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Bis[(1-methyl-1H-tetrazol-5-yl)sulfanyl]methane

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Key indicators: single-crystal X-ray study; T = 296 K; mean $\sigma(N-C) = 0.007$ Å; R factor = 0.063; wR factor = 0.108; data-to-parameter ratio = 13.4.

The molecule of the title compound, $C_5H_8N_8S_2$, lies on a twofold rotation axis that relates on 1-methyltetrazolyl group to the other; the five-membered rings are twisted by $53.1 (1)^{\circ}$.

Related literature

For the synthesis and pharmacological activity of compounds containing tetrazole groups, see: Semenov (2002); Upadhayaya et al. (2004). For a related structure, see: Bronisz (2002).



Experimental

Crystal data

$C_5H_8N_8S_2$	V = 1041.9 (7) Å ³
$M_r = 244.31$	Z = 4
Orthorhombic, Pbcn	Mo $K\alpha$ radiation
a = 6.415 (3) Å	$\mu = 0.49 \text{ mm}^{-1}$
b = 7.314 (3) Å	T = 296 K
c = 22.204 (8) Å	$0.15 \times 0.12 \times 0.08 \text{ mm}$

Data collection

CBruker SMART area-detector 4692 measured reflections diffractometer 936 independent reflections Absorption correction: multi-scan 482 reflections with $I > 2\sigma(I)$ (SADABS; Bruker, 2002) $R_{\rm int} = 0.118$ $T_{\min} = 0.930, T_{\max} = 0.962$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.063$ 70 parameters $wR(F^2) = 0.108$ H-atom parameters constrained $\Delta \rho_{\rm max} = 0.34 \text{ e} \text{ Å}^-$ S = 1.21 $\Delta \rho_{\rm min} = -0.38 \text{ e } \text{\AA}^{-3}$ 936 reflections

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG5140).

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supporting information

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Bis[(1-methyl-1*H*-tetrazol-5-yl)sulfanyl]methane

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S1. Experimental

Sodium hydroxide (1.7 g, 0.043 mol) was added to 5-mercapto-1-methyltetrazole (5 g, 0.043 mol) in dry dimethylsulfoxide (35 ml). The reaction mixture was stirred at 363 K for 1 h. Dichloromethane (3.1 ml, 0.0215 mol) was then added to the solution dropwise with the formation of a grey suspension. The suspension was stirred for 4 h, cooled to room temperature and filtered. The solvent was removed completely under reduced pressure. The residue was recrystallized from ethanol to give a white crystalline product (2.94 g; m.p. 353 - 354 K). Single crystals of the title compound suitable for X-ray diffraction analysis were isolated after a week from a solution in acetone.

S2. Refinement

All H atoms were positioned geometrically (C—H = 0.96 Å for aromatic CH₃ and 0.97 Å for CH₂ groups, respectively) and constrained to ride on their parent atoms with U_{iso} (H) values set to be -1.5 of the carrier atom.



Figure 1

A view of the molecular structure of title compound.



Figure 2

The crystal packing of the title compound.

Bis[(1-methyl-1*H*-tetrazol-5-yl)sulfanyl]methane

Crystal data

 $C_{5}H_{8}N_{8}S_{2}$ $M_{r} = 244.31$ Orthorhombic, *Pbcn* Hall symbol: -P 2n 2ab a = 6.415 (3) Å b = 7.314 (3) Å c = 22.204 (8) Å V = 1041.9 (7) Å³ Z = 4F(000) = 504

Data collection

CBruker SMART area-detector	4692 measured reflections
diffractometer	936 independent reflections
Radiation source: fine-focus sealed tube	482 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.118$
φ and ω scans	$\theta_{\rm max} = 25.1^{\circ}, \ \theta_{\rm min} = 1.8^{\circ}$
Absorption correction: multi-scan	$h = -7 \rightarrow 7$
(SADABS; Bruker, 2002)	$k = -8 \longrightarrow 4$
$T_{\min} = 0.930, \ T_{\max} = 0.962$	$l = -25 \rightarrow 26$

