Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

3-Hydroxy-*N*′-isopropylidene-2-naphthohydrazide

See Mun Lee, Kong Mun Lo, Hapipah Mohd Ali and Seik Weng Ng*

Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia
Correspondence e-mail: seikweng@um.edu.my

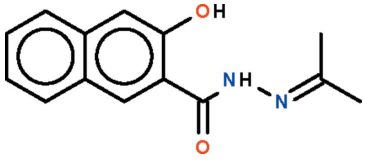
Received 22 November 2009; accepted 24 November 2009

Key indicators: single-crystal X-ray study; T = 123 K; mean σ(C–C) = 0.003 Å; R factor = 0.041; wR factor = 0.113; data-to-parameter ratio = 9.1.

The title Schiff base, $C_{14}H_{14}N_2O_2$, is close to being planar (r.m.s. deviation for the non-hydrogen atoms = 0.052 Å) and an intramolecular N—H···O hydrogen bond generates an S(6) ring. In the crystal, the moleucles are linked by O—H···O hydrogen bonds, giving rise to helical chains propagating along the c axis of the tetragonal unit cell.

Related literature

For the crystal structure of 2′-(2-isopropylidene)-2-hydroxybenzohydrazide monohydrate, see: Kraudelt *et al.* (1996).



Experimental

Crystal data

$C_{14}H_{14}N_2O_2$
M_r = 242.27
Tetragonal, $P\bar{4}2_1c$
a = 13.7343 (3) Å
c = 12.8253 (3) Å
V = 2419.25 (8) Å^3
Z = 8
Mo $K\alpha$ radiation
μ = 0.09 mm^{-1}
T = 123 K
0.35 × 0.15 × 0.05 mm

Data collection

Bruker SMART APEX diffractometer
16460 measured reflections
1566 independent reflections
1341 reflections with $I > 2\sigma(I)$
R_{int} = 0.055

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.041
$wR(F^2)$ = 0.113
S = 1.01
1566 reflections
173 parameters
2 restraints
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.27 e Å^{-3}
$\Delta\rho_{min}$ = −0.22 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
O1—H1o···O2^{i}	0.84 (1)	1.87 (2)	2.648 (2)	153 (3)
N1—H1n···O1	0.88 (1)	1.93 (2)	2.631 (2)	136 (2)

Symmetry code: (i) $-y+\frac{3}{2}, -x+\frac{3}{2}, z+\frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2009).

We thank the University of Malaya (grant No. RG020/09AFR) for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5249).

References

Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
Bruker (2008). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Kraudelt, H., Ludwig, E., Schilde, U. & Uhlemann, E. (1996). *Z. Naturforsch. Teil B*, **51**, 563–566.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Westrip, S. P. (2009). *publCIF*. In preparation.

supporting information

Acta Cryst. (2009). E**65**, o3282 [doi:10.1107/S1600536809050661]

3-Hydroxy-*N*′-isopropylidene-2-naphthohydrazide

See Mun Lee, Kong Mun Lo, Hapipah Mohd Ali and Seik Weng Ng

S1. Experimental

The title Schiff base was obtained as a side product from the reaction between 3-hydroxy-2-naphthoic hydrazide (1 g, 5 mmol) and 4-chlorobenzaldehyde (0.7 g, 5 mmol) in acetone. Colourless irregular chunks of (I) were obtained when the solvent was allowed to evaporate slowly.

S2. Refinement

Anomalous dispersion was negligible and Friedel pairs were merged before refinement.

Carbon-bound H-atoms were placed in calculated positions (C–H 0.95–0.98 Å) and were included in the refinement in the riding model approximation, with *U*(H) set to 1.2–1.5*U*(C). The amino and hydroxy H-atoms were located in a difference Fourier map and were refined with distance restraints of N–H 0.88±0.01 Å and O–H 0.84±0.01 Å.

