Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

rac-*N*,*N*′-Dimethyl-*N*,*N*′-(1,1′-binaphthyl-2,2′-diyl)diformamide

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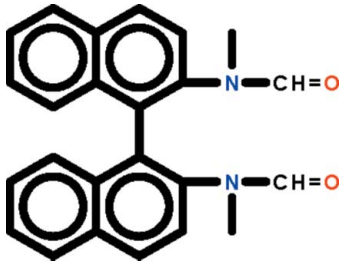
Received 2 November 2011; accepted 3 November 2011

Key indicators: single-crystal X-ray study; T = 113 K; mean σ(C–C) = 0.003 Å; R factor = 0.038; wR factor = 0.101; data-to-parameter ratio = 8.5.

The molecule of the title compound, $C_{24}H_{20}N_2O_2$, lies on a twofold rotation axis that relates one 2-(*N*-methylformamido)naphthyl unit to the other. The *N*-methylformamido substituent is twisted by 54.9 (1)° with respect to the naphthalene fused-ring system; the two fused-ring systems are themselves twisted by 70.3 (1)°.

Related literature

For the synthesis of 2,2′-bis(methylamino)-1,1′-binaphthyl, see: Miyano *et al.* (1984).



Experimental

Crystal data

$C_{24}H_{20}N_2O_2$
M_r = 368.42
Tetragonal, $I4_1$
a = 11.6548 (12) Å
c = 13.9171 (15) Å
V = 1890.4 (3) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.08 mm^{-1}
T = 113 K
0.34 × 0.20 × 0.16 mm

Data collection

Rigaku Saturn CCD area-detector diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku/MSC, 2005)
T_{min} = 0.972, T_{max} = 0.987
11169 measured reflections
1091 independent reflections
1051 reflections with $I > 2\sigma(I)$
R_{int} = 0.044

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.038
$wR(F^2)$ = 0.101
S = 1.07
1091 reflections
129 parameters
1 restraint
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.20 e Å^{-3}
$\Delta\rho_{min}$ = −0.14 e Å^{-3}

Data collection: *CrystalClear* (Rigaku/MSC, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank Xinjiang Normal University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5377).

References

Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
Miyano, S., Nawa, M., Mori, A. & Hashimoto, H. (1984). *Bull. Chem. Soc. Jpn*, **57**, 2171–2176.
Rigaku/MSC (2005). *CrystalClear*. MSC, The Woodlands, Texas, USA, and Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

supporting information

Acta Cryst. (2011). E**67**, o3227 [https://doi.org/10.1107/S160053681104640X]

rac-*N*,*N*′-Dimethyl-*N*,*N*′-(1,1′-binaphthyl-2,2′-diyl)diformamide

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S1. Comment

We report here the methylation of commercially available *racemic* 2,2′-diamino-1,1′-binaphthyl to yield 2,2′-bis(methylamino)-1,1′-binaphthyl. The compound should react with triethyl orthoformate to yield an imidzolium salt. The reaction with ethyl formate, an unintended similar-sounding reagent, gave 2,2′-bis(*N*-methyformido)-1,1′-binaphthyl (Scheme I) instead.

The $C_{24}H_{20}N_2O_2$ molecules lies on a twofold rotation axis that relates one *N*-methylformamido)-1-naphthyl moiety to the other (Fig. 1). The *N*-methylformamido substituent is twisted by 54.9 (1) ° with respect to the naphathalene fused-ring; the two fused-rings are themselves twisted by 70.3 (1) °.

S2. Experimental

2,2′-Bis(methylamino)-1,1′-binaphthyl was prepared by treatment of *racemic* 2,2′-diamino-1,1′-binaphthyl (purchased from Aldrich Chemical Company) with ethyl chloroformate in benzene in the presence of pyridine, followed by reduction with lithium aluminium hydride in tetrahydrofuran (Miyano *et al.*, 1984). An ethyl formate solution (10 ml) of 2,2′-bis(methylamino)-1,1′-binaphthyl (156 mg, 0.50 mmol) was heated at 327 K for 10 h. The solvent was removed under reduced pressure and the residue was purified by column chromatography (eluent: ethyl acetate/petroleum ether 10/1) to give the title compound (158 mg, 86%) as a white solid. Crystals were obtained upon recrystallized from a dichloromethane-hexane mixture.

