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# (*S<sub>p</sub>*)-1-Diphenylphosphanyl-2-{(*S*)-[2-(diphenylphosphanyl)phenyl]hydroxymethyl}ferrocene

# Jan W. Bats,<sup>a</sup>\* Angelino Doppiu,<sup>b</sup> Andreas Rivas Nass<sup>b</sup> and A. Stephen K. Hashmi<sup>c</sup>

<sup>a</sup>Institut für Organische Chemie, Universität Frankfurt, Max-von-Laue-Strasse 7, D-60438 Frankfurt am Main, Germany, <sup>b</sup>Umicore AG & Co. KG, Strategic Research and Development, Precious Metals Chemistry, Rodenbacher Chaussee 4, D-63457 Hanau, Germany, and <sup>c</sup>Organisch-Chemisches Institut, Universität Heidelberg, Im Neuenheimer Feld 270, D-69120 Heidelberg, Germany Correspondence e-mail: bats@chemie.uni-frankfurt.de

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Key indicators: single-crystal X-ray study; T = 162 K; mean  $\sigma$ (C–C) = 0.004 Å; R factor = 0.038; wR factor = 0.083; data-to-parameter ratio = 22.3.

The absolute configuration of the title compound,  $[Fe(C_5H_5)-(C_{36}H_{29}OP_2)]$ , is  $S_p$  at the ferrocene group and S at the asymmetric C atom. Both P atoms have a trigonal-pyramidal conformation. There is a short intramolecular  $C-H\cdots P$  contact with an  $H\cdots P$  distance of 2.56 Å. The hydroxy group is involved in an intramolecular  $O-H\cdots \pi_{phenyl}$  interaction. The crystal packing shows five very weak intermolecular  $C-H\cdots \pi$  ontacts, with  $H\cdots Cg$  distances between 3.26 and 3.39 Å (Cg is the centroid of a phenyl or cyclopentadienyl ring).

### **Related literature**

The preparation of the title compound has been reported by Lotz & Spindler (2005). The stereochemistry of the Taniaphos ligand has been discussed by Ireland *et al.* (2008). For the synthesis of related compounds, see: Ireland *et al.* (2002); Fukuzawa, Yamamoto, Hosaka & Kikuchi (2007). For the crystal structures of related compounds, see: Fukuzawa, Yamamoto & Kikuchi (2007); Ireland *et al.* (1999).



V = 1633.4 (3) Å<sup>3</sup>

Mo  $K\alpha$  radiation  $\mu = 0.59 \text{ mm}^{-1}$ 

25293 measured reflections

9131 independent reflections

7926 reflections with  $I > 2\sigma(I)$ 

H atoms treated by a mixture of

independent and constrained

Absolute structure: Flack (1983), 4069 Friedel pairs

Flack parameter: -0.023 (10)

T = 162 (2) K $0.40 \times 0.40 \times 0.32 \text{ mm}$ 

 $R_{\rm int} = 0.051$ 

refinement

 $\begin{array}{l} \Delta \rho_{\rm max} = 0.33 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta \rho_{\rm min} = -0.23 \ {\rm e} \ {\rm \AA}^{-3} \end{array}$ 

Z = 2

### **Experimental**

### Crystal data

Ν

$Fe(C_5H_5)(C_{36}H_{29}OP_2)$ ]	
$A_r = 660.47$	
Aonoclinic, $P2_1$	
a = 11.6111 (15) Å	
P = 8.6154 (10)  Å	
= 16.481 (2) Å	
$B = 97.807 \ (12)^{\circ}$	

## Data collection

Siemens SMART 1K CCD diffractometer Absorption correction: numerical (SHELXTL; Sheldrick, 2008) T<sub>min</sub> = 0.795, T<sub>max</sub> = 0.845

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.038$	
$wR(F^2) = 0.083$	
S = 1.07	
9131 reflections	
410 parameters	
1 restraint	

# Table 1

H	ydrogei	1-bond	geometry	(A,	0,	)
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$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$\begin{array}{c} C11-H11A\cdots P1\\ O1-H1A\cdots C23 \end{array}$	1.00	2.56	3.153 (2)	118
	0.78 (3)	2.51 (3)	3.217 (3)	152 (2)

Data collection: *SMART* (Siemens, 1995); cell refinement: *SMART*; data reduction: *SAINT* (Siemens, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NC2123).

### References

- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Fukuzawa, S., Yamamoto, M., Hosaka, M. & Kikuchi, S. (2007). Eur. J. Org. Chem. pp. 5540–5545.
- Fukuzawa, S., Yamamoto, M. & Kikuchi, S. (2007). J. Org. Chem. 72, 1514– 1517.
- Ireland, T., Grossheimann, G., Wieser-Jeunesse, C. & Knochel, P. (1999). Angew. Chem. Int. Ed. 38, 3212–3215.
- Ireland, T., Grossheimann, G., Wieser-Jeunesse, C. & Knochel, P. (2008). Angew. Chem. Int. Ed. 47, 3666.
- Ireland, T., Tappe, K., Grossheimann, G. & Knochel, P. (2002). Chem. Eur. J. 8, 843–852.
- Lotz, M. & Spindler, F. (2005). WO Patent 2005/108409 A2.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Siemens (1995). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

# supporting information

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# $(S_p)$ -1-Diphenylphosphanyl-2-{(S)-[2-(diphenylphosphanyl)phenyl]hydroxy-methyl}ferrocene

# Jan W. Bats, Angelino Doppiu, Andreas Rivas Nass and A. Stephen K. Hashmi

# S1. Comment

The structure originally published for the Taniaphos ligands, a chiral ligand technology owned by Umicore and sold *via* Solvias, recently had to be corrected. It was shown that these ligands do not have the  $(R,S_p)$  or  $(S,R_p)$  but the  $(R,R_p)$  or  $(S,S_p)$  configuration, respectively (Fukuzawa, Yamamoto, Hosaka & Kikuchi, *et al.*, 2007; Ireland *et al.*, 2008). The planar chirality of the 1,2-disubstituted ferrocene initially had been assigned incorrectly. Our present investigation confirms the configuration of a special member of this family, bearing a hydroxy group in the side-chain of the ferrocene ring (Lotz & Spindler, 2005).

The molecular structure of the title compound is shown in Fig. 1. The ferrocene group deviates only 4° from an eclipsed conformation. The angle between the planes of the two cyclopentadienyl rings is 4.2 (2)°. Both P atoms have a pyramidal conformation. The lone-pair lobe of atom P1 shows a short intramolecular contact distance of 2.56 Å with the H atom of C11 (Table 1). The hydroxy group is not involved in conventional inter- or intramolecular hydrogen bonding. It shows instead a short intramolecular O—H… $\pi_{phenyl}$  interaction with the phenyl ring labeled C18 through C23. The O—H group is not directed to the center of this phenyl ring, but points mainly to atom C23 (Table 1). There is an intramolecular  $\pi$ … $\pi$  interaction between the benzene rings labeled C12  $\rightarrow$  C17 and C36  $\rightarrow$  C41. The angle between the planes of these rings is 4.8 (2)°. The shortest contact distance is 3.498 (3)Å between C12 and C37. The crystal packing shows five very weak intermolecular C—H… $\pi$  interactions, with H…Cg distances between 3.26 and 3.39 Å (Cg is the centroid of a phenyl or cyclopentadienyl ring).

