Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

2,2'-(3,3'-Dihexyl-2,2'-bithiophene-5,5'diyl)bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolane)

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Received 12 November 2011; accepted 27 November 2011

Key indicators: single-crystal X-ray study; T = 100 K; mean σ (C–C) = 0.002 Å; *R* factor = 0.039; w*R* factor = 0.118; data-to-parameter ratio = 26.3.

In the title molecule, $C_{32}H_{52}B_2O_4S_2$, the two thiophene rings are twisted by 67.34 (2)°. In the crystal, weak $C-H\cdots O$ hydrogen bonds link molecules related by translation along the *a* axis into chains.

Related literature

For potential applications of the title compound, see: Navarro *et al.* (2004); Usta *et al.* (2006); Buszek & Brown (2007); Montes *et al.* (2007). For related structures, see: Decken *et al.* (2008); Kleeberg *et al.* (2009).

Experimental

Crystal data

 $C_{32}H_{52}B_2O_4S_2$ $V = 3355.1 \ (6) \ Å^3$
 $M_r = 586.48$ Z = 4

 Monoclinic, $P2_1/c$ Mo K α radiation

 $a = 11.5004 \ (11) \ Å$ $\mu = 0.19 \ \text{mm}^{-1}$
 $b = 13.6992 \ (13) \ Å$ $T = 100 \ \text{K}$
 $c = 21.300 \ (2) \ Å$ $0.16 \times 0.12 \times 0.10 \ \text{mm}$

Data collection

Bruker APEX CCD diffractometer Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{min} = 0.970, T_{max} = 0.981$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$	371 parameters
$vR(F^2) = 0.118$	H-atom parameters constrained
S = 1.05	$\Delta \rho_{\rm max} = 0.56 \text{ e } \text{\AA}^{-3}$
0743 reflections	$\Delta \rho_{\rm min} = -0.27 \ {\rm e} \ {\rm \AA}^{-3}$

29998 measured reflections

 $R_{\rm int} = 0.026$

9743 independent reflections

8009 reflections with $I > 2\sigma(I)$

Table 1

Hydrogen-bond geometry (Å, °).

2 11 11	$D \cdots A$	$H \cdots A$	D-H	$D - H \cdot \cdot \cdot A$
136	3.2984 (18)	2.53	0.98	$C30-H30C\cdotsO1^{i}$
150	5.2504 (10)	2.55	0.90	

Symmetry code: (i) x - 1, y, z.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

The authors are grateful to Xiangfan University for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5202).

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supporting information

Acta Cryst. (2011). E67, o3512 [https://doi.org/10.1107/S1600536811050938]

2,2'-(3,3'-Dihexyl-2,2'-bithiophene-5,5'-diyl)bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolane)

Lin Huang and Huisheng Li

S1. Comment

Arylboronic acid and their esters are important reactants in the Suzuki-Miyaura cross-coupling reactions (Navarro *et al.*, 2004; Buszek *et al.*, 2007). Lots of functional organic compounds which have broad applications in material chemistry are prepared *via* the named reaction (Montes *et al.*, 2007; Usta *et al.*, 2006). We herein report the crystal structure of the title compound (I).

In (I) (Fig. 1), the geometric parameters of 2-(thiophene-2-yl)-4,4,5,5-tetramethyl- 1,3,2-dioxaborolane fragments are normal and comparable with those observed in related structures (Decken *et al.*, 2008; Kleeberg *et al.*, 2009). Two thiophene rings are twisted at 67.34 (2)°. In the crystal structure, weak intermolecular C—H…O hydrogen bonds (Table 1) link molecules related by translation along axis *a* into chains.

S2. Experimental

The powder form of the title compound (I) was purchased from Aldrich chemical company. Crystals of (I) suitable for X-ray data collection were obtained by slow evaporation of a acetone and MeOH solution in a ratio of 1:2 at room temperature for several days.

S3. Refinement

All H atoms were positioned geometrically (C—H = 0.93–0.97 Å) and refined as riding, allowing for free rotation of the methyl groups. The constraint $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}$ (methyl C) was applied.



Figure 1

The molecular structure of (I) with the atom-numbering scheme. The displacement ellipsoids are drawn at the 30% probability level.

