.5°, indicating that the molecule is non-planar, which perhaps correlates with the intramolecular and intermolecular hydrogen bonds (Table 1).

S2. Experimental

1 mmol of Cobalt acetate was added to a solution of salicylaldehyde-4-aminobenzene sulfonic acid (1 mmol) in 10 ml of 95% ethanol. The mixture was stirred for 2 h at refluxing temperature. Evaporating some ethanol, clear blocks of (I) were obtained after one weeks.

S3. Refinement

The H atoms were placed geometrically [C—H = 0.93 Å, O—H = 0.82 for hydroxy group and O—H = 0.85 Å for water molecules] and refined as riding with U_{iso}(H) = 1.2U_{eq}(carrier).

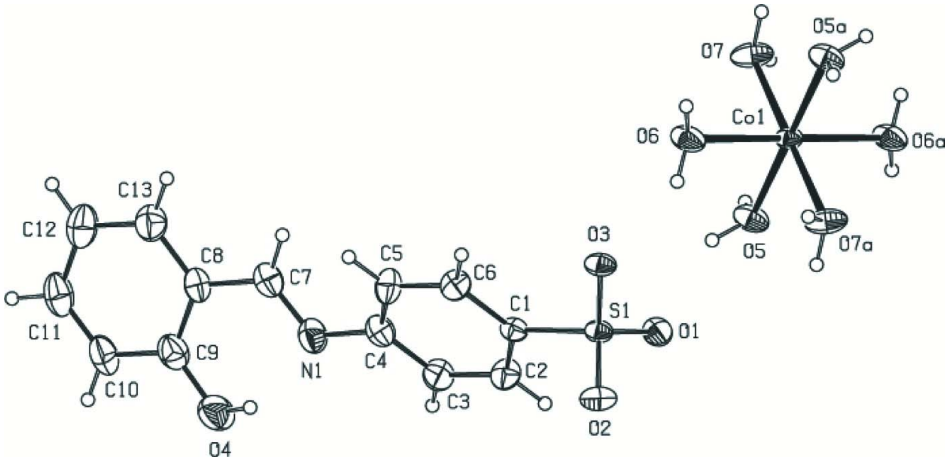

Figure 1

The molecular structure of (I) showing 30% displacement ellipsoids. [Symmetry code: (a) -x +2, -y, -z].

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Crystal data

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$b = 35.211$ (3) Å
$c = 6.9924$ (15) Å
$\beta = 90.186$ (2)°
$V = 1556.4$ (5) Å^3
$Z = 2$
$F(000) = 746$
$D_x = 1.535$ Mg m^{-3}
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 2380 reflections
$\theta = 2.4$–$25.3°$
$\mu = 0.76$ mm^{-1}
$T = 298$ K
Block, light purple
0.40 × 0.35 × 0.15 mm

Data collection

Bruker SMART CCD area-detector diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
φ and ω scans
Absorption correction: multi-scan (*SADABS*; Bruker, 2000)
$T_{min} = 0.752$, $T_{max} = 0.895$
7328 measured reflections
2749 independent reflections
2171 reflections with $I > 2\sigma(I)$
$R_{int} = 0.040$
$\theta_{max} = 25.0°$, $\theta_{min} = 2.3°$
$h = -5 \rightarrow 7$
$k = -41 \rightarrow 37$
$l = -8 \rightarrow 8$

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.071$
$wR(F^2) = 0.141$
$S = 1.17$
2749 reflections
205 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0118P)^2 + 5.6103P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max} = 0.34$ e Å^{-3}
$\Delta\rho_{min} = -0.70$ e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Co1	1.0000	0.0000	0.0000	0.0332 (3)
N1	0.4234 (9)	0.23121 (15)	0.5063 (8)	0.0585 (14)
O1	0.5135 (6)	0.05103 (11)	0.3289 (6)	0.0480 (10)
O2	0.5140 (6)	0.05158 (12)	0.6754 (6)	0.0504 (11)

