

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

(2E)-1-(2-Hydroxy-5-methylphenyl)-3-(4-methoxyphenyl)prop-2-en-1-one

Hoong-Kun Fun,^{a,*} Suhana Arshad,^a B. K. Sarojini,^b
V. Musthafa Khaleel^b and B. Narayana^c

^aX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, ^bDepartment of Chemistry, P. A. College of Engineering, Mangalore 574 153, India, and ^cDepartment of Studies in Chemistry, Mangalore University, Mangalagangothri, Mangalore 574 199, India.

Correspondence e-mail: hkfun@usm.my

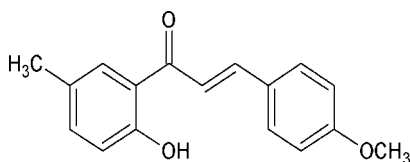
Received 15 April 2011; accepted 21 April 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}—\text{C}) = 0.002$ Å; R factor = 0.047; wR factor = 0.131; data-to-parameter ratio = 21.9.

In the title compound, $\text{C}_{17}\text{H}_{16}\text{O}_3$, the dihedral angle between the aromatic rings is $4.59(7)^\circ$ and an intramolecular $\text{O}—\text{H} \cdots \text{O}$ hydrogen bond generates an $S(6)$ ring. In the crystal, adjacent molecules are linked by $\text{C}—\text{H} \cdots \text{O}$ hydrogen bonds, leading to the formation of $[001]$ supramolecular chains. Weak $\text{C}—\text{H} \cdots \pi$ interactions consolidate the packing.

Related literature

For a related structure and background references to chalcones, see: Fun *et al.* (2010). For related structures, see: Chantrapromma *et al.* (2009, 2010); Fun *et al.* (2009); Horkaew *et al.* (2010); Lu *et al.* (2009); Suwunwong *et al.* (2009); Wang *et al.* (2009, 2010); Jasinski *et al.* (2011). For hydrogen-bond motifs, see: Bernstein *et al.* (1995). For bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{16}\text{O}_3$ $V = 1418.3(4) \text{ \AA}^3$
 $M_r = 268.30$ $Z = 4$
Monoclinic, $P2_1/c$ $\text{Mo K}\alpha$ radiation
 $a = 12.6990(18) \text{ \AA}$ $\mu = 0.09 \text{ mm}^{-1}$
 $b = 8.8022(13) \text{ \AA}$ $T = 296 \text{ K}$
 $c = 13.172(2) \text{ \AA}$ $0.46 \times 0.32 \times 0.18 \text{ mm}$
 $\beta = 105.565(2)^\circ$

Data collection

Bruker SMART APEXII DUO 11493 measured reflections
CCD diffractometer 4090 independent reflections
Absorption correction: multi-scan 2608 reflections with $I > 2\sigma(I)$
(SADABS; Bruker, 2009) $R_{\text{int}} = 0.024$
 $T_{\text{min}} = 0.962$, $T_{\text{max}} = 0.984$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$ H atoms treated by a mixture of
 $wR(F^2) = 0.131$ independent and constrained
 $S = 1.02$ refinement
4090 reflections $\Delta\rho_{\text{max}} = 0.16 \text{ e \AA}^{-3}$
187 parameters $\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry ($\text{\AA}$, $^\circ$).

Cg1 is the centroid of the C1–C6 ring.

$D—H \cdots A$	$D—H$	$H \cdots A$	$D \cdots A$	$D—H \cdots A$
$\text{O1}—\text{H1O1} \cdots \text{O2}$	0.95 (2)	1.65 (2)	2.5112 (18)	149.4 (19)
$\text{C11}—\text{H11A} \cdots \text{O3}^i$	0.93	2.60	3.4317 (17)	149
$\text{C16}—\text{H16C} \cdots \text{Cg1}^{ii}$	0.96	2.81	3.5800 (18)	138

Symmetry codes: (i) $x, -y + \frac{3}{2}, z + \frac{1}{2}$; (ii) $-x, -y, -z$.

Data collection: APEX2 (Bruker, 2009); cell refinement: SAINT (Bruker, 2009); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL and PLATON (Spek, 2009).