2,2'-(3,3'-Dihexyl-2,2'-bithiophene-5,5'-diyl)bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolane)

Crystal data

$C_{32}H_{52}B_{2}O_{4}S_{2}$ $M_{r} = 586.48$ Monoclinic, $P2_{1}/c$ Hall symbol: -P 2ybc $a = 11.5004 (11) \text{ Å}$ $b = 13.6992 (13) \text{ Å}$ $c = 21.300 (2) \text{ Å}$ $\beta = 91.065 (2)^{\circ}$ $V = 3355.1 (6) \text{ Å}^{3}$ $Z = 4$	F(000) = 1272 $D_x = 1.161 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 9936 reflections $\theta = 2.3-31.9^{\circ}$ $\mu = 0.19 \text{ mm}^{-1}$ T = 100 K Block, yellow $0.16 \times 0.12 \times 0.10 \text{ mm}$
Data collection Bruker APEX CCD	20008 measured reflections
diffractometer	9743 independent reflections
Radiation source: fine-focus sealed tube	8009 reflections with $I > 2\sigma(I)$
Graphite monochromator	$R_{\rm int} = 0.026$
φ and ω scans	$\theta_{\rm max} = 30.0^{\circ}, \theta_{\rm min} = 1.8^{\circ}$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$ h = -16 \rightarrow 9 \\ k = -19 \rightarrow 19 $
$T_{\min} = 0.970, \ T_{\max} = 0.981$	$l = -29 \rightarrow 29$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.039$	Hydrogen site location: inferred from
$wR(F^2) = 0.118$	neighbouring sites
S = 1.05	H-atom parameters constrained
9743 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0667P)^2 + 1.0083P]$
371 parameters	where $P = (F_{o}^{2} + 2F_{c}^{2})/3$
0 restraints	$(\Delta/\sigma)_{\rm max} = 0.002$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.56 \text{ e } \text{\AA}^{-3}$
direct methods	$\Delta \rho_{\min} = -0.27 \text{ e} \text{ Å}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (A^2)

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$
B1	1.10475 (12)	0.42742 (10)	0.66320 (6)	0.0165 (2)
B2	0.42238 (12)	0.75653 (10)	0.66220 (6)	0.0177 (2)
C1	1.31526 (14)	0.45484 (11)	0.78413 (7)	0.0302 (3)
H1A	1.3426	0.5109	0.7599	0.045*
H1B	1.3818	0.4142	0.7968	0.045*
H1C	1.2756	0.4781	0.8216	0.045*
C2	1.23132 (11)	0.39499 (9)	0.74398 (6)	0.0191 (2)
C3	1.17597 (13)	0.31513 (11)	0.78332 (6)	0.0252 (3)
H3A	1.1323	0.3449	0.8174	0.038*
H3B	1.2369	0.2728	0.8011	0.038*
H3C	1.1231	0.2763	0.7568	0.038*
C4	1.37052 (12)	0.42478 (12)	0.65253 (7)	0.0292 (3)
H4A	1.3879	0.4033	0.6098	0.044*
H4B	1.4418	0.4234	0.6784	0.044*
H4C	1.3396	0.4914	0.6512	0.044*
C5	1.28072 (11)	0.35677 (9)	0.68070 (6)	0.0181 (2)
C6	1.32667 (12)	0.25288 (10)	0.68271 (6)	0.0245 (3)
H6A	1.2653	0.2088	0.6968	0.037*
H6B	1.3933	0.2492	0.7120	0.037*
H6C	1.3511	0.2335	0.6407	0.037*
C7	0.99473 (10)	0.46694 (8)	0.62913 (5)	0.0160 (2)
C8	0.94547 (11)	0.44202 (9)	0.57185 (6)	0.0167 (2)
H8	0.9786	0.3946	0.5450	0.020*
С9	0.84128 (10)	0.49275 (8)	0.55611 (5)	0.0152 (2)
C10	0.81274 (10)	0.55846 (8)	0.60262 (5)	0.0140 (2)