O3	0.8400 (5)	0.06145 (11)	0.5055 (5)	0.0418 (9)
O4	0.1496 (8)	0.28449 (14)	0.5667 (7)	0.0772 (15)
H4	0.1993	0.2632	0.5809	0.116*
O5	0.7065 (5)	0.02585 (12)	0.0016 (6)	0.0496 (10)
H5A	0.6839	0.0392	−0.0978	0.060*
H5B	0.6850	0.0389	0.1020	0.060*
O6	1.1038 (6)	0.03462 (13)	0.2207 (6)	0.0571 (12)
H6A	1.2332	0.0310	0.2484	0.069*
H6B	1.0279	0.0325	0.3203	0.069*
O7	1.1055 (6)	0.03864 (12)	−0.2024 (6)	0.0575 (12)
H7A	1.0260	0.0392	−0.3007	0.069*
H7B	1.2326	0.0344	−0.2355	0.069*
S1	0.61081 (19)	0.06622 (4)	0.50249 (19)	0.0345 (3)
C1	0.5625 (8)	0.11545 (15)	0.4991 (7)	0.0349 (12)
C2	0.3648 (9)	0.12848 (17)	0.4446 (9)	0.0471 (15)
H2	0.2605	0.1115	0.4059	0.056*
C3	0.3241 (10)	0.16694 (17)	0.4483 (10)	0.0542 (17)
H3	0.1907	0.1758	0.4140	0.065*
C4	0.4787 (10)	0.19236 (17)	0.5021 (10)	0.0522 (15)
C5	0.6758 (10)	0.17953 (18)	0.5569 (9)	0.0545 (17)
H5	0.7794	0.1967	0.5950	0.065*
C6	0.7197 (9)	0.14085 (17)	0.5552 (8)	0.0471 (15)
H6	0.8527	0.1320	0.5911	0.056*
C7	0.5616 (11)	0.25709 (18)	0.4752 (10)	0.0571 (17)
H7	0.6992	0.2500	0.4448	0.069*
C8	0.5093 (11)	0.29731 (17)	0.4862 (10)	0.0542 (16)
C9	0.3064 (12)	0.3093 (2)	0.5366 (10)	0.0624 (19)
C10	0.2661 (13)	0.34768 (19)	0.5538 (10)	0.067 (2)
H10	0.1313	0.3557	0.5881	0.080*
C11	0.4190 (14)	0.3740 (2)	0.5216 (10)	0.071 (2)
H11	0.3882	0.3997	0.5362	0.085*
C12	0.6230 (15)	0.3628 (2)	0.4664 (12)	0.080 (2)
H12	0.7268	0.3808	0.4399	0.095*
C13	0.6654 (12)	0.32435 (19)	0.4525 (10)	0.0661 (19)
H13	0.8009	0.3164	0.4199	0.079*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Co1	0.0218 (5)	0.0452 (6)	0.0327 (5)	−0.0015 (5)	−0.0018 (4)	−0.0005 (5)
N1	0.063 (4)	0.047 (3)	0.066 (4)	0.009 (3)	−0.001 (3)	−0.008 (3)
O1	0.032 (2)	0.060 (3)	0.051 (3)	0.0021 (18)	−0.0070 (18)	−0.011 (2)
O2	0.035 (2)	0.065 (3)	0.051 (3)	0.0022 (19)	0.0071 (18)	0.018 (2)
O3	0.0233 (18)	0.057 (2)	0.045 (2)	0.0035 (17)	0.0008 (16)	0.0017 (19)
O4	0.081 (4)	0.063 (3)	0.087 (4)	0.014 (3)	0.020 (3)	0.006 (3)
O5	0.037 (2)	0.075 (3)	0.037 (2)	0.013 (2)	−0.0008 (17)	0.001 (2)
O6	0.028 (2)	0.095 (3)	0.048 (3)	−0.002 (2)	−0.0009 (18)	−0.023 (2)
O7	0.028 (2)	0.090 (3)	0.055 (3)	−0.002 (2)	−0.0028 (18)	0.026 (2)

