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# Iodido(N-phenylthiourea)bis(triphenylphosphine)copper(I)

# Ruthairat Nimthong,<sup>a</sup> Chaveng Pakawatchai,<sup>a\*</sup> Saowanit Saithong<sup>a</sup> and Jonathan P. H. Charmant<sup>b</sup>

<sup>a</sup>Department of Chemistry, Faculty of Science, Prince of Songkla University, Hat Yai 90112, Thailand, and <sup>b</sup>The School of Chemistry, University of Bristol, Cantock's Close, Bristol BS 1TS, England Correspondence e-mail: chaveng.p@psu.ac.th

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Key indicators: single-crystal X-ray study; T = 173 K; mean  $\sigma$ (C–C) = 0.009 Å; R factor = 0.051; wR factor = 0.108; data-to-parameter ratio = 17.4.

The coordination geometry of the Cu atom in the title compound,  $[CuI(C_7H_8N_2S)(C_{18}H_{15}P)_2]$ , is distorted tetrahedral; it is coordinated by two triphenylphosphine P atoms, one S atom from N-phenylthiourea (ptu) and one I atom. The crystal structure is stabilized by intra- and intermolecular N- $H \cdots I$  and  $N - H \cdots S$  interactions.

## **Related literature**

For related literature, see: Aslanidis et al. (1993, 1998); Bowmaker et al. (1987); Cox et al. (1999); Jianping & Kazuyuki (1996); Karagiannidis et al. (1990); Lecomte et al. (1989): Skoulika et al. (1991).



## **Experimental**

### Crystal data

[CuI(C<sub>7</sub>H<sub>8</sub>N<sub>2</sub>S)(C<sub>18</sub>H<sub>15</sub>P)<sub>2</sub>]  $M_r = 867.19$ Triclinic,  $P\overline{1}$ a = 10.9505 (9) Å b = 18.7294 (15) Å c = 21.3731 (18) Å $\alpha = 67.422 \ (1)^{\circ}$  $\beta = 77.215 (1)^{\circ}$ 

 $\gamma = 73.224 \ (1)^{\circ}$ V = 3844.9 (5) Å<sup>3</sup> Z = 4Mo Ka radiation  $\mu = 1.54 \text{ mm}^{-1}$ T = 173 (2) K  $0.50\,\times\,0.30\,\times\,0.07$  mm  $R_{\rm int} = 0.088$ 

42789 measured reflections

15698 independent reflections

8887 reflections with  $I > 2\sigma(I)$ 

#### Data collection

Bruker SMART Platform diffractometer

Absorption correction: multi-scan (SADABS; Bruker, 2003)  $T_{\rm min} = 0.582, T_{\rm max} = 0.898$ 

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$	901 parameters
$wR(F^2) = 0.108$	H-atom parameters constrained
S = 0.98	$\Delta \rho_{\rm max} = 0.82 \text{ e} \text{ Å}^{-3}$
15698 reflections	$\Delta \rho_{\rm min} = -0.74 \ {\rm e} \ {\rm \AA}^{-3}$

### Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N1 - H1B \cdot \cdot \cdot S2^{i}$	0.88	2.58	3.455 (5)	173
$N3 - H3B \cdot \cdot \cdot S1^{ii}$	0.88	2.63	3.396 (5)	146
$N2 - H2A \cdots I1$	0.88	2.65	3.511 (5)	166
$M4-H4A\cdots I2$	0.88	2.71	3.567 (4)	165

Symmetry codes: (i) x - 1, y + 1, z; (ii) x + 1, y - 1, z.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 2003); data reduction: SAINT and SHELXTL (Sheldrick, 2008); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2408).

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# supporting information

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# Iodido(N-phenylthiourea)bis(triphenylphosphine)copper(I)

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# S1. Comment

Studies of copper(I) complexes with mixed ligand systems containing triphenylphosphine and ligands containing S and N donors, have been increasing because of the versatility of these ligands as well as the different steric characteristics of the phosphine ligands, which can modify the compound geometry (Cox et al., 1999). The title complex, (I), is a monomeric complex which crystallizes in the triclinic system space group  $P\overline{1}$ . The structure consists of two independent [CuI(PPh<sub>3</sub>)<sub>2</sub>(ptu)] molecules in the asymmetric unit. The molecules feature a distorted tetrahedral copper(I) center with two triphenylphosphine P atoms, one S atom from ptu ligand and one iodide atom (Figures 1, 2). A distorted tetrahedral geometry is also found in the copper(I) halide complexes with mixed PPh<sub>3</sub>/sulfur base ligands (Lecomte *et al.*, 1989; Karagiannidis et al., 1990; Skoulika et al., 1991; Jianping & Kazuyuki., 1996; Aslanidis et al., 1998; Cox et al., 1999). The P—Cu—P angle deviates considerably from tetrahedral value at 118.63 (5)° and 122.18 (5)° in molecule A and B, respectively. The Cu—P distances of the molecule A [Cu(1)—P(1), 2.2908 (15)Å and Cu(1)—P(2), 2.3024 (16)Å] and of the molecule B [Cu(2)-P(3), 2.2876 (15)Å and Cu(2)-P(4), 2.2974 (16)Å] are slightly shorter as compared to the observed value for [Cu(PPh<sub>3</sub>)<sub>3</sub>I] (Bowmaker et al., 1987). The observed Cu—S distances of 2.4148 (16) and 2.3942 (15) Å in molecule A and B are consistent with the distances usually found for tetrahedrally coordinated copper(I) with thioamide-sulfur donors (Aslanidis et al., 1993). The mean plane of the phenyl ring (C45-C50) of ptu ligand in the molecule B forms a dihedral angle of 83.9 (2)° with one of phenyl ring (C81–C86) of PPh<sub>3</sub> molecules. Weak intra-molecular hydrogen bonding between the amide group and iodide atom is observed [N2…I1= 3.511 (5) Å, H2A…I1 = 2.65 Å, N2—  $H2A\cdots I1 = 166^{\circ}$  for molecule A and  $N4\cdots I2 = 3.567$  (4) Å,  $H4A\cdots I2 = 2.71$  Å,  $N4\longrightarrow H4A\cdots I2 = 165^{\circ}$  for molecule B]. However, only one C—H<sup>...</sup> $\pi$  interaction between C-sp<sup>2</sup> (C46—H46... $\pi$ ) of the phenyl ring from ptu ligand and the centroid of one phenyl ring (Cg14, C81-C86) of the PPh<sub>3</sub> molecules is observed in molecule B. In addition, the weak inter-molecular interactions between the one H atom of NH<sub>2</sub> group of ptu ligand of molecule A and the thione-S atom of ptu of molecule B [N1...S2 = 3.455 (5) Å, H1B...S2 = 2.58 Å, N1—H1B...S2 = 173 °] are found and vice versa [N3...S1 = 3.396 (5) Å, H3B···S1 = 2.63 Å, N3—H3B···S1 = 146 °]. The intra- and inter-molecular interactions of this complex are shown in Figure 3.

# S2. Experimental

Triphenylphosphine (0.27 g, 1.05 mmol) was dissolved in 30 cm<sup>3</sup> of acetonitrile at 70–75°C. CuI (0.1 g, 0.52 mmol) was added and the mixture was stirred for 2 h. After the formation of a complete clear solution, *N*-phenylthiourea (0.24 g, 1.57 mmol) was added slowly and the reaction mixture was stirring for 3 h. The resulting clear solution was filtered off and left to evaporate at room temperature. The microcrystalline solid, which deposited upon standing for several days, was filtered off and dried *in vacuo*. Single crystals suitable for X-ray diffraction studies were obtained by slow evaporation in acetonitrile. The melting point of the complex is 456-457 K. Elemental analysis, calculated for [CuI(PPh<sub>3</sub>)<sub>2</sub>(ptu)]: C, 59.55; H, 4.43; N, 3.23; S, 3.70%, found: C, 60.28; H, 4.46; N, 3.56; S, 3.93%.

# S3. Refinement

All H atoms atoms were constrained with a riding model for C-*sp*<sup>2</sup> [C—H = 0.95 Å and with  $U_{iso}(H) = 1.2U_{eq}(C)$ ] and for N atoms [N—H = 0.88 Å and with  $U_{iso}(H) = 1.2U_{eq}(N)$ .



# Figure 1

The molecular structure of the molecule A of [CuI(PPh<sub>3</sub>)<sub>2</sub>(ptu)] complex. Thermal ellipsoids are shown at 30% probability level.



# Figure 2

The molecular structure of the molecule B of [CuI(PPh<sub>3</sub>)<sub>2</sub>(ptu)] complex. Thermal ellipsoids are shown at 30% probability level.





The intra- and inter-molecular interactions of [CuI(PPh<sub>3</sub>)<sub>2</sub>(ptu)] complex.

