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trans-di- μ -Acetato- κ^4 O:O'-bis[2-(5phenylisoxazolin-3-yl)phenyl- $\kappa^2 C^1$,N]dipalladium(II)

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Key indicators: single-crystal X-ray study; T = 298 K; mean σ (C–C) = 0.013 Å; *R* factor = 0.049; *wR* factor = 0.143; data-to-parameter ratio = 15.3.

The title compound, $[Pd_2(C_2H_3O_2)_2(C_{15}H_{10}NO)_2]$, crystallized from a dichloromethane/*n*-hexane solution with two crystallographically independent dimeric molecules in the asymmetric unit. Each molecule may be described as a dimer with an *anti* configuration and the cyclometallated fragments in the characteristic open-book disposition, linked by two bridging acetate ligands.

Related literature

For a related palladacycle bridged by acetate ligands, see: Schultz *et al.* (2004). For related literature, see: Dupont *et al.* (2005).



Experimental

Crystal data

 $\begin{bmatrix} Pd_2(C_2H_3O_2)_2(C_{15}H_{10}NO)_2 \end{bmatrix} \\ M_r = 771.37 \\ Monoclinic, P2_1/c \\ a = 14.8160 (6) Å \\ b = 24.2339 (10) Å \\ c = 19.6397 (8) Å \\ \beta = 103.233 (1)^\circ \end{bmatrix}$

Data collection

Bruker SMART APEXII diffractometer Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\rm min} = 0.750, T_{\rm max} = 0.854$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.049$ $wR(F^2) = 0.143$ S = 1.0912098 reflections $V = 6864.4 (5) \text{ Å}^{3}$ Z = 8Mo K\alpha radiation $\mu = 1.09 \text{ mm}^{-1}$ T = 298 (2) K $0.28 \times 0.20 \times 0.15 \text{ mm}$

80599 measured reflections 12098 independent reflections 8103 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.038$

793 parameters H-atom parameters constrained $\Delta \rho_{max} = 0.81$ e Å⁻³ $\Delta \rho_{min} = -0.58$ e Å⁻³

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008) and *PLATON* (Spek, 2003); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CF2190).

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trans-Di- μ -acetato- $\kappa^4 O:O'$ -bis[2-(5-phenylisoxazolin-3-yl)phenyl- $\kappa^2 C^1$,N]dipalladium(II)

Jin Zhou, Qibao Wang and Hongjian Sun

S1. Comment

Recently, much attention is being paid to the use of palladacycles as catalyst precursors (Dupont *et al.*, 2005). Palladacycles bridged by acetate ligands have been reported (Schultz *et al.* 2004). We have prepared a new palladacycle by a C—H bond activation reaction of 3,5-diphenylisoxazole with $Pd(OAc)_2$ in acetic acid. The asymmetric unit contains two molecules of the palladacycle having similar structures, of which only one will be discussed here. The molecular configuration of the complex is a dimeric form of the *anti* isomer with the cyclopalladated units in an arrangement linked by two acetate bridging ligands between the palladium atoms. The palladium atoms are coordinated in a slightly distorted square-planar arrangement by one cyclometallated isoxazole ligand and two bridging acetate ligands. In the crystal structure, the two cyclopalladated isoxazole subunits of the dimers are arranged in an *anti*-fashion. The chelating C, N-bonded isoxazoles are forced to lie above one another in the dimeric molecules because Pd1 and Pd2 are bridged by two *cis-µ*-acetate ligands. The resulting Pd—C (average 1.99 Å) and Pd—N (average 1.98 Å) bond distances fall within the range observed in other palladacycles (Smoliakova *et al.*, 2004). The Pd1…Pd2 distance [2.833 (5) Å] is noticeably larger than the sum of the covalent radii for two square-planar palladium(II) atoms, thus precluding any direct Pd—Pd interaction. The bond lengths of C7—N1 [1.303 (6) Å] and N2—C26 [1.318 (8) Å] are characteristic for C=N double bonds. The Pd—O bond distances *trans* to the carbon donor are slightly longer (*ca* 0.13 Å) than those *trans* to the nitrogen donor, as expected, because of the *trans* lengthening influence of carbon σ -donors.

S2. Experimental

3,5-Diphenylisoxazole (232 mg, 1.05 mmol) and $Pd(OAc)_2$ (224 mg, 1.0 mmol) were added to 5 ml of acetic acid. The reaction mixture was stirred for 2 h at refluxing temperature. After removing all the volatiles, the orange residue was washed with diethyl ether, and was chromatographed on a silica gel column, eluting with CH_2Cl_2/n -hexane. The title compound was dried under vacuum, with a yield of 48% (185 mg). Crystals suitable for X-ray diffraction were obtained from the CH_2Cl_2/n -hexane solution.

S3. Refinement

All H atoms were positioned geometrically and treated as riding on their parent atoms with C—H = 0.93 Å (aromatic), 0.96 (methyl), 0.98 Å (methine), and with $U_{iso}(H) = 1.2$ (1.5 for methyl groups) times $U_{eq}(C)$. Highly disordered and unidentified solvent molecules were treated with the SQUEEZE procedure of PLATON (Spek, 2003), as it proved impossible to model them with discrete atoms.



