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## Structure Reports

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Tetrakis(2-aminopyrazine- $\kappa N^4$ )-dichloridocobalt(II)Wei Kang,<sup>a</sup> Li-Hua Huo,<sup>a</sup> Shan Gao<sup>a</sup> and Seik Weng Ng<sup>b\*</sup><sup>a</sup>College of Chemistry and Materials Science, Heilongjiang University, Harbin 150080, People's Republic of China, and <sup>b</sup>Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

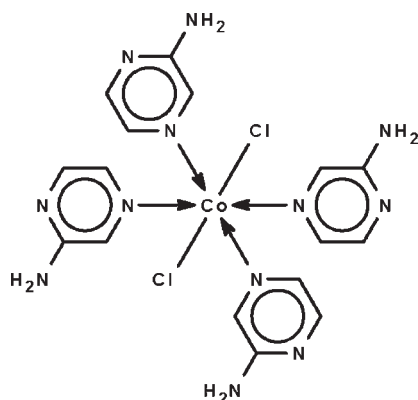
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Key indicators: single-crystal X-ray study;  $T = 293$  K; mean  $\sigma(C-C) = 0.002$  Å;  $R$  factor = 0.025;  $wR$  factor = 0.074; data-to-parameter ratio = 16.3.

The Co<sup>II</sup> atom in the title complex,  $[CoCl_2(C_4H_5N_3)_4]$ , exists in an all-*trans* Cl<sub>2</sub>N<sub>4</sub>Co octahedral geometry. The Co<sup>II</sup> atom lies on a special position of 2 site symmetry. Adjacent molecules are linked by N—H...N and N—H...Cl hydrogen bonds into a three-dimensional network.

## Related literature

For the triclinic modification, see: Csöregi *et al.* (2000).

## Experimental

## Crystal data

 $[CoCl_2(C_4H_5N_3)_4]$  $M_r = 510.27$ Orthorhombic, *Pccn* $a = 7.6347$  (2) Å $b = 15.7341$  (4) Å $c = 18.6074$  (4) Å $V = 2235.22$  (9) Å<sup>3</sup> $Z = 4$ Mo  $K\alpha$  radiation $\mu = 1.04$  mm<sup>-1</sup> $T = 293$  K

0.30 × 0.20 × 0.15 mm

## Data collection

Rigaku R-Axis RAPID IP diffractometer

Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995) $T_{\min} = 0.746$ ,  $T_{\max} = 0.860$ 

20053 measured reflections

2553 independent reflections

2234 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.024$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.025$  $wR(F^2) = 0.074$  $S = 1.06$ 

2553 reflections

157 parameters

4 restraints

H atoms treated by a mixture of independent and constrained refinement

 $\Delta\rho_{\text{max}} = 0.29$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.25$  e Å<sup>-3</sup>

Table 1

Selected bond lengths (Å).

Co1—N1	2.2068 (11)	Co1—Cl1	2.4206 (4)
Co1—N4	2.1941 (11)		

Table 2

Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N3—H1...Cl1 <sup>i</sup>	0.85 (1)	2.36 (1)	3.209 (2)	175 (2)
N3—H2...N5 <sup>ii</sup>	0.86 (1)	2.43 (2)	3.134 (2)	140 (2)
N6—H3...Cl1 <sup>iii</sup>	0.85 (1)	2.42 (1)	3.265 (1)	171 (2)
N6—H4...N2 <sup>iv</sup>	0.86 (1)	2.33 (2)	3.045 (2)	142 (2)

Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x - 1, -y + \frac{1}{2}, z - \frac{1}{2}$ ; (iii)  $-x + 2, y + \frac{1}{2}, -z + \frac{1}{2}$ ; (iv)  $x + \frac{1}{2}, y + \frac{1}{2}, -z$ .

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalClear* (Rigaku/MSK, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2656).

