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## Structure Reports

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# Benzoic acid-2-\{(E)-[(E)-2-(2-pyridyl-methylidene)hydrazin-1-ylidene]methyl\}pyridine (2/1) 

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Key indicators: single-crystal X-ray study; $T=98 \mathrm{~K}$; mean $\sigma(\mathrm{C}-\mathrm{C})=0.002 \AA ; R$ factor $=$ $0.042 ; w R$ factor $=0.117$; data-to-parameter ratio $=12.2$.

## Experimental

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4} \cdot 2 \mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{2}$
$M_{r}=454.48$
Triclinic, $P \overline{1}$
$a=4.4509$ (7) $\AA$
$b=11.3635$ (17) A
$c=12.0612(17) \AA$
$\alpha=108.985$ (6) ${ }^{\circ}$
$\beta=99.830(9)^{\circ}$

## Data collection

Rigaku AFC12/SATURN724 diffractometer
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)
$T_{\text {min }}=0.759, T_{\text {max }}=1.000$

## Refinement

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.042$
$w R\left(F^{2}\right)=0.117$
$S=1.00$
1935 reflections
158 parameters 1 restraint
$\gamma=97.849(10)^{\circ}$
$V=556.16(14) \AA^{3}$
$Z=1$
Mo $K \alpha$ radiation
$\mu=0.09 \mathrm{~mm}^{-1}$
$T=98 \mathrm{~K}$
$0.40 \times 0.29 \times 0.12 \mathrm{~mm}$

2790 measured reflections 1935 independent reflections 1811 reflections with $I>2 \sigma(I)$ $R_{\text {int }}=0.022$

H atoms treated by a mixture of independent and constrained refinement
$\Delta \rho_{\text {max }}=0.20 \mathrm{e}_{\AA^{-3}}$
$\Delta \rho_{\text {min }}=-0.23 \mathrm{e}^{-3}$

The asymmetric unit of the title cocrystal, $\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4} \cdot 2 \mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{2}$, comprises a single molecule of benzoic acid and one halfmolecule of 2-pyridinealdazine situated about a centre of inversion. The carboxyl group is coplanar with the benzene ring to which it is connected $\left[\mathrm{O}-\mathrm{C}-\mathrm{C}-\mathrm{C}=-172.47\right.$ (12) ${ }^{\circ}$ ] and similarly, the 2-pyridinealdazine molecule is planar (r.m.s. deviation of the 16 non- H atoms $=0.017 \AA$ ). In the crystal, molecules are connected into a non-planar three-molecule aggregate [dihedral angle between the benzene and pyridyl ring connected by the hydrogen bond $=61.30(7)^{\circ}$ ] with a twisted Z-shape. Layers of 2-pyridinealdazine molecules in the $a b$ plane are sandwiched by benzoic acid molecules being connected by $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ interactions, the latter involving the carbonyl O atom so that each benzoic acid molecule links three different 2-pyridinealdazine molecules. Interdigitated layers stack along the $c$ axis.

## Related literature

For related studies on co-crystal formation involving the isomeric n-pyridinealdazines, see: Broker et al. (2008); Arman et al. $(2010 a, b)$.


Table 1
Hydrogen-bond geometry ( $\AA{ }^{\circ}{ }^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O} 1-\mathrm{H} 1 o \cdots \mathrm{~N} 1$ | $0.85(2)$ | $1.88(2)$ | $2.7269(16)$ | $177(2)$ |
| C11-H11 $\mathrm{O}^{\mathrm{i}}$ | 0.95 | 2.54 | $3.1811(19)$ | 125 |
| $\mathrm{C}^{\mathrm{i}} 2-\mathrm{H} 12 \cdots 2^{\mathrm{ii}}$ | 0.95 | 2.59 | $3.4647(19)$ | 154 |

Symmetry codes: (i) $-x+1,-y+2,-z+1$; (ii) $x-1, y, z$.
Data collection: CrystalClear (Molecular Structure Corporation \& Rigaku, 2005); cell refinement: CrystalClear; data reduction: CrystalClear; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997) and DIAMOND (Brandenburg, 2006); software used to prepare material for publication: publCIF (Westrip, 2010).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2725).