Figure 1

Molecular structure of the title compound. Displacement ellipsoids are drawn at 30% probability level.

trans-di- μ -Acetato- κ^4 O:O'-bis[2-(5-phenylisoxazolin-3-yl)phenyl- κ^2 C¹,N]dipalladium(II)

Crystal data	
$[Pd_2(C_{15}H_{10}NO)_2(C_2H_3O_2)_2]$	F(000) = 3072
$M_r = 771.37$	$D_{\rm x} = 1.493 {\rm Mg} {\rm m}^{-3}$
Monoclinic, $P2_1/c$	Mo <i>K</i> α radiation, $\lambda = 0.71073$ Å
a = 14.8160 (6) Å	Cell parameters from 6715 reflections
b = 24.2339 (10) Å	$\theta = 2.2 - 24.8^{\circ}$
c = 19.6397 (8) Å	$\mu = 1.09 \text{ mm}^{-1}$
$\beta = 103.233 \ (1)^{\circ}$	T = 298 K
$V = 6864.4 (5) Å^3$	Block, yellow
Z = 8	$0.28 \times 0.20 \times 0.15 \text{ mm}$
Data collection	
Bruker SMART APEXII	80599 measured reflections
diffractometer	12098 independent reflections
Radiation source: fine-focus sealed tube	8103 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.039$
φ and ω scans	$\theta_{\rm max} = 25.0^{\circ}, \theta_{\rm min} = 1.4^{\circ}$
Absorption correction: multi-scan	$h = -17 \rightarrow 17$
(SADABS; Sheldrick, 1996)	$k = 0 \rightarrow 28$
$T_{\min} = 0.750, \ T_{\max} = 0.854$	$l = 0 \rightarrow 23$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.049$	Hydrogen site location: inferred from
$wR(F^2) = 0.143$	neighbouring sites
S = 1.09	H-atom parameters constrained
12098 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0623P)^2 + 7.5732P]$
793 parameters	where $P = (F_{o}^{2} + 2F_{c}^{2})/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.001$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.81 \text{ e} \text{ Å}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.58 \text{ e} \text{ Å}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

	x	У	Ζ	$U_{ m iso}*/U_{ m eq}$
Pd1	0.25845 (3)	0.545234 (17)	0.51711 (2)	0.06594 (13)
Pd2	0.22368 (3)	0.62048 (2)	0.61849 (2)	0.08059 (16)
Pd3	0.27410 (4)	0.36805 (3)	0.24384 (3)	0.1003 (2)
Pd4	0.11212 (4)	0.35074 (3)	0.13585 (3)	0.1050 (2)
C41	0.2228 (4)	0.4531 (3)	0.3289 (3)	0.0803 (16)
O6	0.3757 (2)	0.63129 (15)	0.46051 (19)	0.0714 (9)
C36	0.2313 (5)	0.3043 (4)	0.3647 (5)	0.107 (2)
H36A	0.2458	0.2719	0.3442	0.129*
C7	0.4499 (3)	0.5702 (2)	0.5345 (2)	0.0617 (12)
01	0.1543 (2)	0.59872 (17)	0.45465 (19)	0.0765 (10)
O4	0.1169 (3)	0.5612 (2)	0.6166 (2)	0.0958 (13)
C1	0.3602 (3)	0.4992 (2)	0.5727 (3)	0.0652 (12)
O3	0.1587 (2)	0.49888 (17)	0.5463 (2)	0.0858 (11)
N1	0.3682 (3)	0.58268 (18)	0.4966 (2)	0.0660 (10)
C8	0.5150 (3)	0.6097 (2)	0.5236 (3)	0.0709 (14)
H8A	0.5782	0.6104	0.5436	0.085*
N2	0.3127 (3)	0.5885 (2)	0.6997 (3)	0.0857 (14)
С9	0.4663 (3)	0.6461 (2)	0.4782 (3)	0.0700 (14)
05	0.3106 (3)	0.5401 (2)	0.7351 (2)	0.0970 (13)
C17	0.1105 (4)	0.5155 (3)	0.5869 (3)	0.0850 (17)
O11	0.0447 (3)	0.2742 (2)	0.2375 (3)	0.0999 (13)
O12	0.2618 (3)	0.4951 (2)	0.2398 (2)	0.1017 (13)
C19	0.1241 (4)	0.6422 (3)	0.4752 (4)	0.0794 (16)
C6	0.4493 (3)	0.5203 (2)	0.5754 (2)	0.0647 (13)
C35	0.2383 (4)	0.3549 (3)	0.3329 (3)	0.0902 (19)