## **S2. Experimental**

The preparation of the title compound has been reported by Lotz & Spindler (2005). Crystals were obtained from a solution of the title compound in a mixture of chloroform and n-hexane.

## **S3. Refinement**

H atoms were geometrically positioned using distances:  $C_{planar}$ —H=0.95 Å,  $C_{primary}$ —H=1.00 Å,  $U_{iso}$ (H)=1.2 $U_{eq}$ (C). The H atom of the hydroxy group was taken from a difference Fourier synthesis and was refined with an isotropic thermal parameter. Friedel opposites were not averaged. The absolute configuration was determined from 4069 Friedel pairs.



# Figure 1

The structure of the title compound shown with 50% probability displacement ellipsoids. The H atoms are drawn as small spheres of arbitrary radius.

# (S<sub>p</sub>)-1-Diphenylphosphanyl-2-{(S)-[2- (diphenylphosphanyl)phenyl]hydroxymethyl}ferrocene

F(000) = 688
$D_{\rm x} = 1.343 {\rm Mg} {\rm m}^{-3}$
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 239 reflections
$\theta = 3-23^{\circ}$
$\mu = 0.59 \text{ mm}^{-1}$
T = 162  K
Block, orange
$0.40 \times 0.40 \times 0.32 \text{ mm}$
25293 measured reflections
9131 independent reflections
7926 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.051$
$\theta_{\rm max} = 30.5^{\circ},  \theta_{\rm min} = 1.8^{\circ}$
$h = -16 \rightarrow 16$
$k = -12 \rightarrow 12$
<i>l</i> = −23→23

Refinement

Refinement on $F^2$	Hydrogen site location: inferred from
Least-squares matrix: full	neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.038$	H atoms treated by a mixture of independent
$wR(F^2) = 0.083$	and constrained refinement
S = 1.07	$w = 1/[\sigma^2(F_o^2) + (0.03P)^2 + 0.5P]$
9131 reflections	where $P = (F_o^2 + 2F_c^2)/3$
410 parameters	$(\Delta/\sigma)_{\rm max} = 0.003$
1 restraint	$\Delta  ho_{ m max} = 0.33$ e Å <sup>-3</sup>
Primary atom site location: structure-invariant	$\Delta \rho_{\rm min} = -0.23 \text{ e} \text{ Å}^{-3}$
direct methods	Absolute structure: Flack (1983), 4069 Friedel
Secondary atom site location: difference Fourier	pairs
map	Absolute structure parameter: -0.023 (10)

## Special details

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of  $F^2$  against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on  $F^2$ , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on  $F^2$  are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\hat{A}^2)$ 

	x	у	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
Fe1	0.19359 (2)	-0.01898 (3)	0.741420 (18)	0.02801 (7)
P1	0.62868 (4)	0.27356 (6)	0.85276 (3)	0.02329 (11)
P2	0.32422 (4)	0.33294 (6)	0.71106 (3)	0.02413 (11)
01	0.47730 (15)	-0.14478 (19)	0.81532 (11)	0.0352 (4)
C1	0.35814 (16)	-0.0009(2)	0.71422 (12)	0.0249 (4)
C2	0.30376 (19)	-0.1365 (3)	0.67651 (15)	0.0319 (5)
H2A	0.3304	-0.2402	0.6853	0.038*
C3	0.2027 (2)	-0.0890 (3)	0.62352 (16)	0.0360 (5)
H3A	0.1499	-0.1559	0.5913	0.043*
C4	0.19426 (19)	0.0751 (3)	0.62691 (14)	0.0310 (5)
H4A	0.1349	0.1367	0.5973	0.037*
C5	0.29067 (17)	0.1323 (2)	0.68279 (12)	0.0244 (4)
C6	0.2085 (2)	-0.0518 (3)	0.86549 (16)	0.0466 (7)
H6A	0.2790	-0.0574	0.9021	0.056*
C7	0.1497 (3)	-0.1778 (4)	0.8229 (2)	0.0585 (9)
H7A	0.1742	-0.2831	0.8260	0.070*
C8	0.0498 (3)	-0.1210 (4)	0.7757 (2)	0.0567 (8)
H8A	-0.0053	-0.1812	0.7413	0.068*
C9	0.0444 (2)	0.0418 (4)	0.78764 (17)	0.0445 (6)
H9A	-0.0144	0.1098	0.7628	0.053*
C10	0.1430 (2)	0.0839 (3)	0.84353 (14)	0.0365 (5)
H10A	0.1619	0.1858	0.8630	0.044*
C11	0.46677 (16)	0.0034 (2)	0.77503 (12)	0.0255 (4)

