

# Bis( $\mu$ -4-hydroxybenzoato- $\kappa^2O:O'$ )bis-[triaquabis(4-hydroxybenzoato)- $\kappa O;\kappa^2O,O'$ -terbium(III)] decahydrate

Yi-Min Zhu,<sup>a</sup> Pei-Pei Feng,<sup>a</sup> Yang-Yi Yang<sup>a</sup> and Seik Weng Ng<sup>b\*</sup>

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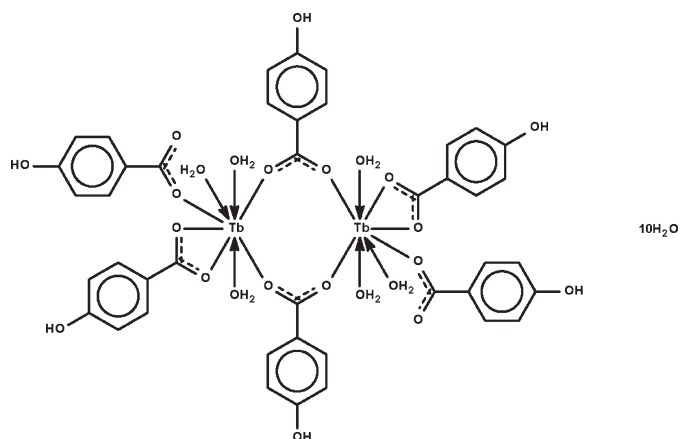
Received 6 December 2009; accepted 7 December 2009

Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(C-C) = 0.003$  Å;  $R$  factor = 0.018;  $wR$  factor = 0.059; data-to-parameter ratio = 14.1.

The title dinuclear compound,  $[Tb_2(C_7H_5O_3)_6(H_2O)_6] \cdot 10H_2O$ , lies on a center of inversion and the two  $Tb^{III}$  atoms are bridged by two 4-hydroxybenzoate anions; each metal atom is further coordinated by one monodentate anion and chelated by the third anion. The eight-coordinate geometry approximates a square antiprism. Hydrogen bonds of the  $O-H \cdots O$  type connect the uncoordinated water molecules to the dinuclear species, forming a three-dimensional network.

## Related literature

For a related structure,  $Tb_2(H_2O)_2(DMF)_2(C_7H_5O_3)_6$ , see: Zhou *et al.* (2008).



## Experimental

### Crystal data

$[Tb_2(C_7H_5O_3)_6(H_2O)_6] \cdot 10H_2O$   
 $M_r = 1428.76$   
 Triclinic,  $P\bar{1}$   
 $a = 10.8308$  (5) Å  
 $b = 11.3337$  (6) Å  
 $c = 11.5128$  (6) Å  
 $\alpha = 90.463$  (1)°  
 $\beta = 101.690$  (1)°

$\gamma = 105.249$  (1)°  
 $V = 1332.37$  (12) Å<sup>3</sup>  
 $Z = 1$   
 Mo  $K\alpha$  radiation  
 $\mu = 2.73$  mm<sup>-1</sup>  
 $T = 173$  K  
 $0.47 \times 0.30 \times 0.19$  mm

### Data collection

Bruker SMART area-detector diffractometer  
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{min} = 0.360$ ,  $T_{max} = 0.625$

12836 measured reflections  
 5756 independent reflections  
 5391 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.018$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.018$   
 $wR(F^2) = 0.059$   
 $S = 1.09$   
 5756 reflections  
 409 parameters  
 27 restraints

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{max} = 0.67$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.64$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O3—H3...O7w <sup>i</sup>	0.84 (1)	1.82 (2)	2.612 (3)	158 (3)
O6—H6...O6w <sup>ii</sup>	0.83 (1)	1.87 (1)	2.678 (3)	165 (4)
O9—H9...O4w <sup>iii</sup>	0.84 (1)	1.98 (2)	2.763 (3)	156 (3)
O1w—H11...O4w	0.84 (1)	2.04 (1)	2.870 (3)	171 (3)
O1w—H12...O9 <sup>iv</sup>	0.83 (1)	1.96 (1)	2.766 (3)	164 (3)
O2w—H22...O5w	0.83 (1)	1.86 (1)	2.678 (3)	167 (3)
O2w—H21...O5w <sup>v</sup>	0.84 (1)	2.20 (2)	2.952 (3)	150 (3)
O3w—H31...O3 <sup>ii</sup>	0.84 (1)	1.95 (1)	2.777 (3)	171 (3)
O3w—H32...O7 <sup>vi</sup>	0.84 (1)	2.25 (2)	2.916 (2)	137 (3)
O4w—H41...O8w <sup>vii</sup>	0.84 (1)	1.94 (1)	2.753 (3)	163 (3)
O4w—H42...O5 <sup>viii</sup>	0.84 (1)	2.10 (1)	2.927 (3)	169 (3)
O5w—H52...O7w	0.85 (1)	2.01 (2)	2.811 (3)	157 (3)
O5w—H51...O8w <sup>ix</sup>	0.86 (1)	1.96 (1)	2.791 (3)	163 (3)
O6w—H61...O2	0.85 (1)	1.90 (1)	2.743 (2)	172 (3)
O6w—H62...O5	0.84 (1)	1.95 (1)	2.785 (2)	171 (3)
O7w—H71...O6w	0.84 (1)	1.90 (1)	2.732 (3)	172 (3)
O7w—H72...O5 <sup>x</sup>	0.83 (1)	1.89 (1)	2.725 (3)	175 (3)
O8w—H81...O1 <sup>xi</sup>	0.84 (1)	2.10 (2)	2.818 (3)	144 (3)
O8w—H82...O6	0.84 (1)	1.90 (1)	2.713 (3)	164 (3)