S1	0.0233 (6)	0.0456 (8)	0.0348 (7)	0.0022 (6)	−0.0004 (5)	0.0012 (6)
C1	0.028 (3)	0.047 (3)	0.030 (3)	0.002 (2)	−0.002 (2)	−0.001 (2)
C2	0.033 (3)	0.051 (4)	0.058 (4)	0.002 (3)	−0.016 (3)	−0.002 (3)
C3	0.041 (3)	0.049 (4)	0.072 (5)	0.011 (3)	−0.005 (3)	−0.001 (3)
C4	0.053 (4)	0.048 (3)	0.056 (4)	0.012 (3)	0.001 (3)	−0.003 (3)
C5	0.061 (4)	0.045 (4)	0.057 (4)	−0.004 (3)	−0.014 (3)	−0.009 (3)
C6	0.038 (3)	0.057 (4)	0.046 (4)	−0.001 (3)	−0.014 (3)	−0.003 (3)
C7	0.061 (4)	0.053 (4)	0.057 (4)	0.012 (3)	−0.001 (3)	−0.007 (3)
C8	0.065 (4)	0.049 (4)	0.049 (4)	0.002 (3)	−0.003 (3)	−0.013 (3)
C9	0.084 (5)	0.057 (4)	0.047 (4)	0.007 (4)	0.008 (4)	−0.002 (3)
C10	0.104 (6)	0.045 (4)	0.051 (4)	0.017 (4)	0.008 (4)	−0.003 (3)
C11	0.107 (6)	0.046 (4)	0.059 (5)	0.019 (4)	−0.010 (4)	−0.002 (4)
C12	0.105 (7)	0.045 (4)	0.089 (6)	0.000 (4)	−0.009 (5)	0.001 (4)
C13	0.067 (5)	0.057 (4)	0.075 (5)	0.003 (4)	−0.005 (4)	−0.007 (4)

Geometric parameters (Å, º)

Co1—O5	2.067 (3)	C1—C6	1.392 (7)
Co1—O5^{i}	2.067 (3)	C2—C3	1.379 (8)
Co1—O6	2.072 (4)	C2—H2	0.9300
Co1—O6^{i}	2.072 (4)	C3—C4	1.377 (8)
Co1—O7^{i}	2.075 (4)	C3—H3	0.9300
Co1—O7	2.075 (4)	C4—C5	1.379 (8)
N1—C7	1.281 (8)	C5—C6	1.390 (8)
N1—C4	1.412 (7)	C5—H5	0.9300
O1—S1	1.460 (4)	C6—H6	0.9300
O2—S1	1.452 (4)	C7—C8	1.456 (8)
O3—S1	1.458 (3)	C7—H7	0.9300
O4—C9	1.338 (8)	C8—C13	1.392 (9)
O4—H4	0.8200	C8—C9	1.397 (9)
O5—H5A	0.8501	C9—C10	1.381 (9)
O5—H5B	0.8500	C10—C11	1.359 (10)
O6—H6A	0.8500	C10—H10	0.9300
O6—H6B	0.8500	C11—C12	1.404 (10)
O7—H7A	0.8500	C11—H11	0.9300
O7—H7B	0.8500	C12—C13	1.384 (9)
S1—C1	1.760 (5)	C12—H12	0.9300
C1—C2	1.384 (7)	C13—H13	0.9300
O5—Co1—O5^{i}	180.0 (2)	C3—C2—H2	120.4
O5—Co1—O6	91.09 (15)	C1—C2—H2	120.4
O5^{i}—Co1—O6	88.91 (15)	C4—C3—C2	120.8 (6)
O5—Co1—O6^{i}	88.91 (15)	C4—C3—H3	119.6
O5^{i}—Co1—O6^{i}	91.09 (15)	C2—C3—H3	119.6
O6—Co1—O6^{i}	180.0 (2)	C3—C4—C5	120.2 (6)
O5—Co1—O7^{i}	89.69 (15)	C3—C4—N1	117.4 (6)
O5^{i}—Co1—O7^{i}	90.31 (15)	C5—C4—N1	122.4 (6)
O6—Co1—O7^{i}	88.81 (17)	C4—C5—C6	119.9 (6)