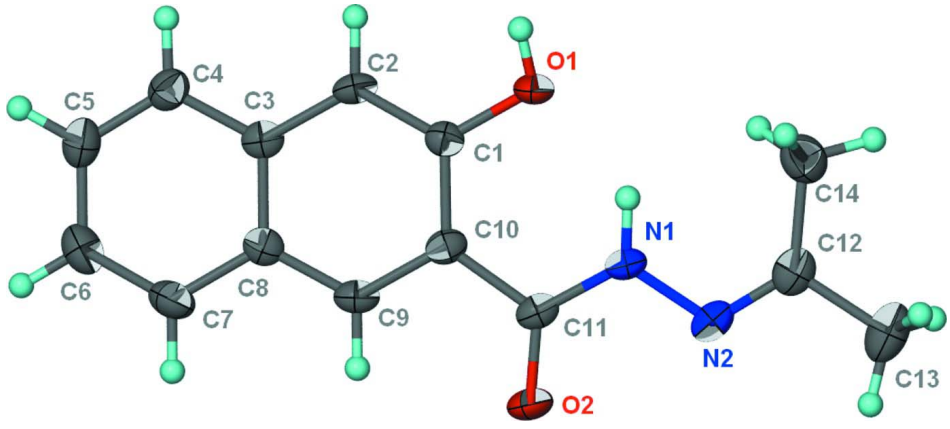


Figure 1

The molecular structure of (I) at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

3-Hydroxy-*N*′-isopropylidene-2-naphthohydrazide

Crystal data

$C_{14}H_{14}N_2O_2$
$M_r = 242.27$
Tetragonal, $P\bar{4}2_1c$
Hall symbol: P -42 n
$a = 13.7343$ (3) Å
$c = 12.8253$ (3) Å
$V = 2419.25$ (8) Å^3
$Z = 8$
$F(000) = 1024$
$D_x = 1.330$ Mg m^{-3}
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 2887 reflections
$\theta = 2.2$–$26.0°$
$\mu = 0.09$ mm^{-1}
$T = 123$ K
Irregular, colourless
0.35 × 0.15 × 0.05 mm

Data collection

Bruker SMART APEX diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
ω scans
16460 measured reflections
1566 independent reflections
1341 reflections with $I > 2\sigma(I)$
$R_{int} = 0.055$
$\theta_{max} = 27.5°$, $\theta_{min} = 2.1°$
$h = -17 \rightarrow 17$
$k = -17 \rightarrow 17$
$l = -16 \rightarrow 16$

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.041$
$wR(F^2) = 0.113$
$S = 1.01$
1566 reflections
173 parameters
2 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H atoms treated by a mixture of independent and constrained refinement
$w = 1/[\sigma^2(F_o^2) + (0.0711P)^2 + 0.6133P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} = 0.001$
$\Delta\rho_{max} = 0.27$ e Å^{-3}
$\Delta\rho_{min} = -0.22$ e Å^{-3}
Absolute structure: Friedel pairs were merged

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
O1	0.89011 (13)	0.64858 (12)	0.81784 (12)	0.0232 (4)
H1O	0.879 (3)	0.649 (3)	0.8825 (10)	0.047 (10)*
O2	0.86201 (14)	0.70322 (12)	0.49816 (12)	0.0268 (4)
N1	0.87509 (16)	0.58826 (14)	0.62381 (16)	0.0213 (4)
H1N	0.875 (2)	0.5750 (19)	0.6910 (9)	0.026 (8)*
N2	0.86924 (15)	0.51398 (15)	0.55082 (15)	0.0244 (5)
C1	0.88285 (16)	0.74227 (16)	0.78288 (18)	0.0185 (5)
C2	0.88364 (17)	0.81988 (17)	0.85018 (18)	0.0203 (5)
H2	0.8892	0.8081	0.9229	0.024*
C3	0.87638 (16)	0.91683 (17)	0.81435 (18)	0.0197 (5)
C4	0.87394 (18)	0.99834 (18)	0.88226 (19)	0.0241 (5)
H4	0.8786	0.9887	0.9554	0.029*
C5	0.86503 (19)	1.09039 (18)	0.8436 (2)	0.0275 (6)
H5	0.8640	1.1441	0.8902	0.033*
C6	0.8573 (2)	1.10692 (18)	0.7353 (2)	0.0297 (6)
H6	0.8505	1.1714	0.7094	0.036*
C7	0.85970 (19)	1.03001 (18)	0.6677 (2)	0.0273 (6)
H7	0.8548	1.0413	0.5949	0.033*
C8	0.86940 (17)	0.93370 (16)	0.70538 (19)	0.0195 (5)
C9	0.86959 (17)	0.85281 (18)	0.63782 (19)	0.0215 (5)
H9	0.8657	0.8640	0.5648	0.026*
C10	0.87513 (17)	0.75829 (17)	0.67298 (18)	0.0192 (5)
C11	0.87014 (17)	0.68105 (17)	0.59108 (18)	0.0196 (5)
C12	0.87295 (18)	0.42740 (19)	0.5869 (2)	0.0257 (5)
C13	0.8674 (2)	0.3471 (2)	0.5088 (2)	0.0374 (7)