Symmetry codes: (i)  $-x+2, -y+1, -z$ ; (ii)  $x-1, y, z$ ; (iii)  $x+1, y+1, z$ ; (iv)  $-x+2, -y+1, -z+1$ ; (v)  $-x+1, -y+1, -z$ ; (vi)  $-x+1, -y+1, -z+1$ ; (vii)  $-x, -y, -z+1$ ; (viii)  $-x+1, -y, -z+1$ ; (ix)  $-x, -y, -z$ ; (x)  $-x+1, -y, -z$ ; (xi)  $x-1, y-1, z$ .

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2706).

## References

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## supporting information

*Acta Cryst.* (2010). E66, m54–m55 [doi:10.1107/S1600536809052489]

**Bis( $\mu$ -4-hydroxybenzoato- $\kappa^2$ O:O')bis[triaquabis(4-hydroxybenzoato)- $\kappa$ O; $\kappa^2$ O,O'-terbium(III)] decahydrate**

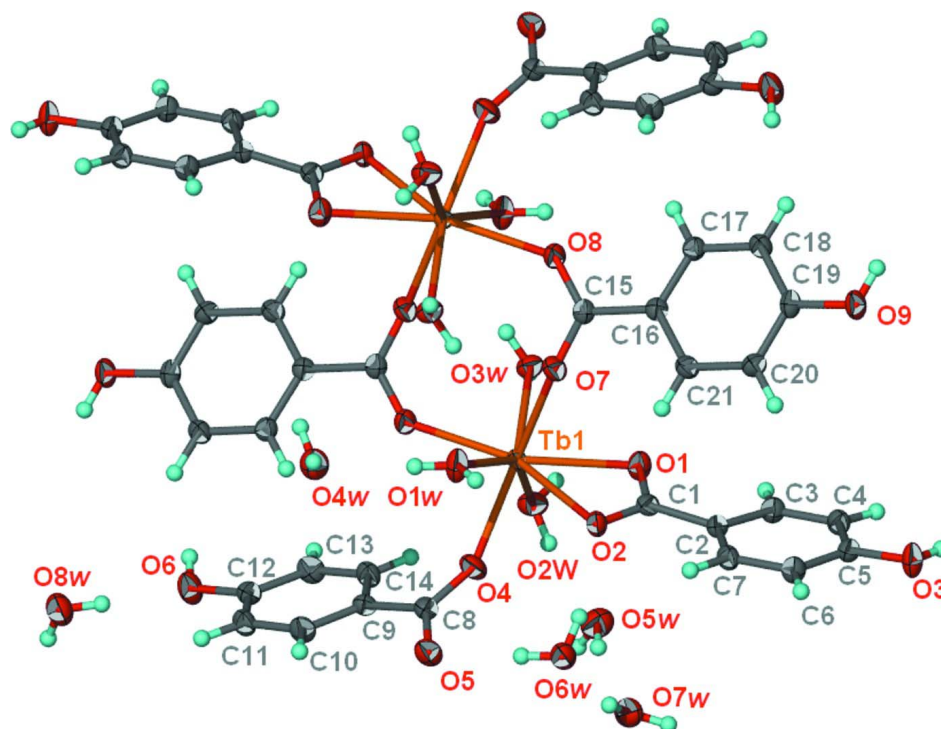
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**S1. Experimental**

4-Hydroxybenzoic acid (0.3 mmol) and sodium hydroxide (0.3 mmol) were dissolved in water (5 ml) and this was mixed with a solution of terbium chloride (0.1 mmol) dissolved in water (5 ml). The white precipitate that formed was removed by filtration. Colorless crystals were isolated from the filtrate after two weeks.

**S2. Refinement**

Carbon-bound hydrogen atoms were generated geometrically and were constrained to ride on their parent atoms [ $C-H = 0.95$ ,  $U_{iso}(H) = 1.2U_{eq}(C)$ ]. The hydroxy and water H-atoms were located in a difference Fourier map, and were refined with distance restraints of  $O-H = 0.84 \pm 0.01$  Å and  $H \cdots H = 1.37 \pm 0.01$  Å. Their temperature factors were tied by a factor of 1.5.



**Figure 1**

Thermal ellipsoid plot (Barbour, 2001) of  $Tb_2(H_2O)_6(C_7H_5O_3)_6 \cdot 10H_2O$  at the 70% probability level; hydrogen atoms are drawn as sphere of arbitrary radius.