## References

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## supporting information

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# Benzoic acid-2-\{(E)-[(E)-2-(2-pyridylmethylidene)hydrazin-1-ylidene]methyl\}pyridine (2/1) 

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## S1. Comment

Co-crystallization experiments with the isomeric n-pyridinealdazines have led to the characterization of several cocrystals (Broker et al., 2008; Arman et al., 2010a; Arman et al., 2010b), and in continuation of these studies, the cocrystallization of benzoic acid and 2-pyridinealdazine was investigated. This lead to the isolation of the title $2 / 1$ cocrystal, (I).
The asymmetric unit in (I) comprises a molecule of benzoic acid, Fig. 1, and half a molecule of 2-pyridinealdazine, with the latter disposed about a centre of inversion, Fig. 2. The constituents of (I) are connected by $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds, Table 1, to generate a centrosymmetric three molecule aggregate, Fig. 3. The benzoic acid molecule is planar as seen in the value of the $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ torsion angle of $-172.47(12)^{\circ}$. Similarly, the 2-pyridinealdazine molecule is planar with the r.m.s. deviation of the 16 non-hydrogen atoms being $0.017 \AA$. However, the three molecule aggregate is not planar as the benzene ring forms a dihedral angle of $61.30(7)^{\circ}$ with the pyridyl ring to which it is hydrogen bonded. Overall, when viewed normal to the plane through 2-pyridinealdazine, the aggregate has the shape of a twisted letter $Z$. In the crystal packing, the 2-pyridinealdazine molecules pack in the $a b$ plane with the benzoic acid molecules sandwiching these, Fig. 4. The connections are mediated by the aforementioned $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bond as well as C $\mathrm{H} \cdots \mathrm{O}$ interactions formed by the carbonyl-O atom; each benzoic acid molecule links three distinct 2-pyridinealdazine molecules. Inter-digitated layers stack along the $c$ axis.

## S2. Experimental

Yellow crystals of (I) were isolated from the $2 / 1$ co-crystallization of benzoic acid (Sigma Aldrich, 0.24 mmol ) and $2-[(E)-[(E)$-2-(pyridin-2-ylmethylidene)hydrazin-1-ylidene]methyl]pyridine (Sigma Aldrich, 0.12 mmol ) in ethanol, m . pt. $351-353 \mathrm{~K}$.
IR assignment $\left(\mathrm{cm}^{-1}\right): 2600($ br $) v(\mathrm{O}-\mathrm{H}) ; 1691 v(\mathrm{C}=\mathrm{O}) ; 1627 v(\mathrm{C}=\mathrm{N}) ; 1469$, 1451, $1416 v(\mathrm{C}-\mathrm{C}($ aromatic $)) ; 1627 v(\mathrm{C}$ $-\mathrm{N}) ; 777 \delta(\mathrm{C}-\mathrm{H})$.

## S3. Refinement

C-bound H -atoms were placed in calculated positions ( $\mathrm{C}-\mathrm{H} 0.95-0.99 \AA$ ) and were included in the refinement in the riding model approximation with $U_{\text {iso }}(\mathrm{H})$ set to $1.2 U_{\text {eq }}(\mathrm{C})$. The O -bound H -atom was located in a difference Fourier map and was refined with a distance restraint of $\mathrm{O}-\mathrm{H} 0.84 \pm 0.01 \AA$, and with $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{O})$. In the final refinement a low angle reflection evidently effected by the beam stop was omitted, i.e. $\binom{0}{1}$.


Figure 1
Molecular structure of benzoic acid found in the structure of (I) showing atom-labelling scheme and displacement ellipsoids at the 50\% probability level


Figure 2
Molecular structure of 2-pyridinealdazine found in the structure of (I) showing atom-labelling scheme and displacement ellipsoids at the $50 \%$ probability level. The molecule is disposed about a centre of inversion and $i=2-x, 1-y, 1-z$.


Figure 3
The three molecule aggregate in (I) highlighting the extended chair conformation. The $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds are shown as orange dashed lines.