The authors thank Universiti Sains Malaysia (USM) for the Research University Grant (No. 1001/PFIZIK/811160). SA thanks the Malaysian Government and USM for the award of a research scholarship. VMK also thanks P. A. College of Engineering for research facilities.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5847).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bernstein, J., Davis, R. E., Shimon, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Bruker (2009). *SADABS, APEX2 and SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Chantrapromma, S., Horkaew, J., Suwunwong, T. & Fun, H.-K. (2009). *Acta Cryst.* **E65**, o2673–o2674.
- Chantrapromma, S., Suwunwong, T., Boonnak, N. & Fun, H.-K. (2010). *Acta Cryst.* **E66**, o312–o313.
- Fun, H.-K., Hemamalini, M., Samshuddin, S., Narayana, B. & Yathirajan, H. S. (2010). *Acta Cryst.* **E66**, o864–o865.
- Fun, H.-K., Suwunwong, T., Boonnak, N. & Chantrapromma, S. (2009). *Acta Cryst.* **E65**, o2168–o2169.
- Horkaew, J., Suwunwong, T., Chantrapromma, S., Karalai, C. & Fun, H.-K. (2010). *Acta Cryst.* **E66**, o800–o801.
- Jasinski, J. P., Butcher, R. J., Yathirajan, H. S., Sarojini, B. K. & Musthafa Khaleel, V. (2011). *Acta Cryst.* **E67**, o756.
- Lu, Y.-H., Wang, G.-Z., Zhou, C.-H. & Zhang, Y.-Y. (2009). *Acta Cryst.* **E65**, o1396.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

* Thomson Reuters ResearcherID: A-3561-2009.

Suwunwong, T., Chantrapromma, S., Karalai, C., Pakdeevanich, P. & Fun, H.-K. (2009). *Acta Cryst. E* **65**, o420–o421.
Wang, G.-Z., Fang, B. & Zhou, C.-H. (2009). *Acta Cryst. E* **65**, o2619.

Wang, X.-L., Wang, G.-Z., Geng, R.-X. & Zhou, C.-H. (2010). *Acta Cryst. E* **66**, o320.

supporting information

Acta Cryst. (2011). E67, o1248–o1249 [https://doi.org/10.1107/S1600536811015054]

(2*E*)-1-(2-Hydroxy-5-methylphenyl)-3-(4-methoxyphenyl)prop-2-en-1-one**Hoong-Kun Fun, Suhana Arshad, B. K. Sarojini, V. Musthafa Khaleel and B. Narayana****S1. Comment**

In continuation of our studies on the crystal structures of chalcones (Fun *et al.*, 2010), we now report the synthesis and crystal structure of the title compound, (I). The structures of some related chalcones *viz*: (Z)-3-(9-anthryl)-1-(4-methoxyphenyl)prop-2-en-1-one (Chantrapromma *et al.*, 2009), (Z)-3-(9-anthryl)-1-(2-thienyl)prop-2-en-1-one (Fun *et al.*, 2009), (E)-3-(anthracen-9-yl)-1-(4-bromophenyl)prop-2-en-1-one (Suwunwong *et al.*, 2009), (Z)-3-(9-anthryl)-1-(4-bromophenyl)-2-(4-nitro-1*H*-imidazol-1-yl)prop-2-en-1-one (Lu *et al.*, 2009), (Z)-3-(9-anthryl)-2-(4-nitro-1*H*-imidazol-1-yl)-1-*p*-tolylprop-2-en-1-one (Wang *et al.*, 2009), (E)-3-(9-anthryl)-1-(4-fluorophenyl)-2-(4-nitro-1*H*-imidazol-1-yl)prop-2-en-1-one (Wang *et al.*, 2010), (E)-3-(anthracen-9-yl)-1-(furan-2-yl)prop-2-en-1-one (Horkaew *et al.*, 2010), and an orthorhombic polymorph of (Z)-3-(9-anthryl)-1-(2-thienyl)prop-2-en-1-one (Chantrapromma *et al.*, 2010) and 2(*E*)-3-(4-hydroxyphenyl)-1-(4-chlorophenyl)prop-2-en-1-one (Jasinski *et al.*, 2011) have been reported.

The molecular structure is shown in Fig. 1. An intramolecular O1—H1O1···O2 hydrogen bond (Table 1) stabilizes the molecular structure and forms an *S*(6) ring motif (Bernstein *et al.*, 1995). The dihedral angle between the phenyl (C1–C6) ring and the methoxy-substituted phenyl (C10–C15) ring is 4.59 (7)°. Bond lengths (Allen *et al.*, 1987) and angles are within normal ranges and are comparable to the related structures (Fun *et al.*, 2010).