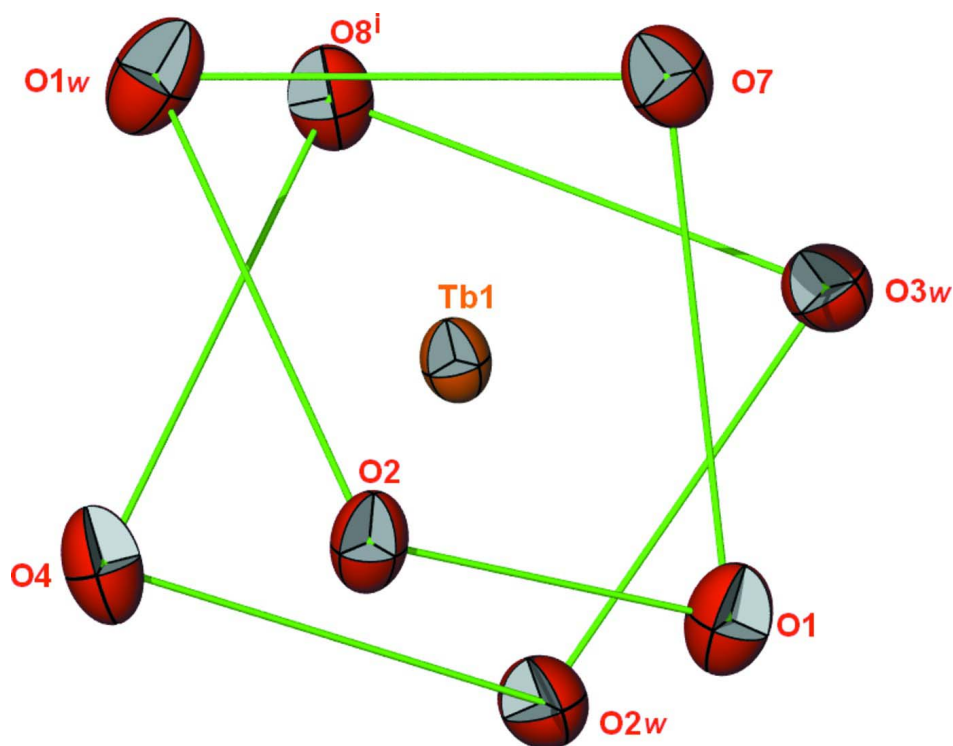


Figure 2

Detail of the coordination environment of the terbium(III) atom.

**Bis( $\mu$ -4-hydroxybenzoato- $\kappa^2$ O:O')bis[triaquabis(4-hydroxybenzoato)- $\kappa$ O; $\kappa^2$ O,O'-terbium(III)] decahydrate**

*Crystal data*

[Tb<sub>2</sub>(C<sub>7</sub>H<sub>5</sub>O<sub>3</sub>)<sub>6</sub>(H<sub>2</sub>O)<sub>6</sub>] $\cdot$ 10H<sub>2</sub>O

$M_r = 1428.76$

Triclinic,  $P\bar{1}$

Hall symbol: -P 1

$a = 10.8308$  (5) Å

$b = 11.3337$  (6) Å

$c = 11.5128$  (6) Å

$\alpha = 90.463$  (1) $^\circ$

$\beta = 101.690$  (1) $^\circ$

$\gamma = 105.249$  (1) $^\circ$

$V = 1332.37$  (12) Å<sup>3</sup>

$Z = 1$

$F(000) = 716$

$D_x = 1.781$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 5392 reflections

$\theta = 2.3$ – $27.0$  $^\circ$

$\mu = 2.73$  mm<sup>-1</sup>

$T = 173$  K

Block, colorless

$0.47 \times 0.30 \times 0.19$  mm

*Data collection*

Bruker SMART area-detector  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(*SADABS*; Sheldrick, 1996)