C4	0.5163 (4)	0.4469 (3)	0.6519(3)	0.0861 (17)
H4A	0.5679	0.4296	0.6795	0.103*
N3	0.2566 (4)	0.4452 (3)	0.2733 (3)	0.0996 (16)
O2	0.1436 (3)	0.66146 (18)	0.5361 (3)	0.0959 (12)
N4	0.0365 (4)	0.3228 (3)	0.2003 (3)	0.0971 (16)
08	0.3198 (4)	0.3906 (3)	0.1518 (4)	0.1283 (19)
C2	0.3520 (4)	0.4515 (2)	0.6092 (3)	0.0804 (16)
H2A	0.2937	0.4368	0.6080	0.096*
C26	0.3982 (5)	0.6078 (3)	0.7161 (3)	0.0907 (19)
C60	-0.0044 (4)	0.3596 (3)	0.2330 (4)	0.0914 (18)
C63	0.0116 (5)	0.2378 (3)	0.3431 (4)	0.108 (2)
C27	0.4539 (5)	0.5722 (4)	0.7614 (4)	0.104 (2)
H27A	0.5169	0.5760	0.7812	0.125*
09	0.1824 (4)	0.2750 (2)	0.1298 (3)	0.1223 (17)
C40	0.2119 (4)	0.4029(3)	0.3649(3)	0.0853(17)
C20	0.3298(4)	0.6723(3)	0.6255(4)	0.0850 (17)
010	0.2898(4)	0.2855(2)	0.2288(3)	0.1221 (17)
C5	0.5273(4)	0.4942(3)	0.6153(3)	0.0763(15)
H5A	0.5861	0.5085	0.6173	0.092*
C42	0.2032(4)	0.5095(3)	0.3326(3)	0.092 0.0872 (17)
H42A	0.1776	0.5266	0.3662	0.105*
C62	0.0057(4)	0.2836 (3)	0.2939(4)	0.0949 (19)
C43	0.2283(4)	0.5344(3)	0.2785(3)	0.0872 (17)
C39	0.1835 (5)	0.4005 (3)	0.4261(4)	0.104 (2)
H39A	0.1675	0.4327	0.4464	0.124*
C10	0.4897 (4)	0.6960 (3)	0.4450(3)	0.0874 (17)
C3	0.4301 (4)	0.4254(3)	0.6475 (3)	0.0889(17)
НЗА	0.4238	0.3927	0.6707	0.107*
C61	-0.0257(4)	0.3358 (3)	0.2921 (4)	0.0950 (19)
H61A	-0.0554	0.3525	0.3236	0.114*
C25	0.4086 (4)	0.6570 (3)	0.6758 (4)	0.094(2)
C59	-0.0077(5)	0.4129 (4)	0.2014 (4)	0.108 (2)
C44	0.2302 (4)	0.5915 (3)	0.2552 (4)	0.0955 (19)
C38	0.1784 (6)	0.3490 (4)	0.4581(4)	0.116 (2)
H38A	0.1589	0.3467	0.4998	0.139*
C21	0.3336 (5)	0.7170 (3)	0.5853 (4)	0.103 (2)
H21A	0.2815	0.7272	0.5515	0.123*
C28	0.3978 (6)	0.5303(4)	0.7713 (4)	0.104 (2)
C45	0.1993 (5)	0.6326 (4)	0.2909 (4)	0.111 (2)
H45A	0.1754	0.6241	0.3294	0.134*
C24	0.4899 (5)	0.6874 (4)	0.6857 (4)	0.113 (2)
H24A	0.5424	0.6773	0.7193	0.136*
C11	0.5732 (5)	0.7217 (3)	0.4697 (4)	0.117 (3)
H11A	0.6149	0.7070	0.5082	0.140*
C54	0.0426 (6)	0.4168 (4)	0.1491 (4)	0.115 (2)
C23	0.4898 (6)	0.7333 (4)	0.6437 (6)	0.133 (3)
H23A	0.5432	0.7547	0.6501	0.159*
C18	0.0554 (5)	0.6765 (3)	0.4213 (4)	0.117 (2)
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H18A	0.0447	0.6587	0.3765	0.175*
H18B	0.0807	0.7126	0.4180	0.175*
H18C	-0.0021	0.6797	0.4356	0.175*
C15	0.4319 (6)	0.7169 (4)	0.3868 (5)	0.162 (4)
H15A	0.3770	0.6988	0.3667	0.194*
C16	0.0389 (5)	0.4753 (4)	0.5994 (5)	0.128 (3)
H16A	0.0050	0.4914	0.6306	0.192*
H16B	0.0690	0.4422	0.6198	0.192*
H16C	-0.0030	0.4666	0.5557	0.192*
C65	-0.0182 (9)	0.2019 (5)	0.4486 (5)	0.151 (4)
H65A	-0.0459	0.2056	0.4864	0.181*
C22	0.4148 (6)	0.7481 (3)	0.5937 (5)	0.120 (3)
H22A	0.4172	0.7786	0.5654	0.144*
C64	-0.0287 (6)	0.2431 (4)	0.3992 (5)	0.125 (3)
H64A	-0.0630	0.2744	0.4037	0.150*
C37	0.2023 (5)	0.3024 (4)	0.4275 (5)	0.120 (3)
H37A	0.1993	0.2684	0.4491	0.143*
C14	0.4553 (7)	0.7657 (5)	0.3574 (6)	0.201 (6)
H14A	0.4147	0.7809	0.3187	0.241*
C46	0.2028 (7)	0.6863 (4)	0.2709 (6)	0.136 (3)
H46A	0.1805	0.7138	0.2957	0.163*
C49	0.2652 (7)	0.6058 (5)	0.1991 (5)	0.147 (3)
H49A	0.2861	0.5784	0.1733	0.176*
C48	0.2699 (9)	0.6601 (6)	0.1805 (7)	0.179 (5)
H48A	0.2956	0.6693	0.1430	0.215*
C58	-0.0549 (7)	0.4573 (4)	0.2205 (6)	0.149 (4)
H58A	-0.0873	0.4536	0.2556	0.179*
C47	0.2378 (8)	0.7000 (5)	0.2160 (6)	0.141 (3)
H47A	0.2399	0.7367	0.2027	0.169*
C68	0.0618 (8)	0.1913 (4)	0.3356 (6)	0.154 (4)
H68A	0.0869	0.1866	0.2967	0.185*
C55	0.0464 (8)	0.4667 (5)	0.1173 (6)	0.171 (4)
H55A	0.0805	0.4707	0.0832	0.206*
C67	0.0736 (10)	0.1514 (5)	0.3883 (8)	0.177 (5)
H67A	0.1109	0.1209	0.3859	0.212*
C66	0.0319 (11)	0.1563 (5)	0.4431 (7)	0.176 (5)
H66A	0.0379	0.1284	0.4763	0.211*
C12	0.5961 (6)	0.7683 (4)	0.4390 (5)	0.143 (4)
H12A	0.6538	0.7844	0.4560	0.172*
C13	0.5374 (7)	0.7909 (5)	0.3852 (6)	0.165 (4)
H13A	0.5523	0.8240	0.3665	0.198*
C29	0.4113 (7)	0.4800 (4)	0.8146 (4)	0.117 (3)
C34	0.3410 (9)	0.4440 (5)	0.8164 (6)	0.156 (4)
H34A	0.2819	0.4514	0.7896	0.188*
C33	0.3558 (12)	0.3960 (6)	0.8580 (7)	0.189 (5)
H33A	0.3075	0.3718	0.8591	0.227*
C30	0.4987 (9)	0.4668 (6)	0.8520 (7)	0.206 (6)
H30A	0.5489	0.4893	0.8504	0.247*