Figure 4
A view of the supramolecular layer in (I) whereby 2-pyridinealdazine molecules are sandwiched by benzoic acid molecules. The $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ contacts are shown as orange and blue dashed lines, respectively.


Figure 5
A view in projection down the $a$ axis showing the stacking of layers comprising three molecule aggregates along $c$. The O $-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds and $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ contacts are shown as orange and blue dashed lines, respectively.

Benzoic acid-2-\{(E)-[(E)-2-(2-pyridylmethylidene)hydrazin-1- ylidene]methyl\}pyridine (2/1)

## Crystal data

$\mathrm{C}_{12} \mathrm{H}_{10} \mathrm{~N}_{4} \cdot 2 \mathrm{C}_{7} \mathrm{H}_{6} \mathrm{O}_{2}$
$M_{r}=454.48$
Triclinic, $P \overline{1}$
Hall symbol: -P 1
$a=4.4509$ (7) $\AA$
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$c=12.0612$ (17) $\AA$
$\alpha=108.985$ (6) ${ }^{\circ}$
$\beta=99.830(9)^{\circ}$

$$
\begin{aligned}
& \gamma=97.849(10)^{\circ} \\
& V=556.16(14) \AA^{3} \\
& Z=1 \\
& F(000)=238 \\
& D_{\mathrm{x}}=1.357 \mathrm{Mg} \mathrm{~m}^{-3} \\
& \text { Mo } K \alpha \text { radiation, }, \lambda=0.71073 \AA \\
& \text { Cell parameters from } 2619 \text { reflections } \\
& \theta=2.1-40.2^{\circ} \\
& \mu=0.09 \mathrm{~mm}^{-1}
\end{aligned}
$$

$T=98 \mathrm{~K}$
Block, yellow

## Data collection

Rigaku AFC12K/SATURN724 diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
$\omega$ scans
Absorption correction: multi-scan
(ABSCOR; Higashi, 1995)
$T_{\text {min }}=0.759, T_{\text {max }}=1.000$

## Refinement

Refinement on $F^{2}$
Least-squares matrix: full
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.042$
$w R\left(F^{2}\right)=0.117$
$S=1.00$
1935 reflections
158 parameters
1 restraint
Primary atom site location: structure-invariant direct methods
$0.40 \times 0.29 \times 0.12 \mathrm{~mm}$

2790 measured reflections
1935 independent reflections
1811 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.022$
$\theta_{\text {max }}=25.0^{\circ}, \theta_{\text {min }}=3.1^{\circ}$
$h=-5 \rightarrow 4$
$k=-13 \rightarrow 13$
$l=-14 \rightarrow 14$

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H atoms treated by a mixture of independent and constrained refinement
$w=1 /\left[\sigma^{2}\left(F_{0}^{2}\right)+(0.0744 P)^{2}+0.1759 P\right]$
where $P=\left(F_{0}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.20 \mathrm{e}_{\AA^{-3}}$
$\Delta \rho_{\text {min }}=-0.23$ e $\AA^{-3}$

## Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two 1.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving 1.s. planes.
Refinement. Refinement of $F^{2}$ against ALL reflections. The weighted $R$-factor $w R$ and goodness of fit $S$ are based on $F^{2}$, conventional $R$-factors $R$ are based on $F$, with $F$ set to zero for negative $F^{2}$. The threshold expression of $F^{2}>2 \sigma\left(F^{2}\right)$ is used only for calculating $R$-factors(gt) etc. and is not relevant to the choice of reflections for refinement. $R$-factors based on $F^{2}$ are statistically about twice as large as those based on $F$, and $R$ - factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\hat{A}^{2}$ )

