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## Structure Reports

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## 1-Isopropylideneamino-1H-tetrazol-5-amine

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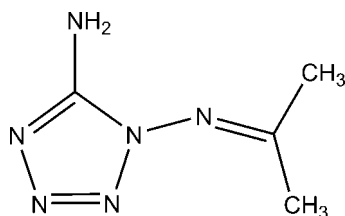
Received 25 June 2009; accepted 29 June 2009

Key indicators: single-crystal X-ray study;  $T = 93$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.036;  $wR$  factor = 0.088; data-to-parameter ratio = 15.0.

The molecule of the title compound,  $\text{C}_4\text{H}_8\text{N}_6$ , assumes an approximately planar structure, the methyl C atoms and the C atom to which they are bonded being out of the mean tetrazole ring plane by 0.108 and 0.139, and 0.144 Å, respectively.  $\pi$ - $\pi$  stacking between parallel tetrazole rings [centroid-centroid distance = 3.4663 (11) Å] is observed in the crystal structure. Intermolecular N-H...N hydrogen bonding further helps to stabilize the crystal structure.

## Related literature

For the preparation of the title compound, see: Gaponnik & Karavai (1984). For general background, see: Galvez-Ruiz *et al.* (2005); Joo *et al.* (2008). For a related structures, see: Lyakhov *et al.* (2005).



## Experimental

## Crystal data

 $\text{C}_4\text{H}_8\text{N}_6$  $M_r = 140.16$ 

Monoclinic,  $P2_1/c$   
 $a = 7.488$  (2) Å  
 $b = 7.4238$  (19) Å  
 $c = 11.997$  (3) Å  
 $\beta = 97.145$  (3)°  
 $V = 661.7$  (3) Å<sup>3</sup>

$Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 93$  K  
 $0.43 \times 0.43 \times 0.33$  mm

## Data collection

Rigaku Saturn724+ diffractometer  
Absorption correction: none  
5172 measured reflections

1520 independent reflections  
1334 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.024$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.088$   
 $S = 1.00$   
1520 reflections  
101 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.28$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.17$  e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N6}-\text{H6A}\cdots\text{N2}^{\text{i}}$	0.890 (16)	2.200 (17)	3.0600 (16)	162.6 (13)
$\text{N6}-\text{H6B}\cdots\text{N1}^{\text{ii}}$	0.876 (15)	2.118 (15)	2.9770 (16)	166.4 (14)

Symmetry codes: (i)  $x, -y + \frac{1}{2}, z + \frac{1}{2}$ ; (ii)  $-x + 2, -y, -z + 1$ .

Data collection: *CrystalClear* (Rigaku, 2008); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2546).

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## supporting information

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## 1-Isopropylideneamino-1*H*-tetrazol-5-amine

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### S1. Comment

1,5-diaminotetrazole and its derivatives have received an increasing interest as a class of nitrogen-rich energetic materials during the last years. These compounds exhibit prospective application in generation of gases, propellants and other combustible and thermally decomposing systems (Galvez-Ruiz *et al.*, 2005; Joo, *et al.* 2008). The title compound had been prepared by Gaponnik & Karavai, but its crystal structure hadn't been reported, therefore, the structure of the title compound has been determined in our present work.

The crystal structure of the title compound is presented in Fig. 1, The bond distances and bond angles in the title compound are similar to the corresponding distances and angles reported by Lyakhov *et al.* (2005). The molecule is almost planar with only C2, C3 and C4 being out of the mean plane of the tetrazole ring by 0.108, 0.144 and 0.139 Å, respectively.

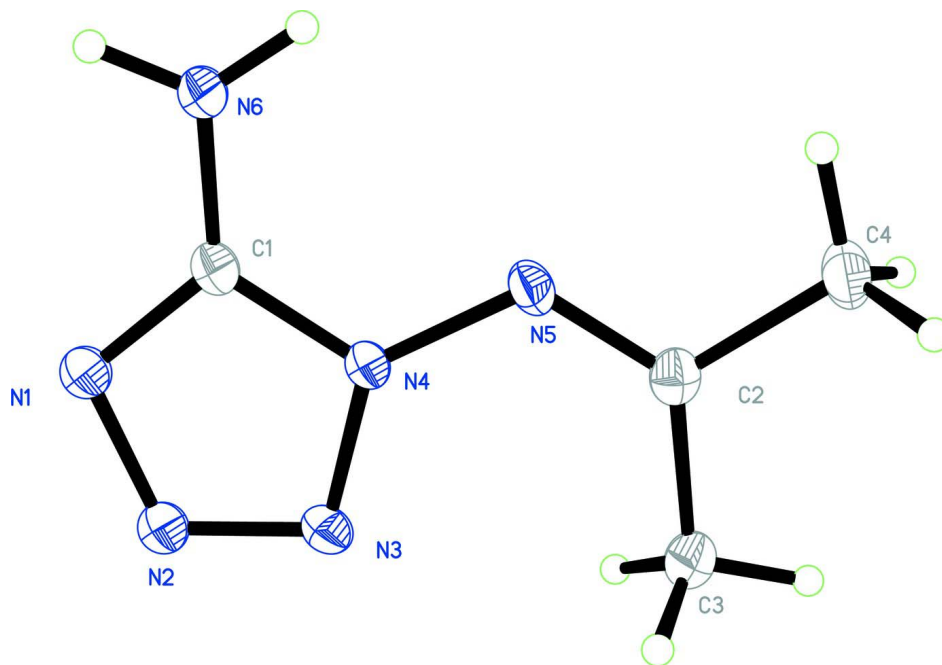
In the crystal structure the molecules are linked to each other via N—H $\cdots$ N hydrogen bonding (Table 1), forming a three dimensional network structure). The offset face-to-face  $\pi$ - $\pi$  contact between the tetrazole rings, related by an inversion center, further helps to stabilize the crystal structure; centroid-centroid distance being 3.4663 (11) Å.