O6^{i}—Co1—O7^{i}	91.19 (17)	C4—C5—H5	120.0
O5—Co1—O7	90.31 (15)	C6—C5—H5	120.0
O5^{i}—Co1—O7	89.69 (15)	C5—C6—C1	119.3 (5)
O6—Co1—O7	91.19 (17)	C5—C6—H6	120.3
O6^{i}—Co1—O7	88.81 (17)	C1—C6—H6	120.3
O7^{i}—Co1—O7	180.0 (3)	N1—C7—C8	121.8 (6)
C7—N1—C4	121.1 (6)	N1—C7—H7	119.1
C9—O4—H4	109.5	C8—C7—H7	119.1
Co1—O5—H5A	112.8	C13—C8—C9	119.2 (6)
Co1—O5—H5B	112.9	C13—C8—C7	119.7 (6)
H5A—O5—H5B	110.4	C9—C8—C7	121.1 (6)
Co1—O6—H6A	112.5	O4—C9—C10	119.2 (7)
Co1—O6—H6B	112.4	O4—C9—C8	121.6 (6)
H6A—O6—H6B	110.2	C10—C9—C8	119.2 (7)
Co1—O7—H7A	112.2	C11—C10—C9	121.5 (7)
Co1—O7—H7B	112.2	C11—C10—H10	119.3
H7A—O7—H7B	110.0	C9—C10—H10	119.3
O2—S1—O3	111.6 (2)	C10—C11—C12	120.5 (7)
O2—S1—O1	112.6 (2)	C10—C11—H11	119.7
O3—S1—O1	112.7 (2)	C12—C11—H11	119.7
O2—S1—C1	106.7 (2)	C13—C12—C11	118.2 (8)
O3—S1—C1	106.6 (2)	C13—C12—H12	120.9
O1—S1—C1	106.1 (2)	C11—C12—H12	120.9
C2—C1—C6	120.5 (5)	C12—C13—C8	121.3 (7)
C2—C1—S1	119.1 (4)	C12—C13—H13	119.3
C6—C1—S1	120.4 (4)	C8—C13—H13	119.3
C3—C2—C1	119.3 (5)		

O2—S1—C1—C2	−77.9 (5)	C2—C1—C6—C5	0.4 (9)
O3—S1—C1—C2	162.7 (4)	S1—C1—C6—C5	−178.0 (5)
O1—S1—C1—C2	42.4 (5)	C4—N1—C7—C8	−177.6 (6)
O2—S1—C1—C6	100.5 (5)	N1—C7—C8—C13	179.6 (7)
O3—S1—C1—C6	−18.9 (5)	N1—C7—C8—C9	1.8 (11)
O1—S1—C1—C6	−139.2 (5)	C13—C8—C9—O4	178.6 (6)
C6—C1—C2—C3	−0.7 (9)	C7—C8—C9—O4	−3.6 (11)
S1—C1—C2—C3	177.7 (5)	C13—C8—C9—C10	−0.7 (10)
C1—C2—C3—C4	1.2 (10)	C7—C8—C9—C10	177.1 (6)
C2—C3—C4—C5	−1.3 (11)	O4—C9—C10—C11	−178.9 (7)
C2—C3—C4—N1	−178.6 (6)	C8—C9—C10—C11	0.4 (11)
C7—N1—C4—C3	−150.6 (7)	C9—C10—C11—C12	1.1 (11)
C7—N1—C4—C5	32.2 (10)	C10—C11—C12—C13	−2.2 (11)
C3—C4—C5—C6	0.9 (11)	C11—C12—C13—C8	2.0 (11)
N1—C4—C5—C6	178.2 (6)	C9—C8—C13—C12	−0.5 (11)
C4—C5—C6—C1	−0.5 (10)	C7—C8—C13—C12	−178.3 (7)

Symmetry code: (i) $-x+2, -y, -z$.

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O4—H4···N1	0.82	1.88	2.588 (7)	143
O5—H5*A*···O2^{ii}	0.85	1.96	2.736 (6)	151
O5—H5*B*···O1	0.85	1.97	2.744 (6)	151
O6—H6*A*···O1^{iii}	0.85	1.99	2.757 (5)	150
O6—H6*B*···O3	0.85	2.03	2.768 (5)	144
O7—H7*A*···O3^{ii}	0.85	1.96	2.759 (5)	157
O7—H7*B*···O2^{iv}	0.85	1.98	2.761 (5)	152
C6—H6···O3	0.93	2.56	2.918 (7)	104

Symmetry codes: (ii) $x, y, z-1$; (iii) $x+1, y, z$; (iv) $x+1, y, z-1$.