C31	0.5108 (13)	0.4185 (10)	0.8929 (11)	0.277 (12)
H31A	0.5696	0.4091	0.9187	0.332*
C57	-0.0535 (11)	0.5061 (6)	0.1877 (7)	0.228 (8)
H57A	-0.0869	0.5359	0.1990	0.274*
C56	-0.0020 (12)	0.5119 (5)	0.1368 (8)	0.247 (8)
H56A	0.0005	0.5460	0.1156	0.296*
07	0.1818 (5)	0.3897 (3)	0.0727 (3)	0.138 (2)
C51	0.2666 (8)	0.3996 (4)	0.0921 (6)	0.127 (3)
C53	0.2457 (7)	0.2581 (4)	0.1773 (5)	0.122 (3)
C52	0.2749 (8)	0.1985 (4)	0.1740 (6)	0.174 (4)
H52A	0.2375	0.1813	0.1331	0.262*
H52B	0.2669	0.1794	0.2150	0.262*
H52C	0.3389	0.1970	0.1718	0.262*
C50	0.3113 (9)	0.4249 (5)	0.0377 (6)	0.196 (5)
H50A	0.2651	0.4309	-0.0047	0.294*
H50B	0.3581	0.4004	0.0286	0.294*
H50C	0.3392	0.4595	0.0546	0.294*
C32	0.4400 (14)	0.3869 (7)	0.8950 (8)	0.192 (7)
H32A	0.4500	0.3562	0.9242	0.230*

Atomic displacement parameters $(Å^2)$

	U^{11}	<i>U</i> ²²	U^{33}	U^{12}	U^{13}	U^{23}
Pd1	0.0508 (2)	0.0751 (3)	0.0743 (3)	-0.00960 (18)	0.01948 (17)	-0.0122 (2)
Pd2	0.0638 (3)	0.0948 (3)	0.0854 (3)	-0.0040 (2)	0.0218 (2)	-0.0239 (2)
Pd3	0.0736 (3)	0.1213 (4)	0.1066 (4)	0.0092 (3)	0.0214 (3)	-0.0357 (3)
Pd4	0.0874 (4)	0.1304 (5)	0.0962 (4)	0.0143 (3)	0.0189 (3)	-0.0161 (3)
C41	0.059 (3)	0.101 (5)	0.084 (4)	-0.008 (3)	0.023 (3)	-0.032 (4)
O6	0.052 (2)	0.087 (3)	0.077 (2)	-0.0084 (17)	0.0183 (16)	0.0039 (19)
C36	0.078 (4)	0.114 (6)	0.110 (6)	0.004 (4)	-0.021 (4)	-0.013 (5)
C7	0.056 (3)	0.074 (3)	0.059 (3)	0.002 (2)	0.021 (2)	-0.008 (2)
01	0.055 (2)	0.091 (3)	0.080 (2)	-0.0064 (19)	0.0069 (17)	-0.007 (2)
04	0.063 (2)	0.129 (4)	0.103 (3)	-0.019 (2)	0.036 (2)	-0.012 (3)
C1	0.061 (3)	0.070 (3)	0.065 (3)	-0.007 (2)	0.016 (2)	-0.018 (3)
03	0.061 (2)	0.092 (3)	0.108 (3)	-0.0222 (19)	0.027 (2)	-0.009 (2)
N1	0.055 (2)	0.074 (3)	0.074 (3)	-0.0017 (19)	0.024 (2)	-0.003 (2)
C8	0.049 (3)	0.091 (4)	0.074 (3)	-0.007 (3)	0.016 (2)	-0.002 (3)
N2	0.079 (3)	0.101 (4)	0.078 (3)	-0.003 (3)	0.019 (3)	-0.016 (3)
C9	0.054 (3)	0.091 (4)	0.069 (3)	-0.011 (3)	0.022 (2)	-0.004 (3)
05	0.092 (3)	0.119 (4)	0.083 (3)	0.007 (3)	0.026 (2)	-0.011 (3)
C17	0.050 (3)	0.116 (5)	0.091 (4)	-0.024 (3)	0.019 (3)	0.000 (4)
011	0.091 (3)	0.099 (3)	0.104 (3)	0.003 (2)	0.011 (3)	-0.016 (3)
O12	0.092 (3)	0.118 (4)	0.103 (3)	-0.009 (3)	0.039 (3)	-0.024 (3)
C19	0.050 (3)	0.091 (4)	0.095 (4)	-0.005 (3)	0.014 (3)	-0.003 (4)
C6	0.064 (3)	0.080 (3)	0.054 (3)	-0.001 (3)	0.021 (2)	-0.018 (3)
C35	0.061 (3)	0.109 (5)	0.089 (4)	0.002 (3)	-0.007 (3)	-0.025 (4)
C4	0.077 (4)	0.092 (4)	0.090 (4)	0.016 (3)	0.021 (3)	0.008 (3)
N3	0.085 (4)	0.106 (4)	0.111 (4)	-0.009 (3)	0.029 (3)	-0.027 (4)