S3. Refinement

Carbon-bound H-atoms were placed in calculated positions (C–H 0.95–0.98 Å) and were included in the refinement in the riding model approximation, with U_{iso}(H) set to 1.2–1.5U_{eq}(C). In the absence of heavy atoms, 1000 Friedel pairs were merged.

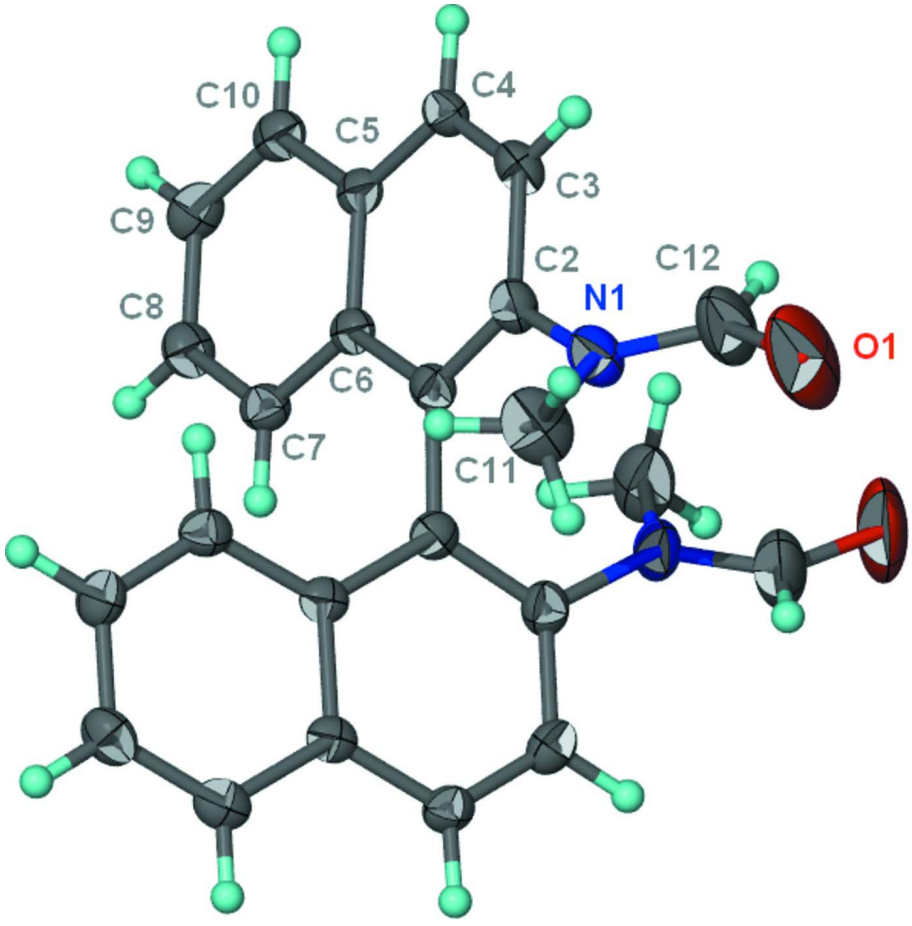

Figure 1

Thermal ellipsoid plot (Barbour, 2001) of $C_{24}H_{20}N_2O_2$ at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius. The unlabeled atoms are related to the labeled one by 2 - *x*, -*y*, *z*.

