

## Dinitrosylbis[tris(4-chlorophenyl)-phosphane]iron

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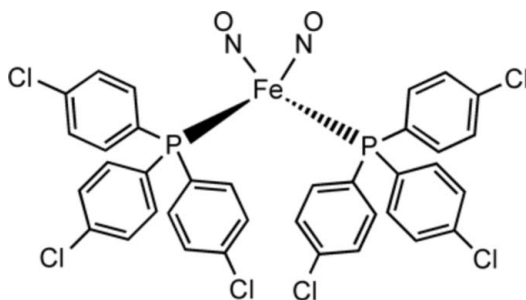
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Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.028;  $wR$  factor = 0.072; data-to-parameter ratio = 16.0.

The title dinitrosyl iron diphosphane complex,  $[\text{Fe}(\text{NO})_2(\text{C}_{18}\text{H}_{12}\text{Cl}_3\text{P})_2]$  or  $\text{Fe}(\text{NO})_2\text{L}_2$  [ $\text{L} = \text{P}(\text{C}_6\text{H}_4\text{-}p\text{-Cl})_3$ ] belongs to the family of metal dinitrosyl compounds with the general formula  $\text{Fe}(\text{NO})_2(\text{L})_x$ , referred to collectively as dinitrosyl iron compounds (DNICs). The iron atom is tetrahedrally coordinated by two phosphane ligands and two NO groups with  $\text{Fe}-\text{N}-\text{O}$  bond angles of  $178.76$  (15) and  $177.67$  (14)°.

### Related literature

For the preparation of the starting compound,  $\text{Fe}(\text{NO})_2(\text{CO})_2$ , see: Eisch & King (1965). For the structures of some related dinitrosyl complexes, see: Li *et al.* (2003); Atkinson *et al.* (1996); Li Kam Wah *et al.* (1989); Albano *et al.* (1974). For general information on metal nitrosyl chemistry, see: Richter-Addo & Legzdins (1992).



### Experimental

#### Crystal data

$[\text{Fe}(\text{NO})_2(\text{C}_{18}\text{H}_{12}\text{Cl}_3\text{P})_2]$	$V = 3638.9$ (18) Å <sup>3</sup>
$M_r = 847.06$	$Z = 4$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 10.340$ (3) Å	$\mu = 0.98$ mm <sup>-1</sup>
$b = 35.025$ (10) Å	$T = 100$ K
$c = 10.589$ (3) Å	$0.38 \times 0.19 \times 0.04$ mm
$\beta = 108.399$ (8)°	

#### Data collection

Bruker APEX CCD diffractometer	24943 measured reflections
Absorption correction: multi-scan ( <i>SADABS</i> ; Sheldrick, 2001)	7073 independent reflections
$T_{\min} = 0.702$ , $T_{\max} = 0.966$	6527 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.023$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.028$	442 parameters
$wR(F^2) = 0.072$	H-atom parameters constrained
$S = 1.00$	$\Delta\rho_{\max} = 0.39$ e Å <sup>-3</sup>
7073 reflections	$\Delta\rho_{\min} = -0.21$ e Å <sup>-3</sup>

Data collection: *SMART* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

We are grateful to the US Department of Education (GAANN Fellowship to MWJ; P200A030196), and the National Science Foundation (CHE-0076640 & CHE-0911537) for funding this work. The authors also thank the National Science Foundation (CHE-0130835) and the University of Oklahoma for funds to acquire the diffractometer and computers used in this work.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2779).

### References

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## supporting information

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**Dinitrosylbis[tris(4-chlorophenyl)phosphane]iron**