O2	0.078 (3)	0.097 (3)	0.108 (3)	0.006(2)	0.012 (2)	-0.020(3)
N4	0.079 (3)	0.105 (4)	0.103 (4)	0.011 (3)	0.012 (3)	-0.011 (4)
08	0.102 (4)	0.159 (5)	0.138 (5)	0.007 (3)	0.058 (4)	-0.041 (4)
C2	0.077 (4)	0.068 (4)	0.105 (4)	-0.009(3)	0.040 (3)	-0.008(3)
C26	0.077 (4)	0.118 (6)	0.076 (4)	0.006 (4)	0.016 (3)	-0.034 (4)
C60	0.070 (4)	0.106 (5)	0.090 (5)	0.013 (3)	0.001 (3)	-0.012(4)
C63	0.107 (5)	0.097 (6)	0.114 (6)	-0.013 (4)	0.016 (4)	-0.018(5)
C27	0.086 (5)	0.137 (7)	0.090 (5)	0.011 (5)	0.018 (4)	-0.032(5)
09	0.115 (4)	0.133 (4)	0.118 (4)	0.025 (3)	0.024 (3)	-0.041 (3)
C40	0.062 (3)	0.111 (5)	0.079 (4)	-0.006(3)	0.007 (3)	-0.024(4)
C20	0.080 (4)	0.084 (4)	0.094 (4)	-0.006(3)	0.026 (3)	-0.033(4)
010	0.110 (4)	0.123 (4)	0.124 (4)	0.033 (3)	0.006 (3)	-0.038(3)
C5	0.060 (3)	0.095 (4)	0.075(3)	-0.001(3)	0.017(3)	-0.008(3)
C42	0.068 (3)	0.112 (5)	0.081 (4)	-0.003(3)	0.017(3)	-0.018(4)
C62	0.073 (4)	0.110 (6)	0.097(5)	0.001 (4)	0.010(3)	-0.014(4)
C43	0.064(3)	0.111 (5)	0.084(4)	-0.003(3)	0.010(3)	-0.015(4)
C39	0.001(5)	0.115 (6)	0.007(5)	-0.005(4)	0.010(3) 0.012(4)	-0.030(4)
C10	0.071(4)	0.113(6)	0.097(3)	-0.015(3)	0.012(1) 0.027(3)	0.016 (4)
C3	0.071(1) 0.087(4)	0.084(4)	0.100 (4)	0.012(3)	0.027(3) 0.030(4)	0.010(1)
C61	0.073(4)	0.116 (6)	0.096(5)	0.003(0)	0.019(3)	-0.004(4)
C25	0.072 (4)	0.113 (5)	0.100(5)	-0.007(4)	0.025(4)	-0.043(4)
C59	0.090(5)	0.125 (6)	0.109 (5)	0.029 (4)	0.023(4)	-0.003(5)
C44	0.073(4)	0.117 (6)	0.096(5)	-0.004(4)	0.017(3)	0.001 (4)
C38	0.109 (6)	0.135(7)	0.097(5)	-0.009(5)	0.008(4)	0.001(5)
C21	0.087 (5)	0.105 (5)	0.116 (5)	-0.006(4)	0.021 (4)	-0.028(5)
C28	0.112 (6)	0.120 (6)	0.084 (5)	0.033 (5)	0.027(4)	-0.018(4)
C45	0.113 (6)	0.111 (6)	0.113 (6)	0.001 (5)	0.031(5)	0.002 (5)
C24	0.085 (5)	0.137 (7)	0.116 (6)	-0.021(5)	0.020 (4)	-0.043(5)
C11	0.074 (4)	0.141 (6)	0.131 (6)	-0.029(4)	0.014 (4)	0.033 (5)
C54	0.106 (5)	0.125 (7)	0.115 (6)	0.027 (5)	0.026 (5)	0.009 (5)
C23	0.103 (6)	0.143 (8)	0.160 (8)	-0.045(6)	0.045 (6)	-0.052(7)
C18	0.092 (5)	0.118 (6)	0.131 (6)	0.014 (4)	0.008 (4)	0.003 (5)
C15	0.112 (6)	0.201 (10)	0.156 (8)	-0.061 (6)	-0.004 (6)	0.094 (8)
C16	0.079 (4)	0.164 (7)	0.148 (7)	-0.048 (5)	0.039 (4)	0.003 (6)
C65	0.202 (11)	0.145 (9)	0.107 (7)	-0.016 (8)	0.038 (7)	0.003 (7)
C22	0.124 (7)	0.102 (5)	0.137 (7)	-0.030(5)	0.036 (6)	-0.025(5)
C64	0.145 (7)	0.119 (7)	0.108 (6)	0.005 (5)	0.022 (5)	-0.015 (5)
C37	0.098 (5)	0.126 (7)	0.114 (6)	0.004 (5)	-0.018 (5)	-0.007(5)
C14	0.116 (7)	0.268 (14)	0.194 (10)	-0.071 (8)	-0.013 (7)	0.138 (10)
C46	0.138 (8)	0.117 (7)	0.144 (8)	0.010 (6)	0.016 (6)	0.014 (6)
C49	0.187 (10)	0.140 (8)	0.135 (7)	0.009 (7)	0.080(7)	0.009 (6)
C48	0.218 (13)	0.181 (12)	0.157 (10)	0.007 (10)	0.082 (9)	0.050 (9)
C58	0.167 (9)	0.130 (7)	0.172 (9)	0.056 (6)	0.084 (7)	0.032 (7)
C47	0.148 (8)	0.134 (8)	0.131 (8)	-0.005 (7)	0.012 (7)	0.038 (7)
C68	0.206 (11)	0.101 (6)	0.169 (9)	0.017 (7)	0.073 (8)	0.005 (7)
C55	0.192 (11)	0.159 (9)	0.194 (11)	0.063 (8)	0.107 (9)	0.041 (8)
C67	0.242 (14)	0.119 (8)	0.185 (12)	0.017 (8)	0.081 (11)	0.010 (9)
C66	0.259 (16)	0.112 (8)	0.140 (9)	0.007 (9)	0.010 (10)	0.010 (7)
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C12	0.087 (5)	0.173 (9)	0.167 (8)	-0.046 (5)	0.025 (5)	0.056 (7)	
C13	0.122 (7)	0.187 (10)	0.189 (10)	-0.048 (7)	0.040 (7)	0.080 (8)	
C29	0.118 (6)	0.150 (8)	0.090 (5)	0.034 (6)	0.040 (5)	-0.011 (5)	
C34	0.163 (10)	0.178 (11)	0.130 (8)	0.039 (9)	0.039 (7)	0.017 (8)	
C33	0.232 (16)	0.186 (12)	0.172 (11)	0.064 (11)	0.091 (11)	0.058 (10)	
C30	0.144 (10)	0.238 (14)	0.226 (13)	0.075 (10)	0.024 (9)	0.071 (12)	
C31	0.175 (15)	0.35 (3)	0.31 (2)	0.122 (17)	0.063 (15)	0.17 (2)	
C57	0.325 (18)	0.177 (11)	0.244 (14)	0.130 (12)	0.192 (14)	0.082 (11)	
C56	0.339 (19)	0.170 (11)	0.296 (17)	0.126 (13)	0.206 (17)	0.117 (12)	
O7	0.114 (4)	0.193 (6)	0.114 (4)	0.019 (4)	0.040 (4)	-0.004 (4)	
C51	0.134 (8)	0.141 (7)	0.124 (7)	0.021 (6)	0.065 (7)	-0.018 (6)	
C53	0.118 (6)	0.135 (7)	0.116 (6)	0.031 (6)	0.030 (5)	-0.032 (6)	
C52	0.214 (11)	0.130 (8)	0.180 (10)	0.058 (8)	0.046 (9)	-0.035 (7)	
C50	0.227 (13)	0.232 (13)	0.165 (9)	0.017 (10)	0.120 (10)	0.014 (9)	
C32	0.243 (18)	0.196 (14)	0.157 (10)	0.124 (14)	0.087 (13)	0.055 (10)	

