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(Z)-2-(5-Chloro-2-oxoindolin-3-ylidene)-N-phenylhydrazinecarbothioamide

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Key indicators: single-crystal X-ray study; T = 100 K; mean σ (C–C) = 0.002 Å; R factor = 0.040; wR factor = 0.101; data-to-parameter ratio = 19.8.

In the title compound, C₁₅H₁₁ClN₄OS, the dihedral angle between the nine-membered 5-chloroindolin-2-one ring system and the benzene ring is 10.00 (6)°. Intramolecular cvclic N-H···O and C-H···S hydrogen-bonding interactions [graph set S(6)] are present in the N-N-C-N chain between the ring systems. In the crystal, molecules form centrosymmetric cyclic dimers through intermolecular N-H···O hydrogen bonds [graph-set $R_2^2(8)$] and are extended by C-H···Cl interactions into infinite chains which propagate along [100].

Related literature

For related structures, see: Ferrari et al. (2002); Pervez et al. (2010); Ramzan et al. (2010). For various biological activities of Schiff bases, see: Bhandari et al. (2008); Bhardwaj et al. (2010); Pandeva et al. (1999); Sridhar et al. (2002); Survavanshi & Pai (2006). For cytotoxic and anticancer activities of isatin and its derivatives, see: Vine et al. (2009). For bond-length data, see; Allen et al. (1987). For graph-set analysis, see Bernstein et al. (1995).



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Experimental

Crystal data

| C ₁₅ H ₁₁ ClN ₄ OS | V = 1460 |
|---|-----------------|
| $M_r = 330.79$ | Z = 4 |
| Monoclinic, $P2_1/c$ | Μο Κα |
| a = 5.7117 (2) Å | $\mu = 0.41$ |
| b = 17.9510 (7) Å | T = 100 |
| c = 14.2455 (5) Å | 0.53×0 |
| $\beta = 91.262 \ (2)^{\circ}$ | |
| | |

Data collection

Bruker APEXII CCD diffractometer Absorption correction: multi-scan (SADABS; Bruker, 2005) $T_{\min} = 0.812, \ T_{\max} = 0.950$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.040$ $wR(F^2) = 0.101$ S = 1.044183 reflections 211 parameters

Table 1

Hydrogen-bond geometry (Å, °).

| $D - H \cdots A$ | D-H | $H \cdots A$ | $D \cdots A$ | $D - \mathbf{H} \cdots A$ |
|---------------------------|----------|--------------|--------------|---------------------------|
| $N1 - H1N1 \cdots O1^{i}$ | 0.89 (2) | 1.98 (2) | 2.8560 (17) | 171 (2) |
| N3−H1 <i>N</i> 3···O1 | 0.86(2) | 2.08(2) | 2.7563 (16) | 135.9 (19) |
| $C2-H2A\cdots Cl1^{ii}$ | 0.93 | 2.81 | 3.6935 (17) | 158 |
| $C11-H11A\cdots S1$ | 0.93 | 2.61 | 3.2423 (14) | 126 |
| | | | | |

Symmetry codes: (i) -x + 2, -y + 1, -z + 1; (ii) -x + 1, -y + 1, -z + 2.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZS2150).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1-19.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555-1573.
- Bhandari, S. V., Bothara, K. G., Raut, M. K., Patil, A. A., Sarkate, A. P. & Mokale, V. J. (2008). Bioorg. Med. Chem. 16, 1822-1831.
- Bhardwaj, S., Kumar, L., Verma, R. & Sing, U. K. (2010). J. Pharm. Res. 3, 2983-2985
- Bruker (2005). APEX2, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin USA

0.25 (9) Å³ radiation mm^{-} Κ $0.16 \times 0.13 \text{ mm}$

15819 measured reflections

 $R_{\rm int} = 0.044$

refinement $\Delta \rho_{\text{max}} = 0.51 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -0.23 \text{ e} \text{ Å}^{-3}$