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**S1. Comment**

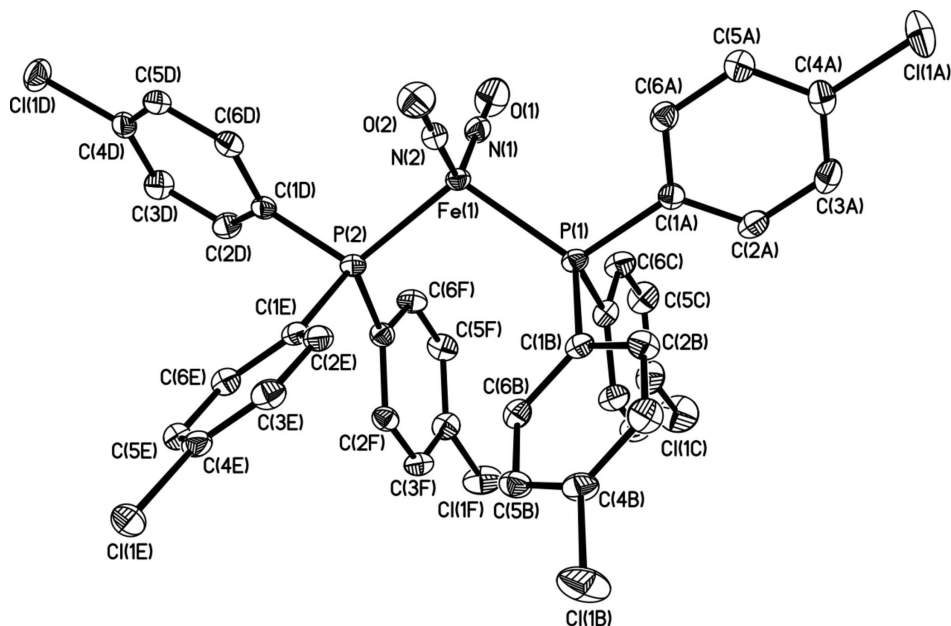
The molecular structure of the title compound is shown in Fig. 1. The molecule possesses a distorted tetrahedral geometry around the iron center. The iron is bound to two nitrosyl groups *via* the nitrogen atoms and to two phosphane ligands *via* the phosphorous atoms. The Fe(NO)<sub>2</sub> group exhibits an *attracto* conformation where the bond angles O...Fe...O < N—Fe—N (Richter-Addo & Legzdins, 1992). The N—Fe—N bond angle is 127.78 (7)° and the interphosphane angle, P—Fe—P, is 106.80 (3)°. The Fe—N—O bond angles are 178.76 (15)° and 177.67 (14)°. For the structures of some related complexes, see: Li *et al.*, 2003, Atkinson *et al.*, 1996, Li Kam Wah *et al.*, 1989, and Albano *et al.* 1974.

**S2. Experimental**

A colorless toluene solution (4 ml) of P(C<sub>6</sub>H<sub>4</sub>-*p*-Cl)<sub>3</sub> (126 mg, 0.35 mmol) was charged with Fe(NO)<sub>2</sub>(CO)<sub>2</sub> (20 μL, 0.18 mmol) (Eisch & King, 1965). The light red solution was heated and stirred under nitrogen. After 20 min the color of the solution had changed to black/brown. The reaction was allowed to proceed for 3.5 h until the infrared spectrum indicated the absence of characteristic carbonyl stretching frequencies for Fe(NO)<sub>2</sub>(CO)<sub>2</sub> ( $\nu_{\text{CO}} = 2090 \text{ cm}^{-1}$  and  $2040 \text{ cm}^{-1}$ ). The reaction mixture was filtered through celite under N<sub>2</sub> and the solvent was subsequently removed under vacuum. Isolated yield of the Fe(NO)<sub>2</sub>L<sub>2</sub> compound: 33%. IR (toluene, cm<sup>-1</sup>):  $\nu_{\text{NO}} = 1722 \text{ s}$  and  $1682 \text{ s}$ . <sup>31</sup>P {<sup>1</sup>H} NMR (CDCl<sub>3</sub>):  $\delta$  60.9 (s) referenced to 85% H<sub>3</sub>PO<sub>4</sub>. Suitable crystals for X-ray diffraction studies were grown by slow evaporation of a chloroform solution of the complex under nitrogen at ambient temperature.

**S3. Refinement**

H atoms were placed using known geometry with C—H (phenyl = 0.95 Å, methyl = 0.98 Å). Displacement parameters of phenyl H atoms were set to 1.2 times the isotropic equivalent for the bonded C.

**Figure 1**

The molecular structure of the title compound. Hydrogen atoms have been omitted for clarity. The displacement ellipsoids are drawn at the 50% probability level.

### Dinitrosylbis[tris(4-chlorophenyl)phosphane]iron

#### Crystal data

[Fe(NO)<sub>2</sub>(C<sub>18</sub>H<sub>12</sub>Cl<sub>3</sub>P)<sub>2</sub>]

*M<sub>r</sub>* = 847.06

Monoclinic, *P*2<sub>1</sub>/*n*

Hall symbol: -*P* 2<sub>1</sub> *n*

*a* = 10.340 (3) Å

*b* = 35.025 (10) Å

*c* = 10.589 (3) Å

$\beta$  = 108.399 (8)°

*V* = 3638.9 (18) Å<sup>3</sup>

*Z* = 4

*F*(000) = 1712

*D<sub>x</sub>* = 1.546 Mg m<sup>-3</sup>

Mo *K*α radiation,  $\lambda$  = 0.71073 Å

Cell parameters from 8898 reflections

$\theta$  = 3.0–26.8°

$\mu$  = 0.98 mm<sup>-1</sup>

*T* = 100 K

Plate, red

0.38 × 0.19 × 0.04 mm

#### Data collection

Bruker APEX CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega$  scans

Absorption correction: multi-scan

(*SADABS*; Sheldrick, 2001)

*T<sub>min</sub>* = 0.702, *T<sub>max</sub>* = 0.966

24943 measured reflections

7073 independent reflections

6527 reflections with *I* > 2σ(*I*)