Geometric parameters (Å, °)

Pd1—N1	1.983 (4)	C61—H61A	0.930
Pd1—C1	1.989 (5)	C25—C24	1.386 (10)
Pd1—O3	2.041 (4)	C59—C58	1.382 (11)
Pd1—O1	2.168 (4)	C59—C54	1.404 (10)
Pd2—N2	1.980 (5)	C44—C45	1.355 (10)
Pd2—C20	1.992 (6)	C44—C49	1.365 (11)
Pd2—O2	2.033 (5)	C38—C37	1.364 (11)
Pd2—O4	2.131 (4)	C38—H38A	0.930
Pd3—C35	1.966 (7)	C21—C22	1.397 (10)
Pd3—N3	1.991 (6)	C21—H21A	0.930
Pd3—O10	2.044 (5)	C28—C29	1.475 (12)
Pd3—O8	2.141 (6)	C45—C46	1.363 (11)
Pd4—C54	1.954 (8)	C45—H45A	0.930
Pd4—N4	1.990 (6)	C24—C23	1.385 (12)
Pd4—O7	2.021 (6)	C24—H24A	0.930
Pd4—O9	2.128 (5)	C11—C12	1.360 (10)
C41—N3	1.315 (8)	C11—H11A	0.930
C41—C42	1.404 (9)	C54—C55	1.368 (12)
C41—C40	1.434 (9)	C23—C22	1.352 (12)
O6—C9	1.355 (6)	С23—Н23А	0.930
O6—N1	1.392 (5)	C18—H18A	0.960
C36—C35	1.391 (10)	C18—H18B	0.960
C36—C37	1.397 (11)	C18—H18C	0.960
С36—Н36А	0.930	C15—C14	1.392 (12)
C7—N1	1.303 (6)	C15—H15A	0.930
C7—C8	1.408 (7)	C16—H16A	0.960
C7—C6	1.454 (7)	C16—H16B	0.960
O1—C19	1.247 (7)	C16—H16C	0.960
O4—C17	1.245 (8)	C65—C66	1.351 (15)
C1—C2	1.380 (8)	C65—C64	1.375 (12)

C1—C6	1.406 (7)	С65—Н65А	0.930
O3—C17	1.251 (7)	C22—H22A	0.930
C8—C9	1.340 (7)	C64—H64A	0.930
C8—H8A	0.930	С37—Н37А	0.930
N2—C26	1.318 (8)	C14—C13	1.360 (13)
N2—O5	1.368 (7)	C14—H14A	0.930
C9—C10	1.454 (8)	C46—C47	1.341 (13)
O5—C28	1.345 (8)	C46—H46A	0.930
C17—C16	1.502 (8)	C49—C48	1.371 (15)
011—N4	1.376 (7)	C49—H49A	0.930
011-C62	1.381 (8)	C48—C47	1.341 (14)
012-C43	1 380 (7)	C48—H48A	0.930
012—N3	1 389 (7)	C58—C57	1.348(13)
C19 - O2	1 254 (7)	C58—H58A	0.930
C19 - C18	1.535(9)	C47—H47A	0.930
C6-C5	1 390 (7)	C68—C67	1 397 (14)
C_{35} C_{40}	1 418 (9)	C68—H68A	0.930
C4-C3	1 364 (8)	C_{55} C_{56}	1410(14)
C4-C5	1 383 (8)	C55—H55A	0.930
$C_4 - H_4 \Delta$	0.930	C67_C66	1 364 (16)
N4-C60	1 325 (8)	$C67 - H67 \Delta$	0.930
08 C51	1.323(0) 1.273(11)	C66 H66A	0.930
$C_2 = C_3$	1.275 (11)	C_{12} C_{13}	0.930 1 324 (12)
$C_2 = C_3$	0.030	C12 = H12A	1.324(12)
$C_2 = 112A$	1.372(10)	C12—III2A C13—H13A	0.930
$C_{20} = C_{27}$	1.372(10) 1.460(10)	C20 C24	1.363(13)
$C_{20} = C_{23}$	1.400(10) 1.306(0)	$C_{29} = C_{34}$	1.303(13) 1.272(13)
C60 - C61	1.390(9) 1.428(10)	$C_{29} = C_{30}$	1.372(13)
C60 - C59	1.428(10) 1.272(11)	C_{24} U_{24A}	1.409 (13)
C(3) = C(4)	1.373(11) 1.277(11)	C34—II34A	0.950
	1.3//(11)	C_{33}	1.312 (18)
$C_{03} = C_{02}$	1.460 (11)	C33—H33A	0.930
$C_{27} = C_{28}$	1.334 (11)	C_{30} U_{30}	1.41 (2)
$C_2/-H_2/A$	0.930	C30—H30A	0.930
09-053	1.231 (9)	$C_{31} = C_{32}$	1.31 (2)
C40—C39	1.363 (9)	C31—H31A	0.930
C20—C21	1.349 (10)	C57—C56	1.398 (15)
C20—C25	1.396 (9)	C57—H57A	0.930
010	1.260 (9)	С56—Н56А	0.930
C5—H5A	0.930	07—C51	1.250 (11)
C42—C43	1.347 (9)	C51—C50	1.511 (13)
C42—H42A	0.930	C53—C52	1.513 (12)
C62—C61	1.347 (9)	С52—Н52А	0.960
C43—C44	1.460 (10)	С52—Н52В	0.960
C39—C38	1.409 (11)	С52—Н52С	0.960
С39—Н39А	0.930	С50—Н50А	0.960
C10—C15	1.360 (10)	С50—Н50В	0.960
C10—C11	1.370 (9)	С50—Н50С	0.960
С3—НЗА	0.930	C32—H32A	0.930