In the crystal packing (Fig. 2), the molecules are linked into infinite one-dimensional chain along the *c*-axis by intermolecular C11—H11A···O3 hydrogen bonds (Table 1). There are also C—H··· π interactions (Table 1) which involves C16 and phenyl ring (*C*_g1 = C1–C6).

S2. Experimental

2-Hydroxy-5-methylacetophenone (1.50 g, 0.01 mol) was mixed with 4-methoxybenzaldehyde (1.36 g, 0.01 mol) and dissolved in ethanol (40 ml). To this solution, 5 ml of KOH (50%) was added at 278 K. The reaction mixture stirred for 6 h and poured on to crushed ice. The pH of this mixture was adjusted to 3–4 with 2 *M* HCl aqueous solution. The resulting crude yellow solid was filtered, washed successively with dilute HCl solution and distilled water and finally recrystallized from ethanol (95%) to give the pure chalcone. Orange blocks of (I) were grown by the slow evaporation of the solution of the compound in ethyl alcohol (*m. p.*: 361 K). Composition: Found (Calculated) for C₁₇H₁₆O₃, C: 76.10 (76.16); H: 6.01 (6.05).

S3. Refinement

H1O1 atom attached to the O atom was located from the difference map and refined freely [O—H = 0.94 (2) Å]. The remaining H atoms were positioned geometrically [C—H = 0.93 or 0.96 Å] and refined using a riding model with *U*_{iso}(H) = 1.2 or 1.5 *U*_{eq}(C). A rotating group model was applied to the methyl groups.

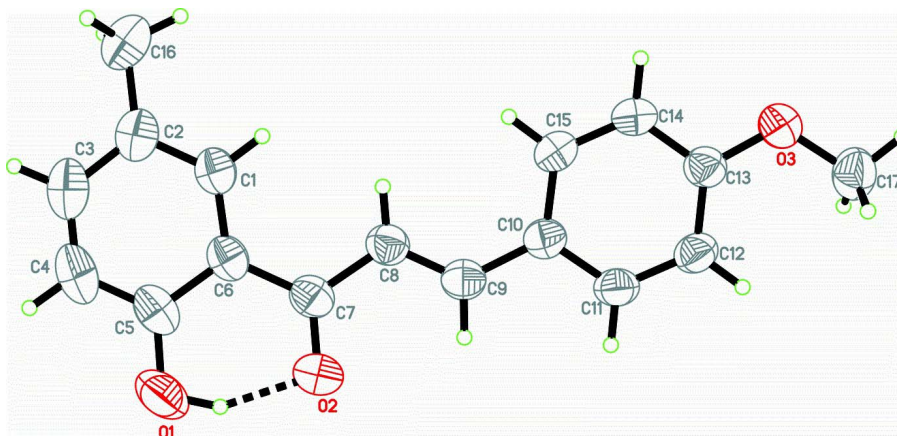


Figure 1

The molecular structure of the title compound, showing 50% probability displacement ellipsoids. The dashed line indicates the intramolecular bond.

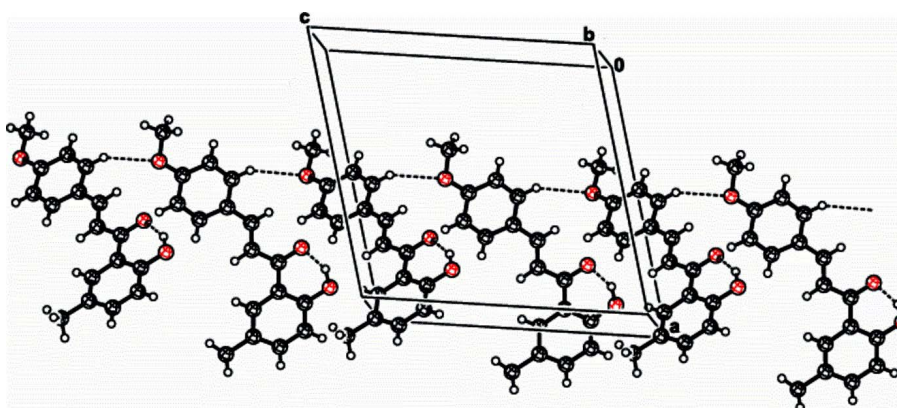


Figure 2

The crystal packing of the title compound, viewed along the *b* axis.