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Crystal data

$C_{24}H_{20}N_2O_2$
M_r = 368.42
Tetragonal, $I4_1$
Hall symbol: I 4bw
a = 11.6548 (12) Å
c = 13.9171 (15) Å
V = 1890.4 (3) Å^3
Z = 4
F(000) = 776
D_x = 1.294 Mg m^{-3}
Mo $K\alpha$ radiation, λ = 0.71070 Å
Cell parameters from 2958 reflections
θ = 2.3–27.1°
μ = 0.08 mm^{-1}
T = 113 K
Prism, colorless
0.34 × 0.20 × 0.16 mm

Data collection

Rigaku Saturn CCD area-detector diffractometer
Radiation source: rotating anode
Confocal monochromator
Detector resolution: 7.31 pixels mm^{-1}
ω and φ scans
Absorption correction: multi-scan (*CrystalClear*; Rigaku/MSC, 2005)
T_{min} = 0.972, T_{max} = 0.987
11169 measured reflections
1091 independent reflections
1051 reflections with $I > 2\sigma(I)$

R_{int} = 0.044
θ_{max} = 27.1°, θ_{min} = 2.3°
h = −14→14
k = −14→14
l = −17→17

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.038
$wR(F^2)$ = 0.101
S = 1.07
1091 reflections
129 parameters
1 restraint
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0702P)^2 + 0.3273P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max}$ = 0.001
$\Delta\rho_{max}$ = 0.20 e Å^{-3}
$\Delta\rho_{min}$ = −0.14 e Å^{-3}
Extinction correction: *SHELXL97* (Sheldrick, 2008), $Fc^*=kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Extinction coefficient: 0.012 (3)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	U_{iso}*/U_{eq}
O1	0.8311 (3)	−0.0321 (2)	0.33342 (16)	0.0772 (9)
N1	0.85923 (17)	0.00005 (15)	0.17381 (13)	0.0284 (5)
C1	0.96952 (16)	0.05699 (16)	0.03106 (14)	0.0187 (4)
C2	0.89065 (18)	0.08195 (17)	0.10211 (15)	0.0215 (4)
C3	0.83669 (17)	0.19135 (17)	0.10642 (15)	0.0220 (4)
H3	0.7816	0.2067	0.1551	0.026*
C4	0.86360 (17)	0.27432 (16)	0.04105 (15)	0.0199 (4)
H4	0.8286	0.3477	0.0459	0.024*
C5	0.94283 (16)	0.25287 (15)	−0.03388 (14)	0.0176 (4)
C6	0.99650 (16)	0.14242 (15)	−0.03944 (14)	0.0172 (4)
C7	1.07543 (16)	0.12335 (17)	−0.11571 (15)	0.0204 (4)
H7	1.1138	0.0516	−0.1202	0.024*
C8	1.09713 (17)	0.20650 (18)	−0.18289 (15)	0.0229 (4)
H8	1.1490	0.1911	−0.2339	0.028*
C9	1.04310 (17)	0.31514 (18)	−0.17716 (16)	0.0241 (5)
H9	1.0591	0.3724	−0.2239	0.029*
C10	0.96813 (17)	0.33714 (17)	−0.10433 (16)	0.0222 (4)
H10	0.9322	0.4102	−0.1007	0.027*
C11	0.8095 (2)	−0.11084 (19)	0.1472 (2)	0.0391 (6)
H11A	0.8619	−0.1726	0.1663	0.059*
H11B	0.7976	−0.1134	0.0775	0.059*
H11C	0.7358	−0.1208	0.1800	0.059*
C12	0.8630 (3)	0.0295 (2)	0.26773 (19)	0.0475 (8)
H12	0.8926	0.1030	0.2836	0.057*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.149 (3)	0.0543 (14)	0.0279 (12)	0.0017 (15)	0.0288 (14)	0.0093 (10)