# Iodido(N-phenylthiourea)bis(triphenylphosphine)copper(I)

Crystal data

## Data collection

Bruker Platform diffractometer Radiation source: sealed X-ray tube Graphite monochromator Detector resolution: 8.192 pixels mm<sup>-1</sup>  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Bruker, 2003)  $T_{\min} = 0.582, T_{\max} = 0.898$  Z = 4 F(000) = 1752  $D_x = 1.498 \text{ Mg m}^{-3}$ Mo K $\alpha$  radiation,  $\lambda = 0.71074 \text{ Å}$ Cell parameters from 8397 reflections  $\theta = 2.4-27.1^{\circ}$   $\mu = 1.54 \text{ mm}^{-1}$  T = 173 KPlate, colorless  $0.50 \times 0.30 \times 0.07 \text{ mm}$ 

42789 measured reflections 15698 independent reflections 8887 reflections with I > 2s(I) $R_{int} = 0.089$  $\theta_{max} = 26.4^\circ, \ \theta_{min} = 1.0^\circ$  $h = -13 \rightarrow 13$  $k = -23 \rightarrow 23$  $l = -26 \rightarrow 26$  Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.050$	Hydrogen site location: inferred from
$wR(F^2) = 0.108$	neighbouring sites
S = 0.98	H-atom parameters constrained
15698 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0284P)^2 + 1.9087P]$
901 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.001$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.82 \text{ e} \text{ Å}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.74 \text{ e} \text{ Å}^{-3}$

# Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(A^2)$ 

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
Cul	0.22484 (6)	0.77723 (4)	0.37656 (3)	0.02320 (17)
Cu2	0.80632 (6)	0.24103 (4)	0.10985 (3)	0.02128 (16)
I1	0.46394 (3)	0.75683 (2)	0.39996 (2)	0.02967 (11)
I2	0.56365 (3)	0.26189 (2)	0.09357 (2)	0.02911 (11)
S1	0.15450 (14)	0.90317 (8)	0.28957 (8)	0.0300 (4)
S2	0.86425 (13)	0.12053 (8)	0.20281 (7)	0.0228 (3)
N1	0.1515 (4)	1.0517 (3)	0.2712 (2)	0.0332 (12)
H1A	0.1840	1.0886	0.2729	0.040*
H1B	0.0778	1.0648	0.2550	0.040*
N2	0.3254 (4)	0.9573 (3)	0.3173 (2)	0.0316 (12)
H2A	0.3623	0.9064	0.3303	0.038*
N3	0.8570 (4)	-0.0299 (2)	0.2387 (2)	0.0316 (12)
H3A	0.8305	-0.0692	0.2371	0.038*
H3B	0.9134	-0.0396	0.2662	0.038*
N4	0.7271 (4)	0.0587 (2)	0.1580 (2)	0.0267 (11)
H4A	0.6980	0.1089	0.1346	0.032*
P1	0.21516 (14)	0.68413 (8)	0.33399 (7)	0.0217 (3)
P2	0.11408 (14)	0.77457 (8)	0.48212 (7)	0.0229 (3)
P3	0.80062 (13)	0.33467 (8)	0.15518 (7)	0.0212 (3)
P4	0.93858 (14)	0.23217 (8)	0.01210 (7)	0.0214 (3)
C1	0.2138 (5)	0.9760 (3)	0.2929 (3)	0.0246 (13)
C2	0.3955 (5)	1.0065 (3)	0.3258 (3)	0.0297 (14)
C3	0.4523 (5)	0.9764 (3)	0.3850 (3)	0.0332 (15)
Н3	0.4354	0.9292	0.4201	0.040*
C4	0.5344 (5)	1.0157 (3)	0.3924 (3)	0.0376 (16)

H4	0.5749	0.9948	0.4327	0.045*
C5	0.5582 (6)	1.0846 (3)	0.3425 (3)	0.0375 (16)
H5	0.6150	1.1111	0.3480	0.045*
C6	0.4991 (6)	1.1147 (3)	0.2848 (3)	0.0366 (16)
H6	0.5137	1.1632	0.2510	0.044*
C7	0.4180 (5)	1.0764 (3)	0.2742 (3)	0.0326 (15)
H7	0.3793	1.0971	0.2333	0.039*
C8	0.3242 (5)	0.6753 (3)	0.2570 (3)	0.0247 (13)
C9	0.4422 (6)	0.6943 (4)	0.2459 (3)	0.0379 (16)
H9	0.4596	0.7179	0.2742	0.045*
C10	0.5348 (6)	0.6796 (4)	0.1941 (3)	0.0454 (18)
H10	0.6159	0.6919	0.1877	0.054*
C11	0.5100 (6)	0.6469 (3)	0.1513 (3)	0.0417 (17)
H11	0.5736	0.6364	0.1157	0.050*
C12	0.3918 (6)	0.6299 (3)	0.1613 (3)	0.0363 (15)
H12	0.3736	0.6082	0.1318	0.044*
C13	0.2983 (5)	0.6438 (3)	0.2136 (3)	0.0284 (14)
H13	0.2171	0.6318	0.2196	0.034*
C14	0.0541 (5)	0.7024 (3)	0.3118 (3)	0.0222 (13)
C15	0.0182 (5)	0.7605 (3)	0.2507 (3)	0.0262 (13)
H15	0.0805	0.7864	0.2184	0.031*
C16	-0.1063 (5)	0.7808 (3)	0.2361 (3)	0.0268 (13)
H16	-0.1293	0.8209	0.1943	0.032*
C17	-0.1984 (5)	0.7431 (3)	0.2824 (3)	0.0317 (15)
H17	-0.2841	0.7569	0.2720	0.038*
C18	-0.1649 (5)	0.6852 (3)	0.3436 (3)	0.0301 (14)
H18	-0.2275	0.6593	0.3755	0.036*
C19	-0.0399 (5)	0.6652 (3)	0.3582 (3)	0.0239 (13)
H19	-0.0175	0.6257	0.4004	0.029*
C20	0.2465 (5)	0.5804 (3)	0.3896 (3)	0.0234 (13)
C21	0.2035 (5)	0.5231 (3)	0.3786 (3)	0.0267 (13)
H21	0.1548	0.5381	0.3421	0.032*
C22	0.2318 (5)	0.4449 (3)	0.4208 (3)	0.0314 (14)
H22	0.2000	0.4068	0.4138	0.038*
C23	0.3049 (6)	0.4212 (3)	0.4727 (3)	0.0387 (16)
H23	0.3255	0.3668	0.5006	0.046*
C24	0.3487 (6)	0.4768 (3)	0.4842 (3)	0.0395 (16)
H24	0.3989	0.4609	0.5203	0.047*
C25	0.3192 (5)	0.5558 (3)	0.4431 (3)	0.0309 (14)
H25	0.3491	0.5938	0.4515	0.037*
C26	-0.0620(5)	0.7959 (3)	0.4911 (3)	0.0250 (13)
C27	-0.1209 (5)	0.8044 (3)	0.4374 (3)	0.0298 (14)
H27	-0.0702	0.8021	0.3957	0.036*
C28	-0.2544 (6)	0.8164 (4)	0.4434 (3)	0.0406 (16)
H28	-0.2947	0.8232	0.4058	0.049*
C29	-0.3271 (6)	0.8183 (3)	0.5047 (3)	0.0399 (16)
H29	-0.4178	0.8253	0.5097	0.048*
C30	-0.2687 (6)	0.8102 (3)	0.5582 (3)	0.0383 (16)

H30	-0.3193	0.8130	0.5998	0.046*
C31	-0.1364 (5)	0.7979 (3)	0.5520 (3)	0.0333 (15)
H31	-0.0965	0.7907	0.5899	0.040*
C32	0.1544 (5)	0.6755 (3)	0.5464 (3)	0.0253 (13)
C33	0.0754 (6)	0.6219 (3)	0.5641 (3)	0.0368 (16)
H33	-0.0039	0.6385	0.5460	0.044*
C34	0.1128 (7)	0.5448 (4)	0.6079 (3)	0.0501 (19)
H34	0.0594	0.5086	0.6190	0.060*
C35	0.2258 (7)	0.5201 (4)	0.6355 (3)	0.0477 (18)
H35	0.2491	0.4674	0.6664	0.057*
C36	0.3056 (7)	0.5711 (4)	0.6186 (3)	0.0454 (17)
H36	0.3845	0.5537	0.6373	0.054*
C37	0.2700 (6)	0.6486 (3)	0.5740 (3)	0.0333 (15)
H37	0.3257	0.6836	0.5623	0.040*
C38	0.1383 (5)	0.8424 (3)	0.5191 (3)	0.0277 (14)
C39	0.1311 (6)	0.8262 (4)	0.5887 (3)	0.0401 (16)
H39	0.1221	0.7755	0.6205	0.048*
C40	0.1373 (7)	0.8861 (4)	0.6111 (4)	0.055 (2)
H40	0.1299	0.8760	0.6587	0.066*
C41	0.1542 (6)	0.9597 (4)	0.5653 (4)	0.054 (2)
H41	0.1597	0.9994	0.5813	0.065*
C42	0.1626 (7)	0.9743 (4)	0.4973 (4)	0.056 (2)
H42	0.1750	1.0243	0.4655	0.067*
C43	0.1534 (6)	0.9167 (4)	0.4743 (3)	0.0407 (17)
H43	0.1575	0.9283	0.4267	0.049*
C44	0.8119 (5)	0.0442 (3)	0.1994 (3)	0.0208 (12)
C45	0.6767 (5)	0.0025 (3)	0.1467 (3)	0.0239 (13)
C46	0.6858 (5)	0.0049 (3)	0.0809 (3)	0.0305 (14)
H46	0.7283	0.0415	0.0446	0.037*
C47	0.6331 (6)	-0.0459 (4)	0.0672 (3)	0.0382 (16)
H47	0.6404	-0.0445	0.0216	0.046*
C48	0.5700 (6)	-0.0984 (4)	0.1198 (3)	0.0386 (16)
H48	0.5331	-0.1328	0.1105	0.046*
C49	0.5607 (5)	-0.1007 (3)	0.1860 (3)	0.0388 (16)
H49	0.5177	-0.1370	0.2223	0.047*
C50	0.6145 (5)	-0.0498 (3)	0.2000 (3)	0.0263 (13)
H50	0.6082	-0.0514	0.2455	0.032*
C51	0.6712 (5)	0.3423 (3)	0.2242 (3)	0.0246 (13)
C52	0.6550 (6)	0.2738 (3)	0.2773 (3)	0.0447 (18)
H52	0.7103	0.2245	0.2765	0.054*
C53	0.5600(7)	0.2748 (4)	0.3319 (3)	0.053 (2)
H53	0.5545	0.2271	0.3693	0.064*
C54	0.4746 (6)	0.3436 (4)	0.3325 (4)	0.0490 (19)
H54	0.4053	0.3436	0.3682	0.059*
C55	0.4896 (6)	0.4126 (4)	0.2810 (4)	0.056 (2)
H55	0.4321	0.4612	0.2818	0.067*
C56	0.5880 (6)	0.4124 (4)	0.2276 (3)	0.0464 (18)
H56	0.5984	0.4612	0.1928	0.056*