(2*E*)-1-(2-Hydroxy-5-methylphenyl)-3-(4-methoxyphenyl)prop-2-en-1-one

Crystal data

$C_{17}H_{16}O_3$

$M_r = 268.30$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1bc$

$a = 12.6990\ (18)\ \text{\AA}$

$b = 8.8022\ (13)\ \text{\AA}$

$c = 13.172\ (2)\ \text{\AA}$

$\beta = 105.565\ (2)^\circ$

$V = 1418.3\ (4)\ \text{\AA}^3$

$Z = 4$

$F(000) = 568$

$D_x = 1.256\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2769 reflections

$\theta = 2.8\text{--}28.6^\circ$

$\mu = 0.09\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Block, orange

$0.46 \times 0.32 \times 0.18\ \text{mm}$

Data collection

Bruker SMART APEXII DUO CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2009)

$T_{\min} = 0.962$, $T_{\max} = 0.984$

11493 measured reflections
 4090 independent reflections
 2608 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.024$

$\theta_{\text{max}} = 29.9^\circ$, $\theta_{\text{min}} = 2.8^\circ$
 $h = -17 \rightarrow 17$
 $k = -10 \rightarrow 12$
 $l = -18 \rightarrow 18$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.131$
 $S = 1.02$
 4090 reflections
 187 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0547P)^2 + 0.1652P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.16 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.16 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.13821 (11)	0.18686 (15)	0.32930 (8)	0.0794 (4)
O2	0.26594 (9)	0.35672 (14)	0.26545 (7)	0.0758 (3)
O3	0.51456 (8)	0.87081 (13)	−0.15126 (7)	0.0666 (3)
C1	0.03317 (10)	0.26022 (14)	0.04622 (10)	0.0486 (3)
H1A	0.0518	0.3125	−0.0078	0.058*
C2	−0.06268 (11)	0.17724 (16)	0.02260 (12)	0.0553 (3)
C3	−0.08898 (13)	0.10217 (19)	0.10552 (15)	0.0704 (4)
H3A	−0.1538	0.0472	0.0921	0.084*
C4	−0.02277 (15)	0.10664 (19)	0.20568 (15)	0.0728 (4)
H4A	−0.0427	0.0543	0.2590	0.087*
C5	0.07432 (12)	0.18853 (16)	0.22906 (11)	0.0581 (4)
C6	0.10376 (10)	0.26897 (14)	0.14840 (10)	0.0468 (3)
C7	0.20644 (11)	0.35632 (16)	0.17338 (10)	0.0502 (3)
C8	0.23943 (10)	0.44232 (15)	0.09183 (10)	0.0488 (3)
H8A	0.1929	0.4469	0.0240	0.059*
C9	0.33518 (10)	0.51410 (15)	0.11356 (10)	0.0478 (3)
H9A	0.3790	0.5037	0.1822	0.057*
C10	0.38013 (9)	0.60674 (14)	0.04375 (9)	0.0442 (3)
C11	0.48166 (10)	0.67433 (16)	0.08326 (10)	0.0501 (3)
H11A	0.5188	0.6589	0.1536	0.060*