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## supporting information

*Acta Cryst.* (2009). E65, m1502 [doi:10.1107/S1600536809045309]

**Tetrakis(2-aminopyrazine- $\kappa N^4$ )dichloridocobalt(II)**

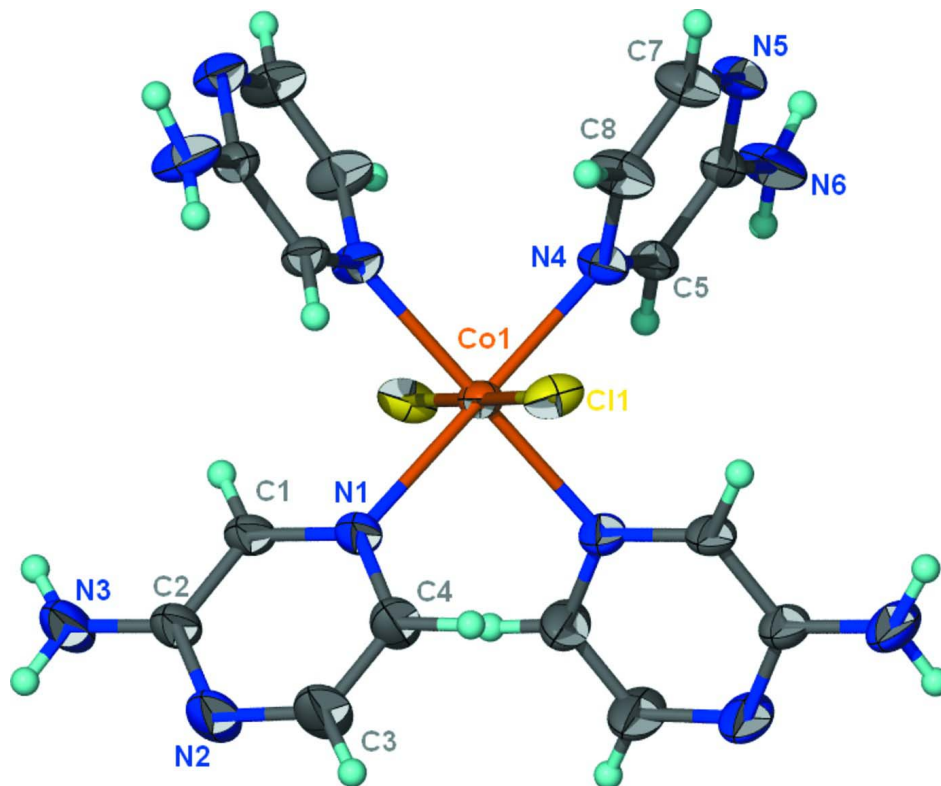
Wei Kang, Li-Hua Huo, Shan Gao and Seik Weng Ng

**S1. Experimental**

Cobalt(II) chloride hexahydrate (0.48 g, 2 mmol) and 2-aminopyrazine (0.19 g, 2 mmol) were dissolved in a small volume of water. Red crystals of the adduct separated from the filtered solution after several days. CH&N elemental analysis. Calc. for  $C_{16}H_{20}Cl_2N_{12}Co$ : C 37.66, H 3.95, N 32.94%; found: C 37.63, H 3.89, N 32.97%.

**S2. Refinement**

Amino-H atoms were located in a difference Fourier map and refined isotropically with a distance restraint of N–H =  $0.86 \pm 0.01$  Å. Carbon-bound H-atoms were placed in calculated positions (C–H = 0.93 Å) and were included in the refinement in the riding model approximation with  $U_{iso}(H)$  set to  $1.2U_{eq}(C)$ .



**Figure 1**

Thermal ellipsoid plot (Barbour, 2001) of  $CoCl_2(C_2H_5N_3)_4$  at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

**Tetrakis(2-aminopyrazine- $\kappa$ N<sup>4</sup>)dichloridocobalt(II)***Crystal data*[CoCl<sub>2</sub>(C<sub>4</sub>H<sub>5</sub>N<sub>3</sub>)<sub>4</sub>] $M_r = 510.27$ Orthorhombic, *Pccn*

Hall symbol: -P 2ab 2ac

 $a = 7.6347$  (2) Å $b = 15.7341$  (4) Å $c = 18.6074$  (4) Å $V = 2235.22$  (9) Å<sup>3</sup> $Z = 4$  $F(000) = 1044$  $D_x = 1.516$  Mg m<sup>-3</sup>Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 15790 reflections

 $\theta = 2.3$ – $27.5^\circ$  $\mu = 1.04$  mm<sup>-1</sup> $T = 293$  K

Block, red

 $0.30 \times 0.20 \times 0.15$  mm*Data collection*

Rigaku RAXIS-RAPID IP

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 $\omega$  scans

Absorption correction: multi-scan

(ABSCOR; Higashi, 1995)