4183 independent reflections

3424 reflections with $I > 2\sigma(I)$

H atoms treated by a mixture of

independent and constrained

- Ferrari, M. B., Pelizzi, C., Pelosi, G. & Rodriguez-Argüelles, M. C. (2002). *Polyhedron*, **21**, 2593–2599.
- Pandeya, S. N., Sriram, D., Nath, G. & Clercq, E. De. (1999). Indian J. Pharm. Sci. 61, 358–361.
- Pervez, H., Yaqub, M., Ramzan, M., Tahir, M. N. & Iqbal, M. S. (2010). Acta Cryst. E66, 01609.
- Ramzan, M., Pervez, H., Yaqub, M. & Tahir, M. N. (2010). Acta Cryst. E66, 02387.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Spek, A. L. (2009). Acta Cryst. D65, 148-155.
- Sridhar, S. K., Pandeya, S. N., Stables, J. P. & Ramesh, A. (2002). *Eur. J. Pharm. Sci.* **16**, 129–132.
- Suryavanshi, J. P. & Pai, N. R. (2006). Indian J. Chem. Sect. B, 45, 1227–1230.
- Vine, K. L., Matesic, L., Locke, J. M., Ranson, M. & Skropeta, D. (2009). Anti-Cancer Agents in Med. Chem. 9, 397–414.

supporting information

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(Z)-2-(5-Chloro-2-oxoindolin-3-ylidene)-N-phenylhydrazinecarbothioamide

Amna Qasem Ali, Naser Eltaher Eltayeb, Siang Guan Teoh, Abdussalam Salhin and Hoong-Kun Fun

S1. Comment

Isatin (2,3-dioxindole) is an endogenous compound identified in humans, and its effect has been studied in a variety of systems (Vine *et al.*, 2009). Biological properties of isatin and its derivatives include a range of actions in the brain and offer protection against certain types of infections, such as anti-bacterial (Suryavanshi & Pai, 2006) antifungal, anticonvulsant, anti-HIV (Pandeya *et al.*, 1999), anti-depressant and anti-inflammatory activities (Bhandari *et al.*, 2008). In this paper we describe the single-crystal X-ray diffraction study of the title compound, $C_{15}H_{11}ClN_4OS$.

In this compound (Fig. 1), the dihedral angle between the nine-membered 5-chloroindolin-2-one ring system and benzene ring is 10.00 (6)°. Atoms C8 in the 5-chloroindolin-2-one ring and C10 in the benzene ring are joined by a chain of four atoms (N2/N3/C9/N4) giving a torsion angle 7.47 (19)° while the the torsion angles C8—N2—N3—C9 and C10 —N4—C9—N3 are 173.15 (13)° and -178.02 (13)°, respectively. The essentially planar conformation of the molecule is maintained by cyclic intramolecular N3—H···O1 and C11—H···S1 hydrogen-bonding interactions [graph set *S*(6) (Bernstein *et al.*, 1995)] (Table 1) together with an *S*(5) N3—H···N2 interaction.

In the crystal the molecules form centrosymmetric cyclic dimers through intermolecular N—H···O hydrogen bonds [graph set $R^2_2(8)$] and are extended by C—H···Cl interactions into infinite chains which propagate along [100] (Fig. 2). Weak C—H··· π interactions are also present: C5—H5A··· $Cg3^{iii} = 3.6321$ (18) Å, where $Cg3^{iii}$ is the centroid of the C10—C15 ring [symmetry code: (iii) -*x* + 1, *y* + 1/2, -*z* + 3/2].

S2. Experimental

The Schiff base have been synthesized by refluxing the reaction mixture of hot ethanolic solution (30 ml) of 4-phenyl-3-thiosemicarbazide (0.01 mol) and hot ethanolic solution (30 ml) of 5-chloroisatin (0.01 mol) for 2 h. The precipitate formed during reflux was filtered, washed with cold ethanol and recrystallized from hot ethanol: yield 97%; m.p. 521.4-521.9 K). The orange crystals were grown in acetone-DMF (3:1) by slow evaporation at room temperature.

S3. Refinement

N bound H atoms were located in a difference Fourier map and were refined freely. The remaining H atoms were positioned geometrically and refined using a riding model with C-H = 0.93 Å and $U_{iso}(H) = 1.2U_{eq}(C)$. The highest residual electron density peak (0.510 eÅ⁻³ is located at 0.87 Å from C3 and the deepest hole (-0.228 eÅ⁻³) is located at 0.56 Å from S1.



Figure 1

The molecular structure of the title compound, with 50% probability displacement ellipsoids and the atom-numbering scheme.



Figure 2

The crystal packing of the title compound viewed down the *a* axis. Hydrogen bonds are shown as dashed lines.

