Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

## 1,1'-(Piperazine-1,4-diyl)dipropan-2-ol

Murat Türkyılmaz, ${ }^{\text {a }}$ Yakup Baran ${ }^{\text {b }}$ and Namık Özdemir ${ }^{\text {c }}$ *<br>${ }^{\text {a }}$ Department of Chemistry, Faculty of Science, Trakya University, 22030-Edirne, Turkey, ${ }^{\mathbf{b}}$ Department of Chemistry, Faculty of Arts and Sciences, Çanakkale Onsekiz Mart University, 17020-Çanakkale, Turkey, and ${ }^{\mathbf{c}}$ Department of Physics, Faculty of Arts and Sciences, Ondokuz Mayıs University, 55139-Samsun, Turkey<br>Correspondence e-mail: namiko@omu.edu.tr

Received 21 April 2011; accepted 15 June 2011
Key indicators: single-crystal X-ray study; $T=273 \mathrm{~K}$; mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$; disorder in main residue; $R$ factor $=0.052 ; w R$ factor $=0.119$; data-to-parameter ratio $=13.1$.

The asymmetric unit of the crystal contains one-fourth of the title compound, $\mathrm{C}_{10} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$, with the centre of the piperazine ring located at a site of $2 / m$ symmetry. The piperazine ring adopts a chair conformation. The methine and methylene C atoms of the 2-hydroxypropyl groups show symmetry-imposed disorder over two equally occupied and mutually exclusive sets of positions. Only intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ contacts are observed.

## Related literature

For the biological properties of piperazine compounds, see: Foroumadi et al. (2007); Upadhayaya et al. (2004); Chen et al. (2006); Cunico et al. (2009); Smits et al. (2008); Penjišević et al. (2007); Becker et al. (2006). For hydrogen-bond graph-set motifs, see: Bernstein et al. (1995). For ring puckering parameters, see: Cremer \& Pople (1975).


## Experimental

## Crystal data

$\mathrm{C}_{10} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$
$M_{r}=2022.30$
Monoclinic, $C 2 / m$
$a=13.838(10) \AA$
$b=7.791(5) \AA$
$c=5.543(4) \AA$
$\beta=97.26(3)^{\circ}$
$V=592.8(7) \AA^{3}$
$Z=2$
Mo $K \alpha$ radiation
$\mu=0.08 \mathrm{~mm}^{-1}$
$T=273 \mathrm{~K}$
$0.40 \times 0.25 \times 0.10 \mathrm{~mm}$

## Data collection

Rigaku R-AXIS RAPID diffractometer
2493 measured reflections
604 independent reflections 441 reflections with $I>2 \sigma(I)$ $R_{\text {int }}=0.139$

## Refinement

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.052$
14 restraints
$w R\left(F^{2}\right)=0.119$
H -atom parameters constrained
$S=1.05$
$\Delta \rho_{\text {max }}=0.28 \mathrm{e}^{-3}$
604 reflections
46 parameters
$\Delta \rho_{\text {min }}=-0.34 \mathrm{e}^{-3}$

Table 1
Hydrogen-bond geometry ( $\AA,^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O} 1-\mathrm{H} 1 \cdots \mathrm{~N} 1$ | 0.82 | 2.22 | $2.696(3)$ | 117 |

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/ MSC, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON (Spek, 2009).

Financial support from the Scientific and Technological Research Council of Turkey research program 1001 grant for 104 T389 is gratefully acknowledged.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FY2011).

## References

Becker, O. M., Dhanoa, D. S., Marantz, Y., Chen, D., Shacham, S., Cheruku, S., Heifetz, A., Mohanty, P., Fichman, M., Sharadendu, A., Nudelman, R., Kauffman, M. \& Noiman, S. (2006). J. Med. Chem. 49, 3116-3135.
Bernstein, J., Davis, R. E., Shimoni, L. \& Chang, N. L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555-1573.
Chen, J. J., Lu, M., Jing, Y. K. \& Dong, J. H. (2006). Bioorg. Med. Chem. 14, 6539-6547.
Cremer, D. \& Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
Cunico, W., Gomes, C. R. B., Moreth, M., Manhanini, D. P., Figueiredo, I. H., Penido, C., Henriques, M. G. M. O., Varotti, F. P. \& Krettli, A. U. (2009). Eur. J. Med. Chem. 44, 1363-1368.
Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
Foroumadi, A., Emami, S., Mansouri, S., Javidnia, A., Saeid-Adeli, N., Shirazi, F. H. \& Shafiee, A. (2007). Eur. J. Med. Chem. 42, 985-992.