H13A	0.8575	0.3747	0.4392	0.056*
H13B	0.8129	0.3040	0.5261	0.056*
H13C	0.9282	0.3098	0.5099	0.056*
C14	0.8808 (2)	0.39944 (19)	0.7001 (2)	0.0322 (6)
H14A	0.9250	0.4445	0.7357	0.048*
H14B	0.9063	0.3330	0.7056	0.048*
H14C	0.8163	0.4026	0.7325	0.048*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0343 (9)	0.0223 (8)	0.0130 (8)	0.0004 (7)	0.0001 (7)	0.0022 (7)
O2	0.0375 (10)	0.0294 (9)	0.0134 (8)	−0.0015 (8)	−0.0012 (8)	−0.0008 (7)
N1	0.0261 (11)	0.0248 (10)	0.0130 (9)	0.0002 (8)	−0.0002 (8)	−0.0008 (8)
N2	0.0253 (10)	0.0272 (10)	0.0205 (10)	−0.0010 (9)	0.0007 (9)	−0.0064 (9)
C1	0.0166 (11)	0.0208 (11)	0.0179 (11)	−0.0003 (9)	0.0005 (9)	0.0015 (9)
C2	0.0195 (12)	0.0284 (12)	0.0130 (10)	−0.0013 (10)	−0.0018 (9)	−0.0009 (9)
C3	0.0139 (10)	0.0254 (11)	0.0198 (11)	−0.0003 (9)	−0.0002 (9)	0.0000 (10)
C4	0.0235 (12)	0.0253 (12)	0.0235 (12)	0.0007 (10)	−0.0026 (10)	−0.0041 (10)
C5	0.0265 (13)	0.0223 (12)	0.0337 (14)	0.0017 (11)	−0.0006 (12)	−0.0053 (11)
C6	0.0296 (13)	0.0227 (12)	0.0368 (14)	0.0015 (10)	0.0010 (12)	0.0052 (11)
C7	0.0313 (14)	0.0279 (13)	0.0226 (12)	0.0020 (11)	−0.0009 (11)	0.0070 (10)
C8	0.0147 (10)	0.0228 (11)	0.0211 (11)	−0.0004 (9)	−0.0016 (9)	0.0014 (9)
C9	0.0201 (11)	0.0288 (12)	0.0155 (10)	−0.0003 (10)	−0.0003 (9)	0.0027 (10)
C10	0.0167 (11)	0.0242 (11)	0.0168 (11)	−0.0018 (9)	−0.0005 (9)	−0.0014 (9)
C11	0.0160 (10)	0.0262 (12)	0.0167 (11)	−0.0017 (9)	0.0004 (9)	−0.0025 (9)
C12	0.0198 (11)	0.0274 (12)	0.0298 (13)	−0.0026 (10)	−0.0002 (11)	−0.0052 (10)
C13	0.0339 (14)	0.0336 (14)	0.0446 (16)	0.0000 (12)	0.0029 (14)	−0.0146 (13)
C14	0.0337 (14)	0.0281 (13)	0.0348 (15)	−0.0026 (11)	−0.0025 (12)	0.0035 (12)

Geometric parameters (Å, º)

O1—C1	1.366 (3)	C6—C7	1.367 (4)
O1—H1O	0.843 (10)	C6—H6	0.9500
O2—C11	1.235 (3)	C7—C8	1.414 (3)
N1—C11	1.344 (3)	C7—H7	0.9500
N1—N2	1.387 (3)	C8—C9	1.409 (3)
N1—H1N	0.880 (10)	C9—C10	1.376 (3)
N2—C12	1.277 (3)	C9—H9	0.9500
C1—C2	1.372 (3)	C10—C11	1.494 (3)
C1—C10	1.430 (3)	C12—C13	1.492 (4)
C2—C3	1.412 (3)	C12—C14	1.505 (4)
C2—H2	0.9500	C13—H13A	0.9800
C3—C8	1.420 (3)	C13—H13B	0.9800
C3—C4	1.419 (3)	C13—H13C	0.9800
C4—C5	1.364 (4)	C14—H14A	0.9800
C4—H4	0.9500	C14—H14B	0.9800
C5—C6	1.411 (4)	C14—H14C	0.9800