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Crystal data

C₁₅H₁₁ClN₄OS $M_r = 330.79$ Monoclinic, $P2_1/c$ Hall symbol: -P 2ybc a = 5.7117 (2) Å b = 17.9510 (7) Å c = 14.2455 (5) Å $\beta = 91.262$ (2)° V = 1460.25 (9) Å³ Z = 4

Data collection

Bruker APEXII CCD diffractometer Radiation source: fine-focus sealed tube Graphite monochromator φ and ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 2005) $T_{\min} = 0.812, T_{\max} = 0.950$

Refinement

Refinement on F^2 Secondary atomLeast-squares matrix: fullmap $R[F^2 > 2\sigma(F^2)] = 0.040$ Hydrogen site loo $wR(F^2) = 0.101$ neighbouring sS = 1.04H atoms treated b4183 reflectionsand constraine211 parameters $w = 1/[\sigma^2(F_o^2) + (w + 1)^2]$ 0 restraintswhere $P = (F_o^2)$ Primary atom site location: structure-invariant $(\Delta/\sigma)_{max} = 0.001$ $d\rho_{max} = 0.51$ e Å

F(000) = 680 $D_x = 1.505 \text{ Mg m}^{-3}$ Melting point = 521.3–521.9 K Mo K α radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 6388 reflections $\theta = 2.3-29.8^{\circ}$ $\mu = 0.41 \text{ mm}^{-1}$ T = 100 KBlock, orange $0.53 \times 0.16 \times 0.13 \text{ mm}$

15819 measured reflections 4183 independent reflections 3424 reflections with $I > 2\sigma(I)$ $R_{int} = 0.044$ $\theta_{max} = 29.9^{\circ}, \theta_{min} = 1.8^{\circ}$ $h = -7 \rightarrow 7$ $k = -16 \rightarrow 25$ $l = -19 \rightarrow 19$

Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H atoms treated by a mixture of independent and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0482P)^2 + 0.4941P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.51$ e Å⁻³ $\Delta\rho_{min} = -0.23$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

| | x | У | Ζ | $U_{ m iso}$ */ $U_{ m eq}$ | |
|------|--------------|--------------|--------------|-----------------------------|--|
| S1 | 0.08445 (7) | 0.29509 (2) | 0.51315 (3) | 0.02175 (10) | |
| C11 | 0.74643 (8) | 0.58625 (3) | 1.01336 (3) | 0.03431 (13) | |
| 01 | 0.71079 (18) | 0.45044 (6) | 0.51237 (8) | 0.0202 (2) | |
| N1 | 0.9178 (2) | 0.52737 (7) | 0.61568 (10) | 0.0192 (3) | |
| N2 | 0.3993 (2) | 0.43359 (7) | 0.68077 (9) | 0.0169 (2) | |
| N3 | 0.3263 (2) | 0.39602 (7) | 0.60374 (9) | 0.0176 (2) | |
| N4 | 0.0298 (2) | 0.34863 (7) | 0.69041 (9) | 0.0162 (2) | |
| C1 | 0.7024 (2) | 0.51529 (8) | 0.74926 (11) | 0.0175 (3) | |
| C2 | 0.6515 (3) | 0.52627 (8) | 0.84293 (11) | 0.0202 (3) | |
| H2A | 0.5189 | 0.5055 | 0.8692 | 0.024* | |
| C3 | 0.8066 (3) | 0.56950 (9) | 0.89592 (11) | 0.0213 (3) | |
| C4 | 1.0055 (3) | 0.60157 (9) | 0.85810 (12) | 0.0226 (3) | |
| H4A | 1.1051 | 0.6301 | 0.8960 | 0.027* | |
| C5 | 1.0559 (3) | 0.59112 (9) | 0.76381 (12) | 0.0218 (3) | |
| H5A | 1.1876 | 0.6125 | 0.7376 | 0.026* | |
| C6 | 0.9031 (2) | 0.54775 (8) | 0.71060 (11) | 0.0177 (3) | |
| C7 | 0.7407 (2) | 0.48082 (8) | 0.58967 (11) | 0.0176 (3) | |
| C8 | 0.5894 (2) | 0.47276 (8) | 0.67398 (10) | 0.0164 (3) | |
| C9 | 0.1396 (2) | 0.34679 (8) | 0.60775 (10) | 0.0163 (3) | |
| C10 | -0.1615 (2) | 0.30657 (8) | 0.72387 (10) | 0.0158 (3) | |
| C11 | -0.3209 (2) | 0.26888 (8) | 0.66549 (11) | 0.0175 (3) | |
| H11A | -0.3036 | 0.2690 | 0.6007 | 0.021* | |
| C12 | -0.5073 (3) | 0.23099 (8) | 0.70597 (12) | 0.0205 (3) | |
| H12A | -0.6127 | 0.2050 | 0.6675 | 0.025* | |
| C13 | -0.5381 (3) | 0.23133 (9) | 0.80179 (12) | 0.0227 (3) | |
| H13A | -0.6635 | 0.2061 | 0.8277 | 0.027* | |
| C14 | -0.3798 (3) | 0.26976 (10) | 0.85915 (12) | 0.0248 (3) | |
| H14A | -0.4001 | 0.2707 | 0.9237 | 0.030* | |
| C15 | -0.1915 (3) | 0.30674 (9) | 0.82049 (11) | 0.0212 (3) | |
| H15A | -0.0849 | 0.3318 | 0.8594 | 0.025* | |
| H1N1 | 1.030 (4) | 0.5395 (14) | 0.5766 (17) | 0.049 (7)* | |
| H1N3 | 0.406 (4) | 0.3978 (11) | 0.5537 (15) | 0.029 (5)* | |
| H1N4 | 0.095 (3) | 0.3761 (11) | 0.7308 (14) | 0.026 (5)* | |
| | | | | | |