C57	0.9462 (5)	0.3192 (3)	0.1916 (3)	0.0199 (12)
C58	1.0621 (5)	0.2804 (3)	0.1645 (3)	0.0268 (13)
H58	1.0629	0.2610	0.1293	0.032*
C59	1.1759 (6)	0.2696 (3)	0.1883 (3)	0.0349 (15)
H59	1.2547	0.2453	0.1678	0.042*
C60	1.1761 (6)	0.2939 (3)	0.2413 (3)	0.0340 (15)
H60	1.2544	0.2854	0.2581	0.041*
C61	1.0622 (6)	0.3306 (3)	0.2698 (3)	0.0326 (15)
H61	1.0621	0.3466	0.3070	0.039*
C62	0.9478 (6)	0.3446 (3)	0.2453 (3)	0.0271 (13)
H62	0.8703	0.3714	0.2647	0.033*
C63	0.7774(5)	0.4363 (3)	0.0956 (3)	0.0225(13)
C64	0.8678(5)	0.4821(3)	0.0775(3)	0.0220(10) 0.0284(14)
H64	0.9438	0.4619	0.0990	0.034*
C65	0.8473 (6)	0.5571 (3)	0.0279(3)	0.031
H65	0.9102	0.5877	0.0150	0.043*
C66	0.7355 (6)	0.5873(3)	-0.0025(3)	0.015 0.0353(15)
H66	0.7216	0.6388	-0.0362	0.0555 (15)
C67	0.6440 (6)	0.5429 (3)	0.0158 (3)	0.042 0.0328 (15)
H67	0.5667	0.5429 (3)	-0.0046	0.0328 (13)
C68	0.6657 (6)	0.3040 0.4678(3)	0.0040	0.039 0.0289 (14)
H68	0.6037 (0)	0.4370	0.0040 (5)	0.0209(14)
C60	1 1067 (5)	0.4370 0.1808 (3)	0.0738	0.035
C09	1.1007(5)	0.1090(3)	0.0278(3)	0.0204(12)
C70 H70	1.2030 (3)	0.2323 (3)	-0.0020(3)	0.0297(14)
H70	1.1000	0.2828	-0.0308	$0.030^{\circ}$
C/I	1.3245 (0)	0.2011 (4)	0.0189 (3)	0.0370(16)
H/1	1.3895	0.2304	-0.0019	0.044*
C72	1.3524 (6)	0.12/9 (4)	0.0696 (3)	0.0351 (15)
H/2	1.4354	0.10/4	0.0841	0.042*
C73	1.2581 (5)	0.0846 (3)	0.0989 (3)	0.0307 (14)
H73	1.2767	0.0338	0.1331	0.037*
C/4	1.1364 (5)	0.1156 (3)	0.0781 (3)	0.0273 (13)
H74	1.0723	0.0857	0.0985	0.033*
C75	0.9415 (5)	0.3227 (3)	-0.0624 (3)	0.0242 (13)
C76	0.8565 (6)	0.3940 (3)	-0.0611 (3)	0.0316 (14)
H76	0.8008	0.3963	-0.0206	0.038*
C77	0.8536 (7)	0.4618 (4)	-0.1195 (3)	0.0461 (18)
H77	0.7960	0.5103	-0.1184	0.055*
C78	0.9325 (7)	0.4593 (4)	-0.1779 (3)	0.0484 (19)
H78	0.9295	0.5060	-0.2172	0.058*
C79	1.0163 (7)	0.3897 (4)	-0.1803 (3)	0.0461 (18)
H79	1.0716	0.3884	-0.2211	0.055*
C80	1.0202 (6)	0.3215 (3)	-0.1234 (3)	0.0366 (15)
H80	1.0771	0.2733	-0.1257	0.044*
C81	0.9045 (5)	0.1720 (3)	-0.0295 (3)	0.0194 (12)
C82	0.9928 (5)	0.1096 (3)	-0.0451 (3)	0.0236 (13)
H82	1.0759	0.0925	-0.0308	0.028*
C83	0.9595 (6)	0.0719 (3)	-0.0819 (3)	0.0313 (14)

# supporting information

Н83	1 0204	0 0291	-0.0925	0.038*
C84	0.8404 (6)	0.0960 (3)	-0.1029(3)	0.0312 (15)
H84	0.8191	0.0699	-0.1280	0.037*
C85	0.7513 (6)	0.1577 (3)	-0.0879 (3)	0.0305 (14)
H85	0.6688	0.1746	-0.1028	0.037*
C86	0.7827 (5)	0.1956 (3)	-0.0506 (3)	0.0287 (14)
H86	0.7208	0.2377	-0.0394	0.034*

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	U <sup>22</sup>	<i>U</i> <sup>33</sup>	<i>U</i> <sup>12</sup>	<i>U</i> <sup>13</sup>	<i>U</i> <sup>23</sup>
Cu1	0.0238 (4)	0.0223 (4)	0.0256 (4)	-0.0074 (3)	-0.0024 (3)	-0.0092 (3)
Cu2	0.0239 (4)	0.0165 (3)	0.0254 (4)	-0.0047(3)	-0.0050(3)	-0.0082 (3)
I1	0.0237 (2)	0.0261 (2)	0.0380 (3)	-0.00612 (18)	-0.00825 (18)	-0.00693 (19)
I2	0.0247 (2)	0.0208 (2)	0.0409 (3)	-0.00426 (17)	-0.01189 (18)	-0.00596 (18)
<b>S</b> 1	0.0375 (9)	0.0223 (8)	0.0324 (9)	-0.0072 (7)	-0.0164 (7)	-0.0048 (7)
S2	0.0289 (8)	0.0168 (7)	0.0249 (8)	-0.0047 (6)	-0.0088 (6)	-0.0070 (6)
N1	0.035 (3)	0.025 (3)	0.044 (3)	-0.006 (2)	-0.017 (2)	-0.010 (2)
N2	0.035 (3)	0.020 (3)	0.044 (3)	-0.010(2)	-0.012 (3)	-0.008(2)
N3	0.044 (3)	0.014 (2)	0.040 (3)	-0.005 (2)	-0.021 (3)	-0.004(2)
N4	0.030 (3)	0.014 (2)	0.036 (3)	-0.001 (2)	-0.017 (2)	-0.005 (2)
P1	0.0254 (8)	0.0174 (7)	0.0224 (8)	-0.0066 (7)	-0.0023 (7)	-0.0061 (6)
P2	0.0250 (8)	0.0230 (8)	0.0232 (8)	-0.0082 (7)	-0.0017 (7)	-0.0095 (7)
P3	0.0237 (8)	0.0158 (7)	0.0251 (8)	-0.0041 (6)	-0.0041 (7)	-0.0078 (6)
P4	0.0250 (8)	0.0178 (8)	0.0236 (8)	-0.0061 (6)	-0.0043 (7)	-0.0077 (7)
C1	0.034 (4)	0.019 (3)	0.019 (3)	-0.003 (3)	-0.005 (3)	-0.005 (3)
C2	0.028 (3)	0.028 (3)	0.034 (4)	-0.004 (3)	-0.005 (3)	-0.012 (3)
C3	0.037 (4)	0.028 (3)	0.036 (4)	-0.014 (3)	-0.009 (3)	-0.005 (3)
C4	0.031 (4)	0.035 (4)	0.050 (4)	-0.011 (3)	-0.008 (3)	-0.015 (3)
C5	0.040 (4)	0.028 (4)	0.051 (4)	-0.016 (3)	-0.006 (3)	-0.015 (3)
C6	0.038 (4)	0.023 (3)	0.045 (4)	-0.013 (3)	0.004 (3)	-0.009 (3)
C7	0.033 (4)	0.028 (3)	0.033 (4)	-0.009 (3)	-0.003 (3)	-0.005 (3)
C8	0.027 (3)	0.016 (3)	0.028 (3)	-0.006 (3)	0.002 (3)	-0.004 (3)
C9	0.042 (4)	0.045 (4)	0.030 (4)	-0.019 (3)	0.009 (3)	-0.018 (3)
C10	0.037 (4)	0.061 (5)	0.047 (4)	-0.024 (4)	0.013 (3)	-0.030 (4)
C11	0.057 (5)	0.029 (4)	0.032 (4)	-0.006 (3)	0.012 (3)	-0.013 (3)
C12	0.052 (4)	0.025 (3)	0.032 (4)	-0.005 (3)	-0.004 (3)	-0.014 (3)
C13	0.030 (3)	0.023 (3)	0.032 (4)	-0.006 (3)	0.000 (3)	-0.011 (3)
C14	0.025 (3)	0.020 (3)	0.026 (3)	-0.002 (3)	0.000 (3)	-0.016 (3)
C15	0.033 (3)	0.023 (3)	0.027 (3)	-0.011 (3)	-0.001 (3)	-0.011 (3)
C16	0.031 (3)	0.022 (3)	0.027 (3)	-0.002 (3)	-0.009 (3)	-0.009 (3)
C17	0.023 (3)	0.036 (4)	0.045 (4)	-0.001 (3)	-0.012 (3)	-0.023 (3)
C18	0.030 (3)	0.030 (3)	0.033 (4)	-0.014 (3)	0.003 (3)	-0.012 (3)
C19	0.037 (4)	0.013 (3)	0.019 (3)	-0.004 (3)	-0.003 (3)	-0.003 (2)
C20	0.020 (3)	0.017 (3)	0.027 (3)	-0.002 (2)	0.003 (3)	-0.006 (3)
C21	0.024 (3)	0.019 (3)	0.033 (4)	-0.003 (3)	-0.001 (3)	-0.009 (3)
C22	0.028 (3)	0.021 (3)	0.043 (4)	-0.010 (3)	0.005 (3)	-0.011 (3)
C23	0.043 (4)	0.016 (3)	0.044 (4)	-0.003(3)	0.007 (3)	-0.006 (3)