$T_{\min} = 0.360$ ,  $T_{\max} = 0.625$

12836 measured reflections

5756 independent reflections

5391 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.018$

$\theta_{\max} = 27.0$  $^\circ$ ,  $\theta_{\min} = 1.8$  $^\circ$

$h = -13$ → $13$

$k = -13$ → $14$

$l = -14$ → $14$

Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.018$

$wR(F^2) = 0.059$

$S = 1.09$

5756 reflections

409 parameters

27 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

H atoms treated by a mixture of independent  
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0407P)^2 + 0.0925P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.67 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.64 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Tb1	0.521099 (9)	0.392335 (9)	0.316629 (8)	0.01393 (5)
O1	0.68247 (16)	0.52500 (16)	0.22113 (15)	0.0219 (4)
O2	0.71938 (16)	0.35118 (15)	0.28025 (15)	0.0189 (3)
O3	1.25530 (17)	0.60503 (18)	0.13235 (17)	0.0243 (4)
H3	1.262 (3)	0.6687 (19)	0.095 (3)	0.036*
O4	0.44616 (17)	0.20553 (16)	0.21445 (16)	0.0233 (4)
O5	0.44536 (17)	0.00991 (16)	0.21119 (16)	0.0236 (4)
O6	-0.15974 (18)	-0.03603 (17)	0.16884 (18)	0.0280 (4)
H6	-0.192 (3)	0.023 (2)	0.171 (3)	0.042*
O7	0.66267 (16)	0.51402 (16)	0.47709 (14)	0.0207 (3)
O8	0.65933 (16)	0.66801 (16)	0.59555 (15)	0.0209 (4)
O9	1.23480 (17)	0.85877 (17)	0.50180 (16)	0.0220 (4)
H9	1.262 (3)	0.9320 (13)	0.529 (3)	0.033*
O1w	0.56576 (17)	0.25542 (17)	0.47360 (16)	0.0241 (4)
H11	0.509 (2)	0.206 (2)	0.502 (3)	0.036*
H12	0.6353 (16)	0.234 (3)	0.487 (3)	0.036*
O2w	0.40382 (18)	0.41807 (17)	0.12181 (15)	0.0233 (4)
H21	0.405 (3)	0.4895 (13)	0.102 (2)	0.035*
H22	0.426 (3)	0.380 (2)	0.071 (2)	0.035*
O3w	0.44922 (17)	0.57723 (16)	0.32029 (15)	0.0201 (3)
H31	0.397 (2)	0.593 (3)	0.2625 (14)	0.030*
H32	0.428 (3)	0.593 (3)	0.3838 (13)	0.030*
O4w	0.39635 (19)	0.08545 (18)	0.59456 (16)	0.0267 (4)
H41	0.358 (3)	0.130 (3)	0.624 (2)	0.040*
H42	0.450 (3)	0.066 (3)	0.6490 (19)	0.040*
O5w	0.4872 (2)	0.3305 (2)	-0.05591 (17)	0.0309 (4)
H51	0.421 (2)	0.294 (3)	-0.111 (2)	0.046*
H52	0.530 (3)	0.278 (2)	-0.037 (3)	0.046*
O6w	0.69870 (17)	0.12258 (17)	0.18465 (16)	0.0235 (4)
H61	0.711 (3)	0.1957 (12)	0.211 (3)	0.035*
H62	0.6246 (17)	0.082 (2)	0.194 (3)	0.035*
O7w	0.6606 (2)	0.18314 (17)	-0.04660 (16)	0.0286 (4)
H71	0.676 (3)	0.160 (3)	0.0230 (11)	0.043*

H72	0.627 (3)	0.1215 (18)	-0.0936 (19)	0.043*
O8w	-0.30680 (19)	-0.22701 (18)	0.26414 (18)	0.0301 (4)
H81	-0.280 (3)	-0.2897 (17)	0.276 (3)	0.045*
H82	-0.249 (2)	-0.172 (2)	0.242 (3)	0.045*
C1	0.7539 (2)	0.4523 (2)	0.2315 (2)	0.0169 (5)
C2	0.8802 (2)	0.4855 (2)	0.19285 (19)	0.0163 (4)
C3	0.9139 (2)	0.5908 (2)	0.1316 (2)	0.0190 (5)
H3A	0.8513	0.6352	0.1057	0.023*
C4	1.0383 (2)	0.6307 (2)	0.1084 (2)	0.0186 (5)
H4	1.0609	0.7020	0.0660	0.022*
C5	1.1296 (2)	0.5662 (2)	0.1474 (2)	0.0177 (5)
C6	1.0956 (2)	0.4574 (2)	0.2042 (2)	0.0188 (5)
H6A	1.1572	0.4112	0.2268	0.023*
C7	0.9713 (2)	0.4181 (2)	0.22698 (19)	0.0178 (5)
H7	0.9475	0.3446	0.2661	0.021*
C8	0.3871 (2)	0.0935 (2)	0.2068 (2)	0.0184 (5)
C9	0.2416 (2)	0.0589 (2)	0.1919 (2)	0.0168 (4)
C10	0.1763 (2)	-0.0536 (2)	0.2258 (2)	0.0206 (5)
H10	0.2237	-0.1109	0.2542	0.025*
C11	0.0414 (2)	-0.0831 (2)	0.2184 (2)	0.0224 (5)
H11A	-0.0027	-0.1592	0.2440	0.027*
C12	-0.0276 (2)	-0.0015 (2)	0.1740 (2)	0.0201 (5)
C13	0.0352 (2)	0.1103 (2)	0.1364 (2)	0.0214 (5)
H13	-0.0135	0.1651	0.1034	0.026*
C14	0.1704 (2)	0.1410 (2)	0.1477 (2)	0.0207 (5)
H14	0.2146	0.2185	0.1251	0.025*
C15	0.7152 (2)	0.6180 (2)	0.53115 (19)	0.0159 (4)
C16	0.8513 (2)	0.6851 (2)	0.52099 (19)	0.0153 (4)
C17	0.9058 (2)	0.8072 (2)	0.56448 (19)	0.0167 (4)
H17	0.8550	0.8494	0.5982	0.020*
C18	1.0337 (2)	0.8673 (2)	0.5587 (2)	0.0187 (5)
H18	1.0704	0.9503	0.5883	0.022*
C19	1.1080 (2)	0.8052 (2)	0.5093 (2)	0.0177 (5)
C20	1.0543 (2)	0.6841 (2)	0.4635 (2)	0.0187 (5)
H20	1.1045	0.6427	0.4282	0.022*
C21	0.9260 (2)	0.6245 (2)	0.4702 (2)	0.0176 (5)
H21A	0.8890	0.5418	0.4398	0.021*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Tb1	0.01190 (7)	0.01180 (7)	0.01807 (7)	0.00220 (4)	0.00446 (4)	0.00005 (4)
O1	0.0188 (8)	0.0190 (9)	0.0318 (9)	0.0082 (7)	0.0104 (7)	0.0067 (7)
O2	0.0174 (8)	0.0138 (8)	0.0276 (8)	0.0042 (6)	0.0098 (6)	0.0035 (6)
O3	0.0152 (8)	0.0248 (10)	0.0348 (10)	0.0051 (7)	0.0099 (7)	0.0120 (8)
O4	0.0198 (8)	0.0152 (9)	0.0338 (9)	0.0032 (7)	0.0051 (7)	-0.0029 (7)
O5	0.0186 (8)	0.0183 (9)	0.0343 (9)	0.0062 (7)	0.0049 (7)	-0.0008 (7)
O6	0.0173 (9)	0.0201 (10)	0.0484 (11)	0.0055 (7)	0.0105 (8)	0.0042 (8)