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\hat{A}^2)

Atomic displacement parameters (\mathring{A}^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|--------------|------------|--------------|---------------|--------------|---------------|
| S1 | 0.02155 (19) | 0.0258 (2) | 0.01799 (19) | -0.00283 (14) | 0.00246 (14) | -0.00359 (14) |
| Cl1 | 0.0431 (3) | 0.0386 (3) | 0.0214 (2) | -0.01947 (19) | 0.00487 (17) | -0.00397 (16) |
| 01 | 0.0156 (5) | 0.0242 (5) | 0.0208 (5) | 0.0009 (4) | 0.0028 (4) | 0.0007 (4) |
| N1 | 0.0136 (6) | 0.0218 (6) | 0.0225 (7) | -0.0018 (5) | 0.0038 (5) | 0.0018 (5) |
| N2 | 0.0141 (5) | 0.0175 (6) | 0.0192 (6) | 0.0004 (4) | 0.0003 (5) | 0.0011 (5) |
| N3 | 0.0148 (6) | 0.0202 (6) | 0.0178 (6) | -0.0015 (4) | 0.0032 (5) | 0.0001 (5) |
| N4 | 0.0140 (5) | 0.0193 (6) | 0.0154 (6) | -0.0035 (4) | 0.0003 (4) | -0.0013 (5) |
| C1 | 0.0140 (6) | 0.0153 (6) | 0.0233 (8) | -0.0001 (5) | 0.0017 (5) | 0.0022 (5) |
| | | | | | | |

supporting information

| C2 | 0.0174 (7) | 0.0198 (7) | 0.0235 (8) | -0.0035 (5) | 0.0021 (6) | 0.0023 (6) |
|-----|------------|------------|------------|-------------|-------------|-------------|
| C3 | 0.0231 (7) | 0.0199 (7) | 0.0209 (7) | -0.0022 (6) | 0.0007 (6) | 0.0009 (6) |
| C4 | 0.0200 (7) | 0.0200 (7) | 0.0277 (8) | -0.0032 (5) | -0.0023 (6) | 0.0022 (6) |
| C5 | 0.0144 (6) | 0.0222 (7) | 0.0288 (8) | -0.0025 (5) | 0.0025 (6) | 0.0016 (6) |
| C6 | 0.0136 (6) | 0.0166 (6) | 0.0231 (7) | 0.0017 (5) | 0.0033 (5) | 0.0026 (5) |
| C7 | 0.0126 (6) | 0.0182 (7) | 0.0221 (7) | 0.0028 (5) | 0.0027 (5) | 0.0046 (5) |
| C8 | 0.0127 (6) | 0.0179 (7) | 0.0188 (7) | 0.0010 (5) | 0.0018 (5) | 0.0026 (5) |
| C9 | 0.0132 (6) | 0.0174 (6) | 0.0183 (7) | 0.0014 (5) | -0.0007 (5) | 0.0023 (5) |
| C10 | 0.0118 (6) | 0.0165 (6) | 0.0192 (7) | 0.0003 (5) | 0.0002 (5) | 0.0017 (5) |
| C11 | 0.0136 (6) | 0.0188 (7) | 0.0200 (7) | 0.0008 (5) | -0.0017 (5) | -0.0002 (5) |
| C12 | 0.0136 (6) | 0.0180 (7) | 0.0298 (8) | -0.0002 (5) | -0.0025 (6) | 0.0007 (6) |
| C13 | 0.0143 (6) | 0.0245 (8) | 0.0295 (9) | -0.0013 (5) | 0.0026 (6) | 0.0059 (6) |
| C14 | 0.0187 (7) | 0.0364 (9) | 0.0192 (8) | -0.0017 (6) | 0.0023 (6) | 0.0058 (7) |
| C15 | 0.0154 (7) | 0.0301 (8) | 0.0182 (7) | -0.0038 (6) | -0.0011 (5) | 0.0005 (6) |
| | | | | | | |