 $T_{\min} = 0.746$ ,  $T_{\max} = 0.860$ 

20053 measured reflections

2553 independent reflections

2234 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.024$  $\theta_{\max} = 27.5^\circ$ ,  $\theta_{\min} = 3.2^\circ$  $h = -9 \rightarrow 9$  $k = -20 \rightarrow 20$  $l = -24 \rightarrow 24$ *Refinement*Refinement on  $F^2$ 

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.025$  $wR(F^2) = 0.074$  $S = 1.06$ 

2553 reflections

157 parameters

4 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

 $w = 1/[\sigma^2(F_o^2) + (0.0439P)^2 + 0.4514P]$ where  $P = (F_o^2 + 2F_c^2)/3$  $(\Delta/\sigma)_{\max} = 0.001$  $\Delta\rho_{\max} = 0.29$  e Å<sup>-3</sup> $\Delta\rho_{\min} = -0.25$  e Å<sup>-3</sup>*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>)*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Co1	0.7500	0.2500	0.103443 (12)	0.02794 (9)
Cl1	0.98632 (5)	0.14744 (2)	0.103952 (17)	0.03970 (11)
N1	0.61989 (16)	0.17473 (8)	0.01880 (6)	0.0348 (3)
N2	0.46194 (19)	0.08976 (11)	-0.09614 (7)	0.0553 (4)
N3	0.1968 (2)	0.09877 (14)	-0.03954 (8)	0.0686 (5)
N4	0.88907 (16)	0.31841 (7)	0.18890 (6)	0.0339 (3)
N5	1.03169 (17)	0.40550 (8)	0.30631 (6)	0.0387 (3)
N6	0.9535 (2)	0.53209 (8)	0.25231 (7)	0.0527 (4)
C1	0.45050 (19)	0.15892 (9)	0.01846 (7)	0.0357 (3)
H1A	0.3823	0.1771	0.0569	0.043*
C2	0.3693 (2)	0.11475 (10)	-0.03920 (7)	0.0413 (3)
C3	0.6338 (2)	0.10573 (14)	-0.09351 (9)	0.0625 (6)
H3A	0.7027	0.0878	-0.1318	0.075*

C4	0.7141 (2)	0.14683 (13)	-0.03790 (9)	0.0509 (4)
H4A	0.8345	0.1557	-0.0391	0.061*
C5	0.88843 (19)	0.40188 (8)	0.19189 (6)	0.0319 (3)
H5	0.8395	0.4324	0.1541	0.038*
C6	0.96002 (18)	0.44673 (8)	0.25094 (7)	0.0325 (3)
C7	1.0327 (2)	0.32043 (10)	0.30155 (8)	0.0453 (4)
H7	1.0823	0.2896	0.3390	0.054*
C8	0.9645 (2)	0.27664 (10)	0.24459 (8)	0.0452 (4)
H8	0.9700	0.2176	0.2440	0.054*
H1	0.136 (3)	0.1094 (14)	-0.0022 (8)	0.078 (7)*
H2	0.155 (3)	0.0704 (11)	-0.0753 (8)	0.063 (6)*
H3	0.978 (3)	0.5580 (12)	0.2913 (7)	0.060 (6)*
H4	0.909 (2)	0.5583 (11)	0.2164 (8)	0.061 (6)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Co1	0.03367 (16)	0.02901 (14)	0.02114 (14)	-0.00229 (9)	0.000	0.000
Cl1	0.0462 (2)	0.0432 (2)	0.02968 (18)	0.01013 (15)	0.00662 (13)	0.00476 (12)
N1	0.0337 (6)	0.0435 (6)	0.0271 (5)	-0.0030 (5)	0.0015 (5)	-0.0079 (5)
N2	0.0407 (8)	0.0851 (11)	0.0401 (7)	-0.0055 (7)	0.0052 (6)	-0.0322 (7)
N3	0.0366 (8)	0.1287 (16)	0.0407 (8)	-0.0134 (9)	0.0027 (6)	-0.0368 (9)
N4	0.0428 (7)	0.0326 (5)	0.0262 (5)	-0.0011 (5)	-0.0058 (5)	0.0003 (4)
N5	0.0447 (7)	0.0433 (6)	0.0281 (5)	-0.0005 (5)	-0.0095 (5)	-0.0004 (5)
N6	0.0912 (12)	0.0339 (6)	0.0331 (6)	-0.0092 (7)	-0.0211 (7)	0.0002 (5)
C1	0.0359 (7)	0.0464 (7)	0.0248 (6)	0.0013 (6)	0.0028 (5)	-0.0084 (6)
C2	0.0352 (8)	0.0588 (9)	0.0300 (6)	-0.0012 (7)	0.0005 (6)	-0.0124 (6)
C3	0.0416 (9)	0.1031 (16)	0.0429 (9)	-0.0042 (10)	0.0106 (7)	-0.0368 (9)
C4	0.0340 (8)	0.0789 (12)	0.0398 (8)	-0.0060 (7)	0.0075 (6)	-0.0206 (8)
C5	0.0399 (7)	0.0323 (6)	0.0236 (6)	-0.0039 (5)	-0.0055 (5)	0.0036 (5)
C6	0.0376 (7)	0.0349 (6)	0.0251 (6)	-0.0057 (5)	-0.0034 (5)	0.0008 (5)
C7	0.0582 (10)	0.0450 (8)	0.0326 (7)	0.0103 (7)	-0.0151 (7)	0.0028 (6)
C8	0.0650 (11)	0.0335 (7)	0.0371 (7)	0.0090 (7)	-0.0136 (7)	0.0015 (6)