O7	0.0160 (8)	0.0188 (9)	0.0244 (8)	0.0005 (7)	0.0032 (6)	-0.0025 (7)
O8	0.0174 (8)	0.0170 (9)	0.0302 (9)	0.0039 (7)	0.0105 (7)	-0.0001 (7)
O9	0.0134 (8)	0.0204 (9)	0.0321 (9)	0.0023 (7)	0.0079 (7)	0.0027 (7)
O1w	0.0184 (8)	0.0241 (10)	0.0337 (9)	0.0082 (7)	0.0108 (7)	0.0114 (7)
O2w	0.0316 (10)	0.0196 (9)	0.0203 (8)	0.0097 (8)	0.0058 (7)	0.0005 (7)
O3w	0.0237 (9)	0.0198 (9)	0.0195 (8)	0.0093 (7)	0.0065 (7)	0.0013 (7)
O4w	0.0262 (10)	0.0251 (10)	0.0301 (9)	0.0082 (8)	0.0067 (7)	0.0042 (8)
O5w	0.0298 (10)	0.0359 (12)	0.0292 (10)	0.0122 (9)	0.0072 (8)	-0.0007 (8)
O6w	0.0200 (9)	0.0194 (9)	0.0330 (10)	0.0057 (7)	0.0092 (7)	0.0001 (7)
O7w	0.0404 (11)	0.0183 (9)	0.0272 (9)	0.0055 (8)	0.0109 (8)	0.0021 (7)
O8w	0.0293 (10)	0.0204 (10)	0.0427 (11)	0.0062 (8)	0.0128 (8)	0.0068 (8)
C1	0.0153 (11)	0.0168 (11)	0.0172 (10)	0.0023 (9)	0.0030 (8)	-0.0022 (9)
C2	0.0156 (10)	0.0159 (11)	0.0170 (10)	0.0029 (9)	0.0047 (8)	0.0000 (8)
C3	0.0185 (11)	0.0181 (12)	0.0224 (11)	0.0074 (9)	0.0057 (9)	0.0039 (9)
C4	0.0207 (11)	0.0155 (11)	0.0206 (11)	0.0047 (9)	0.0066 (9)	0.0040 (9)
C5	0.0162 (11)	0.0183 (12)	0.0185 (11)	0.0028 (9)	0.0057 (8)	-0.0004 (9)
C6	0.0180 (11)	0.0180 (12)	0.0222 (11)	0.0066 (9)	0.0055 (9)	0.0046 (9)
C7	0.0190 (11)	0.0155 (11)	0.0182 (10)	0.0028 (9)	0.0046 (8)	0.0029 (9)
C8	0.0181 (11)	0.0172 (12)	0.0189 (10)	0.0026 (9)	0.0044 (8)	-0.0012 (9)
C9	0.0185 (11)	0.0136 (11)	0.0185 (10)	0.0046 (9)	0.0039 (8)	-0.0004 (8)
C10	0.0203 (11)	0.0158 (12)	0.0276 (12)	0.0078 (9)	0.0054 (9)	0.0029 (9)
C11	0.0222 (12)	0.0147 (12)	0.0318 (13)	0.0040 (10)	0.0102 (10)	0.0027 (10)
C12	0.0186 (11)	0.0184 (12)	0.0230 (11)	0.0036 (9)	0.0055 (9)	-0.0021 (9)
C13	0.0215 (12)	0.0185 (12)	0.0247 (12)	0.0082 (10)	0.0023 (9)	0.0034 (9)
C14	0.0216 (12)	0.0156 (12)	0.0245 (11)	0.0032 (10)	0.0064 (9)	0.0022 (9)
C15	0.0141 (10)	0.0166 (11)	0.0176 (10)	0.0052 (9)	0.0031 (8)	0.0036 (8)
C16	0.0147 (10)	0.0166 (11)	0.0142 (10)	0.0043 (9)	0.0021 (8)	0.0016 (8)
C17	0.0163 (11)	0.0163 (11)	0.0189 (10)	0.0055 (9)	0.0051 (8)	0.0024 (9)
C18	0.0188 (11)	0.0148 (11)	0.0221 (11)	0.0030 (9)	0.0053 (9)	0.0008 (9)
C19	0.0121 (10)	0.0209 (12)	0.0200 (11)	0.0036 (9)	0.0040 (8)	0.0051 (9)
C20	0.0175 (11)	0.0190 (12)	0.0217 (11)	0.0070 (9)	0.0064 (9)	0.0001 (9)
C21	0.0163 (11)	0.0161 (11)	0.0189 (10)	0.0026 (9)	0.0028 (8)	-0.0008 (9)