C11	0.76632 (12)	0.47441 (9)	0.49844 (6)	0.0205 (2)
H11A	0.8136	0.4423	0.4660	0.025*
H11B	0.7386	0.5376	0.4813	0.025*
C12	0.66060 (12)	0.40942 (10)	0.51308 (7)	0.0245(3)
H12A	0.6217	0.4358	0.5505	0.029*
H12B	0.6045	0.4129	0.4773	0.029*
C13	0.69119 (12)	0.30299 (10)	0.52511 (7)	0.0255(3)
H13A	0 7580	0 3000	0.5550	0.031*
H13B	0.7159	0.2730	0.4852	0.031*
C14	0 59126 (13)	0.24350(10)	0.55138(7)	0.0266(3)
H14A	0.5805	0.2622	0.5958	0.032*
H14R	0.5189	0.2604	0.5279	0.032*
C15	0.60942(14)	0.13357(11)	0.5279 0.54790 (8)	0.032 0.0321(3)
H15A	0.6157	0.1142	0.5033	0.0321 (5)
H15R	0.6838	0.1142	0.5694	0.038*
C16	0.51228 (16)	0.07546(12)	0.57741(9)	0.038 0.0380(4)
H16A	0.5080	0.0918	0.6221	0.057*
U16D	0.5080	0.0918	0.5728	0.057*
	0.3279	0.0035	0.5728	0.057*
C17	0.4382 0.71460 (10)	0.0915	0.5504	0.037°
C17	0.71400(10)	0.02709(8) 0.70840(8)	0.00397(3)	0.0142(2)
C10	0.09040(10) 0.50242(11)	0.70849(8)	0.30718(3)	0.0139(2)
U19 U10	0.39343 (11)	0.73830 (9)	0.38403 (0)	0.0173(2)
П19 С20	0.5072	0.8100	0.3040	0.021°
C20	0.53489(10) 0.77521(12)	0.71694 (9)	0.05341(6)	0.0173(2)
C21	0.77521 (12)	0.74408 (9)	0.51653 (6)	0.0201 (2)
HZIA	0.7327	0.7420	0.4/5/	0.024*
H21B	0.8428	0.6996	0.5138	0.024*
022	0.81873 (15)	0.84760 (10)	0.52837(7)	0.0290 (3)
H22A	0.8/55	0.8465	0.5639	0.035*
H22B	0.7523	0.8888	0.5409	0.035*
C23	0.87603 (13)	0.89414 (10)	0.47171 (7)	0.0256 (3)
H23A	0.8210	0.8912	0.4355	0.031*
H23B	0.8911	0.9638	0.4811	0.031*
C24	0.98960 (13)	0.84615 (11)	0.45327 (7)	0.0285 (3)
H24A	0.9740	0.7773	0.4417	0.034*
H24B	1.0435	0.8461	0.4901	0.034*
C25	1.04875 (13)	0.89695 (11)	0.39843 (7)	0.0277 (3)
H25A	1.0532	0.9679	0.4071	0.033*
H25B	1.1293	0.8721	0.3953	0.033*
C26	0.98547 (14)	0.88121 (11)	0.33606 (7)	0.0297 (3)
H26A	0.9815	0.8112	0.3269	0.045*
H26B	1.0275	0.9145	0.3027	0.045*
H26C	0.9065	0.9078	0.3383	0.045*
C27	0.26538 (12)	0.77376 (10)	0.72424 (7)	0.0228 (3)
C28	0.26948 (11)	0.85917 (9)	0.67599 (6)	0.0197 (2)
C29	0.24609 (16)	0.80293 (14)	0.79156 (8)	0.0398 (4)
H29A	0.2499	0.7449	0.8184	0.060*
H29B	0.1694	0.8335	0.7950	0.060*

H29C	0.3064	0.8494	0.8050	0.060*
C30	0.17912 (14)	0.69466 (11)	0.70388 (10)	0.0402 (4)
H30A	0.1920	0.6774	0.6599	0.060*
H30B	0.0996	0.7190	0.7084	0.060*
H30C	0.1904	0.6367	0.7303	0.060*
C31	0.15711 (14)	0.87827 (12)	0.64025 (8)	0.0343 (3)
H31A	0.1682	0.9321	0.6107	0.051*
H31B	0.0962	0.8956	0.6698	0.051*
H31C	0.1340	0.8194	0.6170	0.051*
C32	0.31596 (14)	0.95370 (11)	0.70410 (8)	0.0326 (3)
H32A	0.3881	0.9404	0.7278	0.049*
H32B	0.2582	0.9814	0.7323	0.049*
H32C	0.3316	1.0002	0.6703	0.049*
01	1.13638 (8)	0.45839 (7)	0.72232 (4)	0.02101 (19)
O2	1.17824 (8)	0.35991 (7)	0.63862 (4)	0.01828 (17)
03	0.38074 (8)	0.72931 (7)	0.71924 (5)	0.0244 (2)
O4	0.35654 (8)	0.82553 (7)	0.63193 (4)	0.02210 (19)
S1	0.91171 (3)	0.55548 (2)	0.664168 (13)	0.01627 (7)
S2	0.60692 (3)	0.61359 (2)	0.658712 (14)	0.01632 (7)