*R<sub>int</sub>* = 0.023

$\theta_{\max}$  = 26.0°,  $\theta_{\min}$  = 3.0°

*h* = -11→12

*k* = -43→43

*l* = -13→13

#### Refinement

Refinement on *F*<sup>2</sup>

Least-squares matrix: full

*R* [*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.028

*wR* (*F*<sup>2</sup>) = 0.072

*S* = 1.00

7073 reflections

442 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods  
Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.038P)^2 + 2.2P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.39 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.21 \text{ e } \text{\AA}^{-3}$

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Fe1	0.39351 (2)	0.360112 (6)	0.72317 (2)	0.01475 (7)
N1	0.22480 (15)	0.35677 (4)	0.67163 (15)	0.0211 (3)
O1	0.10393 (13)	0.35496 (4)	0.63323 (16)	0.0374 (4)
N2	0.49768 (15)	0.36032 (4)	0.87866 (14)	0.0183 (3)
O2	0.56832 (14)	0.36016 (4)	0.99132 (12)	0.0286 (3)
P1	0.42791 (4)	0.414501 (12)	0.62695 (4)	0.01567 (10)
C1A	0.37536 (16)	0.45624 (5)	0.70326 (17)	0.0168 (3)
C2A	0.33928 (18)	0.49081 (5)	0.63462 (18)	0.0225 (4)
H2A	0.3363	0.4925	0.5442	0.027*
C3A	0.30787 (18)	0.52255 (5)	0.69749 (19)	0.0244 (4)
H3A	0.2825	0.5459	0.6505	0.029*
C4A	0.31395 (17)	0.51983 (5)	0.82958 (18)	0.0215 (4)
C5A	0.34768 (18)	0.48586 (5)	0.89891 (18)	0.0218 (4)
H5A	0.3503	0.4843	0.9892	0.026*
C6A	0.37756 (17)	0.45423 (5)	0.83540 (17)	0.0197 (3)
H6A	0.3999	0.4308	0.8824	0.024*
C11A	0.27894 (5)	0.559906 (13)	0.91015 (5)	0.03403 (12)
C1B	0.60313 (16)	0.42712 (5)	0.63916 (16)	0.0170 (3)
C2B	0.66819 (18)	0.45927 (5)	0.70917 (17)	0.0215 (4)
H2B	0.6199	0.4757	0.7498	0.026*
C3B	0.80257 (18)	0.46732 (5)	0.71992 (18)	0.0242 (4)
H3B	0.8465	0.4891	0.7683	0.029*
C4B	0.87206 (17)	0.44347 (5)	0.65970 (18)	0.0217 (4)
C5B	0.81105 (17)	0.41094 (5)	0.59145 (17)	0.0209 (4)
H5B	0.8602	0.3945	0.5518	0.025*
C6B	0.67730 (17)	0.40302 (5)	0.58234 (17)	0.0193 (3)
H6B	0.6350	0.3807	0.5365	0.023*
C11B	1.03788 (5)	0.454627 (16)	0.66918 (6)	0.03857 (13)
C1C	0.33064 (17)	0.42091 (5)	0.45080 (16)	0.0185 (3)
C2C	0.38994 (19)	0.42696 (5)	0.35119 (18)	0.0230 (4)
H2C	0.4864	0.4279	0.3735	0.028*
C3C	0.3091 (2)	0.43157 (5)	0.21917 (18)	0.0265 (4)
H3C	0.3502	0.4351	0.1514	0.032*
C4C	0.1692 (2)	0.43096 (5)	0.18746 (18)	0.0269 (4)
C5C	0.10724 (19)	0.42500 (6)	0.28448 (18)	0.0270 (4)
H5C	0.0107	0.4247	0.2618	0.032*
C6C	0.18857 (18)	0.41947 (5)	0.41529 (18)	0.0237 (4)
H6C	0.1469	0.4146	0.4818	0.028*