*Geometric parameters (Å, °)*

Tb1—O4	2.2791 (17)	O8w—H81	0.836 (10)
Tb1—O7	2.3162 (16)	O8w—H82	0.840 (10)
Tb1—O8 <sup>i</sup>	2.3276 (16)	C1—C2	1.480 (3)
Tb1—O2w	2.4042 (17)	C2—C3	1.394 (3)
Tb1—O3w	2.4227 (18)	C2—C7	1.398 (3)
Tb1—O2	2.4276 (16)	C3—C4	1.385 (3)
Tb1—O1	2.4388 (17)	C3—H3A	0.9500
Tb1—O1w	2.4482 (17)	C4—C5	1.386 (3)
O1—C1	1.262 (3)	C4—H4	0.9500
O2—C1	1.280 (3)	C5—C6	1.398 (3)
O3—C5	1.364 (3)	C6—C7	1.383 (3)
O3—H3	0.837 (10)	C6—H6A	0.9500
O4—C8	1.256 (3)	C7—H7	0.9500

O5—C8	1.265 (3)	C8—C9	1.494 (3)
O6—C12	1.370 (3)	C9—C10	1.387 (3)
O6—H6	0.833 (10)	C9—C14	1.395 (4)
O7—C15	1.262 (3)	C10—C11	1.395 (3)
O8—C15	1.262 (3)	C10—H10	0.9500
O8—Tb1 <sup>i</sup>	2.3276 (16)	C11—C12	1.375 (4)
O9—C19	1.368 (3)	C11—H11A	0.9500
O9—H9	0.841 (10)	C12—C13	1.389 (4)
O1w—H11	0.839 (10)	C13—C14	1.392 (3)
O1w—H12	0.834 (10)	C13—H13	0.9500
O2w—H21	0.838 (10)	C14—H14	0.9500
O2w—H22	0.832 (10)	C15—C16	1.499 (3)
O3w—H31	0.835 (10)	C16—C21	1.395 (3)
O3w—H32	0.840 (10)	C16—C17	1.397 (3)
O4w—H41	0.837 (10)	C17—C18	1.389 (3)
O4w—H42	0.840 (10)	C17—H17	0.9500
O5w—H51	0.861 (10)	C18—C19	1.394 (3)
O5w—H52	0.852 (10)	C18—H18	0.9500
O6w—H61	0.846 (10)	C19—C20	1.394 (3)
O6w—H62	0.841 (10)	C20—C21	1.394 (3)
O7w—H71	0.842 (10)	C20—H20	0.9500
O7w—H72	0.833 (10)	C21—H21A	0.9500
O4—Tb1—O7	147.98 (6)	C4—C3—H3A	119.9
O4—Tb1—O8 <sup>i</sup>	86.52 (6)	C2—C3—H3A	119.9
O7—Tb1—O8 <sup>i</sup>	97.15 (6)	C3—C4—C5	119.7 (2)
O4—Tb1—O2w	71.93 (6)	C3—C4—H4	120.2
O7—Tb1—O2w	138.23 (6)	C5—C4—H4	120.2
O8 <sup>i</sup> —Tb1—O2w	97.25 (6)	O3—C5—C4	122.0 (2)
O4—Tb1—O3w	136.40 (6)	O3—C5—C6	117.2 (2)
O7—Tb1—O3w	74.94 (6)	C4—C5—C6	120.8 (2)
O8 <sup>i</sup> —Tb1—O3w	77.43 (6)	C7—C6—C5	119.1 (2)
O2w—Tb1—O3w	70.33 (6)	C7—C6—H6A	120.4
O4—Tb1—O2	76.79 (6)	C5—C6—H6A	120.4
O7—Tb1—O2	83.64 (6)	C6—C7—C2	120.5 (2)
O8 <sup>i</sup> —Tb1—O2	147.37 (6)	C6—C7—H7	119.8
O2w—Tb1—O2	103.72 (6)	C2—C7—H7	119.8
O3w—Tb1—O2	133.16 (6)	O4—C8—O5	122.9 (2)
O4—Tb1—O1	108.71 (6)	O4—C8—C9	117.9 (2)
O7—Tb1—O1	78.16 (6)	O5—C8—C9	119.2 (2)
O8 <sup>i</sup> —Tb1—O1	158.46 (6)	C10—C9—C14	119.2 (2)
O2w—Tb1—O1	74.23 (6)	C10—C9—C8	120.3 (2)
O3w—Tb1—O1	81.06 (6)	C14—C9—C8	120.4 (2)
O2—Tb1—O1	53.69 (6)	C9—C10—C11	120.4 (2)
O4—Tb1—O1w	78.28 (6)	C9—C10—H10	119.8
O7—Tb1—O1w	73.07 (6)	C11—C10—H10	119.8
O8 <sup>i</sup> —Tb1—O1w	70.82 (6)	C12—C11—C10	119.7 (2)
O2w—Tb1—O1w	148.59 (6)	C12—C11—H11A	120.2