 $D_{\rm x} = 1.558 \text{ Mg m}^{-3}$ $D_{\rm m} = 1.558 \text{ Mg m}^{-3}$ $D_{\rm m} \text{ measured by not measured}$ Mo K\$\alpha\$ radiation, \$\lambda\$ = 0.71073 Å Cell parameters from 214 reflections \$\theta\$ = 2.5-18.9° \$\mu\$ = 0.49 mm}^{-1}\$ T = 296 K Flake-like, colourless 0.15 \times 0.12 \times 0.08 mm Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.063$ $wR(F^2) = 0.108$	Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites
S = 1.21	H-atom parameters constrained
936 reflections	$w = 1/[\sigma^2(F_o^2) + (0.P)^2 + 0.7202P]$
70 parameters	where $P = (F_o^2 + 2F_c^2)/3$
0 restraints	$(\Delta/\sigma)_{\rm max} < 0.001$
Primary atom site location: structure-invariant direct methods	$\Delta ho_{ m max} = 0.34$ e Å ⁻³ $\Delta ho_{ m min} = -0.38$ e Å ⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
S1	1.1330 (2)	1.03366 (18)	0.19301 (6)	0.0521 (4)	
N3	0.6134 (7)	0.9428 (6)	0.1121 (2)	0.0590 (13)	
C2	0.9188 (8)	0.9782 (6)	0.1491 (2)	0.0391 (13)	
N4	0.7207 (7)	1.0077 (6)	0.16107 (17)	0.0498 (12)	
N1	0.9358 (7)	0.8966 (6)	0.09509 (17)	0.0463 (11)	
C1	1.1179 (8)	0.8355 (7)	0.0621 (2)	0.0617 (16)	
H1A	1.1692	0.9336	0.0375	0.093*	
H1B	1.0808	0.7338	0.0369	0.093*	
H1C	1.2242	0.7985	0.0899	0.093*	
N2	0.7401 (8)	0.8743 (6)	0.07305 (18)	0.0561 (13)	
C3	1.0000	1.1654 (9)	0.2500	0.050 (2)	
H3A	1.1011	1.2439	0.2697	0.074*	0.50
H3B	0.8989	1.2439	0.2303	0.074*	0.50

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0481 (9)	0.0665 (10)	0.0418 (8)	0.0008 (8)	-0.0041 (7)	-0.0064 (7)
N3	0.049 (3)	0.061 (3)	0.068 (3)	-0.006 (3)	-0.010 (3)	0.001 (3)
C2	0.050 (4)	0.034 (3)	0.033 (3)	-0.004 (3)	-0.002 (2)	0.004 (2)
N4	0.041 (3)	0.061 (3)	0.047 (3)	0.002 (2)	0.004 (2)	0.004 (2)
N1	0.051 (3)	0.053 (3)	0.035 (2)	-0.004 (2)	-0.003 (2)	-0.003 (2)
C1	0.062 (4)	0.075 (4)	0.049 (3)	0.001 (3)	0.004 (3)	-0.012 (3)
N2	0.051 (3)	0.068 (3)	0.050 (3)	-0.004 (3)	-0.008 (3)	0.000 (2)
C3	0.061 (6)	0.054 (5)	0.034 (4)	0.000	-0.012 (4)	0.000

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S1—C2	1.734 (5)	N1—C1	1.450 (6)	
S1—C3	1.805 (4)	C1—H1A	0.9600	
N3—N2	1.289 (5)	C1—H1B	0.9600	
N3—N4	1.372 (5)	C1—H1C	0.9600	
C2—N4	1.316 (6)	C3—S1 ⁱ	1.805 (4)	
C2—N1	1.343 (5)	С3—НЗА	0.9700	
N1—N2	1.357 (5)	С3—Н3В	0.9700	
C2—S1—C3	98.31 (19)	H1A—C1—H1B	109.5	
N2—N3—N4	110.6 (4)	N1—C1—H1C	109.5	
N4—C2—N1	109.4 (4)	H1A—C1—H1C	109.5	
N4—C2—S1	127.8 (4)	H1B—C1—H1C	109.5	
N1-C2-S1	122.8 (4)	N3—N2—N1	107.1 (4)	
C2—N4—N3	105.5 (4)	S1 ⁱ —C3—S1	115.5 (4)	
C2—N1—N2	107.5 (4)	S1 ⁱ —C3—H3A	108.4	
C2—N1—C1	130.8 (5)	S1—C3—H3A	108.4	
N2—N1—C1	121.7 (4)	S1 ⁱ —C3—H3B	108.4	
N1—C1—H1A	109.5	S1—C3—H3B	108.4	
N1—C1—H1B	109.5	НЗА—СЗ—НЗВ	107.5	

Geometric parameters (Å, °)

Symmetry code: (i) -x+2, y, -z+1/2.