Cl1C	0.06876 (6)	0.438658 (17)	0.02359 (5)	0.04204 (14)
P2	0.46109 (4)	0.312563 (12)	0.61848 (4)	0.01492 (9)
C1D	0.40364 (16)	0.26589 (5)	0.65914 (16)	0.0165 (3)
C2D	0.33939 (17)	0.23830 (5)	0.56611 (17)	0.0210 (4)
H2D	0.3153	0.2440	0.4738	0.025*
C3D	0.31013 (18)	0.20251 (5)	0.60686 (18)	0.0228 (4)
H3D	0.2654	0.1838	0.5431	0.027*
C4D	0.34701 (17)	0.19451 (5)	0.74143 (18)	0.0202 (4)
C5D	0.41099 (18)	0.22145 (5)	0.83637 (17)	0.0206 (4)
H5D	0.4361	0.2155	0.9285	0.025*
C6D	0.43760 (17)	0.25712 (5)	0.79447 (17)	0.0198 (3)
H6D	0.4797	0.2760	0.8587	0.024*
Cl1D	0.31377 (5)	0.149324 (12)	0.79346 (5)	0.02760 (11)
C1E	0.64188 (16)	0.30117 (5)	0.65467 (16)	0.0171 (3)
C2E	0.74298 (17)	0.32280 (5)	0.74395 (16)	0.0191 (3)
H2E	0.7184	0.3444	0.7857	0.023*
C3E	0.88021 (18)	0.31309 (5)	0.77269 (17)	0.0218 (4)
H3E	0.9491	0.3278	0.8341	0.026*
C4E	0.91421 (17)	0.28181 (5)	0.71054 (17)	0.0213 (4)
C5E	0.81523 (18)	0.25973 (5)	0.62067 (17)	0.0216 (4)
H5E	0.8405	0.2384	0.5782	0.026*
C6E	0.68019 (18)	0.26927 (5)	0.59438 (17)	0.0204 (4)
H6E	0.6118	0.2540	0.5345	0.025*
Cl1E	1.08460 (4)	0.269062 (13)	0.74297 (5)	0.02791 (11)
C1F	0.39668 (17)	0.31651 (5)	0.43779 (16)	0.0170 (3)
C2F	0.48224 (17)	0.32239 (5)	0.36092 (16)	0.0182 (3)
H2F	0.5780	0.3201	0.4010	0.022*
C3F	0.42994 (18)	0.33155 (5)	0.22701 (17)	0.0209 (4)
H3F	0.4891	0.3360	0.1757	0.025*
C4F	0.29012 (18)	0.33409 (5)	0.16937 (17)	0.0214 (4)
C5F	0.20195 (18)	0.32781 (5)	0.24218 (17)	0.0230 (4)
H5F	0.1061	0.3292	0.2007	0.028*
C6F	0.25575 (17)	0.31942 (5)	0.37631 (17)	0.0209 (4)
H6F	0.1961	0.3156	0.4275	0.025*
Cl1F	0.22499 (5)	0.344935 (16)	0.00079 (4)	0.03281 (12)

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
Fe1	0.01424 (12)	0.01603 (12)	0.01523 (13)	-0.00054 (9)	0.00643 (9)	0.00098 (9)
N1	0.0199 (8)	0.0189 (7)	0.0271 (8)	0.0002 (6)	0.0109 (6)	0.0042 (6)
O1	0.0146 (7)	0.0378 (8)	0.0593 (10)	0.0004 (6)	0.0111 (6)	0.0135 (7)
N2	0.0207 (7)	0.0162 (7)	0.0194 (8)	0.0014 (5)	0.0085 (6)	0.0007 (5)
O2	0.0294 (7)	0.0345 (7)	0.0183 (7)	0.0021 (6)	0.0026 (6)	0.0010 (5)
P1	0.0134 (2)	0.0180 (2)	0.0163 (2)	-0.00109 (16)	0.00563 (16)	0.00173 (16)
C1A	0.0116 (8)	0.0172 (8)	0.0221 (8)	-0.0010 (6)	0.0060 (6)	0.0008 (6)
C2A	0.0216 (9)	0.0238 (9)	0.0234 (9)	-0.0014 (7)	0.0089 (7)	0.0058 (7)
C3A	0.0213 (9)	0.0186 (8)	0.0322 (10)	0.0007 (7)	0.0067 (7)	0.0068 (7)