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
B1	0.0150 (6)	0.0170 (6)	0.0175 (6)	0.0001 (4)	-0.0012 (5)	0.0024 (4)
B2	0.0148 (6)	0.0190 (6)	0.0193 (6)	0.0010 (5)	-0.0016 (5)	-0.0020 (5)
C1	0.0296 (7)	0.0292 (7)	0.0313 (7)	0.0020 (6)	-0.0147 (6)	-0.0049 (6)
C2	0.0177 (6)	0.0207 (5)	0.0188 (5)	0.0035 (4)	-0.0051 (4)	0.0003 (4)
C3	0.0247 (7)	0.0314 (7)	0.0195 (6)	0.0033 (5)	0.0000 (5)	0.0052 (5)
C4	0.0174 (6)	0.0360 (7)	0.0341 (7)	-0.0024 (5)	-0.0016 (5)	0.0115 (6)
C5	0.0137 (5)	0.0219 (5)	0.0186 (5)	0.0025 (4)	-0.0029 (4)	0.0029 (4)
C6	0.0224 (6)	0.0256 (6)	0.0254 (6)	0.0085 (5)	-0.0020 (5)	0.0009 (5)
C7	0.0135 (5)	0.0174 (5)	0.0171 (5)	0.0020 (4)	-0.0002 (4)	0.0009 (4)
C8	0.0160 (5)	0.0173 (5)	0.0169 (5)	0.0024 (4)	0.0013 (4)	0.0002 (4)
C9	0.0151 (5)	0.0155 (5)	0.0148 (5)	0.0001 (4)	-0.0010 (4)	0.0005 (4)
C10	0.0124 (5)	0.0150 (5)	0.0146 (5)	-0.0003 (4)	-0.0006 (4)	0.0009 (4)
C11	0.0240 (6)	0.0213 (5)	0.0159 (5)	0.0017 (5)	-0.0047 (4)	-0.0016 (4)
C12	0.0218 (6)	0.0249 (6)	0.0263 (6)	0.0005 (5)	-0.0074 (5)	-0.0040(5)
C13	0.0241 (7)	0.0259 (6)	0.0263 (6)	-0.0023 (5)	-0.0026 (5)	-0.0022 (5)
C14	0.0244 (7)	0.0275 (6)	0.0277 (7)	-0.0027 (5)	-0.0052 (5)	-0.0028 (5)
C15	0.0267 (7)	0.0267 (7)	0.0427 (8)	-0.0018 (6)	-0.0028 (6)	-0.0002 (6)
C16	0.0362 (9)	0.0270 (7)	0.0508 (10)	-0.0042 (6)	-0.0023 (7)	0.0037 (7)
C17	0.0125 (5)	0.0160 (5)	0.0143 (5)	-0.0002(4)	0.0001 (4)	-0.0011 (4)
C18	0.0162 (5)	0.0164 (5)	0.0151 (5)	0.0016 (4)	0.0004 (4)	-0.0003 (4)
C19	0.0181 (6)	0.0167 (5)	0.0178 (5)	0.0040 (4)	-0.0015 (4)	-0.0003 (4)
C20	0.0151 (5)	0.0182 (5)	0.0185 (5)	0.0019 (4)	-0.0014 (4)	-0.0017 (4)
C21	0.0242 (6)	0.0175 (5)	0.0189 (5)	0.0016 (5)	0.0049 (5)	0.0016 (4)
C22	0.0424 (9)	0.0217 (6)	0.0232 (6)	-0.0055 (6)	0.0115 (6)	-0.0010 (5)
C23	0.0335 (8)	0.0196 (6)	0.0241 (6)	0.0004 (5)	0.0076 (5)	0.0039 (5)