C12	0.52938 (10)	0.76377 (16)	0.02169 (10)	0.0504 (3)
H12A	0.5973	0.8081	0.0504	0.060*
C13	0.47519 (10)	0.78657 (15)	−0.08275 (9)	0.0477 (3)
C14	0.37276 (11)	0.72165 (17)	−0.12373 (10)	0.0581 (4)
H14A	0.3355	0.7385	−0.1939	0.070*
C15	0.32612 (10)	0.63348 (16)	−0.06231 (10)	0.0542 (3)
H15A	0.2577	0.5907	−0.0912	0.065*
C16	−0.13460 (12)	0.16543 (19)	−0.08791 (14)	0.0704 (4)
H16A	−0.1021	0.2204	−0.1347	0.106*
H16B	−0.2051	0.2077	−0.0911	0.106*
H16C	−0.1427	0.0606	−0.1086	0.106*
C17	0.61893 (12)	0.93974 (19)	−0.11419 (12)	0.0645 (4)
H17A	0.6362	0.9942	−0.1708	0.097*
H17B	0.6731	0.8627	−0.0886	0.097*
H17C	0.6181	1.0090	−0.0581	0.097*
H1O1	0.2008 (17)	0.245 (2)	0.3290 (16)	0.107 (7)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.1019 (9)	0.0879 (9)	0.0533 (6)	−0.0055 (7)	0.0295 (6)	0.0145 (6)
O2	0.0760 (7)	0.1013 (9)	0.0451 (5)	−0.0174 (6)	0.0078 (5)	0.0079 (5)
O3	0.0642 (6)	0.0840 (8)	0.0509 (5)	−0.0187 (5)	0.0142 (5)	0.0088 (5)
C1	0.0523 (7)	0.0438 (7)	0.0542 (7)	0.0022 (6)	0.0222 (6)	−0.0008 (6)
C2	0.0514 (7)	0.0456 (8)	0.0729 (9)	0.0012 (6)	0.0239 (6)	−0.0062 (6)
C3	0.0651 (9)	0.0595 (10)	0.0961 (13)	−0.0098 (7)	0.0380 (9)	−0.0011 (9)
C4	0.0863 (11)	0.0626 (10)	0.0845 (11)	−0.0079 (9)	0.0488 (10)	0.0088 (8)
C5	0.0756 (9)	0.0509 (8)	0.0567 (8)	0.0042 (7)	0.0331 (7)	0.0038 (6)
C6	0.0552 (7)	0.0413 (7)	0.0496 (7)	0.0036 (6)	0.0241 (6)	−0.0015 (5)
C7	0.0570 (7)	0.0503 (8)	0.0454 (6)	0.0025 (6)	0.0172 (6)	−0.0010 (5)
C8	0.0537 (7)	0.0499 (7)	0.0428 (6)	−0.0015 (6)	0.0130 (5)	−0.0013 (5)
C9	0.0503 (6)	0.0489 (7)	0.0435 (6)	0.0024 (6)	0.0114 (5)	−0.0028 (5)
C10	0.0445 (6)	0.0453 (7)	0.0427 (6)	0.0028 (5)	0.0113 (5)	−0.0040 (5)
C11	0.0458 (6)	0.0617 (9)	0.0395 (6)	0.0009 (6)	0.0058 (5)	0.0004 (6)
C12	0.0400 (6)	0.0621 (9)	0.0462 (6)	−0.0029 (6)	0.0065 (5)	−0.0027 (6)
C13	0.0483 (6)	0.0521 (8)	0.0434 (6)	−0.0004 (6)	0.0135 (5)	−0.0012 (5)
C14	0.0554 (7)	0.0723 (10)	0.0401 (6)	−0.0100 (7)	0.0016 (6)	0.0034 (6)
C15	0.0465 (7)	0.0632 (9)	0.0481 (7)	−0.0098 (6)	0.0042 (5)	−0.0018 (6)
C16	0.0544 (8)	0.0657 (10)	0.0876 (11)	−0.0045 (7)	0.0131 (8)	−0.0117 (8)
C17	0.0614 (8)	0.0662 (10)	0.0707 (9)	−0.0104 (7)	0.0257 (7)	−0.0025 (7)

Geometric parameters (Å, °)

O1—C5	1.3517 (18)	C9—C10	1.4554 (17)
O1—H1O1	0.94 (2)	C9—H9A	0.9300
O2—C7	1.2449 (15)	C10—C11	1.3881 (17)
O3—C13	1.3624 (15)	C10—C15	1.4011 (17)
O3—C17	1.4198 (17)	C11—C12	1.3814 (18)