C4A	0.0157 (8)	0.0182 (8)	0.0288 (9)	0.0000 (6)	0.0043 (7)	-0.0033 (7)
C5A	0.0213 (9)	0.0218 (9)	0.0209 (9)	-0.0003 (7)	0.0048 (7)	0.0003 (7)
C6A	0.0176 (8)	0.0194 (8)	0.0211 (9)	0.0013 (7)	0.0049 (7)	0.0036 (7)
C11A	0.0377 (3)	0.0223 (2)	0.0359 (3)	0.00912 (19)	0.0028 (2)	-0.00645 (19)
C1B	0.0149 (8)	0.0199 (8)	0.0163 (8)	-0.0008 (6)	0.0051 (6)	0.0038 (6)
C2B	0.0198 (9)	0.0237 (9)	0.0222 (9)	0.0001 (7)	0.0083 (7)	-0.0018 (7)
C3B	0.0199 (9)	0.0250 (9)	0.0260 (9)	-0.0058 (7)	0.0050 (7)	-0.0047 (7)
C4B	0.0126 (8)	0.0293 (9)	0.0234 (9)	-0.0024 (7)	0.0060 (7)	0.0018 (7)
C5B	0.0167 (8)	0.0264 (9)	0.0211 (9)	0.0023 (7)	0.0079 (7)	0.0020 (7)
C6B	0.0190 (8)	0.0182 (8)	0.0199 (8)	-0.0020 (7)	0.0052 (7)	0.0014 (6)
C11B	0.0177 (2)	0.0528 (3)	0.0489 (3)	-0.0128 (2)	0.0158 (2)	-0.0172 (2)
C1C	0.0192 (8)	0.0175 (8)	0.0180 (8)	-0.0016 (7)	0.0049 (7)	0.0016 (6)
C2C	0.0208 (9)	0.0272 (9)	0.0218 (9)	-0.0006 (7)	0.0080 (7)	0.0044 (7)
C3C	0.0299 (10)	0.0311 (10)	0.0198 (9)	0.0011 (8)	0.0099 (8)	0.0050 (7)
C4C	0.0306 (10)	0.0266 (9)	0.0184 (9)	0.0028 (8)	0.0005 (7)	0.0034 (7)
C5C	0.0197 (9)	0.0309 (10)	0.0265 (10)	-0.0001 (7)	0.0016 (7)	0.0036 (8)
C6C	0.0204 (9)	0.0284 (9)	0.0226 (9)	-0.0017 (7)	0.0073 (7)	0.0037 (7)
C11C	0.0398 (3)	0.0589 (3)	0.0200 (2)	0.0084 (2)	-0.0012 (2)	0.0080 (2)
P2	0.0134 (2)	0.0174 (2)	0.0147 (2)	-0.00121 (16)	0.00533 (16)	-0.00021 (15)
C1D	0.0142 (8)	0.0171 (8)	0.0202 (8)	0.0000 (6)	0.0082 (6)	0.0003 (6)
C2D	0.0207 (9)	0.0238 (9)	0.0181 (8)	-0.0016 (7)	0.0058 (7)	0.0003 (7)
C3D	0.0218 (9)	0.0221 (9)	0.0237 (9)	-0.0054 (7)	0.0061 (7)	-0.0036 (7)
C4D	0.0172 (8)	0.0166 (8)	0.0298 (9)	0.0009 (7)	0.0120 (7)	0.0022 (7)
C5D	0.0222 (9)	0.0231 (9)	0.0194 (9)	0.0031 (7)	0.0106 (7)	0.0017 (7)
C6D	0.0209 (9)	0.0208 (8)	0.0191 (8)	0.0006 (7)	0.0081 (7)	-0.0029 (7)
C11D	0.0312 (2)	0.0191 (2)	0.0334 (2)	-0.00247 (17)	0.01136 (19)	0.00508 (17)
C1E	0.0155 (8)	0.0210 (8)	0.0158 (8)	-0.0002 (6)	0.0064 (6)	0.0038 (6)
C2E	0.0185 (8)	0.0215 (8)	0.0188 (8)	0.0000 (7)	0.0083 (7)	0.0017 (7)
C3E	0.0174 (8)	0.0266 (9)	0.0204 (9)	-0.0032 (7)	0.0047 (7)	0.0016 (7)
C4E	0.0156 (8)	0.0268 (9)	0.0230 (9)	0.0038 (7)	0.0084 (7)	0.0094 (7)
C5E	0.0236 (9)	0.0192 (8)	0.0255 (9)	0.0019 (7)	0.0128 (7)	0.0042 (7)
C6E	0.0206 (9)	0.0210 (8)	0.0203 (8)	-0.0016 (7)	0.0074 (7)	0.0014 (7)
C11E	0.0164 (2)	0.0324 (2)	0.0366 (3)	0.00515 (17)	0.01070 (18)	0.00734 (19)
C1F	0.0184 (8)	0.0169 (8)	0.0161 (8)	-0.0005 (6)	0.0059 (6)	-0.0014 (6)
C2F	0.0155 (8)	0.0196 (8)	0.0199 (8)	-0.0013 (6)	0.0062 (7)	-0.0025 (6)
C3F	0.0210 (9)	0.0242 (9)	0.0195 (9)	-0.0040 (7)	0.0094 (7)	-0.0011 (7)
C4F	0.0242 (9)	0.0238 (9)	0.0141 (8)	-0.0017 (7)	0.0031 (7)	-0.0013 (7)
C5F	0.0166 (8)	0.0295 (9)	0.0212 (9)	-0.0012 (7)	0.0036 (7)	-0.0031 (7)
C6F	0.0167 (8)	0.0268 (9)	0.0206 (9)	-0.0024 (7)	0.0077 (7)	-0.0019 (7)
C11F	0.0269 (2)	0.0531 (3)	0.0155 (2)	-0.0027 (2)	0.00256 (17)	0.00332 (19)

*Geometric parameters (Å, °)*

Fe1—N2	1.6589 (15)	C5C—C6C	1.389 (3)
Fe1—N1	1.6594 (16)	C5C—H5C	0.9500
Fe1—P2	2.2316 (6)	C6C—H6C	0.9500
Fe1—P1	2.2410 (7)	P2—C1F	1.8218 (17)
N1—O1	1.188 (2)	P2—C1E	1.8298 (17)