C24	0.048 (4)	0.029 (4)	0.029 (4)	0.004 (3)	-0.010 (3)	-0.002 (3)
C25	0.037 (4)	0.026 (3)	0.028 (3)	-0.001(3)	-0.003 (3)	-0.011 (3)
C26	0.021 (3)	0.022 (3)	0.035 (4)	-0.006 (3)	-0.004 (3)	-0.012 (3)
C27	0.026 (3)	0.033 (3)	0.036 (4)	-0.007 (3)	0.002 (3)	-0.019 (3)
C28	0.039 (4)	0.045 (4)	0.044 (4)	-0.006(3)	-0.019(3)	-0.016 (3)
C29	0.025 (4)	0.037 (4)	0.052 (5)	-0.008(3)	-0.001(3)	-0.012 (3)
C30	0.036 (4)	0.041 (4)	0.030 (4)	-0.007(3)	0.001 (3)	-0.008(3)
C31	0.030 (4)	0.044 (4)	0.029 (4)	-0.008(3)	-0.008(3)	-0.014(3)
C32	0.026 (3)	0.021 (3)	0.027 (3)	-0.005(3)	0.005 (3)	-0.011 (3)
C33	0.034 (4)	0.026 (3)	0.049 (4)	-0.004(3)	0.000 (3)	-0.016(3)
C34	0.059 (5)	0.027 (4)	0.058 (5)	-0.018(4)	0.008 (4)	-0.010(4)
C35	0.057 (5)	0.029(4)	0.042(4)	-0.007(4)	-0.003(4)	0.001 (3)
C36	0.053(4)	0.038(4)	0.039(4)	-0.005(4)	-0.016(3)	-0.004(3)
C37	0.036(4)	0.037(4)	0.031 (4)	-0.008(3)	-0.007(3)	-0.014(3)
C38	0.027(3)	0.030(3)	0.030(4)	-0.009(3)	0.001(3)	-0.014(3)
C39	0.054(4)	0.037(4)	0.033(4)	-0.008(3)	-0.014(3)	-0.012(3)
C40	0.051(1)	0.057(1)	0.055(1)	-0.007(4)	-0.024(4)	-0.035(4)
C41	0.000(3)	0.003(5)	0.031(5)	-0.008(4)	-0.012(4)	-0.060(5)
C42	0.070(5)	0.005(5)	0.071 (6)	-0.027(4)	0.012(1)	-0.038(4)
C43	0.075(4)	0.038(4)	0.071(0)	-0.021(3)	0.009(3)	-0.022(3)
C44	0.025(1) 0.018(3)	0.020(1) 0.017(3)	0.023(3)	-0.003(2)	-0.002(2)	-0.004(3)
C45	0.026(3)	0.017(3)	0.029(3)	-0.006(3)	-0.002(2)	-0.006(3)
C46	0.020(0)	0.029(3)	0.029(4)	-0.006(3)	-0.003(3)	-0.009(3)
C47	0.031(3)	0.025(3)	0.029(1) 0.044(4)	-0.010(3)	-0.006(3)	-0.030(3)
C48	0.034(4)	0.038(4)	0.058(5)	-0.013(3)	-0.011(3)	-0.025(4)
C49	0.029(4)	0.029(4)	0.057(5)	-0.015(3)	-0.001(3)	-0.010(3)
C50	0.027(3)	0.025(1)	0.024(3)	-0.006(3)	-0.003(3)	-0.008(3)
C51	0.031(3)	0.018(3)	0.028(3)	-0.007(3)	-0.006(3)	-0.010(3)
C52	0.052(4)	0.023(4)	0.051 (4)	-0.005(3)	0.016 (4)	-0.019(3)
C53	0.070(5)	0.033(4)	0.049 (5)	-0.023(4)	0.028 (4)	-0.017(4)
C54	0.041 (4)	0.046 (5)	0.057 (5)	-0.010(4)	0.018 (4)	-0.029(4)
C55	0.050 (5)	0.041 (4)	0.057(5)	0.015 (4)	0.009 (4)	-0.023(4)
C56	0.058 (5)	0.025 (4)	0.035(4)	0.008(3)	0.000(3)	-0.003(3)
C57	0.027(3)	0.013(3)	0.019 (3)	-0.006(2)	-0.004(2)	-0.003(2)
C58	0.032(3)	0.018 (3)	0.031 (3)	-0.004(3)	-0.003(3)	-0.012(3)
C59	0.024 (3)	0.039 (4)	0.039 (4)	-0.003(3)	-0.005(3)	-0.012(3)
C60	0.040 (4)	0.026 (3)	0.035 (4)	-0.009(3)	-0.020(3)	-0.001(3)
C61	0.047 (4)	0.025(3)	0.025 (3)	-0.008(3)	-0.012(3)	-0.004(3)
C62	0.038(4)	0.023(3)	0.021(3)	-0.010(3)	-0.002(3)	-0.005(3)
C63	0.032 (3)	0.018 (3)	0.023(3)	-0.006(3)	-0.006(3)	-0.010(3)
C64	0.033 (3)	0.026 (3)	0.027(3)	-0.007(3)	-0.008(3)	-0.007(3)
C65	0.033 (4)	0.024 (3)	0.048 (4)	-0.012(3)	0.001 (3)	-0.009(3)
C66	0.047 (4)	0.017 (3)	0.034 (4)	-0.002(3)	-0.004(3)	-0.005(3)
C67	0.034 (4)	0.023 (3)	0.039 (4)	0.001 (3)	-0.017(3)	-0.007(3)
C68	0.041 (4)	0.019 (3)	0.032 (4)	-0.010(3)	-0.009 (3)	-0.009 (3)
C69	0.023 (3)	0.024 (3)	0.020 (3)	-0.008 (3)	-0.002 (2)	-0.013 (3)
C70	0.039 (4)	0.027 (3)	0.032 (4)	-0.012 (3)	-0.004(3)	-0.015 (3)
C71	0.034 (4)	0.033 (4)	0.054 (4)	-0.015 (3)	-0.006 (3)	-0.019 (3)
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C72	0.027 (3)	0.044 (4)	0.045 (4)	0.003 (3)	-0.014 (3)	-0.029 (3)	
C73	0.033 (4)	0.029 (3)	0.030 (4)	0.006 (3)	-0.010 (3)	-0.015 (3)	
C74	0.035 (4)	0.030 (3)	0.024 (3)	-0.011 (3)	-0.002 (3)	-0.015 (3)	
C75	0.033 (3)	0.022 (3)	0.020 (3)	-0.009(3)	-0.007 (3)	-0.006 (3)	
C76	0.043 (4)	0.023 (3)	0.030 (4)	0.000 (3)	-0.010 (3)	-0.013 (3)	
C77	0.073 (5)	0.022 (4)	0.041 (4)	-0.006 (3)	-0.012 (4)	-0.009 (3)	
C78	0.083 (6)	0.025 (4)	0.039 (4)	-0.027 (4)	-0.014 (4)	0.002 (3)	
C79	0.072 (5)	0.040 (4)	0.027 (4)	-0.028 (4)	0.004 (3)	-0.007 (3)	
C80	0.054 (4)	0.026 (3)	0.030 (4)	-0.009 (3)	-0.010 (3)	-0.008 (3)	
C81	0.023 (3)	0.016 (3)	0.016 (3)	-0.006 (2)	-0.003 (2)	-0.001 (2)	
C82	0.032 (3)	0.017 (3)	0.025 (3)	-0.011 (3)	-0.006 (3)	-0.005 (3)	
C83	0.048 (4)	0.020 (3)	0.028 (3)	-0.011 (3)	0.000 (3)	-0.009 (3)	
C84	0.059 (4)	0.018 (3)	0.025 (3)	-0.019 (3)	-0.010 (3)	-0.005 (3)	
C85	0.039 (4)	0.030 (3)	0.026 (3)	-0.018 (3)	-0.015 (3)	0.000 (3)	
C86	0.041 (4)	0.019 (3)	0.025 (3)	-0.011 (3)	-0.004 (3)	-0.004 (3)	