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C24	0.0293 (7)	0.0315 (7)	0.0248 (6)	0.0027 (6)	0.0009 (5)	0.0078 (5)
C25	0.0238 (7)	0.0341 (7)	0.0252 (7)	-0.0029 (5)	0.0022 (5)	0.0031 (5)
C26	0.0304 (8)	0.0345 (7)	0.0245 (7)	-0.0019 (6)	0.0052 (6)	-0.0020 (5)
C27	0.0176 (6)	0.0220 (6)	0.0291 (6)	0.0064 (5)	0.0063 (5)	0.0062 (5)
C28	0.0164 (6)	0.0199 (5)	0.0228 (6)	0.0041 (4)	0.0038 (4)	0.0009 (4)
C29	0.0375 (9)	0.0540 (10)	0.0284 (7)	0.0162 (8)	0.0117 (6)	0.0098 (7)
C30	0.0246 (7)	0.0231 (7)	0.0732 (12)	-0.0018 (6)	0.0112 (8)	0.0030 (7)
C31	0.0265 (7)	0.0380 (8)	0.0382 (8)	0.0115 (6)	-0.0042 (6)	0.0049 (6)
C32	0.0288 (8)	0.0230 (6)	0.0464 (9)	-0.0026 (6)	0.0106 (7)	-0.0076 (6)
01	0.0206 (4)	0.0220 (4)	0.0202 (4)	0.0071 (3)	-0.0061 (3)	-0.0023 (3)
O2	0.0148 (4)	0.0229 (4)	0.0171 (4)	0.0036 (3)	-0.0029 (3)	0.0000 (3)
O3	0.0172 (4)	0.0299 (5)	0.0261 (5)	0.0092 (4)	0.0052 (4)	0.0081 (4)
O4	0.0223 (5)	0.0248 (4)	0.0193 (4)	0.0089 (4)	0.0043 (3)	0.0022 (3)
S1	0.01508 (14)	0.01852 (14)	0.01509 (13)	0.00182 (10)	-0.00279 (10)	-0.00195 (10)
S2	0.01368 (14)	0.01725 (13)	0.01810 (14)	0.00051 (10)	0.00189 (10)	0.00161 (10)

Geometric parameters (Å, °)

B1—O2	1.3638 (16)	C16—H16A	0.9800
B1—O1	1.3715 (16)	C16—H16B	0.9800
B1—C7	1.5449 (18)	C16—H16C	0.9800
B2—O4	1.3654 (16)	C17—C18	1.3767 (16)
B2—O3	1.3660 (17)	C17—S2	1.7265 (12)
B2—C20	1.5408 (18)	C18—C19	1.4225 (16)
C1—C2	1.5181 (18)	C18—C21	1.5030 (17)
C1—H1A	0.9800	C19—C20	1.3715 (17)
C1—H1B	0.9800	С19—Н19	0.9500
C1—H1C	0.9800	C20—S2	1.7218 (12)
C2—O1	1.4631 (14)	C21—C22	1.5233 (18)
C2—C3	1.5244 (19)	C21—H21A	0.9900
C2—C5	1.5627 (18)	C21—H21B	0.9900
С3—НЗА	0.9800	C22—C23	1.5254 (19)
С3—Н3В	0.9800	C22—H22A	0.9900
С3—НЗС	0.9800	C22—H22B	0.9900
C4—C5	1.5223 (18)	C23—C24	1.520 (2)
C4—H4A	0.9800	С23—Н23А	0.9900
C4—H4B	0.9800	С23—Н23В	0.9900
C4—H4C	0.9800	C24—C25	1.5301 (19)
C5—O2	1.4681 (14)	C24—H24A	0.9900
C5—C6	1.5184 (18)	C24—H24B	0.9900
С6—Н6А	0.9800	C25—C26	1.518 (2)
С6—Н6В	0.9800	C25—H25A	0.9900
С6—Н6С	0.9800	С25—Н25В	0.9900
C7—C8	1.3785 (16)	C26—H26A	0.9800
C7—S1	1.7223 (12)	С26—Н26В	0.9800
C8—C9	1.4199 (16)	C26—H26C	0.9800
С8—Н8	0.9500	С27—ОЗ	1.4654 (15)
C9—C10	1.3826 (16)	C27—C29	1.509 (2)