O3w—Tb1—O1w	130.93 (6)	C10—C11—H11A	120.2
O2—Tb1—O1w	78.37 (6)	O6—C12—C11	116.8 (2)
O1—Tb1—O1w	126.13 (6)	O6—C12—C13	122.2 (2)
C1—O1—Tb1	93.28 (14)	C11—C12—C13	121.0 (2)
C1—O2—Tb1	93.32 (14)	C14—C13—C12	119.1 (2)
C5—O3—H3	107 (2)	C14—C13—H13	120.5
C8—O4—Tb1	150.98 (16)	C12—C13—H13	120.5
C12—O6—H6	114 (2)	C13—C14—C9	120.6 (2)
C15—O7—Tb1	150.17 (16)	C13—C14—H14	119.7
C15—O8—Tb1 <sup>i</sup>	136.88 (15)	C9—C14—H14	119.7
C19—O9—H9	113 (2)	O8—C15—O7	123.1 (2)
Tb1—O1w—H11	125.7 (19)	O8—C15—C16	117.9 (2)
Tb1—O1w—H12	121.8 (19)	O7—C15—C16	119.0 (2)
H11—O1w—H12	109.2 (16)	C21—C16—C17	119.4 (2)
Tb1—O2w—H21	118 (2)	C21—C16—C15	119.8 (2)
Tb1—O2w—H22	110 (2)	C17—C16—C15	120.8 (2)
H21—O2w—H22	110.7 (16)	C18—C17—C16	120.3 (2)
Tb1—O3w—H31	121 (2)	C18—C17—H17	119.8
Tb1—O3w—H32	114.7 (19)	C16—C17—H17	119.8
H31—O3w—H32	109.5 (16)	C17—C18—C19	119.7 (2)
H41—O4w—H42	108.8 (16)	C17—C18—H18	120.1
H51—O5w—H52	105.4 (15)	C19—C18—H18	120.1
H61—O6w—H62	108.3 (16)	O9—C19—C18	122.6 (2)
H71—O7w—H72	108.9 (16)	O9—C19—C20	116.8 (2)
H81—O8w—H82	109.5 (16)	C18—C19—C20	120.6 (2)
O1—C1—O2	119.7 (2)	C21—C20—C19	119.2 (2)
O1—C1—C2	120.5 (2)	C21—C20—H20	120.4
O2—C1—C2	119.7 (2)	C19—C20—H20	120.4
C3—C2—C7	119.6 (2)	C20—C21—C16	120.7 (2)
C3—C2—C1	120.0 (2)	C20—C21—H21A	119.7
C7—C2—C1	120.2 (2)	C16—C21—H21A	119.7
C4—C3—C2	120.2 (2)		
O4—Tb1—O1—C1	-56.96 (14)	C3—C4—C5—C6	3.5 (3)
O7—Tb1—O1—C1	90.56 (14)	O3—C5—C6—C7	176.6 (2)
O8 <sup>i</sup> —Tb1—O1—C1	169.90 (15)	C4—C5—C6—C7	-3.4 (3)
O2w—Tb1—O1—C1	-121.11 (14)	C5—C6—C7—C2	0.5 (3)
O3w—Tb1—O1—C1	166.92 (14)	C3—C2—C7—C6	2.3 (3)
O2—Tb1—O1—C1	-0.22 (12)	C1—C2—C7—C6	-171.8 (2)
O1w—Tb1—O1—C1	31.99 (16)	Tb1—O4—C8—O5	-108.1 (3)
O4—Tb1—O2—C1	125.77 (14)	Tb1—O4—C8—C9	72.0 (4)
O7—Tb1—O2—C1	-79.74 (13)	O4—C8—C9—C10	-156.2 (2)
O8 <sup>i</sup> —Tb1—O2—C1	-173.07 (12)	O5—C8—C9—C10	23.8 (3)
O2w—Tb1—O2—C1	58.44 (14)	O4—C8—C9—C14	21.5 (3)
O3w—Tb1—O2—C1	-17.32 (16)	O5—C8—C9—C14	-158.5 (2)
O1—Tb1—O2—C1	0.22 (12)	C14—C9—C10—C11	-1.5 (4)
O1w—Tb1—O2—C1	-153.71 (14)	C8—C9—C10—C11	176.2 (2)
O7—Tb1—O4—C8	66.9 (4)	C9—C10—C11—C12	1.9 (4)