N2—O2	1.1860 (19)	P2—C1D	1.8362 (17)
P1—C1B	1.8294 (18)	C1D—C2D	1.390 (2)
P1—C1C	1.8322 (18)	C1D—C6D	1.398 (2)
P1—C1A	1.8338 (17)	C2D—C3D	1.390 (2)
C1A—C6A	1.394 (2)	C2D—H2D	0.9500
C1A—C2A	1.401 (2)	C3D—C4D	1.383 (3)
C2A—C3A	1.386 (3)	C3D—H3D	0.9500
C2A—H2A	0.9500	C4D—C5D	1.385 (2)
C3A—C4A	1.383 (3)	C4D—C11D	1.7451 (17)
C3A—H3A	0.9500	C5D—C6D	1.382 (2)
C4A—C5A	1.384 (2)	C5D—H5D	0.9500
C4A—C11A	1.7390 (18)	C6D—H6D	0.9500
C5A—C6A	1.381 (2)	C1E—C2E	1.392 (2)
C5A—H5A	0.9500	C1E—C6E	1.405 (2)
C6A—H6A	0.9500	C2E—C3E	1.396 (2)
C1B—C6B	1.398 (2)	C2E—H2E	0.9500
C1B—C2B	1.398 (2)	C3E—C4E	1.380 (3)
C2B—C3B	1.387 (2)	C3E—H3E	0.9500
C2B—H2B	0.9500	C4E—C5E	1.392 (3)
C3B—C4B	1.382 (3)	C4E—C11E	1.7434 (18)
C3B—H3B	0.9500	C5E—C6E	1.376 (2)
C4B—C5B	1.389 (3)	C5E—H5E	0.9500
C4B—C11B	1.7303 (18)	C6E—H6E	0.9500
C5B—C6B	1.384 (2)	C1F—C2F	1.394 (2)
C5B—H5B	0.9500	C1F—C6F	1.400 (2)
C6B—H6B	0.9500	C2F—C3F	1.386 (2)
C1C—C2C	1.393 (2)	C2F—H2F	0.9500
C1C—C6C	1.398 (2)	C3F—C4F	1.383 (2)
C2C—C3C	1.393 (3)	C3F—H3F	0.9500
C2C—H2C	0.9500	C4F—C5F	1.385 (3)
C3C—C4C	1.378 (3)	C4F—C11F	1.7394 (18)
C3C—H3C	0.9500	C5F—C6F	1.384 (2)
C4C—C5C	1.387 (3)	C5F—H5F	0.9500
C4C—C11C	1.7390 (18)	C6F—H6F	0.9500
N2—Fe1—N1	127.78 (7)	C6C—C5C—H5C	120.5
N2—Fe1—P2	106.82 (5)	C5C—C6C—C1C	121.17 (17)
N1—Fe1—P2	104.22 (5)	C5C—C6C—H6C	119.4
N2—Fe1—P1	107.52 (5)	C1C—C6C—H6C	119.4
N1—Fe1—P1	102.22 (5)	C1F—P2—C1E	104.31 (8)
P2—Fe1—P1	106.80 (3)	C1F—P2—C1D	106.01 (8)
O1—N1—Fe1	178.76 (15)	C1E—P2—C1D	98.14 (7)
O2—N2—Fe1	177.67 (14)	C1F—P2—Fe1	113.40 (6)
C1B—P1—C1C	104.80 (8)	C1E—P2—Fe1	121.20 (6)
C1B—P1—C1A	101.59 (8)	C1D—P2—Fe1	111.88 (6)
C1C—P1—C1A	101.95 (8)	C2D—C1D—C6D	118.87 (15)
C1B—P1—Fe1	117.97 (5)	C2D—C1D—P2	124.77 (13)
C1C—P1—Fe1	116.88 (6)	C6D—C1D—P2	116.16 (12)