****			0.01.66	0.001.4
HIIA	0.4583	0.0857	0.8166	0.031*
C12	0.57475 (16)	0.0380 (2)	0.73482 (12)	0.0232 (4)
C13	0.59482 (18)	-0.0521 (2)	0.66787 (13)	0.0306 (5)
H13A	0.5415	-0.1319	0.6489	0.037*
C14	0.69159 (18)	-0.0268 (3)	0.62852 (13)	0.0345 (5)
H14A	0.7051	-0.0905	0.5837	0.041*
C15	0.76792 (19)	0.0911 (3)	0.65477 (14)	0.0341 (5)
H15A	0.8330	0.1109	0.6271	0.041*
C16	0.74938 (18)	0.1810 (3)	0.72182 (13)	0.0272 (4)
H16A	0.8024	0.2621	0.7394	0.033*
C17	0.65449 (17)	0.1546 (2)	0.76401 (12)	0.0224 (4)
C18	0.64664 (17)	0.1267 (2)	0.93453 (12)	0.0247 (4)
C19	0.59727 (19)	0.1590 (3)	1.00546 (13)	0.0309 (5)
H19A	0.5589	0.2553	1.0102	0.037*
C20	0.6035 (2)	0.0530 (3)	1.06872 (14)	0.0380 (5)
H20A	0.5703	0.0774	1.1168	0.046*
C21	0.6582 (2)	-0.0893(3)	1.06242 (14)	0.0345 (5)
H21A	0.6609	-0.1630	1.1055	0.041*
C22	0.70842 (19)	-0.1230(3)	0.99310(13)	0.0312 (5)
H22A	0.7463	-0.2198	0.9888	0.037*
C23	0.70385 (16)	-0.0159(3)	0.92957 (12)	0.0284 (4)
H23A	0.7397	-0.0395	0.8825	0.034*
C24	0.76505 (18)	0.3831 (3)	0.87543 (13)	0.0284(4)
C25	0.8662 (2)	0.3235 (3)	0.91878 (17)	0.0450 (6)
H25A	0.8679	0.2187	0.9369	0.054*
C26	0.9648 (2)	0.4150 (4)	0.93611 (19)	0.0574 (8)
H26A	1.0332	0.3731	0.9663	0.069*
C27	0.9636 (3)	0 5670 (4)	0 90944 (17)	0.0511 (8)
H27A	1.0310	0.6298	0.9215	0.061*
C28	0 8645 (3)	0.6274 (3)	0.86535 (19)	0.0492(7)
H28A	0.8642	0.7313	0.8460	0.059*
C29	0.3012 0.7647 (2)	0.5367 (3)	0.84906 (16)	0.039
H29A	0.6960	0.5798	0.8198	0.0309 (3)
C30	0.0900 0.19045 (17)	0.3790 0.4249(2)	0.65968 (13)	0.074
C31	0.19045(17)	0.4392(3)	0.05900(15) 0.70517(14)	0.0200(4) 0.0334(5)
H31A	0.0981 (2)	0.4392 (3)	0.7612	0.0334 (3)
C32	-0.00866(10)	0.4057 (3)	0.66960 (15)	0.070 0.0300(5)
U32	-0.0713	0.4937 (3)	0.00900 (15)	0.0399 (3)
1152A C22	0.0713	0.5027	0.7010 0.59954 (17)	0.048
	-0.0244(2) -0.0075	0.5420 (5)	0.56654 (17)	0.0431(0) 0.052*
ПЭЭА С24	-0.0975	0.3009	0.3042 0.54212 (16)	$0.032^{\circ}$
C34	0.0073 (2)	0.5515 (4)	0.34313 (10)	0.04/3(/)
H34A	0.0572	0.5635	0.4875	0.05/*
	0.1/429 (18)	0.4/34 (3)	0.57875(13)	0.030/(3)
нээА	0.2309	0.40/0	0.54/5	U.U44 <sup>*</sup>
036	0.43404 (17)	0.3780 (2)	0.64374 (12)	0.0249 (4)
C37	0.45399 (18)	0.2857 (3)	0.57774 (13)	0.0301 (4)
H37A	0.4039	0.2006	0.5620	0.036*
C38	0.54639 (19)	0.3165 (3)	0.53441 (14)	0.0348 (5)

H38A	0.5595	0.2522	0.4898	0.042*	
C39	0.6192 (2)	0.4415 (3)	0.55664 (15)	0.0377 (6)	
H39A	0.6822	0.4630	0.5271	0.045*	
C40	0.6003 (2)	0.5345 (3)	0.62161 (17)	0.0378 (5)	
H40A	0.6501	0.6203	0.6365	0.045*	
C41	0.50852 (17)	0.5034 (2)	0.66566 (14)	0.0301 (4)	
H41A	0.4965	0.5675	0.7107	0.036*	
H1A	0.532 (2)	-0.146 (3)	0.8487 (17)	0.034 (7)*	

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Fe1	0.02189 (13)	0.02740 (15)	0.03497 (16)	-0.00588 (13)	0.00474 (11)	0.00120 (13)
P1	0.0218 (2)	0.0242 (2)	0.0240 (2)	-0.00007 (19)	0.00343 (19)	-0.0001 (2)
P2	0.0264 (2)	0.0262 (3)	0.0196 (2)	-0.0037 (2)	0.00265 (18)	-0.0002(2)
01	0.0310 (8)	0.0341 (9)	0.0392 (9)	-0.0060(7)	0.0002 (7)	0.0144 (7)
C1	0.0224 (8)	0.0251 (10)	0.0276 (9)	-0.0054 (8)	0.0049 (7)	-0.0012 (8)
C2	0.0293 (11)	0.0288 (11)	0.0380 (12)	-0.0044 (9)	0.0062 (9)	-0.0060 (9)
C3	0.0340 (12)	0.0360 (12)	0.0370 (13)	-0.0084 (10)	0.0006 (10)	-0.0089 (10)
C4	0.0283 (11)	0.0328 (12)	0.0300 (11)	-0.0047 (9)	-0.0024 (9)	-0.0023 (9)
C5	0.0230 (9)	0.0262 (10)	0.0241 (10)	-0.0055 (8)	0.0032 (7)	0.0005 (8)
C6	0.0403 (12)	0.063 (2)	0.0393 (13)	0.0053 (12)	0.0143 (10)	0.0198 (12)
C7	0.0666 (19)	0.0405 (15)	0.078 (2)	-0.0039 (14)	0.0434 (17)	0.0172 (15)
C8	0.0426 (15)	0.0584 (19)	0.074 (2)	-0.0285 (14)	0.0247 (14)	-0.0143 (16)
C9	0.0246 (11)	0.0599 (17)	0.0501 (15)	0.0002 (11)	0.0098 (10)	0.0011 (13)
C10	0.0364 (12)	0.0422 (14)	0.0328 (12)	-0.0073 (10)	0.0113 (10)	0.0005 (10)
C11	0.0260 (9)	0.0249 (11)	0.0262 (9)	-0.0024 (8)	0.0053 (7)	0.0023 (8)
C12	0.0219 (9)	0.0245 (9)	0.0230 (9)	0.0006 (7)	0.0022 (7)	0.0039 (7)
C13	0.0302 (10)	0.0309 (12)	0.0304 (11)	-0.0023 (8)	0.0031 (8)	-0.0042 (8)
C14	0.0369 (10)	0.0417 (12)	0.0267 (10)	0.0016 (11)	0.0103 (8)	-0.0050 (11)
C15	0.0276 (10)	0.0458 (13)	0.0311 (11)	0.0004 (9)	0.0120 (9)	0.0007 (10)
C16	0.0238 (10)	0.0305 (11)	0.0280 (10)	-0.0031 (8)	0.0058 (8)	0.0013 (9)
C17	0.0206 (9)	0.0258 (10)	0.0206 (9)	0.0028 (7)	0.0018 (7)	0.0031 (7)
C18	0.0213 (9)	0.0288 (10)	0.0233 (9)	-0.0035 (8)	0.0008 (7)	0.0009 (8)
C19	0.0348 (11)	0.0328 (11)	0.0265 (10)	0.0005 (9)	0.0092 (9)	-0.0043 (9)
C20	0.0468 (13)	0.0450 (13)	0.0246 (11)	0.0002 (11)	0.0140 (10)	-0.0016 (10)
C21	0.0383 (12)	0.0403 (13)	0.0248 (11)	-0.0008 (10)	0.0045 (9)	0.0090 (9)
C22	0.0314 (10)	0.0345 (11)	0.0269 (11)	0.0053 (9)	0.0008 (8)	0.0033 (8)
C23	0.0285 (9)	0.0334 (10)	0.0241 (9)	0.0056 (10)	0.0066 (7)	0.0033 (9)
C24	0.0297 (10)	0.0302 (11)	0.0257 (10)	-0.0069 (8)	0.0056 (8)	-0.0048 (8)
C25	0.0334 (12)	0.0499 (15)	0.0488 (15)	-0.0136 (12)	-0.0047 (10)	0.0138 (12)
C26	0.0378 (14)	0.083 (2)	0.0483 (16)	-0.0260 (15)	-0.0056 (12)	0.0108 (15)
C27	0.0493 (16)	0.0645 (19)	0.0433 (15)	-0.0342 (14)	0.0194 (13)	-0.0207 (14)
C28	0.0579 (17)	0.0324 (13)	0.0644 (18)	-0.0186 (12)	0.0343 (15)	-0.0121 (12)
C29	0.0401 (12)	0.0304 (11)	0.0434 (13)	-0.0024 (9)	0.0167 (10)	-0.0038 (10)
C30	0.0267 (9)	0.0261 (10)	0.0278 (10)	-0.0013 (8)	0.0041 (8)	-0.0010 (8)
C31	0.0383 (12)	0.0312 (12)	0.0330 (11)	0.0020 (9)	0.0130 (9)	0.0015 (9)
C32	0.0330 (11)	0.0404 (13)	0.0489 (14)	0.0020 (11)	0.0149 (10)	0.0006 (12)