C1—C2	1.3815 (18)	C11—H11A	0.9300
C1—C6	1.4048 (18)	C12—C13	1.3778 (17)
C1—H1A	0.9300	C12—H12A	0.9300
C2—C3	1.392 (2)	C13—C14	1.3903 (18)
C2—C16	1.500 (2)	C14—C15	1.3656 (19)
C3—C4	1.361 (2)	C14—H14A	0.9300
C3—H3A	0.9300	C15—H15A	0.9300
C4—C5	1.390 (2)	C16—H16A	0.9600
C4—H4A	0.9300	C16—H16B	0.9600
C5—C6	1.4082 (18)	C16—H16C	0.9600
C6—C7	1.4729 (18)	C17—H17A	0.9600
C7—C8	1.4641 (18)	C17—H17B	0.9600
C8—C9	1.3315 (17)	C17—H17C	0.9600
C8—H8A	0.9300		
C5—O1—H1O1	106.0 (13)	C11—C10—C9	119.05 (11)
C13—O3—C17	118.65 (11)	C15—C10—C9	123.64 (11)
C2—C1—C6	122.82 (12)	C12—C11—C10	122.26 (11)
C2—C1—H1A	118.6	C12—C11—H11A	118.9
C6—C1—H1A	118.6	C10—C11—H11A	118.9
C1—C2—C3	117.18 (14)	C13—C12—C11	119.26 (11)
C1—C2—C16	121.70 (13)	C13—C12—H12A	120.4
C3—C2—C16	121.11 (14)	C11—C12—H12A	120.4
C4—C3—C2	122.06 (15)	O3—C13—C12	124.53 (11)
C4—C3—H3A	119.0	O3—C13—C14	115.97 (11)
C2—C3—H3A	119.0	C12—C13—C14	119.50 (12)
C3—C4—C5	120.69 (14)	C15—C14—C13	120.86 (12)
C3—C4—H4A	119.7	C15—C14—H14A	119.6
C5—C4—H4A	119.7	C13—C14—H14A	119.6
O1—C5—C4	118.37 (13)	C14—C15—C10	120.80 (12)
O1—C5—C6	122.08 (14)	C14—C15—H15A	119.6
C4—C5—C6	119.54 (14)	C10—C15—H15A	119.6
C1—C6—C5	117.70 (12)	C2—C16—H16A	109.5
C1—C6—C7	122.80 (11)	C2—C16—H16B	109.5
C5—C6—C7	119.49 (12)	H16A—C16—H16B	109.5
O2—C7—C8	119.66 (12)	C2—C16—H16C	109.5
O2—C7—C6	119.25 (12)	H16A—C16—H16C	109.5
C8—C7—C6	121.09 (11)	H16B—C16—H16C	109.5
C9—C8—C7	120.81 (12)	O3—C17—H17A	109.5
C9—C8—H8A	119.6	O3—C17—H17B	109.5
C7—C8—H8A	119.6	H17A—C17—H17B	109.5
C8—C9—C10	128.35 (12)	O3—C17—H17C	109.5
C8—C9—H9A	115.8	H17A—C17—H17C	109.5
C10—C9—H9A	115.8	H17B—C17—H17C	109.5
C11—C10—C15	117.30 (12)		
C6—C1—C2—C3	1.0 (2)	O2—C7—C8—C9	3.8 (2)
C6—C1—C2—C16	−177.83 (13)	C6—C7—C8—C9	−176.22 (12)

C1—C2—C3—C4	−1.3 (2)	C7—C8—C9—C10	−178.59 (12)
C16—C2—C3—C4	177.45 (14)	C8—C9—C10—C11	178.78 (13)
C2—C3—C4—C5	0.5 (3)	C8—C9—C10—C15	−0.7 (2)
C3—C4—C5—O1	−178.70 (15)	C15—C10—C11—C12	−0.5 (2)
C3—C4—C5—C6	0.7 (2)	C9—C10—C11—C12	−179.97 (12)
C2—C1—C6—C5	0.22 (19)	C10—C11—C12—C13	−0.4 (2)
C2—C1—C6—C7	179.43 (12)	C17—O3—C13—C12	0.6 (2)
O1—C5—C6—C1	178.32 (12)	C17—O3—C13—C14	−179.83 (13)
C4—C5—C6—C1	−1.1 (2)	C11—C12—C13—O3	−179.21 (13)
O1—C5—C6—C7	−0.9 (2)	C11—C12—C13—C14	1.2 (2)
C4—C5—C6—C7	179.72 (13)	O3—C13—C14—C15	179.21 (13)
C1—C6—C7—O2	−178.73 (13)	C12—C13—C14—C15	−1.2 (2)
C5—C6—C7—O2	0.47 (19)	C13—C14—C15—C10	0.3 (2)
C1—C6—C7—C8	1.27 (19)	C11—C10—C15—C14	0.5 (2)
C5—C6—C7—C8	−179.54 (12)	C9—C10—C15—C14	179.98 (13)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O1—H1O1 $\cdots$ O2	0.95 (2)	1.65 (2)	2.5112 (18)	149.4 (19)
C11—H11A $\cdots$ O3 ⁱ	0.93	2.60	3.4317 (17)	149
C16—H16C $\cdots$ Cg1 ⁱⁱ	0.96	2.81	3.5800 (18)	138

Symmetry codes: (i) $x, -y+3/2, z+1/2$; (ii) $-x, -y, -z$.