*Geometric parameters (Å, °)*

Co1—N1 <sup>i</sup>	2.2068 (11)	N5—C6	1.3348 (18)
Co1—N1	2.2068 (11)	N5—C7	1.341 (2)
Co1—N4	2.1941 (11)	N6—C6	1.3441 (18)
Co1—N4 <sup>i</sup>	2.1941 (11)	N6—H3	0.852 (9)
Co1—Cl1 <sup>i</sup>	2.4206 (4)	N6—H4	0.856 (9)
Co1—Cl1	2.4206 (4)	C1—C2	1.4207 (19)
N1—C1	1.3170 (19)	C1—H1A	0.9300
N1—C4	1.3501 (19)	C3—C4	1.366 (2)
N2—C2	1.3332 (19)	C3—H3A	0.9300
N2—C3	1.337 (2)	C4—H4A	0.9300
N3—C2	1.341 (2)	C5—C6	1.4157 (18)
N3—H1	0.853 (9)	C5—H5	0.9300

N3—H2	0.864 (9)	C7—C8	1.367 (2)
N4—C5	1.3145 (17)	C7—H7	0.9300
N4—C8	1.3556 (18)	C8—H8	0.9300
N4—Co1—N4 <sup>i</sup>	87.12 (6)	C6—N6—H4	118.8 (14)
N4—Co1—N1 <sup>i</sup>	92.07 (4)	H3—N6—H4	121 (2)
N4 <sup>i</sup> —Co1—N1 <sup>i</sup>	176.67 (4)	N1—C1—C2	121.63 (13)
N4—Co1—N1	176.67 (4)	N1—C1—H1A	119.2
N4 <sup>i</sup> —Co1—N1	92.07 (4)	C2—C1—H1A	119.2
N1 <sup>i</sup> —Co1—N1	88.93 (6)	N2—C2—N3	117.51 (14)
N4—Co1—Cl1 <sup>i</sup>	91.77 (3)	N2—C2—C1	120.88 (15)
N4 <sup>i</sup> —Co1—Cl1 <sup>i</sup>	87.91 (3)	N3—C2—C1	121.59 (13)
N1 <sup>i</sup> —Co1—Cl1 <sup>i</sup>	88.89 (3)	N2—C3—C4	123.88 (15)
N1—Co1—Cl1 <sup>i</sup>	91.43 (3)	N2—C3—H3A	118.1
N4—Co1—Cl1	87.91 (3)	C4—C3—H3A	118.1
N4 <sup>i</sup> —Co1—Cl1	91.77 (3)	N1—C4—C3	120.47 (15)
N1 <sup>i</sup> —Co1—Cl1	91.43 (3)	N1—C4—H4A	119.8
N1—Co1—Cl1	88.89 (3)	C3—C4—H4A	119.8
Cl1 <sup>i</sup> —Co1—Cl1	179.552 (18)	N4—C5—C6	121.97 (12)
C1—N1—C4	117.25 (12)	N4—C5—H5	119.0
C1—N1—Co1	123.15 (9)	C6—C5—H5	119.0
C4—N1—Co1	119.50 (10)	N5—C6—N6	119.10 (12)
C2—N2—C3	115.85 (13)	N5—C6—C5	121.00 (12)
C2—N3—H1	119.6 (16)	N6—C6—C5	119.89 (12)
C2—N3—H2	117.9 (14)	N5—C7—C8	123.50 (13)
H1—N3—H2	122 (2)	N5—C7—H7	118.2
C5—N4—C8	116.97 (12)	C8—C7—H7	118.2