# supporting information

C33	0.0271 (11)	0.0507 (15)	0.0502 (15)	0.0051 (10)	0.0009 (10)	0.0023 (12)	
C34	0.0368 (13)	0.071 (2)	0.0337 (13)	0.0077 (12)	0.0017 (10)	0.0095 (12)	
C35	0.0302 (10)	0.0532 (14)	0.0276 (10)	0.0050 (11)	0.0068 (8)	0.0037 (11)	
C36	0.0249 (9)	0.0246 (10)	0.0247 (10)	0.0011 (8)	0.0016 (7)	0.0041 (8)	
C37	0.0311 (10)	0.0330 (11)	0.0256 (10)	-0.0017 (9)	0.0018 (8)	0.0027 (9)	
C38	0.0355 (11)	0.0439 (13)	0.0258 (10)	0.0052 (10)	0.0070 (9)	0.0047 (9)	
C39	0.0322 (11)	0.0399 (14)	0.0432 (13)	0.0044 (9)	0.0136 (10)	0.0164 (10)	
C40	0.0315 (11)	0.0270 (11)	0.0553 (15)	-0.0037 (9)	0.0073 (10)	0.0118 (10)	
C41	0.0283 (9)	0.0239 (11)	0.0379 (11)	0.0009 (8)	0.0042 (8)	0.0041 (9)	

Geometric parameters (Å, °)

Fe1—C1	2.0269 (18)	C16—C17	1.399 (3)
Fe1—C7	2.030 (3)	C16—H16A	0.9500
Fe1—C8	2.033 (3)	C18—C19	1.398 (3)
Fe1—C2	2.044 (2)	C18—C23	1.405 (3)
Fe1—C6	2.048 (3)	C19—C20	1.380 (3)
Fe1—C5	2.049 (2)	C19—H19A	0.9500
Fe1—C3	2.051 (3)	C20—C21	1.391 (3)
Fe1—C9	2.053 (2)	C20—H20A	0.9500
Fe1—C4	2.055 (2)	C21—C22	1.382 (3)
Fe1—C10	2.057 (2)	C21—H21A	0.9500
P1—C24	1.837 (2)	C22—C23	1.391 (3)
P1-C18	1.839 (2)	C22—H22A	0.9500
P1—C17	1.843 (2)	С23—Н23А	0.9500
P2—C5	1.818 (2)	C24—C25	1.387 (3)
P2—C36	1.842 (2)	C24—C29	1.393 (3)
P2-C30	1.844 (2)	C25—C26	1.388 (4)
O1—C11	1.437 (3)	С25—Н25А	0.9500
O1—H1A	0.78 (3)	C26—C27	1.381 (4)
C1—C2	1.429 (3)	C26—H26A	0.9500
C1—C5	1.446 (3)	C27—C28	1.376 (4)
C1—C11	1.501 (3)	С27—Н27А	0.9500
C2—C3	1.424 (3)	C28—C29	1.394 (3)
C2—H2A	0.9500	C28—H28A	0.9500
C3—C4	1.419 (3)	С29—Н29А	0.9500
С3—НЗА	0.9500	C30—C35	1.386 (3)
C4—C5	1.437 (3)	C30—C31	1.395 (3)
C4—H4A	0.9500	C31—C32	1.386 (3)
C6—C10	1.415 (4)	C31—H31A	0.9500
C6—C7	1.416 (5)	C32—C33	1.382 (4)
C6—H6A	0.9500	С32—Н32А	0.9500
C7—C8	1.395 (5)	C33—C34	1.385 (3)
С7—Н7А	0.9500	С33—Н33А	0.9500
C8—C9	1.418 (4)	C34—C35	1.392 (3)
C8—H8A	0.9500	С34—Н34А	0.9500
C9—C10	1.416 (4)	С35—Н35А	0.9500
С9—Н9А	0.9500	C36—C37	1.392 (3)