N1	0.0414 (11)	0.0224 (9)	0.0214 (10)	−0.0013 (8)	0.0095 (8)	0.0025 (7)
C1	0.0213 (9)	0.0166 (9)	0.0182 (9)	−0.0009 (7)	−0.0013 (7)	−0.0022 (7)
C2	0.0268 (10)	0.0200 (9)	0.0179 (9)	−0.0023 (8)	0.0009 (8)	−0.0008 (8)
C3	0.0232 (10)	0.0226 (10)	0.0203 (9)	−0.0009 (7)	0.0022 (8)	−0.0053 (8)
C4	0.0229 (9)	0.0173 (9)	0.0196 (10)	0.0014 (7)	−0.0029 (8)	−0.0049 (7)
C5	0.0174 (9)	0.0166 (9)	0.0190 (9)	−0.0011 (7)	−0.0042 (7)	−0.0016 (8)
C6	0.0172 (9)	0.0163 (9)	0.0179 (9)	−0.0010 (7)	−0.0035 (7)	−0.0020 (7)
C7	0.0197 (9)	0.0195 (9)	0.0220 (10)	0.0011 (7)	−0.0021 (8)	−0.0013 (8)
C8	0.0219 (9)	0.0255 (10)	0.0215 (10)	−0.0019 (8)	0.0035 (8)	−0.0003 (8)
C9	0.0253 (10)	0.0211 (10)	0.0259 (11)	−0.0051 (8)	−0.0012 (8)	0.0058 (8)
C10	0.0231 (10)	0.0188 (10)	0.0246 (10)	−0.0011 (7)	−0.0052 (8)	−0.0012 (8)
C11	0.0523 (15)	0.0288 (12)	0.0361 (13)	−0.0120 (10)	0.0130 (12)	0.0041 (10)
C12	0.084 (2)	0.0345 (14)	0.0237 (12)	0.0020 (13)	0.0099 (13)	0.0022 (10)

Geometric parameters (Å, º)

O1—C12	1.220 (4)	C5—C6	1.433 (2)
N1—C12	1.352 (3)	C6—C7	1.422 (3)
N1—C2	1.429 (3)	C7—C8	1.370 (3)
N1—C11	1.464 (3)	C7—H7	0.9500
C1—C2	1.381 (3)	C8—C9	1.416 (3)
C1—C6	1.433 (3)	C8—H8	0.9500
C1—C1^{i}	1.507 (4)	C9—C10	1.363 (3)
C2—C3	1.423 (3)	C9—H9	0.9500
C3—C4	1.364 (3)	C10—H10	0.9500
C3—H3	0.9500	C11—H11A	0.9800
C4—C5	1.415 (3)	C11—H11B	0.9800
C4—H4	0.9500	C11—H11C	0.9800
C5—C10	1.419 (3)	C12—H12	0.9500
C12—N1—C2	119.83 (19)	C8—C7—C6	121.21 (18)
C12—N1—C11	118.8 (2)	C8—C7—H7	119.4
C2—N1—C11	120.98 (19)	C6—C7—H7	119.4
C2—C1—C6	119.34 (16)	C7—C8—C9	120.79 (19)
C2—C1—C1^{i}	119.98 (15)	C7—C8—H8	119.6
C6—C1—C1^{i}	120.61 (15)	C9—C8—H8	119.6
C1—C2—C3	120.87 (18)	C10—C9—C8	119.68 (19)
C1—C2—N1	122.00 (17)	C10—C9—H9	120.2
C3—C2—N1	117.12 (18)	C8—C9—H9	120.2
C4—C3—C2	120.35 (18)	C9—C10—C5	121.14 (18)
C4—C3—H3	119.8	C9—C10—H10	119.4
C2—C3—H3	119.8	C5—C10—H10	119.4
C3—C4—C5	121.08 (17)	N1—C11—H11A	109.5
C3—C4—H4	119.5	N1—C11—H11B	109.5
C5—C4—H4	119.5	H11A—C11—H11B	109.5
C4—C5—C10	121.50 (16)	N1—C11—H11C	109.5
C4—C5—C6	118.90 (17)	H11A—C11—H11C	109.5
C10—C5—C6	119.59 (18)	H11B—C11—H11C	109.5

C7—C6—C5	117.58 (17)	O1—C12—N1	124.4 (3)
C7—C6—C1	122.99 (16)	O1—C12—H12	117.8
C5—C6—C1	119.42 (17)	N1—C12—H12	117.8

Symmetry code: (i) $-x+2, -y, z$.