C9—C11	1.5086 (16)	C27—C30	1.526 (2)
C10—C17	1.4696 (16)	C27—C28	1.5587 (18)
C10—S1	1.7206 (12)	C28—O4	1.4598 (15)
C11—C12	1.5434 (19)	C28—C31	1.510(2)
C11—H11A	0.9900	C28—C32	1.5196 (19)
C11—H11B	0.9900	С29—Н29А	0.9800
C12—C13	1.5204 (19)	С29—Н29В	0.9800
C12—H12A	0.9900	С29—Н29С	0.9800
C12—H12B	0.9900	C30—H30A	0.9800
C13—C14	1.524 (2)	C30—H30B	0.9800
C13—H13A	0.9900	C30—H30C	0.9800
C13—H13B	0.9900	C31—H31A	0.9800
C14—C15	1.522 (2)	C31—H31B	0.9800
C14—H14A	0.9900	C31—H31C	0.9800
C14—H14B	0.9900	С32—Н32А	0.9800
C15—C16	1.518 (2)	C32—H32B	0.9800
C15—H15A	0.9900	С32—Н32С	0.9800
C15—H15B	0.9900		
O2—B1—O1	114.00 (11)	C18—C17—C10	128.00 (11)
O2—B1—C7	124.42 (11)	C18—C17—S2	111.58 (9)
O1—B1—C7	121.59 (11)	C10—C17—S2	120.36 (8)
O4—B2—O3	114.12 (11)	C17—C18—C19	111.02 (11)
O4—B2—C20	121.21 (11)	C17—C18—C21	125.73 (11)
O3—B2—C20	124.66 (11)	C19—C18—C21	123.22 (10)
C2—C1—H1A	109.5	C20—C19—C18	114.97 (11)
C2—C1—H1B	109.5	С20—С19—Н19	122.5
H1A—C1—H1B	109.5	C18—C19—H19	122.5
C2—C1—H1C	109.5	С19—С20—В2	125.70 (11)
H1A—C1—H1C	109.5	C19—C20—S2	109.71 (9)
H1B—C1—H1C	109.5	B2—C20—S2	124.59 (9)
01	108.66 (10)	C18—C21—C22	112.57 (10)
01-C2-C3	106.41 (11)	C18—C21—H21A	109.1
C1-C2-C3	110.22 (11)	C22—C21—H21A	109.1
01	102.00 (9)	C18—C21—H21B	109.1
C1-C2-C5	115.42 (12)	C22—C21—H21B	109.1
C_{3} — C_{2} — C_{5}	113.32 (11)	H21A—C21—H21B	107.8
C2—C3—H3A	109 5	$C_{21} - C_{22} - C_{23}$	113 81 (11)
$C_2 = C_3 = H_3B$	109.5	C21—C22—H22A	108.8
H_{3A} C_{3} H_{3B}	109.5	C^{23} C^{22} H^{22A}	108.8
$C^2 - C^3 - H^3C$	109.5	C_{21} C_{22} H_{22B}	108.8
H_{3A} $-C_{3}$ $-H_{3C}$	109.5	C_{23} C_{22} H_{22B}	108.8
H3B-C3-H3C	109.5	$H_{22}^{2} = H_{22}^{2} = H_{$	107.7
C5—C4—H4A	109.5	C_{24} C_{23} C_{22}	114,17 (12)
C5—C4—H4B	109.5	C24—C23—H23A	108 7
H4A—C4—H4B	109.5	C22—C23—H23A	108.7
C5-C4-H4C	109.5	C24—C23—H23B	108 7
H4A—C4—H4C	109.5	C22—C23—H23B	108.7

H4B—C4—H4C	109.5	H23A—C23—H23B	107.6
O2—C5—C6	108.64 (10)	C23—C24—C25	113.38 (12)
O2—C5—C4	106.51 (10)	C23—C24—H24A	108.9
C6—C5—C4	110.27 (11)	C25—C24—H24A	108.9
O2—C5—C2	102.49 (9)	C23—C24—H24B	108.9
C6—C5—C2	114.94 (10)	C25—C24—H24B	108.9
C4—C5—C2	113.27 (11)	H24A—C24—H24B	107.7
С5—С6—Н6А	109.5	C26—C25—C24	113.00 (12)
С5—С6—Н6В	109.5	С26—С25—Н25А	109.0
H6A—C6—H6B	109.5	С24—С25—Н25А	109.0
С5—С6—Н6С	109.5	C26—C25—H25B	109.0
H6A—C6—H6C	109.5	С24—С25—Н25В	109.0
H6B—C6—H6C	109.5	H25A—C25—H25B	107.8
C8—C7—B1	130.43 (11)	С25—С26—Н26А	109.5
C8—C7—S1	109.62 (9)	C25—C26—H26B	109.5
B1—C7—S1	119.94 (9)	H26A—C26—H26B	109.5
C7—C8—C9	114.77 (11)	С25—С26—Н26С	109.5
С7—С8—Н8	122.6	H26A—C26—H26C	109.5
С9—С8—Н8	122.6	H26B—C26—H26C	109.5
С10—С9—С8	111.10 (10)	O3—C27—C29	109.17 (12)
C10—C9—C11	123.51 (11)	O3—C27—C30	105.59 (11)
C8—C9—C11	125.30 (11)	C29—C27—C30	110.59 (14)
C9—C10—C17	128.56 (10)	O3—C27—C28	103.03 (10)
C9—C10—S1	111.59 (9)	C29—C27—C28	115.76 (12)
C17—C10—S1	119.84 (8)	C30—C27—C28	111.90 (12)
C9—C11—C12	111.84 (10)	O4—C28—C31	108.74 (11)
C9—C11—H11A	109.2	O4—C28—C32	106.37 (11)
C12—C11—H11A	109.2	C31—C28—C32	110.03 (12)
C9—C11—H11B	109.2	O4—C28—C27	102.50 (9)
C12—C11—H11B	109.2	C31—C28—C27	115.23 (12)
H11A—C11—H11B	107.9	C32—C28—C27	113.25 (12)
C13—C12—C11	113.99 (11)	С27—С29—Н29А	109.5
C13—C12—H12A	108.8	С27—С29—Н29В	109.5
C11—C12—H12A	108.8	H29A—C29—H29B	109.5
C13—C12—H12B	108.8	С27—С29—Н29С	109.5
C11—C12—H12B	108.8	H29A—C29—H29C	109.5
H12A—C12—H12B	107.6	H29B—C29—H29C	109.5
C12—C13—C14	113.64 (12)	С27—С30—Н30А	109.5
С12—С13—Н13А	108.8	С27—С30—Н30В	109.5
C14—C13—H13A	108.8	H30A-C30-H30B	109.5
C12—C13—H13B	108.8	С27—С30—Н30С	109.5
C14—C13—H13B	108.8	H30A—C30—H30C	109.5
H13A—C13—H13B	107.7	H30B—C30—H30C	109.5
C15—C14—C13	113.96 (13)	C28—C31—H31A	109.5
C15—C14—H14A	108.8	C28—C31—H31B	109.5
C13—C14—H14A	108.8	H31A—C31—H31B	109.5
C15—C14—H14B	108.8	C28—C31—H31C	109.5
C13—C14—H14B	108.8	H31A—C31—H31C	109.5