C10—H10A	0.9500	C36—C41	1.400 (3)
C11—C12	1.525 (3)	C37—C38	1.393 (3)
C11—H11A	1.0000	С37—Н37А	0.9500
C12—C13	1.394 (3)	C38—C39	1.387 (3)
C12—C17	1.406 (3)	C38—H38A	0.9500
C13—C14	1.389 (3)	C39—C40	1.378 (4)
С13—Н13А	0.9500	С39—Н39А	0.9500
C14—C15	1.379 (3)	C40—C41	1.395 (3)
C14—H14A	0.9500	C40—H40A	0.9500
C15—C16	1.390 (3)	C41—H41A	0.9500
C15—H15A	0.9500		
C1—Fe1—C7	121.78 (11)	С8—С9—Н9А	126.4
C1—Fe1—C8	158.24 (11)	Fe1—C9—H9A	126.2
C7—Fe1—C8	40.16 (14)	C6—C10—C9	108.3 (2)
C1—Fe1—C2	41.10 (8)	C6-C10-Fe1	69.50 (15)
C7—Fe1—C2	104.11 (11)	C9-C10-Fe1	69.69 (14)
C8—Fe1—C2	121.82 (11)	C6C10H10A	125.9
C1—Fe1—C6	106.09 (9)	C9—C10—H10A	125.9
C7—Fe1—C6	40.62 (13)	Fe1—C10—H10A	126.5
C8—Fe1—C6	67.92 (13)	01—C11—C1	107.29 (16)
C2—Fe1—C6	118.70 (10)	O1—C11—C12	110.55 (16)
C1—Fe1—C5	41.54 (8)	C1—C11—C12	112.37 (16)
C7—Fe1—C5	160.76 (12)	01—C11—H11A	108.9
C8—Fe1—C5	158.57 (12)	C1—C11—H11A	108.9
C2—Fe1—C5	69.20 (9)	C12—C11—H11A	108.9
C6—Fe1—C5	125.55 (10)	C13—C12—C17	119.70 (18)
C1—Fe1—C3	68.98 (9)	C13—C12—C11	118.09 (18)
C7—Fe1—C3	118.61 (12)	C17—C12—C11	122.20 (17)
C8—Fe1—C3	106.64 (12)	C14—C13—C12	121.0 (2)
C2—Fe1—C3	40.70 (10)	C14—C13—H13A	119.5
C6—Fe1—C3	153.77 (11)	C12—C13—H13A	119.5
C5—Fe1—C3	68.76 (9)	C15—C14—C13	119.7 (2)
C1—Fe1—C9	158.57 (10)	C15—C14—H14A	120.2
C7—Fe1—C9	68.04 (12)	C13—C14—H14A	120.2
C8—Fe1—C9	40.62 (12)	C14—C15—C16	119.85 (19)
C2—Fe1—C9	160.10 (10)	C14—C15—H15A	120.1
C6—Fe1—C9	68.02 (11)	C16—C15—H15A	120.1
C5—Fe1—C9	123.97 (10)	C15—C16—C17	121.5 (2)
C3—Fe1—C9	125.55 (11)	C15—C16—H16A	119.3
C1—Fe1—C4	69.23 (9)	C17—C16—H16A	119.3
C7—Fe1—C4	154.99 (12)	C16—C17—C12	118.18 (18)
C8—Fe1—C4	122.09 (12)	C16—C17—P1	121.89 (16)
C2—Fe1—C4	68.54 (10)	C12—C17—P1	119.88 (14)
C6—Fe1—C4	163.94 (10)	C19—C18—C23	118.26 (19)
C5—Fe1—C4	40.98 (8)	C19—C18—P1	117.21 (17)
C3—Fe1—C4	40.43 (9)	C23—C18—P1	124.51 (15)
C9—Fe1—C4	110.42 (11)	C20—C19—C18	121.0 (2)
			. /

C1Fe1C10	122 04 (9)	C20_C19_H19A	119.5
C7 Fe1 $C10$	67.90(12)	$C_{18}$ $C_{19}$ $H_{19A}$	119.5
$C_{1}^{2} = C_{1}^{2} = C_{1}^{2}$	67.81(11)	$C_{10} = C_{10} = C_{10}$	119.3 120.3(2)
$C_{2}$ $F_{2}$ $C_{10}$	15554(10)	$C_{19} = C_{20} = C_{21}$	120.3 (2)
$C_2$ — $ref$ — $C_{10}$	100.04(10)	$C_{1}$ $C_{20}$ $H_{20A}$	119.0
$C_{0}$ FeI $C_{10}$	40.32(10)	$C_{21} = C_{20} = H_{20A}$	119.6
C3 = FeI = CI0	110.58 (9)	$C_{22} = C_{21} = C_{20}$	119.0 (2)
$C_3$ —FeI—CI0	103.34 (10)	C22—C21—H21A	120.2
C9—Fel—C10	40.29 (10)	C20—C21—H21A	120.2
C4—Fe1—C10	128.20 (10)	C21—C22—C23	120.4 (2)
C24—P1—C18	101.26 (10)	C21—C22—H22A	119.8
C24—P1—C17	102.60 (9)	C23—C22—H22A	119.8
C18—P1—C17	100.64 (9)	C22—C23—C18	120.43 (18)
C5—P2—C36	100.83 (9)	С22—С23—Н23А	119.8
C5—P2—C30	98.80 (10)	C18—C23—H23A	119.8
C36—P2—C30	103.98 (9)	C25—C24—C29	118.6 (2)
C11—O1—H1A	110 (2)	C25—C24—P1	124.09 (18)
C2—C1—C5	107.90 (17)	C29—C24—P1	117.28 (18)
C2—C1—C11	126.3 (2)	C24—C25—C26	120.8 (3)
C5—C1—C11	125.77 (18)	C24—C25—H25A	119.6
C2-C1-Fe1	70.09 (11)	C26—C25—H25A	119.6
C5-C1-Fe1	70.06 (11)	$C_{27}$ $C_{26}$ $C_{25}$	1200(3)
C11-C1-Fe1	125 88 (13)	$C_{27} = C_{26} = H_{26A}$	120.0 (3)
$C_3 - C_2 - C_1$	1081(2)	$C_{25}$ $C_{26}$ $H_{26A}$	120.0
$C_3 = C_2 = C_1$	60.02(13)	$C_{23} = C_{20} = H_{20} R$	120.0 110.0(3)
$C_1 = C_2 = F_{c_1}$	69.92(13)	$C_{28} = C_{27} = C_{20}$	119.9 (3)
$C_1 = C_2 = F_2 C_1$	126.0	$C_{20}$ $C_{27}$ $H_{27}$	120.0
$C_3 = C_2 = H_2 A$	120.0	$C_{20} = C_{2} = C_{20}$	120.0
CI = C2 = H2A	126.0	$C_2/-C_{28}-C_{29}$	120.2 (3)
Fel—C2—H2A	126.9	C27—C28—H28A	119.9
C4—C3—C2	108.5 (2)	С29—С28—Н28А	119.9
C4—C3—Fel	69.94 (15)	C24—C29—C28	120.4 (3)
C2—C3—Fe1	69.38 (14)	С24—С29—Н29А	119.8
С4—С3—НЗА	125.7	С28—С29—Н29А	119.8
С2—С3—НЗА	125.7	C35—C30—C31	118.5 (2)
Fe1—C3—H3A	126.5	C35—C30—P2	124.53 (16)
C3—C4—C5	108.4 (2)	C31—C30—P2	116.88 (17)
C3—C4—Fe1	69.63 (15)	C32—C31—C30	120.8 (2)
C5—C4—Fe1	69.30 (12)	С32—С31—Н31А	119.6
C3—C4—H4A	125.8	C30—C31—H31A	119.6
C5—C4—H4A	125.8	C33—C32—C31	120.2 (2)
Fe1—C4—H4A	126.8	С33—С32—Н32А	119.9
C4—C5—C1	107.12 (18)	С31—С32—Н32А	119.9
C4-C5-P2	127 69 (17)	$C_{32}$ $C_{33}$ $C_{34}$	119 5 (2)
C1 - C5 - P2	127.09(17) 125.19(15)	$C_{32}$ $C_{33}$ $H_{33A}$	120.2
C4-C5-Fe1	69 72 (12)	C34_C33_H33A	120.2
$C1 = C5 = Fe^1$	68 39 (11)	$C_{33}$ $C_{34}$ $C_{35}$	120.2 120.2(2)
$C_1 = C_2 = C_1$ $D_2 = C_2 = E_2 1$	126.60 (11)	$C_{33} = C_{34} = C_{35}$	120.2(2)
12 - 03 - 101	120.00(11) 107.5(2)	$C_{25} = C_{24} = H_{24} \wedge H$	117.7
$C_{10} = C_{0} = C_{10}$	10/.3(2)	$C_{33} = C_{34} = C_{34}$	119.9
C10-C6-Fel	/0.18 (14)	0.30-0.35-0.34	120.7 (2)