C1A—P1—Fe1	111.46 (6)	C3D—C2D—C1D	120.61 (16)
C6A—C1A—C2A	118.81 (16)	C3D—C2D—H2D	119.7
C6A—C1A—P1	119.25 (12)	C1D—C2D—H2D	119.7
C2A—C1A—P1	121.86 (13)	C4D—C3D—C2D	119.12 (16)
C3A—C2A—C1A	120.56 (16)	C4D—C3D—H3D	120.4
C3A—C2A—H2A	119.7	C2D—C3D—H3D	120.4
C1A—C2A—H2A	119.7	C3D—C4D—C5D	121.53 (16)
C4A—C3A—C2A	119.19 (16)	C3D—C4D—C11D	119.44 (13)
C4A—C3A—H3A	120.4	C5D—C4D—C11D	119.03 (14)
C2A—C3A—H3A	120.4	C6D—C5D—C4D	118.73 (16)
C3A—C4A—C5A	121.27 (16)	C6D—C5D—H5D	120.6
C3A—C4A—C11A	119.36 (14)	C4D—C5D—H5D	120.6
C5A—C4A—C11A	119.36 (14)	C5D—C6D—C1D	121.11 (16)
C6A—C5A—C4A	119.29 (17)	C5D—C6D—H6D	119.4
C6A—C5A—H5A	120.4	C1D—C6D—H6D	119.4
C4A—C5A—H5A	120.4	C2E—C1E—C6E	118.86 (15)
C5A—C6A—C1A	120.86 (16)	C2E—C1E—P2	121.49 (13)
C5A—C6A—H6A	119.6	C6E—C1E—P2	119.62 (12)
C1A—C6A—H6A	119.6	C1E—C2E—C3E	120.62 (16)
C6B—C1B—C2B	118.49 (15)	C1E—C2E—H2E	119.7
C6B—C1B—P1	119.31 (13)	C3E—C2E—H2E	119.7
C2B—C1B—P1	122.14 (13)	C4E—C3E—C2E	118.94 (16)
C3B—C2B—C1B	120.61 (16)	C4E—C3E—H3E	120.5
C3B—C2B—H2B	119.7	C2E—C3E—H3E	120.5
C1B—C2B—H2B	119.7	C3E—C4E—C5E	121.64 (16)
C4B—C3B—C2B	119.51 (16)	C3E—C4E—C11E	120.21 (14)
C4B—C3B—H3B	120.2	C5E—C4E—C11E	118.15 (14)
C2B—C3B—H3B	120.2	C6E—C5E—C4E	118.92 (16)
C3B—C4B—C5B	121.25 (16)	C6E—C5E—H5E	120.5
C3B—C4B—C11B	119.12 (14)	C4E—C5E—H5E	120.5
C5B—C4B—C11B	119.63 (14)	C5E—C6E—C1E	121.00 (16)
C6B—C5B—C4B	118.71 (16)	C5E—C6E—H6E	119.5
C6B—C5B—H5B	120.6	C1E—C6E—H6E	119.5
C4B—C5B—H5B	120.6	C2F—C1F—C6F	118.51 (15)
C5B—C6B—C1B	121.40 (16)	C2F—C1F—P2	122.38 (13)
C5B—C6B—H6B	119.3	C6F—C1F—P2	118.41 (13)
C1B—C6B—H6B	119.3	C3F—C2F—C1F	121.13 (16)
C2C—C1C—C6C	118.56 (16)	C3F—C2F—H2F	119.4
C2C—C1C—P1	123.91 (13)	C1F—C2F—H2F	119.4
C6C—C1C—P1	117.52 (13)	C4F—C3F—C2F	118.83 (16)
C3C—C2C—C1C	120.60 (17)	C4F—C3F—H3F	120.6
C3C—C2C—H2C	119.7	C2F—C3F—H3F	120.6
C1C—C2C—H2C	119.7	C3F—C4F—C5F	121.61 (16)
C4C—C3C—C2C	119.60 (17)	C3F—C4F—C11F	118.62 (14)
C4C—C3C—H3C	120.2	C5F—C4F—C11F	119.76 (14)
C2C—C3C—H3C	120.2	C6F—C5F—C4F	118.90 (16)
C3C—C4C—C5C	121.08 (17)	C6F—C5F—H5F	120.6
C3C—C4C—C11C	119.40 (15)	C4F—C5F—H5F	120.6