C5—N4—Co1	121.27 (9)	N4—C8—C7	120.67 (14)
C8—N4—Co1	121.45 (10)	N4—C8—H8	119.7
C6—N5—C7	115.87 (12)	C7—C8—H8	119.7
C6—N6—H3	119.0 (14)		
N4 <sup>i</sup> —Co1—N1—C1	45.73 (12)	C3—N2—C2—N3	-179.1 (2)
N1 <sup>i</sup> —Co1—N1—C1	-131.09 (13)	C3—N2—C2—C1	2.7 (3)
Cl1 <sup>i</sup> —Co1—N1—C1	-42.23 (11)	N1—C1—C2—N2	-1.9 (2)
Cl1—Co1—N1—C1	137.45 (11)	N1—C1—C2—N3	-179.93 (17)
N4 <sup>i</sup> —Co1—N1—C4	-138.03 (13)	C2—N2—C3—C4	-1.6 (3)
N1 <sup>i</sup> —Co1—N1—C4	45.15 (12)	C1—N1—C4—C3	1.4 (3)
Cl1 <sup>i</sup> —Co1—N1—C4	134.01 (13)	Co1—N1—C4—C3	-175.03 (16)
Cl1—Co1—N1—C4	-46.30 (13)	N2—C3—C4—N1	-0.5 (4)
N4 <sup>i</sup> —Co1—N4—C5	-124.52 (13)	C8—N4—C5—C6	-1.1 (2)
N1 <sup>i</sup> —Co1—N4—C5	52.25 (11)	Co1—N4—C5—C6	172.58 (10)
Cl1 <sup>i</sup> —Co1—N4—C5	-36.70 (11)	C7—N5—C6—N6	179.54 (15)
Cl1—Co1—N4—C5	143.61 (11)	C7—N5—C6—C5	0.7 (2)
N4 <sup>i</sup> —Co1—N4—C8	48.91 (11)	N4—C5—C6—N5	0.1 (2)
N1 <sup>i</sup> —Co1—N4—C8	-134.32 (13)	N4—C5—C6—N6	-178.74 (15)
Cl1 <sup>i</sup> —Co1—N4—C8	136.73 (12)	C6—N5—C7—C8	-0.5 (3)
Cl1—Co1—N4—C8	-42.96 (12)	C5—N4—C8—C7	1.4 (2)

C4—N1—C1—C2	-0.3 (2)	Co1—N4—C8—C7	-172.29 (13)
Co1—N1—C1—C2	176.01 (11)	N5—C7—C8—N4	-0.6 (3)

Symmetry code: (i)  $-x+3/2, -y+1/2, z$ .

*Hydrogen-bond geometry* ( $\text{\AA}, ^\circ$ )

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N3—H1 $\cdots$ C11 <sup>ii</sup>	0.85 (1)	2.36 (1)	3.209 (2)	175 (2)
N3—H2 $\cdots$ N5 <sup>iii</sup>	0.86 (1)	2.43 (2)	3.134 (2)	140 (2)
N6—H3 $\cdots$ C11 <sup>iv</sup>	0.85 (1)	2.42 (1)	3.265 (1)	171 (2)
N6—H4 $\cdots$ N2 <sup>v</sup>	0.86 (1)	2.33 (2)	3.045 (2)	142 (2)

Symmetry codes: (ii)  $x-1, y, z$ ; (iii)  $x-1, -y+1/2, z-1/2$ ; (iv)  $-x+2, y+1/2, -z+1/2$ ; (v)  $x+1/2, y+1/2, -z$ .