|  | $x$ | $y$ | $z$ | $U_{\text {iso }}{ }^{*} / U_{\text {eq }}$ |
| :--- | :--- | :--- | :--- | :--- |
| O1 | $0.6343(3)$ | $0.67141(10)$ | $0.18059(9)$ | $0.0287(3)$ |
| H1o | $0.584(5)$ | $0.692(2)$ | $0.2479(12)$ | $0.050(6)^{*}$ |
| O2 | $0.7598(3)$ | $0.88248(10)$ | $0.23009(9)$ | $0.0315(3)$ |
| C1 | $0.7523(3)$ | $0.77587(13)$ | $0.16270(12)$ | $0.0235(3)$ |
| C2 | $0.8828(3)$ | $0.74881(14)$ | $0.05272(12)$ | $0.0227(3)$ |
| C3 | $1.0455(3)$ | $0.85068(14)$ | $0.03269(13)$ | $0.0278(4)$ |
| H3 | 1.0702 | 0.9349 | 0.0878 | $0.033^{*}$ |
| C4 | $1.1716(4)$ | $0.82926(15)$ | $-0.06766(14)$ | $0.0311(4)$ |
| H4 | 1.2846 | 0.8987 | -0.0809 | $0.037^{*}$ |
| C5 | $1.1331(4)$ | $0.70640(15)$ | $-0.14890(13)$ | $0.0289(4)$ |
| H5 | 1.2192 | 0.6919 | -0.2178 | $0.035^{*}$ |
| C6 | $0.9695(4)$ | $0.60477(15)$ | $-0.12974(13)$ | $0.0294(4)$ |
| H6 | 0.9428 | 0.5208 | -0.1856 | $0.035^{*}$ |


| C7 | $0.8444(3)$ | $0.62603(14)$ | $-0.02863(13)$ | $0.0263(3)$ |
| :--- | :--- | :--- | :--- | :--- |
| H7 | 0.7326 | 0.5564 | -0.0152 | $0.032^{*}$ |
| N1 | $0.4834(3)$ | $0.73022(11)$ | $0.39929(10)$ | $0.0214(3)$ |
| N2 | $0.9261(3)$ | $0.54922(10)$ | $0.52702(10)$ | $0.0211(3)$ |
| C8 | $0.6169(3)$ | $0.69350(12)$ | $0.48783(12)$ | $0.0190(3)$ |
| C9 | $0.6033(3)$ | $0.75148(13)$ | $0.60760(12)$ | $0.0219(3)$ |
| H9 | 0.7002 | 0.7236 | 0.6683 | $0.026^{*}$ |
| C10 | $0.4469(3)$ | $0.85009(13)$ | $0.63643(12)$ | $0.0233(3)$ |
| H10 | 0.4365 | 0.8918 | 0.7175 | $0.028^{*}$ |
| C11 | $0.3052(3)$ | $0.88738(13)$ | $0.54546(13)$ | $0.0231(3)$ |
| H11 | 0.1932 | 0.9541 | 0.5627 | $0.028^{*}$ |
| C12 | $0.3304(3)$ | $0.82529(13)$ | $0.42883(13)$ | $0.0230(3)$ |
| H12 | 0.2344 | 0.8515 | 0.3668 | $0.028^{*}$ |
| C13 | $0.7833(3)$ | $0.58938(13)$ | $0.45006(12)$ | $0.0209(3)$ |
| H13 | 0.7846 | 0.5516 | 0.3674 | $0.025^{*}$ |