Geometric parameters (Å, °)

| S1—C9 | 1.6606 (15) | C3—C4 | 1.392 (2) |
|-------------|-------------|-------------|-------------|
| Cl1—C3 | 1.7416 (17) | C4—C5 | 1.393 (2) |
| 01—C7 | 1.2374 (18) | C4—H4A | 0.9300 |
| N1—C7 | 1.3573 (19) | C5—C6 | 1.383 (2) |
| N1—C6 | 1.405 (2) | С5—Н5А | 0.9300 |
| N1—H1N1 | 0.88 (3) | С7—С8 | 1.502 (2) |
| N2—C8 | 1.2992 (18) | C10—C15 | 1.391 (2) |
| N2—N3 | 1.3462 (18) | C10-C11 | 1.395 (2) |
| N3—C9 | 1.3872 (18) | C11—C12 | 1.398 (2) |
| N3—H1N3 | 0.85 (2) | C11—H11A | 0.9300 |
| N4—C9 | 1.3467 (19) | C12—C13 | 1.380 (2) |
| N4—C10 | 1.4190 (18) | C12—H12A | 0.9300 |
| N4—H1N4 | 0.84 (2) | C13—C14 | 1.389 (2) |
| C1—C2 | 1.386 (2) | C13—H13A | 0.9300 |
| C1—C6 | 1.409 (2) | C14—C15 | 1.388 (2) |
| C1—C8 | 1.456 (2) | C14—H14A | 0.9300 |
| C2—C3 | 1.389 (2) | C15—H15A | 0.9300 |
| C2—H2A | 0.9300 | | |
| | | | |
| C7—N1—C6 | 111.28 (13) | O1—C7—N1 | 126.99 (14) |
| C7—N1—H1N1 | 121.6 (16) | O1—C7—C8 | 126.77 (13) |
| C6—N1—H1N1 | 127.0 (16) | N1—C7—C8 | 106.23 (13) |
| C8—N2—N3 | 117.04 (13) | N2—C8—C1 | 125.93 (14) |
| N2—N3—C9 | 120.67 (13) | N2—C8—C7 | 127.47 (14) |
| N2—N3—H1N3 | 120.1 (14) | C1—C8—C7 | 106.56 (12) |
| C9—N3—H1N3 | 118.8 (14) | N4—C9—N3 | 113.18 (13) |
| C9—N4—C10 | 131.11 (13) | N4—C9—S1 | 129.76 (11) |
| C9—N4—H1N4 | 113.9 (14) | N3—C9—S1 | 117.06 (11) |
| C10—N4—H1N4 | 114.6 (14) | C15—C10—C11 | 119.87 (13) |
| C2—C1—C6 | 120.49 (14) | C15—C10—N4 | 116.35 (13) |
| C2—C1—C8 | 133.22 (13) | C11—C10—N4 | 123.73 (13) |