C7—C6—Fe1	69.01 (16)	C30—C35—H35A	1197
C10-C6-H6A	126.2	C34—C35—H35A	119.7
C7-C6-H6A	126.2	$C_{37} - C_{36} - C_{41}$	118 68 (19)
$E_{\rm el}$ $C_{\rm el}$ $H_{\rm el}$	126.1	$C_{37}$ $C_{36}$ $P_{2}$	123 54 (16)
C8-C7-C6	108.4(3)	$C_{41}$ $C_{36}$ $P_{2}$	123.34(10) 117.47(16)
$C_{8}$ $C_{7}$ $E_{81}$	70.03(17)	$C_{1}^{2} = C_{30}^{2} = 12$	117.47(10) 120.9(2)
C6 = C7 = Fe1	70.03(17) 70.37(15)	$C_{30} = C_{37} = C_{38}$	120.9 (2)
	125.8	$C_{30} = C_{37} = H_{37A}$	119.0
$C_{0} = C_{1} = H_{1}$	125.0	$C_{30} = C_{37} = H_{37} = H_{37}$	119.0
$C_0 - C_1 - H_1 A$	125.0	$C_{39} = C_{30} = C_{37}$	119.8 (2)
$\frac{1}{1} \frac{1}{1} \frac{1}$	123.4	C37 C28 H28A	120.1
$C_{7} = C_{8} = C_{9}$	106.0(3)	$C_{3} = C_{30} = C_{30}$	120.1
	09.81(13)	C40 - C39 - C38	120.0 (2)
C9—C8—Fel	/0.43 (15)	C28 C20 H20A	120.0
C = C = H = A	125.7	C38—C39—H39A	120.0
C9—C8—H8A	125.7	$C_{39} - C_{40} - C_{41}$	120.4 (2)
Fel—C8—H8A	125.6	C39—C40—H40A	119.8
C10—C9—C8	107.2 (3)	C41—C40—H40A	119.8
C10—C9—Fe1	70.02 (14)	C40—C41—C36	120.2 (2)
C8—C9—Fe1	68.95 (16)	C40—C41—H41A	119.9
С10—С9—Н9А	126.4	C36—C41—H41A	119.9
	74.00 (10)		102.02 (10)
C/-FeI-CI-C2	/4.09 (18)	$C_2$ —FeI— $C_7$ — $C_8$	-123.03(18)
C8—FeI— $CI$ — $C2$	44.9 (3)	$C_6$ —FeI— $C_7$ — $C_8$	119.1 (3)
C6—Fel— $C1$ — $C2$	115.44 (15)	C5—Fel— $C7$ — $C8$	170.0 (3)
C5—FeI— $C1$ — $C2$	-118.63 (17)	C3—Fel—C7—C8	-81.9 (2)
C3—Fe1—C1—C2	-37.34 (14)	C9—Fe1—C7—C8	37.71 (18)
C9—Fe1—C1—C2	-173.9 (3)	C4—Fe1—C7—C8	-53.8 (3)
C4—Fe1—C1—C2	-80.76 (14)	C10—Fe1—C7—C8	81.33 (19)
C10—Fe1—C1—C2	156.36 (14)	C1—Fe1—C7—C6	77.20 (18)
C7—Fe1—C1—C5	-167.29 (15)	C8—Fe1—C7—C6	-119.1 (3)
C8—Fe1—C1—C5	163.5 (3)	C2—Fe1—C7—C6	117.87 (16)
C2—Fe1—C1—C5	118.63 (17)	C5—Fe1—C7—C6	50.9 (4)
C6—Fe1—C1—C5	-125.93 (13)	C3—Fe1—C7—C6	158.99 (15)
C3—Fe1—C1—C5	81.28 (13)	C9—Fe1—C7—C6	-81.39 (17)
C9—Fe1—C1—C5	-55.3 (3)	C4—Fe1—C7—C6	-172.8 (2)
C4—Fe1—C1—C5	37.87 (12)	C10—Fe1—C7—C6	-37.76 (15)
C10—Fe1—C1—C5	-85.01 (14)	C6—C7—C8—C9	0.2 (3)
C7—Fe1—C1—C11	-47.0 (2)	Fe1—C7—C8—C9	-60.0 (2)
C8—Fe1—C1—C11	-76.2 (4)	C6—C7—C8—Fe1	60.16 (18)
C2—Fe1—C1—C11	-121.0 (2)	C1—Fe1—C8—C7	40.0 (4)
C6—Fe1—C1—C11	-5.6 (2)	C2—Fe1—C8—C7	73.1 (2)
C5—Fe1—C1—C11	120.3 (2)	C6—Fe1—C8—C7	-37.87 (19)
C3—Fe1—C1—C11	-158.4 (2)	C5—Fe1—C8—C7	-171.0 (3)
C9—Fe1—C1—C11	65.1 (4)	C3—Fe1—C8—C7	114.9 (2)
C4—Fe1—C1—C11	158.2 (2)	C9—Fe1—C8—C7	-119.4 (3)
C10—Fe1—C1—C11	35.3 (2)	C4—Fe1—C8—C7	156.26 (18)
C5—C1—C2—C3	-1.0(2)	C10—Fe1—C8—C7	-81.6 (2)
C11—C1—C2—C3	179.57 (19)	C1—Fe1—C8—C9	159.4 (2)
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Fe1—C1—C2—C3	59.08 (16)	C7—Fe1—C8—C9	119.4 (3)
C5-C1-C2-Fe1	-60.13 (13)	C2—Fe1—C8—C9	-167.49 (16)
C11-C1-C2-Fe1	120.49 (19)	C6—Fe1—C8—C9	81.5 (2)
C1—Fe1—C2—C3	-119.74 (19)	C5—Fe1—C8—C9	-51.6 (4)
C7—Fe1—C2—C3	117.71 (17)	C3—Fe1—C8—C9	-125.72 (19)
C8—Fe1—C2—C3	78.19 (19)	C4—Fe1—C8—C9	-84.3 (2)
C6—Fe1—C2—C3	158.71 (15)	C10—Fe1—C8—C9	37.84 (17)
C5—Fe1—C2—C3	-81.22 (15)	C7—C8—C9—C10	-0.2(3)
C9—Fe1—C2—C3	53.7 (4)	Fe1—C8—C9—C10	-59.81 (17)
C4—Fe1—C2—C3	-37.14 (14)	C7—C8—C9—Fe1	59.61 (19)
C10—Fe1—C2—C3	-174.9(2)	C1—Fe1—C9—C10	-40.6(3)
C7—Fe1—C2—C1	-122.56 (16)	C7—Fe1—C9—C10	81.27 (19)
C8 - Fe1 - C2 - C1	-162.07(16)	C8 - Fe1 - C9 - C10	1186(3)
C6-Fe1-C2-C1	-81.55(16)	$C_{2}$ Fe1 - C9 - C10	151.3 (3)
$C_{5}$ Fe1 $C_{2}$ $C_{1}$	38 52 (12)	C6-Fe1-C9-C10	37 31 (16)
$C_3 = F_{e1} = C_2 = C_1$	119 74 (19)	C5-Fe1-C9-C10	-81.64(17)
C9-Fe1-C2-C1	173 5 (3)	$C_{3}$ Fe1 $C_{9}$ $C_{10}$	-16846(15)
C4-Fe1-C2-C1	82 59 (14)	C4—Fe1—C9—C10	-12554(15)
C10—Fe1—C2—C1	-552(3)	C1—Fe1—C9—C8	-1591(3)
C1 - C2 - C3 - C4	0.7(3)	C7—Fe1—C9—C8	-373(2)
Fe1-C2-C3-C4	59 15 (18)	$C_{2}$ Fe1 $C_{2}$ $C_{2}$ $C_{2}$ $C_{3}$ $C_{4}$ $C_{4}$	37.5(2)
C1-C2-C3-Fe1	-5840(15)	C6-Fe1-C9-C8	-81.2(2)
C1 - Fe1 - C3 - C4	-82.25(15)	$C_{5}$ Fe1 $C_{9}$ $C_{8}$	159.80(19)
C7—Fe1—C3—C4	162.08 (16)	$C_{3}$ Fe1 $C_{9}$ $C_{8}$	73.0(2)
C8 - Fe1 - C3 - C4	120.29(17)	C4-Fe1-C9-C8	1159(2)
$C_{2}$ Fe1 $C_{3}$ $C_{4}$	-120.29(17)	C10—Fe1—C9—C8	-1186(3)
C6-Fe1-C3-C4	-166.0(2)	C7 - C6 - C10 - C9	-0.1(3)
$C_{5}$ Fe1 $C_{3}$ $C_{4}$	-37.56(14)	$F_{e1}$ $-C_{e10}$	59.08 (16)
C9-Fe1-C3-C4	79 76 (18)	C7 - C6 - C10 - Fe1	-59.14(17)
C10-Fe1-C3-C4	52 6 (4)	$C_{8} - C_{9} - C_{10} - C_{6}$	0.2(3)
$C_1$ Fel $C_3$ $C_2$	37.70(13)	$F_{e1} = C_{e1} = C_{e1} = C_{e1}$	-58.97(16)
$C7_{Fe1}$	-77.97(17)	$C_{8} = C_{9} = C_{10} = C_{0}$	59 13 (19)
$C_{1}^{2} = C_{1}^{2} = C_{2}^{2} = C_{2}^{2}$	-110 76 (16)	$C_1 = C_1 = C_1 = C_1$	-7658(17)
C6-Fe1-C3-C2	-461(3)	C7 - Fe1 - C10 - C6	70.30(17)
$C_{0} = 101 = C_{0} = C_{2}$	40.1(5) 82 30 (14)	$C_{1}^{2} = C_{1}^{2} = C_{1$	38.04 (17) 81.55 (19)
$C_{3}$ $C_{1}$ $C_{3}$ $C_{2}$	-160.29(14)	$C_{2} = C_{10} = C_{10} = C_{0}$	-370(3)
$C_{2} = 101 = 0.000$	100.29(14) 120.0(2)	$C_2 = F_{c1} = C_{10} = C_0$	-121.39(15)
$C_{+-1}C_{1} - C_{3} - C_{2}$	120.0(2) 172.5(3)	$C_{3} = F_{e1} = C_{10} = C_{0}$	121.39(13) 1548(3)
$C_{10}^{-1}$	-0.2(3)	$C_{3}$ $C_{10}$ $C_$	134.8(3)
$C_2 - C_3 - C_4 - C_5$	0.2 (3) 58 64 (16)	$C_{4} = C_{10} = C_{10} = C_{10}$	-164.20(15)
$C_{1}^{2} = C_{1}^{2} = C_{4}^{2} = C_{5}^{2}$	-58.04(10)	$C_{1} = C_{1} = C_{10} = C_{0}$	164.29(13)
$C_2 = C_3 = C_4 = Fer$	-36.60(16)	$C_{1} = F_{0} = C_{10} = C_{9}$	-81.66(10)
C1 - Fe1 - C4 - C3	-20.7(2)	$C^{2} = C^{2} = C^{2$	-31.00(19) -38.14(18)
$C^{2} = Fe_{1} = C_{4} = C_{3}$	-39.7(3)	$C_{0} = F_{0} = C_{10} = C_{9}$	-36.14(18) -156.7(2)
$C_{2} = C_{1} = C_{4} = C_{2}$	(17) 27 28 (14)	$C_{2} = C_{1} = C_{10} = C_{9}$	130.7(2)
$C_2$ —rei— $C_4$ — $C_5$	37.30(14) 157.2(2)	$C_{0} = C_{1} = C_{1} = C_{1} = C_{1} = C_{2}$	119.7(2)
$C_{0} = Fe_{1} = C_{4} = C_{3}$	13/.3(3) 1200(2)	$C_{3}$ = Fe1 = C10 = C9	118.92 (10)
$C_{3} = Fe_{1} = C_{4} = C_{3}$	120.0(2) 121.22(10)	$C_{4} = C_{10} = C_{10} = C_{9}$	33.1(4)
C9—re1—C4—C3	-121.32(10)	C4—re1—C10—C9	/0.01 (18)