Atomic displacement parameters $\left(\AA^{2}\right)$

|  | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| O1 | $0.0405(6)$ | $0.0251(6)$ | $0.0230(5)$ | $0.0084(5)$ | $0.0124(5)$ | $0.0085(4)$ |
| O2 | $0.0457(7)$ | $0.0254(6)$ | $0.0247(6)$ | $0.0126(5)$ | $0.0110(5)$ | $0.0070(5)$ |
| C1 | $0.0261(7)$ | $0.0244(7)$ | $0.0208(7)$ | $0.0084(6)$ | $0.0028(5)$ | $0.0093(6)$ |
| C2 | $0.0231(7)$ | $0.0261(7)$ | $0.0200(7)$ | $0.0083(6)$ | $0.0018(5)$ | $0.0098(6)$ |
| C3 | $0.0333(8)$ | $0.0248(8)$ | $0.0231(7)$ | $0.0064(6)$ | $0.0035(6)$ | $0.0070(6)$ |
| C4 | $0.0336(8)$ | $0.0315(8)$ | $0.0299(8)$ | $0.0025(7)$ | $0.0065(6)$ | $0.0151(7)$ |
| C5 | $0.0302(8)$ | $0.0372(8)$ | $0.0238(7)$ | $0.0118(6)$ | $0.0087(6)$ | $0.0135(6)$ |
| C6 | $0.0349(8)$ | $0.0277(8)$ | $0.0262(8)$ | $0.0103(6)$ | $0.0091(6)$ | $0.0075(6)$ |
| C7 | $0.0314(8)$ | $0.0236(7)$ | $0.0259(7)$ | $0.0084(6)$ | $0.0081(6)$ | $0.0098(6)$ |
| N1 | $0.0212(6)$ | $0.0202(6)$ | $0.0235(6)$ | $0.0037(5)$ | $0.0055(5)$ | $0.0088(5)$ |
| N2 | $0.0200(6)$ | $0.0203(6)$ | $0.0242(6)$ | $0.0063(5)$ | $0.0077(5)$ | $0.0073(5)$ |
| C8 | $0.0165(6)$ | $0.0182(6)$ | $0.0232(7)$ | $0.0018(5)$ | $0.0060(5)$ | $0.0087(5)$ |
| C9 | $0.0205(7)$ | $0.0234(7)$ | $0.0231(7)$ | $0.0037(5)$ | $0.0059(5)$ | $0.0100(6)$ |
| C10 | $0.0251(7)$ | $0.0223(7)$ | $0.0230(7)$ | $0.0036(6)$ | $0.0100(6)$ | $0.0069(6)$ |
| C11 | $0.0214(7)$ | $0.0183(7)$ | $0.0312(8)$ | $0.0052(5)$ | $0.0092(6)$ | $0.0091(6)$ |
| C12 | $0.0220(7)$ | $0.0217(7)$ | $0.0272(7)$ | $0.0050(5)$ | $0.0040(5)$ | $0.0117(6)$ |
| C13 | $0.0194(6)$ | $0.0204(7)$ | $0.0237(7)$ | $0.0036(5)$ | $0.0076(5)$ | $0.0078(6)$ |
|  |  |  |  |  |  |  |

Geometric parameters ( $\AA$, ${ }^{\circ}$ )

| $\mathrm{O} 1-\mathrm{C} 1$ | $1.3292(18)$ | $\mathrm{N} 1-\mathrm{C} 12$ | $1.3381(18)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 1-\mathrm{H} 1 \mathrm{o}$ | $0.846(9)$ | $\mathrm{N} 1-\mathrm{C} 8$ | $1.3449(18)$ |
| $\mathrm{O} 2-\mathrm{C} 1$ | $1.2117(17)$ | $\mathrm{N} 2-\mathrm{C} 13$ | $1.2757(18)$ |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.496(2)$ | $\mathrm{N} 2-\mathrm{N} 2^{\mathrm{i}}$ | $1.408(2)$ |
| $\mathrm{C} 2-\mathrm{C} 7$ | $1.388(2)$ | $\mathrm{C} 8-\mathrm{C} 9$ | $1.3949(19)$ |
| $\mathrm{C} 2-\mathrm{C} 3$ | $1.391(2)$ | $\mathrm{C} 8-\mathrm{C} 13$ | $1.4698(18)$ |
| $\mathrm{C} 3-\mathrm{C} 4$ | $1.384(2)$ | $\mathrm{C} 9-\mathrm{C} 10$ | $1.380(2)$ |
| C3-H3 | 0.9500 | $\mathrm{C} 9-\mathrm{H} 9$ | 0.9500 |
| $\mathrm{C} 4-\mathrm{C} 5$ | $1.388(2)$ | $\mathrm{C} 10-\mathrm{C} 11$ | $1.386(2)$ |