Geometric parameters (Å, °)

Cu1—P1	2.2908 (15)	C34—H34	0.9500
Cu1—P2	2.3024 (16)	C35—C36	1.373 (8)
Cu1—S1	2.4148 (16)	С35—Н35	0.9500
Cu1—I1	2.6658 (8)	C36—C37	1.393 (8)
Cu2—P3	2.2876 (15)	С36—Н36	0.9500
Cu2—P4	2.2974 (16)	С37—Н37	0.9500
Cu2—S2	2.3942 (15)	C38—C43	1.385 (8)
Cu2—I2	2.6534 (8)	C38—C39	1.387 (7)
S1—C1	1.701 (5)	C39—C40	1.399 (8)
S2—C44	1.718 (5)	С39—Н39	0.9500
N1—C1	1.329 (6)	C40—C41	1.388 (9)
N1—H1A	0.8800	C40—H40	0.9500
N1—H1B	0.8800	C41—C42	1.358 (9)
N2	1.335 (6)	C41—H41	0.9500
N2—C2	1.438 (7)	C42—C43	1.381 (8)
N2—H2A	0.8800	C42—H42	0.9500
N3—C44	1.328 (6)	C43—H43	0.9500
N3—H3A	0.8800	C45—C46	1.371 (7)
N3—H3B	0.8800	C45—C50	1.382 (7)
N4—C44	1.324 (6)	C46—C47	1.391 (7)
N4—C45	1.434 (6)	C46—H46	0.9500
N4—H4A	0.8800	C47—C48	1.381 (8)
P1—C20	1.824 (5)	C47—H47	0.9500
P1—C14	1.830 (5)	C48—C49	1.380 (8)
P1—C8	1.841 (5)	C48—H48	0.9500
P2—C32	1.833 (5)	C49—C50	1.402 (7)
P2—C26	1.834 (5)	C49—H49	0.9500
P2—C38	1.836 (5)	C50—H50	0.9500
P3—C63	1.821 (5)	C51—C52	1.376 (7)
P3—C51	1.824 (6)	C51—C56	1.383 (7)

P3—C57	1.827 (5)	C52—C53	1.382 (8)
P4—C75	1.830 (5)	С52—Н52	0.9500
P4—C69	1.833 (5)	C53—C54	1.360 (8)
P4—C81	1.834 (5)	С53—Н53	0.9500
C2—C3	1.379 (7)	C54—C55	1.363 (9)
C2—C7	1.396 (7)	C54—H54	0.9500
C3—C4	1.383 (7)	C55—C56	1.386 (8)
C3—H3	0.9500	C55—H55	0.9500
C4-C5	1 373 (8)	C56—H56	0.9500
C4—H4	0.9500	C57—C58	1 393 (7)
C5-C6	1 366 (8)	C57 - C62	1.393(7) 1 407(7)
C5—H5	0.9500	$C_{58} - C_{59}$	1.107(7) 1 382(7)
C6-C7	1 391 (7)	C58—H58	0.9500
С6—Н6	0.9500	C59—C60	1.374(7)
C7—H7	0.9500	C59_H59	0.9500
$C_{8}$	1 385 (7)	C60_C61	1 376 (8)
$C_8 C_{13}$	1.385 (7)	C60 H60	0.9500
$C_{9}$ $C_{10}$	1.380 (7)	C61 - C62	1 381 (7)
C9 H9	0.9500	C61 H61	0.9500
	1 385 (8)	C62 H63	0.9500
C10 H10	0.0500	C62 - C64	1.388(7)
$C_{11}$ $C_{12}$	0.9500	C63 - C68	1.305(7)
C11_U11	1.575 (8)	C64 - C65	1.393(7)
	0.9300	C64 = U64	1.367 (7)
C12 - C13	1.388 (7)		0.9300
C12—H12	0.9500	$C_{00} = C_{00}$	1.383 (8)
С13—Н13	0.9500	С65—Н65	0.9500
	1.391 (7)		1.380 (7)
	1.399 (7)		0.9500
	1.3// (/)	C67—C68	1.3//(/)
	0.9500	C6/—H6/	0.9500
C16—C17	1.389 (7)	C68—H68	0.9500
C16—H16	0.9500	C69—C74	1.392 (7)
	1.384 (8)	C69—C70	1.401 (7)
	0.9500	C/0_C/1	1.381 (8)
C18—C19	1.384 (7)	С70—Н70	0.9500
C18—H18	0.9500	C71—C72	1.383 (8)
С19—Н19	0.9500	C71—H71	0.9500
C20—C25	1.392 (7)	C72—C73	1.387 (8)
C20—C21	1.402 (7)	С72—Н72	0.9500
C21—C22	1.379 (7)	C73—C74	1.388 (7)
C21—H21	0.9500	С73—Н73	0.9500
C22—C23	1.370 (8)	С74—Н74	0.9500
C22—H22	0.9500	C75—C76	1.396 (7)
C23—C24	1.384 (8)	C75—C80	1.398 (7)
С23—Н23	0.9500	C76—C77	1.396 (8)
C24—C25	1.385 (7)	С76—Н76	0.9500
C24—H24	0.9500	С77—С78	1.361 (8)
С25—Н25	0.9500	С77—Н77	0.9500

C26—C27	1.373 (7)	C78—C79	1.373 (9)
C26—C31	1.383 (7)	С78—Н78	0.9500
C27—C28	1.398 (8)	C79—C80	1.383 (8)
С27—Н27	0.9500	С79—Н79	0.9500
C28—C29	1.382 (8)	C80—H80	0.9500
C28—H28	0.9500	C81—C82	1.384 (7)
C29—C30	1.369 (8)	C81—C86	1.395 (7)
С29—Н29	0.9500	C82—C83	1.398 (7)
C30—C31	1.384 (8)	C82—H82	0.9500
C30—H30	0.9500	C83—C84	1.368 (8)
C31—H31	0.9500	C83—H83	0.9500
$C_{32} - C_{37}$	1 393 (7)	C84—C85	1 374 (8)
$C_{32} = C_{33}$	1 401 (7)	C84—H84	0.9500
$C_{33}$ $C_{34}$	1 387 (8)	C85—C86	1 399 (7)
C33_H33	0.9500	C85—H85	0.9500
$C_{34}$ $C_{35}$	1 369 (9)	C86—H86	0.9500
034-035	1.509 (9)	000-1180	0.9500
P1_Cu1_P2	118 63 (5)	C36_C35_H35	110.0
P1Cu1S1	104 71 (6)	$C_{35}$ $C_{35}$ $C_{35}$ $C_{37}$	119.6 (6)
$P_2 = C_{11} = S_1$	111.08 (6)	$C_{35} = C_{36} = C_{37}$	120.2
P1  Cu1  U1	111.00(0) 110.24(4)	C37 C36 H36	120.2
$P_2 = C_{11} = 11$	110.24(4) 100.32(4)	$C_{37} = C_{30} = 1150$	120.2 121.2(5)
12 - Cu1 - 11	100.32 (4) 112.06 (4)	$C_{32} = C_{37} = C_{30}$	121.2(3)
$B_1 = C_{11} = B_1$	112.00(4) 122.18(5)	$C_{32} = C_{37} = H_{37}$	119.4
$P_3 = C_{12} = P_4$	122.18(3)	$C_{30} = C_{3} = C_{30}$	119.4
$P_3 = Cu_2 = S_2$	101.05(5)	$C_{43} = C_{38} = C_{39}$	118.9 (5)
P4—Cu2—S2	109.15 (5)	C43—C38—P2	116.9 (4)
P3—Cu2—I2	104.23 (4)	C39—C38—P2	124.0 (4)
P4—Cu2—I2	109.85 (4)	$C_{38} = C_{39} = C_{40}$	118.7 (6)
S2—Cu2—I2	109.05 (4)	C38—C39—H39	120.6
Cl—Sl—Cul	111.92 (19)	С40—С39—Н39	120.6
C44—S2—Cu2	112.02 (19)	C41—C40—C39	121.5 (6)
C1—N1—H1A	120.0	C41—C40—H40	119.2
C1—N1—H1B	120.0	C39—C40—H40	119.2
H1A—N1—H1B	120.0	C42—C41—C40	119.0 (6)
C1—N2—C2	130.9 (5)	C42—C41—H41	120.5
C1—N2—H2A	114.6	C40—C41—H41	120.5
C2—N2—H2A	114.6	C41—C42—C43	120.3 (7)
C44—N3—H3A	120.0	C41—C42—H42	119.9
C44—N3—H3B	120.0	C43—C42—H42	119.9
H3A—N3—H3B	120.0	C42—C43—C38	121.6 (6)
C44—N4—C45	128.0 (4)	C42—C43—H43	119.2
C44—N4—H4A	116.0	C38—C43—H43	119.2
C45—N4—H4A	116.0	N4—C44—N3	119.9 (5)
C20—P1—C14	103.9 (2)	N4—C44—S2	120.4 (4)
C20—P1—C8	99.8 (2)	N3—C44—S2	119.7 (4)
C14—P1—C8	104.6 (2)	C46—C45—C50	120.6 (5)
C20—P1—Cu1	117.67 (19)	C46—C45—N4	118.2 (5)
C14—P1—Cu1	110.39 (17)	C50—C45—N4	121.0 (5)