O8 <sup>i</sup> —Tb1—O4—C8	-31.0 (3)	C10—C11—C12—O6	179.8 (2)
O2 <sub>w</sub> —Tb1—O4—C8	-129.8 (3)	C10—C11—C12—C13	-0.1 (4)
O3 <sub>w</sub> —Tb1—O4—C8	-98.7 (3)	O6—C12—C13—C14	178.0 (2)
O2—Tb1—O4—C8	120.8 (3)	C11—C12—C13—C14	-2.1 (4)
O1—Tb1—O4—C8	164.6 (3)	C12—C13—C14—C9	2.4 (4)
O1 <sub>w</sub> —Tb1—O4—C8	40.1 (3)	C10—C9—C14—C13	-0.7 (4)
O4—Tb1—O7—C15	169.6 (3)	C8—C9—C14—C13	-178.4 (2)
O8 <sup>i</sup> —Tb1—O7—C15	-95.6 (3)	Tb1 <sup>i</sup> —O8—C15—O7	14.8 (4)
O2 <sub>w</sub> —Tb1—O7—C15	13.8 (3)	Tb1 <sup>i</sup> —O8—C15—C16	-163.99 (16)
O3 <sub>w</sub> —Tb1—O7—C15	-20.7 (3)	Tb1—O7—C15—O8	81.1 (4)
O2—Tb1—O7—C15	117.3 (3)	Tb1—O7—C15—C16	-100.2 (3)
O1—Tb1—O7—C15	63.1 (3)	O8—C15—C16—C21	167.8 (2)
O1 <sub>w</sub> —Tb1—O7—C15	-163.0 (3)	O7—C15—C16—C21	-11.1 (3)
Tb1—O1—C1—O2	0.4 (2)	O8—C15—C16—C17	-10.6 (3)
Tb1—O1—C1—C2	-176.78 (18)	O7—C15—C16—C17	170.5 (2)
Tb1—O2—C1—O1	-0.4 (2)	C21—C16—C17—C18	-0.9 (3)
Tb1—O2—C1—C2	176.80 (18)	C15—C16—C17—C18	177.6 (2)
O1—C1—C2—C3	-7.7 (3)	C16—C17—C18—C19	0.0 (3)
O2—C1—C2—C3	175.1 (2)	C17—C18—C19—O9	-179.5 (2)
O1—C1—C2—C7	166.3 (2)	C17—C18—C19—C20	1.2 (4)
O2—C1—C2—C7	-10.8 (3)	O9—C19—C20—C21	179.2 (2)
C7—C2—C3—C4	-2.3 (3)	C18—C19—C20—C21	-1.5 (4)
C1—C2—C3—C4	171.8 (2)	C19—C20—C21—C16	0.5 (3)
C2—C3—C4—C5	-0.6 (4)	C17—C16—C21—C20	0.6 (3)
C3—C4—C5—O3	-176.5 (2)	C15—C16—C21—C20	-177.8 (2)

Symmetry code: (i)  $-x+1, -y+1, -z+1$ .

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O3—H3 $\cdots$ O7 <sub>w</sub> <sup>ii</sup>	0.84 (1)	1.82 (2)	2.612 (3)	158 (3)
O6—H6 $\cdots$ O6 <sub>w</sub> <sup>iii</sup>	0.83 (1)	1.87 (1)	2.678 (3)	165 (4)
O9—H9 $\cdots$ O4 <sub>w</sub> <sup>iv</sup>	0.84 (1)	1.98 (2)	2.763 (3)	156 (3)
O1 <sub>w</sub> —H11 $\cdots$ O4 <sub>w</sub>	0.84 (1)	2.04 (1)	2.870 (3)	171 (3)
O1 <sub>w</sub> —H12 $\cdots$ O9 <sup>v</sup>	0.83 (1)	1.96 (1)	2.766 (3)	164 (3)
O2 <sub>w</sub> —H22 $\cdots$ O5 <sub>w</sub>	0.83 (1)	1.86 (1)	2.678 (3)	167 (3)
O2 <sub>w</sub> —H21 $\cdots$ O5 <sub>w</sub> <sup>vi</sup>	0.84 (1)	2.20 (2)	2.952 (3)	150 (3)
O3 <sub>w</sub> —H31 $\cdots$ O3 <sup>iii</sup>	0.84 (1)	1.95 (1)	2.777 (3)	171 (3)
O3 <sub>w</sub> —H32 $\cdots$ O7 <sup>i</sup>	0.84 (1)	2.25 (2)	2.916 (2)	137 (3)
O4 <sub>w</sub> —H41 $\cdots$ O8 <sub>w</sub> <sup>vii</sup>	0.84 (1)	1.94 (1)	2.753 (3)	163 (3)
O4 <sub>w</sub> —H42 $\cdots$ O5 <sup>viii</sup>	0.84 (1)	2.10 (1)	2.927 (3)	169 (3)
O5 <sub>w</sub> —H52 $\cdots$ O7 <sub>w</sub>	0.85 (1)	2.01 (2)	2.811 (3)	157 (3)
O5 <sub>w</sub> —H51 $\cdots$ O8 <sub>w</sub> <sup>ix</sup>	0.86 (1)	1.96 (1)	2.791 (3)	163 (3)
O6 <sub>w</sub> —H61 $\cdots$ O2	0.85 (1)	1.90 (1)	2.743 (2)	172 (3)
O6 <sub>w</sub> —H62 $\cdots$ O5	0.84 (1)	1.95 (1)	2.785 (2)	171 (3)
O7 <sub>w</sub> —H71 $\cdots$ O6 <sub>w</sub>	0.84 (1)	1.90 (1)	2.732 (3)	172 (3)
O7 <sub>w</sub> —H72 $\cdots$ O5 <sup>x</sup>	0.83 (1)	1.89 (1)	2.725 (3)	175 (3)

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O8 <sub>w</sub> —H81...O1 <sup>xi</sup>	0.84 (1)	2.10 (2)	2.818 (3)	144 (3)
O8 <sub>w</sub> —H82...O6	0.84 (1)	1.90 (1)	2.713 (3)	164 (3)

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Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x+2, -y+1, -z$ ; (iii)  $x-1, y, z$ ; (iv)  $x+1, y+1, z$ ; (v)  $-x+2, -y+1, -z+1$ ; (vi)  $-x+1, -y+1, -z$ ; (vii)  $-x, -y, -z+1$ ; (viii)  $-x+1, -y, -z+1$ ; (ix)  $-x, -y, -z$ ; (x)  $-x+1, -y, -z$ ; (xi)  $x-1, y-1, z$ .