H14A - C14 - H14B	107.7	H31B_C31_H31C	109 5
$C_{16} - C_{15} - C_{14}$	113 33 (14)	$C_{28} = C_{32} = H_{32} \Delta$	109.5
$C_{16} = C_{15} = C_{14}$	108.0	$C_{28} = C_{32} = H_{32R}$	109.5
$C_{10} = C_{15} = H_{15A}$	108.9	$H_{32A} = C_{32} = H_{32B}$	109.5
$C_{14} = C_{15} = M_{15}R$	108.9	1132A - C32 - 1152B	109.5
$C_{10} = C_{15} = H_{15} B$	108.9	$H_{224} = C_{22} = H_{22C}$	109.5
H_{15} H_{15} H_{15}	106.9	$H_{22}^{} C_{22}^{} H_{22}^{} H_{22}^{} C_{22}^{} H_{22}^{} H_{2$	109.5
C15 C16 U16A	107.7	$H_{22} = C_{22} = H_{22} = H$	109.3
C15 - C16 - H16A	109.5	BI = OI = C2	100.85(9)
	109.5	BI-02-C3	106.36 (9)
HI6A - CI6 - HI6B	109.5	$B_2 = 0_3 = 0_2 / 0_2 / 0_2 = 0_2 / 0_2 / 0_2 / 0_2 = 0_2 / 0_2 / 0_2 / 0_2 = 0_2 / 0_2 / 0_2 / 0_2 $	106.50 (10)
C15—C16—H16C	109.5	B2	107.21 (10)
H16A—C16—H16C	109.5		92.91 (6)
H16B—C16—H16C	109.5	C20—S2—C17	92.72 (6)
01 02 05 02	29.20 (11)	C17 C18 C21 C22	121 22 (14)
01 - 02 - 03 - 02	-28.30(11)	C17 - C18 - C21 - C22	-121.33(14)
C1 = C2 = C3 = O2	-145.90(11)	C19 - C18 - C21 - C22	30.35 (10)
C3-C2-C5-02	85.67 (12)	C18—C21—C22—C23	-167.13 (12)
01	-145.95 (11)	C21—C22—C23—C24	-66.75 (18)
C1—C2—C5—C6	96.45 (14)	C22—C23—C24—C25	-177.15 (12)
C3—C2—C5—C6	-31.98 (15)	C23—C24—C25—C26	-71.53 (17)
O1—C2—C5—C4	86.04 (12)	O3—C27—C28—O4	25.36 (13)
C1—C2—C5—C4	-31.56 (15)	C29—C27—C28—O4	144.44 (12)
C3—C2—C5—C4	-159.99 (11)	C30—C27—C28—O4	-87.62 (13)
O2—B1—C7—C8	3.4 (2)	O3—C27—C28—C31	143.28 (12)
O1—B1—C7—C8	-176.39 (12)	C29—C27—C28—C31	-97.64 (16)
O2—B1—C7—S1	-177.69 (10)	C30-C27-C28-C31	30.30 (17)
O1—B1—C7—S1	2.55 (17)	O3—C27—C28—C32	-88.81 (13)
B1—C7—C8—C9	178.53 (12)	C29—C27—C28—C32	30.27 (17)
S1—C7—C8—C9	-0.50 (14)	C30—C27—C28—C32	158.21 (12)
C7—C8—C9—C10	0.85 (15)	O2—B1—O1—C2	-9.18 (14)
C7—C8—C9—C11	-175.73 (11)	C7—B1—O1—C2	170.61 (11)
C8—C9—C10—C17	177.90 (11)	C1-C2-O1-B1	145.43 (12)
C11—C9—C10—C17	-5.45 (19)	C3—C2—O1—B1	-95.90 (12)
C8—C9—C10—S1	-0.80(13)	C5—C2—O1—B1	23.08 (12)
C11—C9—C10—S1	175.85 (9)	O1—B1—O2—C5	-10.37(14)
C10-C9-C11-C12	-77.10(15)	C7—B1—O2—C5	169.85 (11)
C8-C9-C11-C12	99.07 (14)	C6-C5-O2-B1	145.86 (11)
C9-C11-C12-C13	-72.44(14)	C4-C5-O2-B1	-95 37 (12)
$C_{11} - C_{12} - C_{13} - C_{14}$	169 15 (11)	$C_{2} = C_{5} = O_{2} = B_{1}$	23 82 (12)
C12 - C13 - C14 - C15	165 11 (12)	$04 - B^2 - 0^3 - C^{27}$	9.09(15)
$C_{12} = C_{13} = C_{14} = C_{15} = C_{16}$	176.88 (13)	C_{20} B_{2} C_{27} C_{27}	-17244(12)
$C_{10}^{$	-68.23(18)	$C_{20} = D_{2} = 03 = C_{27}$	-14477(13)
$C_{10} = C_{10} = C_{17} = C_{18}$	110.28(13)	C_{2}^{3} C_{2}^{7} C_{3}^{3} C_{2}^{3} C_{2	144.77(13)
C_{0} C_{10} C_{17} C_{10} C_{10} C_{17} C_{20}	114 69 (12)	$C_{20} = C_{27} = C_{3} = B_{2}$	-21 22 (13)
$S_{1} = C_{10} = C_{17} = S_{2}$	-66 71 (12)	$\bigcirc 20 \bigcirc 27 \bigcirc 03 \bigcirc 04 \bigcirc 020$	21.22(13) 8 41 (15)
51 - 010 - 017 - 52	-177.50(11)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.41(13) -170 12(11)
$C_{10} - C_{17} - C_{10} - C_{19}$	1/7.30(11) 0.21(12)	$C_{20} = D_2 = - U_4 = - U_{20}$	1/0.13(11) 1/2.22(12)
52-01/-018-019	-0.21(13)	$C_{22} = C_{28} = C_{4} = B_{2}$	-143.23(12)
C10—C17—C18—C21	0.4 (2)	C32—C28—O4—B2	98.31 (13)

