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Poly[(μ_4 -tetrazole-1-acetato- $\kappa^4N^3:N^4:O:-O'$)silver(I)]Shi-Jie Li,^a Hao Wang,^a Wen-Dong Song,^{b*} Shi-Wei Hu^a and Pei-Wen Qin^c

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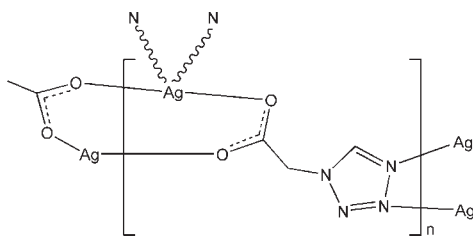
Received 29 December 2009; accepted 11 January 2010

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(C-C) = 0.012$ Å; R factor = 0.062; wR factor = 0.177; data-to-parameter ratio = 13.8.

In the title complex, $[Ag(C_3H_3N_4O_2)]_n$, the Ag^I atom is four-coordinated in a slightly distorted tetrahedral coordination geometry by two N atoms from two tetrazole-1-acetate (tza) ligands and two O atoms from the other two tza ligands. The tza ligand bridges two Ag atoms through the carboxylate O atoms and simultaneously binds to the other two Ag atoms through the tetrazole N atoms, forming a two-dimensional network parallel to (100).

Related literature

For the diverse coordination modes and potential applications of metal complexes with tetrazole derivatives, see: Stagni *et al.* (2006); Ye *et al.* (2006).



Experimental

Crystal data

 $[Ag(C_3H_3N_4O_2)]$ $M_r = 234.96$

Triclinic, $P\bar{1}$
 $a = 5.1584$ (10) Å
 $b = 7.7805$ (16) Å
 $c = 7.8711$ (16) Å
 $\alpha = 109.40$ (3)°
 $\beta = 98.87$ (3)°
 $\gamma = 104.85$ (3)°

$V = 277.92$ (14) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 3.56$ mm⁻¹
 $T = 293$ K
 $0.25 \times 0.23 \times 0.21$ mm

Data collection

Rigaku/MSM Mercury CCD diffractometer
 Absorption correction: multi-scan (REQAB; Jacobson, 1998)
 $T_{min} = 0.470$, $T_{max} = 0.522$

2722 measured reflections
 1267 independent reflections
 1150 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.056$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.177$
 $S = 1.23$
 1267 reflections

92 parameters
 H-atom parameters constrained
 $\Delta\rho_{max} = 2.15$ e Å⁻³
 $\Delta\rho_{min} = -0.97$ e Å⁻³

Table 1

Selected bond lengths (Å).

Ag1—O1	2.330 (7)	Ag1—N3 ⁱⁱ	2.494 (9)
Ag1—O2 ⁱ	2.282 (7)	Ag1—N4 ⁱⁱⁱ	2.442 (8)

Symmetry codes: (i) $-x + 1, -y, -z$; (ii) $x, y - 1, z - 1$; (iii) $-x + 1, -y, -z + 1$.

Data collection: *CrystalStructure* (Rigaku/MSM, 2002); cell refinement: *CrystalStructure*; data reduction: *CrystalStructure*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEPII* (Johnson, 1976) and *DIAMOND* (Brandenburg, 1999); software used to prepare material for publication: *SHELXL97*.

The authors acknowledge Guang Dong Ocean University for supporting this work.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HY2270).

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supporting information

Acta Cryst. (2010). E66, m160 [https://doi.org/10.1107/S1600536810001236]

Poly[(μ_4 -tetrazole-1-acetato- κ^4 N³:N⁴:O:O')silver(I)]**Shi-Jie Li, Hao Wang, Wen-Dong Song, Shi-Wei Hu and Pei-Wen Qin****S1. Comment**

In recent years, organic ligands with a tetrazole functional group have been greatly used in coordination chemistry for construction of metal-organic frameworks due to their diverse coordination modes and potential applications in varied fields (Stagni *et al.*, 2006; Ye *et al.*, 2006). The reaction of tetrazole-1-acetic acid (Htza) with AgNO₃ in an alkaline aqueous solution yielded a new Ag^I coordination polymer, whose crystal structure is reported here.