Penjišević, J., Šukalović, V., Andrić, D., Kostić-Rajačić, S., Šoškić, V. \& Roglić, G. (2007). Arch. Pharm. Chem. Life Sci. 340, 456-465.

Rigaku (1998). PROCESS-AUTO. Rigaku Corporation, Tokyo, Japan.
Rigaku/MSC (2004). CrystalStructure. Rigaku/MSC, The Woodlands, Texas, USA.
Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
Smits, R. A., Lim, H. D., Hanzer, A., Zuiderveld, O. P., Guaita, E., Adami, M., Coruzzi, G., Leurs, R. \& Esch, I. J. P. (2008). J. Med. Chem. 51, 2457-2467. Spek, A. L. (2009). Acta Cryst. D65, 148-155.
Upadhayaya, R. S., Sinha, N., Jain, S., Kishore, N., Chandra, R. \& Arora, S. K. (2004). Bioorg. Med. Chem. 12, 2225-2238.

## supporting information

Acta Cryst. (2011). E67, o1758 [doi:10.1107/S1600536811023397]

## 1,1'-(Piperazine-1,4-diyl)dipropan-2-ol

Murat Türkyılmaz, Yakup Baran and Namık Özdemir

## S1. Comment

Piperazine-based research has attracted considerable attention in recent years. A broad range of compounds displaying antibacterial (Foroumadi et al., 2007), antifungal (Upadhayaya et al., 2004), anticancer (Chen et al., 2006), antiparasitic (Cunico et al., 2009), antihistamin (Smits et al., 2008), psychotolytic (Penjišević et al., 2007), and antidepressive activities (Becker et al., 2006) have been found to contain this versatile core. In view of these important properties, we have undertaken the X-ray diffraction study of the title compound.
The structure of the title compound is shown in Fig. 1. The structure contains one central piperazine ring
( $\mathrm{N} 1 / \mathrm{C} 4 / \mathrm{C} 4 / \mathrm{i} 11^{\mathrm{i}} / \mathrm{C} 4 \mathrm{ii} / \mathrm{C} 4{ }^{\mathrm{iii}}$ ) with two propanol moieties substituted at the two N atoms of the piperazine ring. The centre of the ring located at a site of $2 / m$ symmetry. The $\mathrm{N}, \mathrm{O}$ and methyl C atoms are located on the mirror plane, while atoms C 2 and C 3 show symmetry-imposed disorder.
The interatomic distances and angles in the compound show no anomalies. The piperazine ring adopts a chair conformation, as is evident from the puckering parameters (Cremer \& Pople, 1975): $\mathrm{Q}_{\mathrm{T}}=1.0333$ (10) $\AA, \mathrm{q}_{2}=0.8812$ (9) $\AA, \mathrm{q}_{3}=0.5396(6) \AA, \theta=58.52(2)^{\circ}$ and $\varphi_{2}=30.00(5)^{\circ}$ for the atom sequence $\mathrm{N} 1 / \mathrm{C} 4 / \mathrm{C} 4{ }^{\mathrm{i}} / \mathrm{N} 1^{\mathrm{i}} / \mathrm{C} 44^{\mathrm{ii}} / \mathrm{C} 4{ }^{\mathrm{iii}}$. Atoms N 1 and $\mathrm{N} 1^{\mathrm{i}}$ are on opposite sides of the $\mathrm{C} 4 / \mathrm{C} 4^{\mathrm{i}} / \mathrm{C} 4^{\mathrm{ii}} / \mathrm{C} 4^{\mathrm{iii}}$ plane and are both displaced from it by 0.2424 (30) $\AA$.
The molecular structure of the title compound contains two intramolecular $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}$ contacts, which form a fivemembered ring with graph-set descriptor $S(5)$ (Bernstein et al., 1995). No intermolecular hydrogen bonds are observed in the crystal structure. Van der Waals forces stabilize the packing.

## S2. Experimental

Piperazine ( $1.50 \mathrm{~g}, 17.40 \mathrm{mmol}$ ) was dissolved in 50 ml argon saturated methanol. Methanol solution of 2.88 g ( 50.00 mmol ) propylene oxide was added to the piperazine solution at room temperature. The solution was left under magnetic stirrer for 24 h . The solution volume was reduced by rotary evaporator and the oily product was left for crystallization.