supporting information

S2—C17—C18—C21	$177.72 (10) \\ 0.33 (15) \\ -177.65 (11) \\ 179.11 (11) \\ -0.30 (14) \\ 15.00 (19) \\ -163.37 (13) \\ -165.68 (10) \\ 15.01 (12) \\ -165.61 (12) \\ $	C27—C28—O4—B2	-20.80 (13)
C17—C18—C19—C20		C9—C10—S1—C7	0.46 (9)
C21—C18—C19—C20		C17—C10—S1—C7	-178.37 (10)
C18—C19—C20—B2		C8—C7—S1—C10	0.02 (10)
C18—C19—C20—S2		B1—C7—S1—C10	-179.12 (10)
O4—B2—C20—C19		C19—C20—S2—C17	0.15 (10)
O3—B2—C20—C19		B2—C20—S2—C17	-179.27 (11)
O4—B2—C20—S2		C18—C17—S2—C20	0.04 (9)
O4—B2—C20—S2 O3—B2—C20—S2	15.95 (18)	C10-C17-S2-C20	177.57 (10)

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	Н…А	D···A	<i>D</i> —H··· <i>A</i>
C30—H30C…O1 ⁱ	0.98	2.53	3.2984 (18)	136

Symmetry code: (i) x-1, y, z.