In the title complex, the Ag^I atom is four-coordinated in a slightly distorted tetrahedral coordination geometry by two N atoms and two O atoms from four different tza ligands (Table 1), as illustrated in Fig. 1. The adjacent Ag^I atoms are co-bridged by tza ligands. The tza ligand acts as a tetradentate ligand, bridging two Ag atoms through its carboxylate O atoms, while simultaneously binding to the other two Ag atoms through two N atoms of the tetrazole group, forming a two-dimensional network parallel to (1 0 0).

S2. Experimental

A mixture of AgNO₃ (0.073 g, 0.5 mmol) and Htza (0.990 g, 0.5 mmol) in 15 ml of H₂O solution was sealed in an autoclave equipped with a Teflon liner (20 ml) and then heated at 373 K for 4 d. Crystals of the title compound were obtained by slow evaporation of the solvent at room temperature.

S3. Refinement

H atoms were placed at calculated positions and treated as riding on the parent C atoms, with C—H = 0.93 (CH) and 0.97 (CH₂) Å and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The highest residual electron density was found 1.40 Å from N4 and the deepest hole 1.12 Å from Ag1.

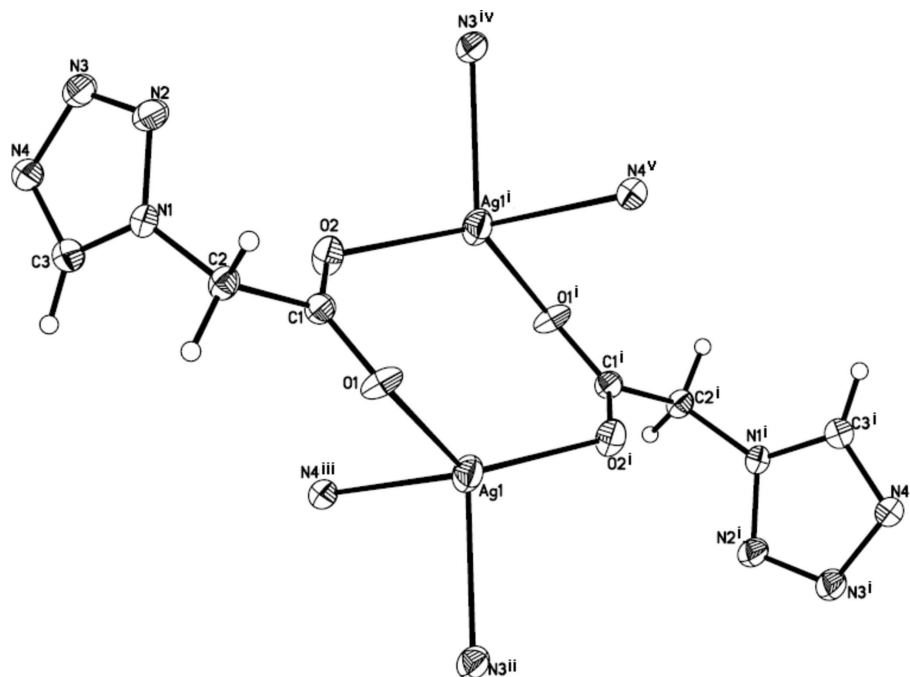


Figure 1

The asymmetric unit of the title compound, with symmetrically related atoms to complete the Ag coordination.

Displacement ellipsoids are drawn at the 30% probability level. [Symmetry codes: (i) $1-x, -y, -z$; (ii) $x, -1+y, -1+z$; (iii) $1-x, -y, 1-z$; (iv) $1-x, 1-y, 1-z$; (v) $x, y, -1+z$.]