### S2. Experimental

The title compound was prepared according to the literature method (Gaponnik & Karavai, 1984). The purity of the compound was checked by determining its melting point, m.p. 445–446 K. Crystals suitable for X-ray structure determination were obtained by slow evaporation of an acetone solution at room temperature.

### S3. Refinement

Amino H atoms were located in a difference Fourier maps and were refined isotropically. Other H-atoms were placed in calculated positions with C—H = 0.98 Å, and refined in riding mode with  $U_{\text{iso}} = 1.2U_{\text{eq}}(\text{C})$ .

**Figure 1**

The molecular structure of the title compound with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

### 1-Isopropylideneamino-1H<i>t</i>-tetrazol-5-amine

#### Crystal data

$C_4H_8N_6$   
 $M_r = 140.16$   
 Monoclinic,  $P2_1/c$   
 Hall symbol:  $-P\ 2ybc$   
 $a = 7.488\ (2)\ \text{\AA}$   
 $b = 7.4238\ (19)\ \text{\AA}$   
 $c = 11.997\ (3)\ \text{\AA}$   
 $\beta = 97.145\ (3)^\circ$   
 $V = 661.7\ (3)\ \text{\AA}^3$   
 $Z = 4$

$F(000) = 296$   
 $D_x = 1.407\ \text{Mg m}^{-3}$   
 Melting point: 445 K  
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$   
 Cell parameters from 2076 reflections  
 $\theta = 3.1\text{--}27.5^\circ$   
 $\mu = 0.10\ \text{mm}^{-1}$   
 $T = 93\ \text{K}$   
 Block, colourless  
 $0.43 \times 0.43 \times 0.33\ \text{mm}$

#### Data collection

Rigaku Saturn724+  
 diffractometer  
 Radiation source: Rotating Anode  
 Graphite monochromator  
 Detector resolution:  $28.5714\ \text{pixels mm}^{-1}$   
 multi-scan  
 5172 measured reflections

1520 independent reflections  
 1334 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.024$   
 $\theta_{\text{max}} = 27.5^\circ$ ,  $\theta_{\text{min}} = 3.2^\circ$   
 $h = -9 \rightarrow 9$   
 $k = -8 \rightarrow 9$   
 $l = -15 \rightarrow 15$

#### Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.088$

$S = 1.00$   
 1520 reflections  
 101 parameters  
 0 restraints

H atoms treated by a mixture of independent  
and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.045P)^2 + 0.16P]$   
where  $P = (F_o^2 + 2F_c^2)/3$

$$(\Delta/\sigma)_{\max} < 0.001$$

$$\Delta\rho_{\max} = 0.28 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$$

#### Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

#### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.90734 (13)	0.19832 (13)	0.44509 (8)	0.0177 (2)
N2	0.84796 (13)	0.34792 (13)	0.38501 (8)	0.0182 (2)
N3	0.78344 (13)	0.46815 (13)	0.44617 (8)	0.0181 (2)
N4	0.79984 (12)	0.39605 (12)	0.55284 (7)	0.0144 (2)
N5	0.74880 (12)	0.46344 (12)	0.65193 (7)	0.0165 (2)
N6	0.91309 (15)	0.12338 (14)	0.63884 (8)	0.0223 (2)
C1	0.87689 (14)	0.23122 (15)	0.55029 (9)	0.0153 (2)
C2	0.66745 (14)	0.61599 (15)	0.65364 (9)	0.0169 (2)
C3	0.61800 (16)	0.74455 (16)	0.55860 (10)	0.0215 (3)
H3A	0.7275	0.7992	0.5366	0.026*
H3B	0.5397	0.8391	0.5825	0.026*
H3C	0.5545	0.6795	0.4945	0.026*
C4	0.62021 (16)	0.66948 (17)	0.76627 (10)	0.0232 (3)
H4A	0.6608	0.5760	0.8213	0.028*
H4B	0.4895	0.6838	0.7624	0.028*
H4C	0.6794	0.7837	0.7891	0.028*
H6A	0.887 (2)	0.156 (2)	0.7063 (14)	0.033 (4)*
H6B	0.972 (2)	0.025 (2)	0.6261 (13)	0.037 (4)*

#### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0210 (5)	0.0176 (5)	0.0146 (5)	0.0012 (4)	0.0026 (4)	-0.0006 (4)
N2	0.0214 (5)	0.0174 (5)	0.0161 (5)	0.0008 (4)	0.0036 (4)	-0.0003 (4)
N3	0.0230 (5)	0.0186 (5)	0.0133 (4)	0.0000 (4)	0.0045 (4)	0.0006 (4)
N4	0.0175 (5)	0.0144 (5)	0.0117 (4)	0.0008 (3)	0.0030 (3)	-0.0010 (3)
N5	0.0192 (5)	0.0183 (5)	0.0127 (4)	-0.0001 (4)	0.0042 (4)	-0.0031 (4)
N6	0.0337 (6)	0.0202 (5)	0.0137 (5)	0.0093 (4)	0.0062 (4)	0.0008 (4)
C1	0.0152 (5)	0.0167 (5)	0.0141 (5)	-0.0007 (4)	0.0026 (4)	-0.0019 (4)
C2	0.0151 (5)	0.0167 (5)	0.0190 (6)	-0.0019 (4)	0.0025 (4)	-0.0029 (4)
C3	0.0243 (6)	0.0185 (6)	0.0220 (6)	0.0043 (4)	0.0045 (5)	0.0002 (5)
C4	0.0261 (6)	0.0244 (6)	0.0196 (6)	0.0046 (5)	0.0043 (5)	-0.0055 (5)

## Geometric parameters (Å, °)

N1—C1	1.3326 (14)	N6—H6B	0.878 (17)
N1—N2	1.3678 (13)	C2—C4	1.4924 (16)
N2—N3	1.2870 (13)	C2—C3	1.4977 (16)
N3—N4	1.3784 (13)	C3—H3A	0.9800
N4—C1	1.3549 (14)	C3—H3B	0.9800
N4—N5	1.3866 (12)	C3—H3C	0.9800
N5—C2	1.2873 (15)	C4—H4A	0.9800
N6—C1	1.3309 (14)	C4—H4B	0.9800
N6—H6A	0.891 (17)	C4—H4C	0.9800
C1—N1—N2	105.52 (9)	N5—C2—C3	128.40 (10)
N3—N2—N1	112.53 (9)	C4—C2—C3	117.11 (10)
N2—N3—N4	105.28 (9)	C2—C3—H3A	109.5
C1—N4—N3	108.59 (9)	C2—C3—H3B	109.5
C1—N4—N5	120.58 (9)	H3A—C3—H3B	109.5
N3—N4—N5	130.82 (9)	C2—C3—H3C	109.5
C2—N5—N4	120.83 (9)	H3A—C3—H3C	109.5
C1—N6—H6A	121.1 (10)	H3B—C3—H3C	109.5
C1—N6—H6B	114.8 (10)	C2—C4—H4A	109.5
H6A—N6—H6B	123.9 (14)	C2—C4—H4B	109.5
N6—C1—N1	127.15 (11)	H4A—C4—H4B	109.5
N6—C1—N4	124.77 (10)	C2—C4—H4C	109.5
N1—C1—N4	108.08 (9)	H4A—C4—H4C	109.5
N5—C2—C4	114.48 (10)	H4B—C4—H4C	109.5
C1—N1—N2—N3	-0.23 (12)	N2—N1—C1—N4	0.45 (11)
N1—N2—N3—N4	-0.08 (12)	N3—N4—C1—N6	179.91 (10)
N2—N3—N4—C1	0.37 (11)	N5—N4—C1—N6	-1.31 (16)
N2—N3—N4—N5	-178.25 (10)	N3—N4—C1—N1	-0.52 (12)
C1—N4—N5—C2	-176.51 (10)	N5—N4—C1—N1	178.26 (9)
N3—N4—N5—C2	1.97 (17)	N4—N5—C2—C4	179.59 (9)
N2—N1—C1—N6	-179.99 (11)	N4—N5—C2—C3	-1.35 (17)

## Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N6—H6A $\cdots$ N2 <sup>i</sup>	0.890 (16)	2.200 (17)	3.0600 (16)	162.6 (13)
N6—H6B $\cdots$ N1 <sup>ii</sup>	0.876 (15)	2.118 (15)	2.9770 (16)	166.4 (14)

Symmetry codes: (i)  $x, -y+1/2, z+1/2$ ; (ii)  $-x+2, -y, -z+1$ .