N1—Pd1—C1	79.3 (2)	C45—C44—C43	120.1 (7)
N1—Pd1—O3	171.81 (17)	C49—C44—C43	122.0 (8)
C1—Pd1—O3	92.49 (19)	C37—C38—C39	119.5 (8)
N1—Pd1—O1	97.02 (16)	C37—C38—H38A	120.3
C1—Pd1—O1	176.35 (18)	C39—C38—H38A	120.3
O3—Pd1—O1	91.16 (15)	C20—C21—C22	121.0 (8)
N2—Pd2—C20	79.7 (3)	C20—C21—H21A	119.5
N2—Pd2—O2	172.7 (2)	C22—C21—H21A	119.5
C20—Pd2—O2	93.1 (3)	O5—C28—C27	109.8 (7)
N2—Pd2—O4	96.0 (2)	O5—C28—C29	116.0 (8)
C20—Pd2—O4	175.7 (3)	C27—C28—C29	134.1 (9)
O2—Pd2—O4	91.14 (18)	C44—C45—C46	120.8 (9)
C35—Pd3—N3	79.3 (3)	C44—C45—H45A	119.6
C35—Pd3—O10	91.8 (3)	C46—C45—H45A	119.6
N3—Pd3—O10	171.0 (3)	C23—C24—C25	117.4 (8)
C35—Pd3—O8	173.7 (3)	C23—C24—H24A	121.3
N3—Pd3—O8	95.3 (2)	C25—C24—H24A	121.3
O10—Pd3—O8	93.5 (2)	C12—C11—C10	121.1 (7)
C54—Pd4—N4	78.7 (3)	C12—C11—H11A	119.4
C54—Pd4—O7	93.3 (3)	C10-C11-H11A	119.4
N4—Pd4—O7	171.9 (3)	C55—C54—C59	118.7 (8)
C54—Pd4—O9	174.1 (3)	C55—C54—Pd4	125.8 (7)
N4—Pd4—O9	95.5 (2)	C59—C54—Pd4	115.2 (6)
O7—Pd4—O9	92.5 (2)	C22—C23—C24	122.3 (8)
N3—C41—C42	107.9 (6)	С22—С23—Н23А	118.8
N3—C41—C40	113.1 (6)	С24—С23—Н23А	118.8
C42—C41—C40	138.9 (6)	C19—C18—H18A	109.5
C9—O6—N1	106.4 (4)	C19—C18—H18B	109.5
C35—C36—C37	119.7 (8)	H18A—C18—H18B	109.5
С35—С36—Н36А	120.1	C19—C18—H18C	109.5
С37—С36—Н36А	120.1	H18A—C18—H18C	109.5
N1—C7—C8	109.5 (5)	H18B—C18—H18C	109.5
N1—C7—C6	112.7 (4)	C10—C15—C14	119.8 (8)
C8—C7—C6	137.8 (5)	C10—C15—H15A	120.1
C19—O1—Pd1	125.7 (4)	C14—C15—H15A	120.1
C17—O4—Pd2	125.8 (4)	C17—C16—H16A	109.5
C2—C1—C6	118.6 (5)	C17—C16—H16B	109.5
C2—C1—Pd1	127.6 (4)	H16A—C16—H16B	109.5
C6—C1—Pd1	113.8 (4)	C17—C16—H16C	109.5
C17—O3—Pd1	123.9 (4)	H16A—C16—H16C	109.5
C7—N1—O6	108.3 (4)	H16B—C16—H16C	109.5
C7—N1—Pd1	118.4 (4)	C66—C65—C64	120.8 (11)
O6—N1—Pd1	131.0 (3)	С66—С65—Н65А	119.6
C9—C8—C7	105.4 (5)	С64—С65—Н65А	119.6
С9—С8—Н8А	127.3	C23—C22—C21	119.0 (9)
С7—С8—Н8А	127.3	C23—C22—H22A	120.5
C26—N2—O5	108.1 (6)	C21—C22—H22A	120.5