C5—H5	0.9500

C1—O1—H1O	108 (2)	C9—C8—C3	118.5 (2)
C11—N1—N2	118.94 (19)	C7—C8—C3	119.7 (2)
C11—N1—H1N	120.1 (18)	C10—C9—C8	122.8 (2)
N2—N1—H1N	120.6 (18)	C10—C9—H9	118.6
C12—N2—N1	116.0 (2)	C8—C9—H9	118.6
O1—C1—C2	121.7 (2)	C9—C10—C1	118.2 (2)
O1—C1—C10	118.27 (19)	C9—C10—C11	115.9 (2)
C2—C1—C10	120.1 (2)	C1—C10—C11	125.9 (2)
C1—C2—C3	121.8 (2)	O2—C11—N1	122.7 (2)
C1—C2—H2	119.1	O2—C11—C10	120.5 (2)
C3—C2—H2	119.1	N1—C11—C10	116.8 (2)
C2—C3—C8	118.6 (2)	N2—C12—C13	116.3 (2)
C2—C3—C4	123.1 (2)	N2—C12—C14	126.2 (2)
C8—C3—C4	118.3 (2)	C13—C12—C14	117.5 (2)
C5—C4—C3	120.7 (2)	C12—C13—H13A	109.5
C5—C4—H4	119.7	C12—C13—H13B	109.5
C3—C4—H4	119.7	H13A—C13—H13B	109.5
C4—C5—C6	120.9 (2)	C12—C13—H13C	109.5
C4—C5—H5	119.5	H13A—C13—H13C	109.5
C6—C5—H5	119.5	H13B—C13—H13C	109.5
C7—C6—C5	119.9 (2)	C12—C14—H14A	109.5
C7—C6—H6	120.1	C12—C14—H14B	109.5
C5—C6—H6	120.1	H14A—C14—H14B	109.5
C6—C7—C8	120.6 (2)	C12—C14—H14C	109.5
C6—C7—H7	119.7	H14A—C14—H14C	109.5
C8—C7—H7	119.7	H14B—C14—H14C	109.5
C9—C8—C7	121.9 (2)		

C11—N1—N2—C12	−179.0 (2)	C7—C8—C9—C10	−177.4 (2)
O1—C1—C2—C3	179.9 (2)	C3—C8—C9—C10	0.8 (4)
C10—C1—C2—C3	0.4 (4)	C8—C9—C10—C1	−1.2 (4)
C1—C2—C3—C8	−0.9 (4)	C8—C9—C10—C11	177.5 (2)
C1—C2—C3—C4	178.0 (2)	O1—C1—C10—C9	−178.9 (2)
C2—C3—C4—C5	−178.8 (2)	C2—C1—C10—C9	0.6 (4)
C8—C3—C4—C5	0.1 (4)	O1—C1—C10—C11	2.6 (4)
C3—C4—C5—C6	0.5 (4)	C2—C1—C10—C11	−177.9 (2)
C4—C5—C6—C7	−0.6 (4)	N2—N1—C11—O2	−1.1 (4)
C5—C6—C7—C8	0.3 (4)	N2—N1—C11—C10	179.0 (2)
C6—C7—C8—C9	178.4 (2)	C9—C10—C11—O2	0.3 (3)
C6—C7—C8—C3	0.2 (4)	C1—C10—C11—O2	178.9 (2)
C2—C3—C8—C9	0.3 (4)	C9—C10—C11—N1	−179.8 (2)
C4—C3—C8—C9	−178.6 (2)	C1—C10—C11—N1	−1.2 (4)
C2—C3—C8—C7	178.5 (2)	N1—N2—C12—C13	−179.6 (2)
C4—C3—C8—C7	−0.4 (4)	N1—N2—C12—C14	1.4 (4)

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
O1—H1o···O2^{i}	0.84 (1)	1.87 (2)	2.648 (2)	153 (3)
N1—H1n···O1	0.88 (1)	1.93 (2)	2.631 (2)	136 (2)

Symmetry code: (i) −*y*+3/2, −*x*+3/2, *z*+1/2.