C8—P1—Cu1	118.67 (17)	C45—C46—C47	120.2 (5)
C32—P2—C26	102.7 (2)	C45—C46—H46	119.9
C32—P2—C38	105.0 (3)	C47—C46—H46	119.9
C26—P2—C38	101.0 (2)	C48—C47—C46	119.9 (6)
$C_{32} = P_{2} = C_{11}$	111.06(17)	C48—C47—H47	120.1
$C_{26}$ $P_{2}$ $C_{11}$	117 76 (19)	C46-C47-H47	120.1
$C_{20}$ $P_{2}$ $C_{11}$	117.57 (18)	$C_{40}$ $C_{48}$ $C_{47}$	120.1 110.0(5)
$C_{50} = 12 - C_{01}$	102.3(2)	$C_{40} = C_{40} = C_{47}$	119.9 (3)
$C_{03} = 13 = C_{51}$	102.3(2) 104.5(2)	C47 = C48 = 1148	120.1
$C_{03} = F_{3} = C_{37}$	104.3(2) 104.1(2)	C47 - C40 - C40	120.1
$C_{31}$ $P_{3}$ $C_{37}$	104.1(2)	C48 - C49 - C30	120.3 (3)
C63—P3—Cu2	115.18 (17)	C48—C49—H49	119.8
C51—P3—Cu2	115.14 (17)	С50—С49—Н49	119.8
C57—P3—Cu2	114.08 (17)	C45—C50—C49	119.1 (5)
C75—P4—C69	105.1 (2)	C45—C50—H50	120.5
C75—P4—C81	97.9 (2)	С49—С50—Н50	120.5
C69—P4—C81	105.0 (2)	C52—C51—C56	117.1 (5)
C75—P4—Cu2	118.66 (19)	C52—C51—P3	118.1 (4)
C69—P4—Cu2	111.57 (17)	C56—C51—P3	124.8 (4)
C81—P4—Cu2	116.81 (17)	C51—C52—C53	121.5 (6)
N1—C1—N2	119.0 (5)	С51—С52—Н52	119.2
N1—C1—S1	121.2 (4)	С53—С52—Н52	119.2
N2—C1—S1	119.8 (4)	C54—C53—C52	120.5 (6)
C3—C2—C7	121.0 (5)	С54—С53—Н53	119.8
$C_3 - C_2 - N_2$	116 3 (5)	C52—C53—H53	119.8
$C_{7}$ $C_{2}$ $N_{2}$	1224(5)	$C_{52} = C_{53} = H_{55}$	119.1 (6)
$C_{2}$ $C_{3}$ $C_{4}$	122.1(5) 1191(5)	$C_{53}$ $C_{54}$ $H_{54}$	120.5
$C_2 = C_3 = C_4$	120 4	C55 C54 H54	120.5
$C_2 = C_3 = H_3$	120.4	$C_{55} - C_{54} - C_{55} - C_{56}$	120.5
C4 - C3 - H3	120.4	$C_{54} = C_{55} = C_{50}$	120.0 (0)
$C_{3}$	121.1 (0)	С54—С55—Н55	119.7
C5—C4—H4	119.5	С56—С55—Н55	119.7
C3—C4—H4	119.5	C51—C56—C55	121.0 (6)
C6—C5—C4	119.2 (5)	C51—C56—H56	119.5
С6—С5—Н5	120.4	С55—С56—Н56	119.5
C4—C5—H5	120.4	C58—C57—C62	118.1 (5)
C5—C6—C7	121.9 (6)	C58—C57—P3	118.2 (4)
С5—С6—Н6	119.0	C62—C57—P3	123.6 (4)
С7—С6—Н6	119.0	C59—C58—C57	120.8 (5)
C6—C7—C2	117.7 (6)	С59—С58—Н58	119.6
С6—С7—Н7	121.2	С57—С58—Н58	119.6
С2—С7—Н7	121.2	C60—C59—C58	120.5 (6)
C9—C8—C13	119.0 (5)	C60—C59—H59	119.7
C9—C8—P1	117.8 (4)	С58—С59—Н59	119.7
C13—C8—P1	122.9 (4)	C59—C60—C61	119.5 (6)
C10-C9-C8	120.8 (6)	C59—C60—H60	120.2
$C_{10}$ $C_{9}$ $C_{0}$ $H_{9}$	110.6	C61 C60 H60	120.2
	110.6	C60 C61 C62	120.2
$C_0 = C_1 C_1 C_1 C_1 C_1 C_1 C_1 C_1 C_1 C_1$	117.0	$C_{00} = C_{01} = C_{02}$	121.0(3)
$C_{2} = C_{10} = C_{11}$	120.4 (0)		119.5
C9-C10-H10	119.8	C02-C01-H01	119.5

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С11—С10—Н10	119.8	C61—C62—C57	120.0 (5)
C12—C11—C10	118.7 (6)	С61—С62—Н62	120.0
C12—C11—H11	120.7	С57—С62—Н62	120.0
C10—C11—H11	120.7	C64—C63—C68	118.7 (5)
C11—C12—C13	121.5 (6)	C64—C63—P3	123.5 (4)
C11—C12—H12	119.2	C68—C63—P3	117.7 (4)
C13—C12—H12	119.2	C65—C64—C63	120.2 (5)
C8—C13—C12	119.6 (5)	С65—С64—Н64	119.9
С8—С13—Н13	120.2	С63—С64—Н64	119.9
C12—C13—H13	120.2	C66—C65—C64	120.1 (5)
C15—C14—C19	118.1 (5)	С66—С65—Н65	120.0
C15—C14—P1	120.0 (4)	С64—С65—Н65	120.0
C19—C14—P1	121.5 (4)	C67—C66—C65	120.3 (5)
C16—C15—C14	121.0 (5)	С67—С66—Н66	119.9
С16—С15—Н15	119.5	С65—С66—Н66	119.9
C14—C15—H15	119.5	C68—C67—C66	119.5 (5)
C15-C16-C17	120 3 (5)	С68—С67—Н67	120.2
C15—C16—H16	119.8	С66—С67—Н67	120.2
C17—C16—H16	119.8	C67 - C68 - C63	120.2 121.2(5)
C18 - C17 - C16	119.7 (5)	C67 - C68 - H68	119.4
$C_{18} - C_{17} - C_{10}$	120.2	C63 - C68 - H68	119.4
$C_{16} = C_{17} = H_{17}$	120.2	C74 $C69$ $C70$	119.4
$C_{10} = C_{17} = M_{17}$	110.0 (5)	C74 - C69 - C70	118.2(3)
C17 - C18 - C19	119.9 (3)	C70 - C60 - P4	110.2(4)
C10 - C18 - H18	120.1	C71 - C70 - C70	123.1(4)
C19—C18—H18	120.1	C/1 = C/0 = C69	120.4 (5)
C18 - C19 - C14	121.0 (5)	C/I_C/0_H/0	119.8
С18—С19—Н19	119.5	C69—C70—H70	119.8
С14—С19—Н19	119.5	C70—C71—C72	120.9 (5)
C25—C20—C21	118.2 (5)	С70—С71—Н71	119.5
C25—C20—P1	120.0 (4)	С72—С71—Н71	119.5
C21—C20—P1	121.7 (4)	C71—C72—C73	119.4 (5)
C22—C21—C20	120.1 (5)	С71—С72—Н72	120.3
C22—C21—H21	120.0	С73—С72—Н72	120.3
C20—C21—H21	120.0	С72—С73—С74	119.9 (5)
C23—C22—C21	121.2 (5)	С72—С73—Н73	120.1
C23—C22—H22	119.4	С74—С73—Н73	120.1
C21—C22—H22	119.4	C73—C74—C69	121.2 (5)
C22—C23—C24	119.6 (5)	С73—С74—Н74	119.4
С22—С23—Н23	120.2	С69—С74—Н74	119.4
С24—С23—Н23	120.2	C76—C75—C80	118.3 (5)
C23—C24—C25	119.9 (6)	C76—C75—P4	119.9 (4)
C23—C24—H24	120.0	C80—C75—P4	121.6 (4)
C25—C24—H24	120.0	C75—C76—C77	119.7 (6)
C24—C25—C20	121.0 (5)	C75—C76—H76	120.1
$C_{24}$ $C_{25}$ $H_{25}$	119 5	C77—C76—H76	120.1
$C_{20}$ $C_{25}$ $H_{25}$	119.5	C78 - C77 - C76	120.8 (6)
$C_{27}$ $C_{26}$ $C_{31}$	119.1 (5)	C78—C77—H77	119.6
$C_{27} = C_{20} = C_{31}$	117.1(3) 1100(4)	C76 C77 H77	110.6
$C_2 / - C_2 O - \Gamma_2$	117.0 (4)	$C / 0 - C / / - \Pi / /$	117.0