## S3. Refinement

H atoms were positioned geometrically and treated using a riding model, fixing the bond lengths at $0.82,0.93,0.97$ and $0.96 \AA$ for $\mathrm{OH}, \mathrm{CH}, \mathrm{CH}_{2}$ and $\mathrm{CH}_{3}$ groups, respectively. The isotropic displacement parameters of the H atoms were constrained at $1.2 U_{\text {eq }}$ of their parent atom (1.5 $U_{\text {eq }}$ for methyl and OH groups). Atoms C 2 and C 3 showed symmetryimposed disorder and were refined anisotropically using ADP restraints (SIMU and DELU) and half occupancy.


## Figure 1

ORTEP-3 drawing of the title compound, showing $30 \%$ probability displacement ellipsoids and the atomic numbering scheme. For the sake of clarity, H atoms have been omitted. One of the disorder components is drawn with dashed bonds. [Symmetry codes: (i) $1-x, y, 1-z$; (ii) $1-x, 1-y, 1-z$; (iii) $x, 1-y, z$.]

## 1,1'-(Piperazine-1,4-diyl)dipropan-2-ol

## Crystal data

## $\mathrm{C}_{10} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$

$M_{r}=202.30$
Monoclinic, $C 2 / m$
Hall symbol: -C 2 y
$a=13.838$ (10) $\AA$
$b=7.791$ (5) $\AA$
$c=5.543$ (4) $\AA$
$\beta=97.26(3)^{\circ}$
$V=592.8(7) \AA^{3}$
$Z=2$

## Data collection

Rigaku R-AXIS RAPID
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 10.00 pixels $\mathrm{mm}^{-1}$
$\omega$ scans
2493 measured reflections

## Refinement

Refinement on $F^{2}$
Least-squares matrix: full
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.052$
$w R\left(F^{2}\right)=0.119$
$S=1.05$
604 reflections
46 parameters
14 restraints
Primary atom site location: structure-invariant direct methods
$F(000)=224$
$D_{\mathrm{x}}=1.133 \mathrm{Mg} \mathrm{m}^{-3}$
Mo K $\alpha$ radiation, $\lambda=0.71073 \AA$
Cell parameters from 3092 reflections
$\theta=3.0-30.0^{\circ}$
$\mu=0.08 \mathrm{~mm}^{-1}$
$T=273 \mathrm{~K}$
Prism, colourless
$0.40 \times 0.25 \times 0.10 \mathrm{~mm}$

604 independent reflections
441 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.139$
$\theta_{\text {max }}=25.5^{\circ}, \theta_{\text {min }}=3.0^{\circ}$
$h=-16 \rightarrow 16$
$k=-9 \rightarrow 9$
$l=-5 \rightarrow 6$

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{0}{ }^{2}\right)+(0.0375 P)^{2}+0.2898 P\right]$
where $P=\left(F_{0}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.28 \mathrm{e} \AA^{-3}$
$\Delta \rho_{\text {min }}=-0.34 \mathrm{e} \AA^{-3}$

## Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.
Refinement. Refinement of $F^{2}$ against ALL reflections. The weighted $R$-factor $w R$ and goodness of fit $S$ are based on $F^{2}$, conventional $R$-factors $R$ are based on $F$, with $F$ set to zero for negative $F^{2}$. The threshold expression of $F^{2}>\sigma\left(F^{2}\right)$ is used only for calculating $R$-factors(gt) etc. and is not relevant to the choice of reflections for refinement. $R$-factors based on $F^{2}$ are statistically about twice as large as those based on $F$, and $R$ - factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\AA^{2}$ )

|  | $x$ | $y$ | $z$ | $U_{\text {iso }} / U_{\text {eq }}$ | Occ. $(<1)$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| O1 | $0.78313(12)$ | 0.5000 | $0.4674(3)$ | $0.0706(7)$ |  |
| H1 | 0.7297 | 0.5197 | 0.3907 | $0.106^{*}$ | 0.50 |
| N1 | $0.60084(15)$ | 0.5000 | $0.5912(4)$ | $0.0715(7)$ |  |
| C1 | $0.86326(18)$ | 0.5000 | $0.8733(5)$ | $0.0754(9)$ |  |
| H1A | 0.9169 | 0.4390 | 0.8197 | $0.113^{*}$ | 0.50 |
| H1B | 0.8573 | 0.4690 | 1.0384 | $0.113^{*}$ | 0.50 |
| H1C | 0.8746 | 0.6213 | 0.8640 | $0.113^{*}$ | 0.50 |
| C2 | $0.7712(2)$ | $0.4545(5)$ | $0.7137(5)$ | $0.0603(12)$ | 0.50 |
| H2 | 0.7597 | 0.3307 | 0.7233 | $0.072^{*}$ | 0.50 |
| C3 | $0.6841(2)$ | $0.5495(5)$ | $0.7803(6)$ | $0.0609(11)$ | 0.50 |
| H3A | 0.6952 | 0.6724 | 0.7782 | $0.073^{*}$ | 0.50 |
| H3B | 0.6705 | 0.5166 | 0.9413 | $0.073^{*}$ | 0.50 |
| C4 | $0.54111(14)$ | $0.6506(3)$ | $0.6021(3)$ | $0.0751(7)$ |  |
| H4A | 0.5806 | 0.7526 | 0.5917 | $0.090^{*}$ |  |
| H4B | 0.5150 | 0.6533 | 0.7565 | $0.090^{*}$ |  |