C26—N2—Pd2	118.9 (5)	C63—C64—C65	120.0 (9)
05-N2-Pd2	131.0 (4)	C63—C64—H64A	120.0
C8-C9-O6	110.3 (5)	C65—C64—H64A	120.0
C8—C9—C10	134.3 (5)	C_{38} — C_{37} — C_{36}	121.7 (9)
06-C9-C10	115 4 (5)	C38—C37—H37A	119.2
$C_{28} = 05 = N_{2}^{2}$	106.8 (6)	C36—C37—H37A	119.2
04-C17-03	126.2(5)	C_{13} C_{14} C_{15}	120.0(9)
04-C17-C16	118.8 (6)	C13—C14—H14A	120.0
03-C17-C16	115.0(7)	C15— $C14$ — $H14A$	120.0
N4-011-C62	106 5 (5)	C47 - C46 - C45	120.0 121.0(10)
C43 = 012 = N3	105.9(5)	C47 - C46 - H46A	119 5
01-C19-02	127.0 (6)	C45—C46—H46A	119.5
01 - C19 - C18	1177(6)	C44-C49-C48	120.6(10)
$0^{2}-C_{19}-C_{18}$	115 3 (6)	C44— $C49$ — $H49A$	119 7
C_{5} C_{6} C_{1}	120.6 (5)	C48 - C49 - H49A	119.7
C_{5} C_{6} C_{7}	125.4(5)	C47 - C48 - C49	120.6(11)
$C_1 - C_6 - C_7$	1120.1(5) 114.0(5)	C47 - C48 - H48A	119.7
$C_{36} = C_{35} = C_{40}$	117.8 (7)	C49-C48-H48A	119.7
$C_{36} = C_{35} = P_{d3}$	127 3 (6)	C_{57} C_{58} C_{59}	119.7
C40-C35-Pd3	114 7 (6)	C57—C58—H58A	120.3
$C_{3}-C_{4}-C_{5}$	120.2 (6)	C59—C58—H58A	120.3
C3—C4—H4A	119.9	C48 - C47 - C46	1191(11)
C5-C4-H4A	119.9	C48—C47—H47A	120.4
C41 - N3 - O12	109.9 (5)	C46—C47—H47A	120.4
C41 - N3 - Pd3	118.5 (5)	C63—C68—C67	117.8 (10)
012—N3—Pd3	130.9 (4)	C63—C68—H68A	121.1
C19—O2—Pd2	122.8 (4)	С67—С68—Н68А	121.1
C60—N4—O11	108.2 (6)	C54—C55—C56	119.0 (10)
C60—N4—Pd4	117.7 (5)	С54—С55—Н55А	120.5
O11—N4—Pd4	129.4 (4)	С56—С55—Н55А	120.5
C51—O8—Pd3	124.9 (6)	C66—C67—C68	121.5 (12)
C3—C2—C1	120.3 (5)	С66—С67—Н67А	119.2
C3—C2—H2A	119.9	С68—С67—Н67А	119.2
C1—C2—H2A	119.9	C65—C66—C67	119.4 (12)
N2—C26—C27	109.8 (7)	С65—С66—Н66А	120.3
N2-C26-C25	111.8 (6)	С67—С66—Н66А	120.3
C27—C26—C25	138.0 (7)	C13—C12—C11	120.9 (8)
N4—C60—C61	109.9 (7)	C13—C12—H12A	119.6
N4—C60—C59	111.9 (7)	C11—C12—H12A	119.6
C61—C60—C59	138.0 (7)	C12—C13—C14	119.8 (9)
C64—C63—C68	120.4 (9)	С12—С13—Н13А	120.1
C64—C63—C62	119.3 (8)	C14—C13—H13A	120.1
C68—C63—C62	120.2 (9)	C34—C29—C30	118.0 (11)
C28—C27—C26	105.4 (7)	C34—C29—C28	122.7 (9)
С28—С27—Н27А	127.3	C30—C29—C28	119.2 (11)
С26—С27—Н27А	127.3	C29—C34—C33	121.6 (13)
C53—O9—Pd4	122.8 (5)	C29—C34—H34A	119.2
C39—C40—C35	121.9 (7)	C33—C34—H34A	119.2

C39—C40—C41	124.1 (7)	C32—C33—C34	117.5 (16)
C35—C40—C41	113.8 (6)	С32—С33—Н33А	121.2
C21—C20—C25	119.1 (6)	С34—С33—Н33А	121.2
C21—C20—Pd2	127.1 (5)	C29—C30—C31	118.8 (15)
C25—C20—Pd2	113.6 (6)	С29—С30—Н30А	120.6
C53—O10—Pd3	125.2 (5)	С31—С30—Н30А	120.6
C4—C5—C6	119.2 (5)	C32—C31—C30	120.4 (16)
С4—С5—Н5А	120.4	С32—С31—Н31А	119.8
С6—С5—Н5А	120.4	С30—С31—Н31А	119.8
C43—C42—C41	107.3 (6)	C58—C57—C56	120.0 (10)
C43—C42—H42A	126.4	С58—С57—Н57А	120.0
C41—C42—H42A	126.4	С56—С57—Н57А	120.0
C61—C62—O11	109.8 (7)	C57—C56—C55	120.8 (11)
C61—C62—C63	134.4 (7)	С57—С56—Н56А	119.6
O11—C62—C63	115.7 (7)	С55—С56—Н56А	119.6
C42—C43—O12	109.0 (6)	C51—O7—Pd4	121.4 (6)
C42—C43—C44	134.4 (7)	O7—C51—O8	128.1 (9)
O12—C43—C44	116.5 (6)	O7—C51—C50	115.8 (10)
C40—C39—C38	119.3 (7)	O8—C51—C50	116.2 (10)
С40—С39—Н39А	120.4	O9—C53—O10	126.7 (8)
С38—С39—Н39А	120.4	O9—C53—C52	117.4 (9)
C15—C10—C11	118.2 (6)	O10—C53—C52	115.8 (9)
C15—C10—C9	121.3 (6)	С53—С52—Н52А	109.5
C11—C10—C9	120.5 (6)	С53—С52—Н52В	109.5
C4—C3—C2	121.1 (6)	H52A—C52—H52B	109.5
С4—С3—Н3А	119.5	С53—С52—Н52С	109.5
С2—С3—НЗА	119.5	H52A—C52—H52C	109.5
C62—C61—C60	105.6 (6)	H52B—C52—H52C	109.5
C62—C61—H61A	127.2	С51—С50—Н50А	109.5
C60—C61—H61A	127.2	С51—С50—Н50В	109.5
C24—C25—C20	121.2 (8)	H50A—C50—H50B	109.5
C24—C25—C26	123.7 (8)	С51—С50—Н50С	109.5
C20—C25—C26	115.2 (6)	H50A—C50—H50C	109.5
C58—C59—C54	122.1 (9)	H50B-C50-H50C	109.5
C58—C59—C60	123.8 (8)	C31—C32—C33	123.5 (17)
C54—C59—C60	114.1 (7)	C31—C32—H32A	118.2
C45—C44—C49	117.8 (8)	С33—С32—Н32А	118.2