C31—C26—P2	121.7 (4)	С77—С78—С79	120.3 (6)
C26—C27—C28	120.9 (6)	С77—С78—Н78	119.9
С26—С27—Н27	119.6	С79—С78—Н78	119.9
C28—C27—H27	119.6	C78—C79—C80	120.1 (6)
C29—C28—C27	119.2 (6)	C78—C79—H79	120.0
$C_{29}$ $C_{28}$ $H_{28}$	120.4	$C_{80}$ $C_{79}$ $H_{79}$	120.0
$C_{27} C_{28} H_{28}$	120.4	$C_{70} C_{70} C_{70} C_{75}$	120.0
$C_{20} = C_{20} = C_{20}$	120.4	$C_{79} = C_{80} = C_{79}$	120.8 (0)
$C_{30}$ $C_{29}$ $C_{28}$	120.1 (0)	C75 C80 H80	119.0
C30—C29—H29	120.0	C/3 - C80 - H80	119.6
С28—С29—Н29	120.0	C82—C81—C86	119.1 (5)
C29—C30—C31	120.4 (6)	C82—C81—P4	124.6 (4)
С29—С30—Н30	119.8	C86—C81—P4	116.1 (4)
С31—С30—Н30	119.8	C81—C82—C83	119.8 (5)
C26—C31—C30	120.4 (5)	С81—С82—Н82	120.1
С26—С31—Н31	119.8	С83—С82—Н82	120.1
C30—C31—H31	119.8	C84—C83—C82	120.8 (5)
C37—C32—C33	117.9 (5)	С84—С83—Н83	119.6
C37—C32—P2	121.1 (4)	С82—С83—Н83	119.6
$C_{33}$ $C_{32}$ $P_{2}$	1207(4)	C83 - C84 - C85	120.3(5)
$C_{33}$ $C_{32}$ $C_{32}$ $C_{32}$	120.7(4)	$C^{83}$ $C^{84}$ $H^{84}$	110.0
$C_{34} = C_{33} = C_{32}$	120.2 (0)	$C_{00} = C_{00} = 1104$	119.9
$C_{22}$ $C_{22}$ $H_{22}$	119.9	$C_{0}^{0} = C_{0}^{0} = C_{0$	119.9
C32—C33—H33	119.9	$C_{84}$ $C_{85}$ $C_{86}$ $C$	119.6 (5)
C35—C34—C33	120.9 (6)	С84—С85—Н85	120.2
C35—C34—H34	119.6	С86—С85—Н85	120.2
С33—С34—Н34	119.6	C81—C86—C85	120.4 (5)
C34—C35—C36	120.2 (6)	C81—C86—H86	119.8
С34—С35—Н35	119.9	С85—С86—Н86	119.8
P1—Cu1—S1—C1	-155.0(2)	P2-C32-C33-C34	174.0 (5)
P2—Cu1—S1—C1	75.8 (2)	C32—C33—C34—C35	1.1 (10)
I1—Cu1—S1—C1	-35.5(2)	C33—C34—C35—C36	-1.6(10)
P3—Cu2—S2—C44	154.2 (2)	C34—C35—C36—C37	0.9 (10)
$P4-Cu^2-S^2-C44$	-755(2)	$C_{33}$ $C_{32}$ $C_{37}$ $C_{36}$	-0.9(8)
$12 - Cu^2 - S^2 - C44$	445(2)	$P_{2}$ $C_{32}$ $C_{37}$ $C_{36}$	-1747(5)
$P_2 = C_{11} = P_1 = C_{20}$	-49.5(2)	$C_{35}$ $C_{36}$ $C_{37}$ $C_{32}$	0.1(0)
12 - Cu1 - 11 - C20	+9.5(2) -174.04(10)	$C_{33} = C_{30} = C_{37} = C_{32}$	0.4(9)
$S_1 = C_{11} = F_1 = C_{20}$	-1/4.04(19)	$C_{32}$ $F_{2}$ $C_{30}$ $C_{43}$ $C_{43}$	100.0(5)
$\Pi - CuI - PI - C20$	65.3 (2)	$C_{26} = P_2 = C_{38} = C_{43}$	-92.9 (5)
P2—Cu1—P1—C14	69.5 (2)	Cu1—P2—C38—C43	36.6 (5)
S1—Cu1—P1—C14	-55.1 (2)	C32—P2—C38—C39	-25.2 (6)
I1—Cu1—P1—C14	-175.77 (19)	C26—P2—C38—C39	81.3 (6)
P2—Cu1—P1—C8	-169.8 (2)	Cu1—P2—C38—C39	-149.2 (5)
S1—Cu1—P1—C8	65.6 (2)	C43—C38—C39—C40	1.0 (9)
I1—Cu1—P1—C8	-55.1 (2)	P2-C38-C39-C40	-173.1 (5)
P1—Cu1—P2—C32	52.3 (2)	C38—C39—C40—C41	-1.9 (10)
S1—Cu1—P2—C32	173.66 (19)	C39—C40—C41—C42	1.1 (11)
I1—Cu1—P2—C32	-67.70 (19)	C40—C41—C42—C43	0.6 (11)
P1—Cu1—P2—C26	-65.6(2)	C41—C42—C43—C38	-1.5 (11)
S1-Cu1-P2-C26	55 7 (2)	$C_{39}$ $C_{38}$ $C_{43}$ $C_{42}$	06(9)
$51 \ 0 \ 1 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 2 \ 0 \ 0$	55.7 (2)	0.57 0.50 0 - 0 - 0 - 0 - 12	0.0 (7)