Atomic displacement parameters $\left(\AA^{2}\right)$

|  | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| O1 | $0.0609(11)$ | $0.0975(15)$ | $0.0541(11)$ | 0.000 | $0.0103(8)$ | 0.000 |
| N1 | $0.0463(11)$ | $0.118(2)$ | $0.0498(12)$ | 0.000 | $0.0031(9)$ | 0.000 |
| C1 | $0.0527(15)$ | $0.103(2)$ | $0.0677(17)$ | 0.000 | $-0.0030(13)$ | 0.000 |
| C2 | $0.0519(15)$ | $0.075(3)$ | $0.0533(16)$ | $0.0024(15)$ | $0.0027(13)$ | $-0.0010(15)$ |
| C3 | $0.0483(14)$ | $0.084(3)$ | $0.0490(14)$ | $0.0021(14)$ | $0.0016(12)$ | $-0.0047(14)$ |
| C4 | $0.0807(13)$ | $0.0883(15)$ | $0.0563(11)$ | $-0.0195(11)$ | $0.0084(9)$ | $-0.0038(11)$ |

## Geometric parameters $\left(\AA,{ }^{\circ}\right)$

| $\mathrm{O} 1-\mathrm{C} 2^{\mathrm{i}}$ | $1.440(4)$ | $\mathrm{C} 1-\mathrm{H} 1 \mathrm{~B}$ | 0.9600 |
| :--- | :--- | :--- | :--- |
| $\mathrm{O} 1-\mathrm{C} 2$ | $1.440(4)$ | $\mathrm{C} 1-\mathrm{H} 1 \mathrm{C}$ | 0.9600 |
| $\mathrm{O} 1-\mathrm{H} 1$ | 0.8200 | $\mathrm{C} 2-\mathrm{C} 3$ | $1.499(4)$ |
| $\mathrm{N} 1-\mathrm{C} 4^{\mathrm{i}}$ | $1.441(3)$ | $\mathrm{C} 2-\mathrm{H} 2$ | 0.9800 |
| $\mathrm{~N} 1-\mathrm{C} 4$ | $1.441(3)$ | $\mathrm{C} 3-\mathrm{H} 3 \mathrm{~A}$ | 0.9700 |
| $\mathrm{~N} 1-\mathrm{C} 3$ | $1.507(3)$ | $\mathrm{C} 3-\mathrm{H} 3 \mathrm{~B}$ | 0.9700 |
| $\mathrm{~N} 1-\mathrm{C} 3^{\mathrm{i}}$ | $1.507(3)$ | $\mathrm{C} 4-\mathrm{C} 4^{\mathrm{ii}}$ | $1.500(4)$ |
| $\mathrm{C} 1-\mathrm{C} 2$ | $1.499(4)$ | $\mathrm{C} 4-\mathrm{H} 4 \mathrm{~A}$ | 0.9700 |