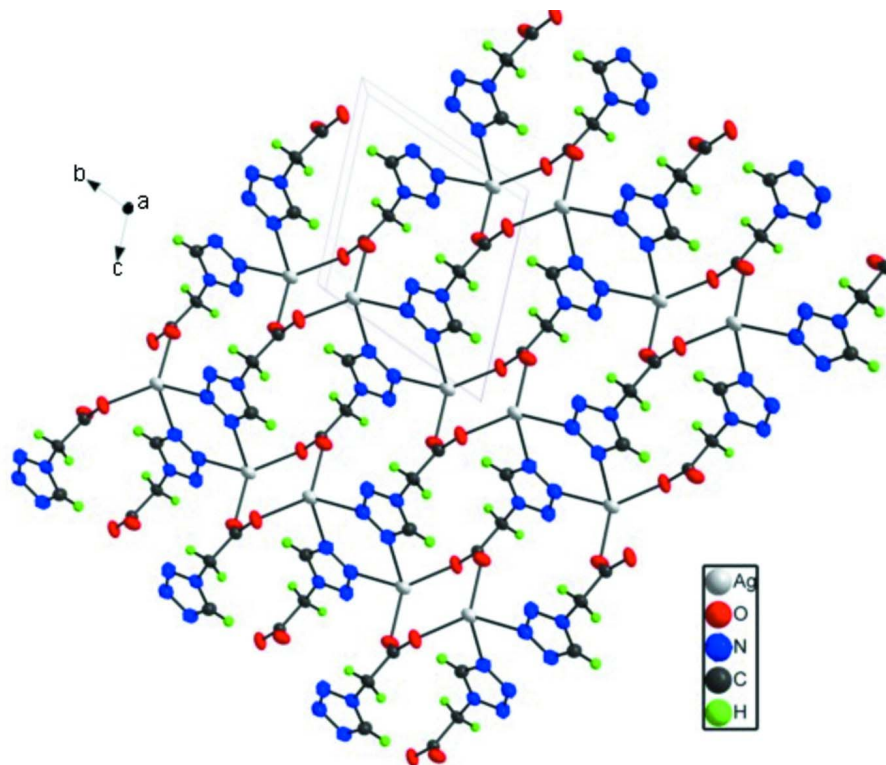


Figure 2

A view of the layer structure of the title compound.

Poly[(μ_4 -tetrazole-1-acetato- $\kappa^4 N^3:N^4:O:O'$)silver(I)]

Crystal data

[Ag(C₃H₃N₄O₂)]

$M_r = 234.96$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

$a = 5.1584$ (10) Å

$b = 7.7805$ (16) Å

$c = 7.8711$ (16) Å

$\alpha = 109.40$ (3)°

$\beta = 98.87$ (3)°

$\gamma = 104.85$ (3)°

$V = 277.92$ (14) Å³

$Z = 2$

$F(000) = 224$

$D_x = 2.808$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 3600 reflections

$\theta = 1.4$ – 28°

$\mu = 3.56$ mm⁻¹

$T = 293$ K

Block, blue

$0.25 \times 0.23 \times 0.21$ mm

Data collection

Rigaku/MSM Mercury CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scans

Absorption correction: multi-scan
(REQAB; Jacobson, 1998)

$T_{\min} = 0.470$, $T_{\max} = 0.522$

2722 measured reflections

1267 independent reflections

1150 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.056$

$\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 3.2^\circ$

$h = -6 \rightarrow 6$

$k = -9 \rightarrow 10$

$l = -10 \rightarrow 9$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.177$
 $S = 1.23$
 1267 reflections
 92 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods
 Secondary atom site location: difference Fourier
 map

Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0519P)^2 + 3.2858P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 2.15 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.97 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL97* (Sheldrick,
 2008), $F_c^* = kFc[1 + 0.001x \text{Fc}^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.052 (15)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Ag1	0.61189 (17)	-0.19436 (12)	-0.05077 (10)	0.0387 (5)
O1	0.9281 (15)	0.0639 (10)	0.2082 (10)	0.0365 (16)
O2	0.6195 (14)	0.1949 (12)	0.3227 (10)	0.0360 (16)
N1	0.9120 (16)	0.3243 (11)	0.6770 (10)	0.0277 (16)
N2	0.872 (2)	0.4946 (12)	0.7248 (12)	0.0376 (19)
N3	0.7161 (19)	0.4973 (12)	0.8396 (12)	0.0363 (19)
N4	0.656 (2)	0.3322 (13)	0.8697 (12)	0.0351 (18)
C1	0.8444 (18)	0.1645 (13)	0.3374 (12)	0.0265 (17)
C2	1.0509 (18)	0.2615 (13)	0.5323 (12)	0.0267 (17)
H2A	1.1368	0.1710	0.5545	0.032*
H2B	1.1964	0.3719	0.5373	0.032*
C3	0.781 (2)	0.2254 (15)	0.7646 (14)	0.034 (2)
H3	0.7775	0.1022	0.7545	0.041*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ag1	0.0430 (6)	0.0441 (6)	0.0259 (5)	0.0145 (4)	0.0071 (3)	0.0105 (3)
O1	0.033 (3)	0.031 (3)	0.034 (4)	0.013 (3)	0.010 (3)	-0.003 (3)
O2	0.028 (3)	0.055 (4)	0.029 (3)	0.020 (3)	0.008 (3)	0.016 (3)
N1	0.033 (4)	0.030 (4)	0.021 (3)	0.011 (3)	0.008 (3)	0.009 (3)
N2	0.051 (5)	0.027 (4)	0.033 (4)	0.013 (4)	0.017 (4)	0.008 (3)
N3	0.045 (5)	0.030 (4)	0.030 (4)	0.010 (4)	0.012 (4)	0.008 (3)
N4	0.048 (5)	0.034 (4)	0.030 (4)	0.020 (4)	0.017 (4)	0.014 (3)
C1	0.026 (4)	0.028 (4)	0.025 (4)	0.010 (3)	0.005 (3)	0.009 (3)
C2	0.023 (4)	0.032 (4)	0.022 (4)	0.010 (3)	0.003 (3)	0.007 (3)
C3	0.041 (5)	0.033 (5)	0.032 (5)	0.014 (4)	0.013 (4)	0.015 (4)

Geometric parameters (\AA , $^\circ$)

Ag1—O1	2.330 (7)	N1—C2	1.453 (11)
Ag1—O2 ⁱ	2.282 (7)	N2—N3	1.297 (12)
Ag1—N3 ⁱⁱ	2.494 (9)	N3—N4	1.350 (12)

Ag1—N4 ⁱⁱⁱ	2.442 (8)	N4—C3	1.331 (13)
O1—C1	1.270 (11)	C1—C2	1.540 (12)
O2—C1	1.238 (11)	C2—H2A	0.9700
N1—C3	1.324 (12)	C2—H2B	0.9700
N1—N2	1.331 (12)	C3—H3	0.9300
O2 ⁱ —Ag1—O1	129.2 (3)	C3—N4—N3	105.1 (8)
O2 ⁱ —Ag1—N4 ⁱⁱⁱ	118.7 (3)	C3—N4—Ag1 ⁱⁱⁱ	117.8 (6)
O1—Ag1—N4 ⁱⁱⁱ	95.0 (3)	N3—N4—Ag1 ⁱⁱⁱ	137.0 (6)
O2 ⁱ —Ag1—N3 ⁱⁱ	102.2 (3)	O2—C1—O1	127.3 (9)
O1—Ag1—N3 ⁱⁱ	118.0 (3)	O2—C1—C2	117.2 (8)
N4 ⁱⁱⁱ —Ag1—N3 ⁱⁱ	86.0 (3)	O1—C1—C2	115.5 (8)
C1—O1—Ag1	120.6 (6)	N1—C2—C1	111.1 (7)
C1—O2—Ag1 ⁱ	121.2 (6)	N1—C2—H2A	109
C3—N1—N2	109.2 (8)	C1—C2—H2A	109
C3—N1—C2	128.9 (8)	N1—C2—H2B	109
N2—N1—C2	121.4 (8)	C1—C2—H2B	109
N3—N2—N1	106.3 (8)	H2A—C2—H2B	108
N2—N3—N4	111.0 (8)	N1—C3—N4	108.4 (9)
N2—N3—Ag1 ^{iv}	112.1 (6)	N1—C3—H3	126
N4—N3—Ag1 ^{iv}	136.9 (6)	N4—C3—H3	126

Symmetry codes: (i) $-x+1, -y, -z$; (ii) $x, y-1, z-1$; (iii) $-x+1, -y, -z+1$; (iv) $x, y+1, z+1$.