I1P2C26	174 36 (19)	P2_C38_C43_C42	1751(5)
$P1_{1}_{1}_{1}_{1}_{1}_{2}_{2}_{2}_{1}_{3}_{3}$	1732(2)	C45 NA C44 N3	-26(8)
$S_1 = C_{11} = P_2 = C_{38}$	-65.5(2)	C45 NA $C44$ S2	177.4(4)
11  Cy1  D2  C28	53.3(2)	$C_{12} = N_1 - C_{14} - S_2$	177.4(4) -12.2(5)
$P_{1} = C_{1} = F_{2} = C_{30}$	55.2 (2)	Cu2 = S2 = C44 = N4	-13.3(3)
P4-Cu2-P3-C63	51.5 (2)	$Cu_2 = S_2 = C_4 = N_3$	100.7 (4)
$S_2 = Cu_2 = P_3 = C_{63}$	1/3.1 (2)	C44—N4—C45—C46	-125.8 (6)
12—Cu2—P3—C63	-73.5(2)	C44—N4—C45—C50	57.6(8)
P4—Cu2—P3—C51	170.3 (2)	C50-C45-C46-C47	-0.4 (8)
S2—Cu2—P3—C51	-68.1 (2)	N4—C45—C46—C47	-177.0(5)
I2—Cu2—P3—C51	45.3 (2)	C45—C46—C47—C48	0.7 (9)
P4—Cu2—P3—C57	-69.39 (19)	C46—C47—C48—C49	-0.6 (9)
S2—Cu2—P3—C57	52.28 (19)	C47—C48—C49—C50	0.3 (9)
I2—Cu2—P3—C57	165.61 (18)	C46—C45—C50—C49	0.1 (8)
P3—Cu2—P4—C75	-48.3 (2)	N4—C45—C50—C49	176.6 (5)
S2—Cu2—P4—C75	-166.35 (19)	C48—C49—C50—C45	0.0 (9)
I2—Cu2—P4—C75	74.13 (19)	C63—P3—C51—C52	176.1 (5)
P3—Cu2—P4—C69	74.04 (19)	C57—P3—C51—C52	-75.3 (5)
S2—Cu2—P4—C69	-44.03 (18)	Cu2—P3—C51—C52	50.3 (5)
I2—Cu2—P4—C69	-163.55 (17)	C63—P3—C51—C56	-4.5 (6)
P3—Cu2—P4—C81	-165.18 (19)	C57—P3—C51—C56	104.1 (5)
S2—Cu2—P4—C81	76.8 (2)	Cu2—P3—C51—C56	-130.2(5)
I2—Cu2—P4—C81	-42.8(2)	C56—C51—C52—C53	-0.2(10)
$C_{2} = N_{2} = C_{1} = N_{1}$	0.8(9)	P3-C51-C52-C53	179.2 (5)
$C_2 = N_2 = C_1 = S_1$	-1797(5)	$C_{51} - C_{52} - C_{53} - C_{54}$	40(11)
Cu1 = S1 = C1 = N1	-1526(4)	$C_{52} = C_{53} = C_{54} = C_{55}$	-49(11)
$C_{\rm H}$ S1 C1 N2	132.0(+)	$C_{52} = C_{53} = C_{54} = C_{55} = C_{56}$	21(11)
$C_{1} = S_{1} = C_{1} = N_{2}$	20.1(5)	$C_{53} = C_{54} = C_{55} = C_{50}$	-26(10)
C1 N2 C2 C7	139.7(0)	$C_{32} = C_{31} = C_{30} = C_{33}$	2.0(10)
C1 - N2 - C2 - C7	-4/.4(9)	$P_{3} = C_{31} = C_{30} = C_{33}$	178.0(3)
$C_{1} = C_{2} = C_{3} = C_{4}$	-1.1(9)	$C_{34} = C_{35} = C_{30} = C_{31}$	1.7(11)
$N_2 - C_2 - C_3 - C_4$	1/1.9 (5)	$C_{63}$ $P_{3}$ $C_{57}$ $C_{58}$	-98.3 (4)
$C_2 - C_3 - C_4 - C_5$	1.1 (9)	C51 - P3 - C57 - C58	154.7 (4)
C3—C4—C5—C6	0.3 (9)	Cu2—P3—C57—C58	28.3 (4)
C4—C5—C6—C7	-1.8 (9)	C63—P3—C57—C62	81.5 (5)
C5—C6—C7—C2	1.8 (9)	C51—P3—C57—C62	-25.6 (5)
C3—C2—C7—C6	-0.3 (9)	Cu2—P3—C57—C62	-151.9 (4)
N2—C2—C7—C6	-172.9 (5)	C62—C57—C58—C59	-2.2 (8)
C20—P1—C8—C9	-99.4 (5)	P3—C57—C58—C59	177.6 (4)
C14—P1—C8—C9	153.3 (4)	C57—C58—C59—C60	3.0 (8)
Cu1—P1—C8—C9	29.8 (5)	C58—C59—C60—C61	-1.4 (8)
C20—P1—C8—C13	74.6 (5)	C59—C60—C61—C62	-1.0 (8)
C14—P1—C8—C13	-32.7 (5)	C60—C61—C62—C57	1.8 (8)
Cu1—P1—C8—C13	-156.3 (4)	C58—C57—C62—C61	-0.2 (7)
C13—C8—C9—C10	-2.4 (9)	P3—C57—C62—C61	-180.0 (4)
P1-C8-C9-C10	171.8 (5)	C51—P3—C63—C64	117.5 (5)
C8—C9—C10—C11	1.4 (10)	C57—P3—C63—C64	9.2 (5)
C9—C10—C11—C12	0.3 (10)	Cu2—P3—C63—C64	-116.8(4)
C10—C11—C12—C13	-0.9 (9)	C51—P3—C63—C68	-66.1 (5)
C9—C8—C13—C12	1.8 (8)	C57—P3—C63—C68	-174.4 (4)
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P1—C8—C13—C12	-172.1 (4)	Cu2—P3—C63—C68	59.6 (5)
C11—C12—C13—C8	-0.2 (9)	C68—C63—C64—C65	-1.1 (8)
C20-P1-C14-C15	-153.0 (4)	P3—C63—C64—C65	175.3 (4)
C8—P1—C14—C15	-48.8 (5)	C63—C64—C65—C66	1.4 (9)
Cu1—P1—C14—C15	80.0 (4)	C64—C65—C66—C67	-0.4 (9)
C20—P1—C14—C19	33.9 (5)	C65—C66—C67—C68	-1.0(9)
C8—P1—C14—C19	138.1 (4)	C66—C67—C68—C63	1.3 (9)
Cu1—P1—C14—C19	-93.2 (4)	C64—C63—C68—C67	-0.3 (8)
C19—C14—C15—C16	0.1 (8)	P3—C63—C68—C67	-176.8(4)
P1-C14-C15-C16	-173.3 (4)	C75—P4—C69—C74	-173.9 (4)
C14—C15—C16—C17	-0.6 (8)	C81—P4—C69—C74	-71.1 (4)
C15—C16—C17—C18	0.6 (8)	Cu2—P4—C69—C74	56.3 (4)
C16—C17—C18—C19	-0.2 (8)	C75—P4—C69—C70	14.3 (5)
C17—C18—C19—C14	-0.3 (8)	C81—P4—C69—C70	117.1 (4)
C15—C14—C19—C18	0.3 (7)	Cu2—P4—C69—C70	-115.5 (4)
P1—C14—C19—C18	173.6 (4)	C74—C69—C70—C71	-0.7 (8)
C14—P1—C20—C25	-146.2(4)	P4—C69—C70—C71	171.1 (4)
C8—P1—C20—C25	106.0 (5)	C69—C70—C71—C72	-0.2(9)
Cu1—P1—C20—C25	-23.8 (5)	C70—C71—C72—C73	1.2 (9)
C14—P1—C20—C21	36.4 (5)	C71—C72—C73—C74	-1.2(8)
C8—P1—C20—C21	-71.5 (5)	C72—C73—C74—C69	0.2 (8)
Cu1—P1—C20—C21	158.8 (4)	C70—C69—C74—C73	0.7 (8)
C25—C20—C21—C22	0.9 (8)	P4—C69—C74—C73	-171.5 (4)
P1—C20—C21—C22	178.4 (4)	C69—P4—C75—C76	-130.9(4)
C20—C21—C22—C23	-1.9 (8)	C81—P4—C75—C76	121.1 (4)
C21—C22—C23—C24	1.7 (9)	Cu2—P4—C75—C76	-5.4 (5)
C22—C23—C24—C25	-0.5 (9)	C69—P4—C75—C80	54.2 (5)
C23—C24—C25—C20	-0.4 (9)	C81—P4—C75—C80	-53.8 (5)
C21—C20—C25—C24	0.2 (8)	Cu2—P4—C75—C80	179.7 (4)
P1-C20-C25-C24	-177.3 (4)	C80—C75—C76—C77	-1.0 (8)
C32—P2—C26—C27	-115.7 (4)	P4—C75—C76—C77	-176.1 (4)
C38—P2—C26—C27	136.1 (4)	C75—C76—C77—C78	0.2 (9)
Cu1—P2—C26—C27	6.7 (5)	C76—C77—C78—C79	0.0 (10)
C32—P2—C26—C31	59.6 (5)	C77—C78—C79—C80	0.5 (10)
C38—P2—C26—C31	-48.7 (5)	C78—C79—C80—C75	-1.4 (9)
Cu1—P2—C26—C31	-178.1 (4)	C76—C75—C80—C79	1.6 (8)
C31—C26—C27—C28	1.2 (8)	P4—C75—C80—C79	176.6 (4)
P2-C26-C27-C28	176.6 (4)	C75—P4—C81—C82	107.0 (5)
C26—C27—C28—C29	-1.1 (9)	C69—P4—C81—C82	-1.0 (5)
C27—C28—C29—C30	1.4 (9)	Cu2—P4—C81—C82	-125.2 (4)
C28—C29—C30—C31	-1.7 (9)	C75—P4—C81—C86	-68.4 (4)
C27—C26—C31—C30	-1.5 (8)	C69—P4—C81—C86	-176.4 (4)
P2-C26-C31-C30	-176.8 (4)	Cu2—P4—C81—C86	59.4 (4)
C29—C30—C31—C26	1.8 (9)	C86—C81—C82—C83	0.6 (8)
C26—P2—C32—C37	-156.6 (4)	P4—C81—C82—C83	-174.7 (4)
C38—P2—C32—C37	-51.4 (5)	C81—C82—C83—C84	0.1 (8)
Cu1—P2—C32—C37	76.7 (5)	C82—C83—C84—C85	-0.1 (8)
C26—P2—C32—C33	29.8 (5)	C83—C84—C85—C86	-0.4 (8)

C38—P2—C32—C33	135.0 (4)	C82—C81—C86—C85	-1.1 (8)
Cu1—P2—C32—C33	-97.0 (4)	P4—C81—C86—C85	174.6 (4)
C37—C32—C33—C34	0.1 (8)	C84—C85—C86—C81	1.1 (8)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	H···A	D···A	D—H···A	
N1—H1 <i>B</i> ····S2 <sup>i</sup>	0.88	2.58	3.455 (5)	173	
N3—H3 <i>B</i> ····S1 <sup>ii</sup>	0.88	2.63	3.396 (5)	146	
N2—H2A…I1	0.88	2.65	3.511 (5)	166	
N4—H4 <i>A</i> …I2	0.88	2.71	3.567 (4)	165	

Symmetry codes: (i) *x*-1, *y*+1, *z*; (ii) *x*+1, *y*-1, *z*.