C5C—C4C—C11C	119.51 (15)	C5F—C6F—C1F	121.01 (16)
C4C—C5C—C6C	118.95 (17)	C5F—C6F—H6F	119.5
C4C—C5C—H5C	120.5	C1F—C6F—H6F	119.5
N2—Fe1—P1—C1B	50.95 (8)	N2—Fe1—P2—C1F	-161.85 (8)
N1—Fe1—P1—C1B	-172.53 (8)	N1—Fe1—P2—C1F	60.70 (8)
P2—Fe1—P1—C1B	-63.40 (7)	P1—Fe1—P2—C1F	-47.03 (6)
N2—Fe1—P1—C1C	177.35 (8)	N2—Fe1—P2—C1E	-36.62 (8)
N1—Fe1—P1—C1C	-46.13 (8)	N1—Fe1—P2—C1E	-174.07 (8)
P2—Fe1—P1—C1C	63.01 (7)	P1—Fe1—P2—C1E	78.20 (6)
N2—Fe1—P1—C1A	-65.99 (8)	N2—Fe1—P2—C1D	78.31 (8)
N1—Fe1—P1—C1A	70.53 (8)	N1—Fe1—P2—C1D	-59.14 (8)
P2—Fe1—P1—C1A	179.66 (6)	P1—Fe1—P2—C1D	-166.87 (6)
C1B—P1—C1A—C6A	-99.51 (14)	C1F—P2—C1D—C2D	7.28 (17)
C1C—P1—C1A—C6A	152.44 (13)	C1E—P2—C1D—C2D	-100.22 (15)
Fe1—P1—C1A—C6A	27.00 (15)	Fe1—P2—C1D—C2D	131.37 (14)
C1B—P1—C1A—C2A	77.33 (15)	C1F—P2—C1D—C6D	-177.80 (12)
C1C—P1—C1A—C2A	-30.73 (16)	C1E—P2—C1D—C6D	74.70 (14)
Fe1—P1—C1A—C2A	-156.16 (12)	Fe1—P2—C1D—C6D	-53.71 (14)
C6A—C1A—C2A—C3A	0.7 (3)	C6D—C1D—C2D—C3D	-0.4 (3)
P1—C1A—C2A—C3A	-176.11 (13)	P2—C1D—C2D—C3D	174.39 (13)
C1A—C2A—C3A—C4A	0.7 (3)	C1D—C2D—C3D—C4D	-0.6 (3)
C2A—C3A—C4A—C5A	-1.5 (3)	C2D—C3D—C4D—C5D	0.7 (3)
C2A—C3A—C4A—C11A	178.34 (13)	C2D—C3D—C4D—C11D	-178.72 (13)
C3A—C4A—C5A—C6A	0.9 (3)	C3D—C4D—C5D—C6D	0.3 (3)
C11A—C4A—C5A—C6A	-178.98 (13)	C11D—C4D—C5D—C6D	179.70 (13)
C4A—C5A—C6A—C1A	0.6 (3)	C4D—C5D—C6D—C1D	-1.3 (3)
C2A—C1A—C6A—C5A	-1.4 (2)	C2D—C1D—C6D—C5D	1.4 (2)
P1—C1A—C6A—C5A	175.54 (13)	P2—C1D—C6D—C5D	-173.83 (13)
C1C—P1—C1B—C6B	-69.90 (15)	C1F—P2—C1E—C2E	126.39 (14)
C1A—P1—C1B—C6B	-175.73 (13)	C1D—P2—C1E—C2E	-124.70 (14)
Fe1—P1—C1B—C6B	62.15 (14)	Fe1—P2—C1E—C2E	-2.92 (16)
C1C—P1—C1B—C2B	113.22 (15)	C1F—P2—C1E—C6E	-55.50 (15)
C1A—P1—C1B—C2B	7.39 (16)	C1D—P2—C1E—C6E	53.40 (14)
Fe1—P1—C1B—C2B	-114.73 (13)	Fe1—P2—C1E—C6E	175.18 (11)
C6B—C1B—C2B—C3B	1.1 (3)	C6E—C1E—C2E—C3E	0.3 (2)
P1—C1B—C2B—C3B	177.96 (14)	P2—C1E—C2E—C3E	178.42 (13)
C1B—C2B—C3B—C4B	0.6 (3)	C1E—C2E—C3E—C4E	0.4 (2)
C2B—C3B—C4B—C5B	-1.7 (3)	C2E—C3E—C4E—C5E	-0.3 (3)
C2B—C3B—C4B—C11B	177.93 (14)	C2E—C3E—C4E—C11E	179.48 (13)
C3B—C4B—C5B—C6B	1.2 (3)	C3E—C4E—C5E—C6E	-0.6 (3)
C11B—C4B—C5B—C6B	-178.48 (13)	C11E—C4E—C5E—C6E	179.69 (13)
C4B—C5B—C6B—C1B	0.5 (3)	C4E—C5E—C6E—C1E	1.3 (2)
C2B—C1B—C6B—C5B	-1.6 (2)	C2E—C1E—C6E—C5E	-1.2 (2)
P1—C1B—C6B—C5B	-178.62 (13)	P2—C1E—C6E—C5E	-179.31 (13)
C1B—P1—C1C—C2C	10.70 (17)	C1E—P2—C1F—C2F	-19.21 (16)
C1A—P1—C1C—C2C	116.27 (15)	C1D—P2—C1F—C2F	-122.22 (14)
Fe1—P1—C1C—C2C	-121.97 (14)	Fe1—P2—C1F—C2F	114.65 (13)

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C1B—P1—C1C—C6C	-169.53 (13)	C1E—P2—C1F—C6F	170.49 (13)
C1A—P1—C1C—C6C	-63.97 (15)	C1D—P2—C1F—C6F	67.48 (15)
Fe1—P1—C1C—C6C	57.80 (15)	Fe1—P2—C1F—C6F	-55.65 (14)
C6C—C1C—C2C—C3C	0.2 (3)	C6F—C1F—C2F—C3F	1.1 (2)
P1—C1C—C2C—C3C	179.96 (14)	P2—C1F—C2F—C3F	-169.24 (13)
C1C—C2C—C3C—C4C	1.4 (3)	C1F—C2F—C3F—C4F	-1.2 (3)
C2C—C3C—C4C—C5C	-1.4 (3)	C2F—C3F—C4F—C5F	0.1 (3)
C2C—C3C—C4C—C11C	177.41 (14)	C2F—C3F—C4F—C11F	-179.12 (13)
C3C—C4C—C5C—C6C	-0.2 (3)	C3F—C4F—C5F—C6F	1.1 (3)
C11C—C4C—C5C—C6C	-179.01 (15)	C11F—C4F—C5F—C6F	-179.67 (14)
C4C—C5C—C6C—C1C	1.8 (3)	C4F—C5F—C6F—C1F	-1.2 (3)
C2C—C1C—C6C—C5C	-1.8 (3)	C2F—C1F—C6F—C5F	0.2 (3)
P1—C1C—C6C—C5C	178.39 (15)	P2—C1F—C6F—C5F	170.88 (14)

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