C10—Fe1—C4—C3	-163.36 (15)	C2-C1-C11-O1	-25.1 (3)
C1—Fe1—C4—C5	-38.37 (12)	C5-C1-C11-O1	155.63 (18)
C7—Fe1—C4—C5	-159.7 (2)	Fe1-C1-C11-O1	65.5 (2)
C8—Fe1—C4—C5	162.46 (15)	C2-C1-C11-C12	96.6 (2)
C2—Fe1—C4—C5	-82.57 (14)	C5-C1-C11-C12	-82.7 (2)
C6—Fe1—C4—C5	37.4 (4)	Fe1—C1—C11—C12	-172.79 (14)
C3—Fe1—C4—C5	-120.0 (2)	O1—C11—C12—C13	68.2 (2)
C9—Fe1—C4—C5	118.72 (14)	C1-C11-C12-C13	-51.7 (3)
C10—Fe1—C4—C5	76.69 (16)	O1—C11—C12—C17	-110.6(2)
C3—C4—C5—C1	-0.5 (3)	C1—C11—C12—C17	129.5 (2)
Fe1-C4-C5-C1	58.36 (13)	C17—C12—C13—C14	-1.1 (3)
C3—C4—C5—P2	179.97 (17)	C11—C12—C13—C14	-179.9 (2)
Fe1—C4—C5—P2	-121.18 (17)	C12—C13—C14—C15	-1.4 (4)
C3—C4—C5—Fe1	-58.85 (18)	C13—C14—C15—C16	1.8 (4)
C2-C1-C5-C4	0.9 (2)	C14—C15—C16—C17	0.1 (3)
C11—C1—C5—C4	-179.67 (18)	C15-C16-C17-C12	-2.5 (3)
Fe1—C1—C5—C4	-59.20 (14)	C15-C16-C17-P1	-179.75 (17)
C2—C1—C5—P2	-179.50 (16)	C13—C12—C17—C16	3.0 (3)
C11—C1—C5—P2	-0.1 (3)	C11—C12—C17—C16	-178.27 (18)
Fe1—C1—C5—P2	120.36 (16)	C13—C12—C17—P1	-179.75 (16)
C2-C1-C5-Fe1	60.14 (13)	C11—C12—C17—P1	-1.0 (3)
C11-C1-C5-Fe1	-120.47 (18)	C24—P1—C17—C16	-11.30 (19)
C36—P2—C5—C4	-98.5 (2)	C18—P1—C17—C16	-115.52 (17)
C30—P2—C5—C4	7.7 (2)	C24—P1—C17—C12	171.52 (16)
C36—P2—C5—C1	82.08 (18)	C18—P1—C17—C12	67.30 (17)
C30—P2—C5—C1	-171.75 (17)	C24—P1—C18—C19	95.15 (18)
C36—P2—C5—Fe1	169.93 (13)	C17—P1—C18—C19	-159.55 (17)
C30-P2-C5-Fe1	-83.90 (14)	C24—P1—C18—C23	-86.39 (19)
C1—Fe1—C5—C4	118.94 (17)	C17—P1—C18—C23	18.90 (19)
C7—Fe1—C5—C4	153.5 (3)	C23-C18-C19-C20	-0.6 (3)
C8—Fe1—C5—C4	-44.3 (3)	P1-C18-C19-C20	177.94 (18)
C2—Fe1—C5—C4	80.82 (14)	C18—C19—C20—C21	-0.8 (4)
C6—Fe1—C5—C4	-168.09 (14)	C19—C20—C21—C22	1.3 (4)
C3—Fe1—C5—C4	37.08 (14)	C20—C21—C22—C23	-0.5 (4)
C9—Fe1—C5—C4	-82.29 (16)	C21—C22—C23—C18	-1.0 (3)
C10—Fe1—C5—C4	-125.33 (14)	C19—C18—C23—C22	1.5 (3)
C7—Fe1—C5—C1	34.6 (4)	P1-C18-C23-C22	-176.96 (16)
C8—Fe1—C5—C1	-163.3 (3)	C18—P1—C24—C25	23.5 (2)
C2—Fe1—C5—C1	-38.12 (12)	C17—P1—C24—C25	-80.3 (2)
C6—Fe1—C5—C1	72.98 (15)	C18—P1—C24—C29	-154.66 (17)
C3—Fe1—C5—C1	-81.86 (13)	C17—P1—C24—C29	101.60 (18)
C9—Fe1—C5—C1	158.78 (13)	C29—C24—C25—C26	0.4 (4)
C4—Fe1—C5—C1	-118.94 (17)	P1-C24-C25-C26	-177.7 (2)
C10—Fe1—C5—C1	115.73 (13)	C24—C25—C26—C27	-0.6 (4)
C1—Fe1—C5—P2	-118.55 (19)	C25—C26—C27—C28	-0.3 (4)
C7—Fe1—C5—P2	-84.0 (4)	C26—C27—C28—C29	1.4 (4)
C8—Fe1—C5—P2	78.2 (3)	C25—C24—C29—C28	0.7 (3)
C2—Fe1—C5—P2	-156.66 (18)	P1—C24—C29—C28	178.92 (19)