| C4-H4 | 0.9500 |
| :---: | :---: |
| C5-C6 | 1.385 (2) |
| C5-H5 | 0.9500 |
| C6-C7 | 1.390 (2) |
| C6-H6 | 0.9500 |
| C7-H7 | 0.9500 |
| $\mathrm{C} 1-\mathrm{O} 1-\mathrm{H} 1 \mathrm{o}$ | 109.1 (15) |
| $\mathrm{O} 2-\mathrm{C} 1-\mathrm{O} 1$ | 123.44 (13) |
| $\mathrm{O} 2-\mathrm{C} 1-\mathrm{C} 2$ | 123.29 (14) |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2$ | 113.26 (12) |
| $\mathrm{C} 7-\mathrm{C} 2-\mathrm{C} 3$ | 119.91 (13) |
| C7-C2-C1 | 121.78 (14) |
| C3-C2-C1 | 118.31 (13) |
| $\mathrm{C} 4-\mathrm{C} 3-\mathrm{C} 2$ | 119.96 (14) |
| $\mathrm{C} 4-\mathrm{C} 3-\mathrm{H} 3$ | 120.0 |
| C2-C3-H3 | 120.0 |
| C3-C4-C5 | 120.05 (14) |
| C3-C4-H4 | 120.0 |
| C5-C4-H4 | 120.0 |
| C6-C5-C4 | 120.19 (14) |
| C6-C5-H5 | 119.9 |
| C4-C5-H5 | 119.9 |
| C5-C6-C7 | 119.84 (14) |
| C5-C6-H6 | 120.1 |
| C7-C6- H 6 | 120.1 |
| C2-C7-C6 | 120.06 (14) |
| C2-C7-H7 | 120.0 |
| $\mathrm{O} 2-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 7$ | -173.81 (14) |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 7$ | 7.72 (19) |
| $\mathrm{O} 2-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | 6.0 (2) |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | -172.47 (12) |
| C7-C2-C3-C4 | -0.7 (2) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{C} 4$ | 179.52 (13) |
| C2-C3-C4-C5 | 0.6 (2) |
| C3-C4-C5-C6 | -0.2 (2) |
| C4-C5-C6-C7 | -0.2 (2) |
| C3-C2-C7-C6 | 0.3 (2) |
| C1-C2-C7-C6 | -179.93 (13) |
| C5-C6-C7-C2 | 0.2 (2) |


| $\mathrm{C} 10-\mathrm{H} 10$ | 0.9500 |
| :--- | :--- |
| $\mathrm{C} 11-\mathrm{C} 12$ | $1.385(2)$ |
| $\mathrm{C} 11-\mathrm{H} 11$ | 0.9500 |
| $\mathrm{C} 12-\mathrm{H} 12$ | 0.9500 |
| $\mathrm{C} 13-\mathrm{H} 13$ | 0.9500 |

C6-C7-H7 120.0
$\mathrm{C} 12-\mathrm{N} 1-\mathrm{C} 8 \quad 117.83$ (12)
$\mathrm{C} 13-\mathrm{N} 2 — \mathrm{~N} 2^{\mathrm{i}} \quad 111.87$ (13)
N1-C8-C9 122.45 (12)
N1—C8-C13 115.34 (12)
122.20 (12)
118.85 (13)
120.6
120.6
119.04 (12)
120.5
120.5
118.57 (12)
120.7
120.7
123.25 (13)
118.4
118.4
120.75 (12)
119.6
119.6
-0.65 (19)
-179.57 (11)
0.1 (2)
178.93 (12)
0.8 (2)
-1.0 (2)
0.4 (2)
0.5 (2)
-179.13 (12)
178.03 (12)
-0.9 (2)

Symmetry code: (i) $-x+2,-y+1,-z+1$.

Hydrogen-bond geometry (A, ${ }^{\circ}$ )

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O}-\mathrm{H} 10 \cdots \mathrm{~N} 1$ | $0.85(2)$ | $1.88(2)$ | $2.7269(16)$ | $177(2)$ |

## supporting information

| $\mathrm{C} 11 — \mathrm{H} 11 \cdots \mathrm{O}^{2 \mathrm{ii}}$ | 0.95 | 2.54 | $3.1811(19)$ | 125 |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 12 — \mathrm{H} 12 \cdots 2^{\mathrm{iii}}$ | 0.95 | 2.59 | $3.4647(19)$ | 154 |

Symmetry codes: (ii) $-x+1,-y+2,-z+1$; (iii) $x-1, y, z$.