| C6—C1—C8 | 106.28 (13) | C10—C11—C12 | 118.89 (14) |
|--------------|--------------|-----------------|--------------|
| C1—C2—C3 | 117.25 (14) | C10-C11-H11A | 120.6 |
| C1—C2—H2A | 121.4 | C12—C11—H11A | 120.6 |
| C3—C2—H2A | 121.4 | C13—C12—C11 | 121.37 (14) |
| C2—C3—C4 | 122.48 (15) | C13—C12—H12A | 119.3 |
| C2—C3—Cl1 | 118.78 (12) | C11—C12—H12A | 119.3 |
| C4—C3—Cl1 | 118.71 (12) | C12—C13—C14 | 119.25 (14) |
| C3—C4—C5 | 120.36 (15) | C12—C13—H13A | 120.4 |
| C3—C4—H4A | 119.8 | C14—C13—H13A | 120.4 |
| С5—С4—Н4А | 119.8 | C15—C14—C13 | 120.28 (15) |
| C6—C5—C4 | 117.56 (14) | C15—C14—H14A | 119.9 |
| С6—С5—Н5А | 121.2 | C13—C14—H14A | 119.9 |
| С4—С5—Н5А | 121.2 | C14—C15—C10 | 120.34 (14) |
| C5—C6—N1 | 128.57 (14) | C14—C15—H15A | 119.8 |
| C5—C6—C1 | 121.85 (15) | C10—C15—H15A | 119.8 |
| N1C6C1 | 109.57 (13) | | |
| | | | |
| C8—N2—N3—C9 | 173.15 (13) | C6—C1—C8—N2 | 179.26 (14) |
| C6—C1—C2—C3 | -0.6 (2) | C2—C1—C8—C7 | -177.07 (15) |
| C8—C1—C2—C3 | 177.83 (15) | C6—C1—C8—C7 | 1.51 (15) |
| C1—C2—C3—C4 | 0.5 (2) | O1—C7—C8—N2 | -1.1 (2) |
| C1—C2—C3—Cl1 | 178.76 (11) | N1—C7—C8—N2 | 179.62 (14) |
| C2—C3—C4—C5 | 0.1 (2) | O1—C7—C8—C1 | 176.64 (14) |
| Cl1—C3—C4—C5 | -178.24 (12) | N1—C7—C8—C1 | -2.68 (15) |
| C3—C4—C5—C6 | -0.4 (2) | C10—N4—C9—N3 | -178.02 (13) |
| C4—C5—C6—N1 | -178.16 (14) | C10—N4—C9—S1 | 1.7 (2) |
| C4—C5—C6—C1 | 0.3 (2) | N2—N3—C9—N4 | 7.47 (19) |
| C7—N1—C6—C5 | 176.63 (15) | N2—N3—C9—S1 | -172.31 (10) |
| C7—N1—C6—C1 | -1.99 (16) | C9—N4—C10—C15 | 162.39 (15) |
| C2-C1-C6-C5 | 0.2 (2) | C9—N4—C10—C11 | -20.3 (2) |
| C8—C1—C6—C5 | -178.57 (13) | C15-C10-C11-C12 | -0.8 (2) |
| C2-C1-C6-N1 | 178.95 (13) | N4—C10—C11—C12 | -178.04 (13) |
| C8—C1—C6—N1 | 0.15 (15) | C10-C11-C12-C13 | 1.1 (2) |
| C6—N1—C7—O1 | -176.46 (14) | C11—C12—C13—C14 | -0.4 (2) |
| C6—N1—C7—C8 | 2.86 (16) | C12—C13—C14—C15 | -0.6 (2) |
| N3—N2—C8—C1 | -177.80 (13) | C13—C14—C15—C10 | 0.9 (2) |
| N3—N2—C8—C7 | -0.5 (2) | C11—C10—C15—C14 | -0.2 (2) |
| C2—C1—C8—N2 | 0.7 (3) | N4—C10—C15—C14 | 177.23 (14) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H··· <i>A</i> | D—H | Н…А | D····A | <i>D</i> —H··· <i>A</i> |
|------------------------------------|----------|------------|-------------|-------------------------|
| N1—H1 <i>N</i> 1···O1 ⁱ | 0.89 (2) | 1.98 (2) | 2.8560 (17) | 171 (2) |
| N3—H1 <i>N</i> 3…O1 | 0.86 (2) | 2.08 (2) | 2.7563 (16) | 135.9 (19) |
| N4—H1 <i>N</i> 4····N2 | 0.84 (2) | 2.156 (18) | 2.6099 (17) | 113.8 (16) |

| | | | supporting | ; information |
|-------------------------------------|------|------|-------------|---------------|
| C2—H2 <i>A</i> ···C11 ⁱⁱ | 0.93 | 2.81 | 3.6935 (17) | 158 |
| C11—H11A····S1 | 0.93 | 2.61 | 3.2423 (14) | 126 |

Symmetry codes: (i) -*x*+2, -*y*+1, -*z*+1; (ii) -*x*+1, -*y*+1, -*z*+2.