C6—Fe1—C5—P2	-4557(19)	C27—C28—C29—C24	-1.6(4)
C3—Fe1—C5—P2	159.59 (17)	C5 - P2 - C30 - C35	-87.0 (2)
C9—Fe1—C5—P2	40.23 (18)	C36—P2—C30—C35	16.6 (2)
C4—Fe1—C5—P2	122.5 (2)	C5—P2—C30—C31	89.93 (18)
C10—Fe1—C5—P2	-2.82 (17)	C36—P2—C30—C31	-166.51 (17)
C1—Fe1—C6—C10	120.89 (14)	C35—C30—C31—C32	2.1 (3)
C7—Fe1—C6—C10	-118.7 (2)	P2-C30-C31-C32	-174.96 (19)
C8—Fe1—C6—C10	-81.27 (17)	C30—C31—C32—C33	-1.4 (4)
C2—Fe1—C6—C10	163.48 (14)	C31—C32—C33—C34	0.1 (4)
C5—Fe1—C6—C10	79.59 (16)	C32—C33—C34—C35	0.4 (4)
C3—Fe1—C6—C10	-164.1 (2)	C31—C30—C35—C34	-1.6 (4)
C9—Fe1—C6—C10	-37.29 (15)	P2-C30-C35-C34	175.2 (2)
C4—Fe1—C6—C10	50.3 (4)	C33—C34—C35—C30	0.4 (4)
C1—Fe1—C6—C7	-120.38 (17)	C5—P2—C36—C37	14.0 (2)
C8—Fe1—C6—C7	37.46 (18)	C30—P2—C36—C37	-88.04 (19)
C2—Fe1—C6—C7	-77.79 (18)	C5—P2—C36—C41	-159.45 (16)
C5—Fe1—C6—C7	-161.68 (16)	C30—P2—C36—C41	98.54 (17)
C3—Fe1—C6—C7	-45.4 (3)	C41—C36—C37—C38	0.3 (3)
C9—Fe1—C6—C7	81.44 (18)	P2-C36-C37-C38	-173.05 (17)
C4—Fe1—C6—C7	169.0 (3)	C36—C37—C38—C39	-0.5 (3)
C10—Fe1—C6—C7	118.7 (2)	C37—C38—C39—C40	0.2 (3)
C10—C6—C7—C8	-0.1 (3)	C38—C39—C40—C41	0.3 (4)
Fe1—C6—C7—C8	-59.9 (2)	C39—C40—C41—C36	-0.6 (3)
C10-C6-C7-Fe1	59.88 (16)	C37—C36—C41—C40	0.3 (3)
C1—Fe1—C7—C8	-163.71 (16)	P2—C36—C41—C40	174.00 (17)

# Hydrogen-bond geometry (Å, °)

D—H···A	D—H	Н…А	D····A	D—H…A
С11—Н11А…Р1	1.00	2.56	3.153 (2)	118
O1—H1A···C23	0.78 (3)	2.51 (3)	3.217 (3)	152 (2)