| $\mathrm{C} 1-\mathrm{C} 2{ }^{\text {i }}$ | 1.499 (4) |
| :---: | :---: |
| $\mathrm{C} 1-\mathrm{H} 1 \mathrm{~A}$ | 0.9600 |
| $\mathrm{C} 2 \mathrm{i}-\mathrm{O} 1-\mathrm{H} 1$ | 103.9 |
| $\mathrm{C} 2-\mathrm{O} 1-\mathrm{H} 1$ | 109.5 |
| $\mathrm{C} 4 \mathrm{i}^{\text {- }} 10-\mathrm{C} 4$ | 109.0 (2) |
| C4i-N1-C3 | 124.8 (2) |
| $\mathrm{C} 4-\mathrm{N} 1-\mathrm{C} 3$ | 98.90 (17) |
| $\mathrm{C} 4{ }^{\text {i }}$ - $\mathrm{N} 1-\mathrm{C} 3^{\text {i }}$ | 98.90 (17) |
| $\mathrm{C} 4-\mathrm{N} 1-\mathrm{C} 3^{\text {i }}$ | 124.8 (2) |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{H} 1 \mathrm{~A}$ | 109.5 |
| $\mathrm{C} 2 \mathrm{i}-\mathrm{C} 1-\mathrm{H} 1 \mathrm{~A}$ | 124.6 |
| C2-C1-H1B | 109.5 |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{H} 1 \mathrm{~B}$ | 116.9 |
| $\mathrm{H} 1 \mathrm{~A}-\mathrm{C} 1-\mathrm{H} 1 \mathrm{~B}$ | 109.5 |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{H} 1 \mathrm{C}$ | 109.5 |
| $\mathrm{C} 2-\mathrm{C} 1-\mathrm{H} 1 \mathrm{C}$ | 82.4 |
| $\mathrm{H} 1 \mathrm{~A}-\mathrm{C} 1-\mathrm{H} 1 \mathrm{C}$ | 109.5 |
| H1B-C1-H1C | 109.5 |
| $\mathrm{O} 1-\mathrm{C} 2-\mathrm{C} 1$ | 108.1 (2) |
| $\mathrm{C} 2 \mathrm{i}-\mathrm{O} 1-\mathrm{C} 2-\mathrm{C} 1$ | -70.1 (2) |
| $\mathrm{C} 2 \mathrm{i}-\mathrm{O} 1-\mathrm{C} 2-\mathrm{C} 3$ | 52.0 (2) |
| $\mathrm{C} 2 \mathrm{i}-\mathrm{C} 1-\mathrm{C} 2-\mathrm{O} 1$ | 69.7 (2) |
| $\mathrm{C} 2 \mathrm{i}-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | -49.2 (3) |
| $\mathrm{O} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{N} 1$ | 56.3 (3) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3-\mathrm{N} 1$ | 175.4 (2) |

C4—H4B

O1-C2-C3
C1-C2-C3
$\mathrm{O} 1-\mathrm{C} 2-\mathrm{H} 2$
$\mathrm{C} 1-\mathrm{C} 2-\mathrm{H} 2$
$\mathrm{C} 3-\mathrm{C} 2-\mathrm{H} 2$
$\mathrm{C} 2-\mathrm{C} 3-\mathrm{N} 1$
$\mathrm{C} 2-\mathrm{C} 3-\mathrm{H} 3 \mathrm{~A}$
N1-C3-H3A
$\mathrm{C} 2-\mathrm{C} 3-\mathrm{H} 3 \mathrm{~B}$
N1-C3-H3B
H3A-C3-H3B
$\mathrm{N} 1-\mathrm{C} 4-\mathrm{C} 4^{\mathrm{ii}}$
N1-C4-H4A
C4ii-C4-H4A
N1-C4-H4B
$\mathrm{C} 4{ }^{\mathrm{ii}}-\mathrm{C} 4-\mathrm{H} 4 \mathrm{~B}$
$\mathrm{H} 4 \mathrm{~A}-\mathrm{C} 4-\mathrm{H} 4 \mathrm{~B}$
C4-N1-C3-C2
$\mathrm{C} 4-\mathrm{N} 1-\mathrm{C} 3-\mathrm{C} 2$
C3i-N1-C3-C2
$\mathrm{C} 4-\mathrm{N} 1-\mathrm{C} 4-\mathrm{C} 4{ }^{\mathrm{ii}}$
$\mathrm{C} 3-\mathrm{N} 1-\mathrm{C} 4-\mathrm{C} 4{ }^{\text {ii }}$
$\mathrm{C} 3-\mathrm{N} 1-\mathrm{C} 4-\mathrm{C} 4^{\mathrm{ii}}$
0.9700
107.7 (2)
112.8 (3)
109.4
109.4
109.4
105.7 (2)
110.6
110.6
110.6
110.6
108.7
110.64 (15)
109.5
109.5
109.5
109.5
108.1
85.4 (3)
-153.8 (2)
52.8 (3)
-58.1 (3)
170.2 (2)
-174.2 (2)

Symmetry codes: (i) $x,-y+1, z$; (ii) $-x+1, y,-z+1$.

Hydrogen-bond geometry ( $A,{ }^{\circ}$ )

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{O} 1 — \mathrm{H} 1 \cdots \mathrm{~N} 1$ | 0.82 | 2.22 | $2.696(3)$ | 117 |

