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4,4-Difluoro-8-(4-iodophenyl)-1,3,5,7-tetramethyl-3a-aza-4a-azonia-4-borata-s-indacene

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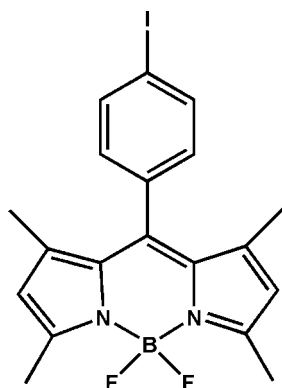
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Key indicators: single-crystal X-ray study; $T = 293\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.008\text{ \AA}$; R factor = 0.062; wR factor = 0.166; data-to-parameter ratio = 14.6.

In the title compound, $\text{C}_{19}\text{H}_{18}\text{BF}_2\text{IN}_2$, which is a boron-dipyrromethene (BODIPY) derivative, the BODIPY mean plane forms dihedral angles of $88.95(4)$ and $78.21(3)^\circ$ with the F/B/F and 4-iodophenyl planes, respectively.

Related literature

For the crystal structures of related boron-dipyrromethene derivatives, see: Zhou (2010); Chen & Jiang (2011); Hinkle *et al.* (2011); Cui *et al.* (2012).



Experimental

Crystal data

$\text{C}_{19}\text{H}_{18}\text{BF}_2\text{IN}_2$
 $M_r = 450.06$
 Monoclinic, $P2_1/c$
 $a = 12.1004(3)\text{ \AA}$
 $b = 8.1992(2)\text{ \AA}$
 $c = 18.0607(4)\text{ \AA}$
 $\beta = 90.577(3)^\circ$

$V = 1791.77(8)\text{ \AA}^3$
 $Z = 4$
 Cu $K\alpha$ radiation
 $\mu = 14.24\text{ mm}^{-1}$
 $T = 293\text{ K}$
 $0.20 \times 0.18 \times 0.16\text{ mm}$

Data collection

Bruker SMART 1000 CCD area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 2007)
 $T_{\min} = 0.102$, $T_{\max} = 0.110$

6829 measured reflections
 3348 independent reflections
 2770 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.044$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.166$
 $S = 1.05$
 3348 reflections

230 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 1.33\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -1.36\text{ e \AA}^{-3}$

Data collection: SMART (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5238).

References

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supporting information

Acta Cryst. (2012). E68, o1302 [https://doi.org/10.1107/S1600536812004072]

4,4-Difluoro-8-(4-iodophenyl)-1,3,5,7-tetramethyl-3a-aza-4a-azonia-4-borata-s-indacene**Yongling Sun****S1. Comment**

In our search for new potential boron-dipyrromethene (BODIPY) fluorescent dyes (Chen & Jiang, 2011), we have obtained the title compound (I). Herewith we report its crystal structure.

In (I) (Fig. 1), all bond lengths and angles are normal in relation to those observed in the related boron-dipyrromethene derivatives (Zhou, 2010; Hinkle *et al.*, 2011; Cui *et al.*, 2012). The C—C and C—N bond lengths within BODIPY fragment are in the range of 1.371–1.422 and 1.337–1.402 Å, respectively, without any clear distinction between single and double bonds, indicating strongly delocalized π -system. The C₉BN₂ fragment is essentially flat, with the maximum deviation from the least-squares mean plane of 0.065 (3) Å. The dihedral angle between the F—B—F plane and the BODIPY mean plane is 88.95 (4)°. Due to the presence of two methyl groups attached to C1 and C7 atoms in BODIPY fragment, the dihedral angles between the BODIPY mean plane and 4-iodophenyl fragment is 78.21 (3)°.

S2. Experimental

To the mixture of *p*-iodobenzyldehyde (231 mg, 1 mmol) and 2,4-dimethylpyrrole (190 mg, 2.00 mmol) dissolved in CH₂Cl₂ (150 ml), one drop of TFA was added. After the resulting mixture was stirred for five hours at room temperature under N₂ atmosphere, a solution of DDQ (227 mg, 1 mmol) in CH₂Cl₂ (60 ml) was added and the reaction mixture was further stirred for another 10 min. After the addition of *N,N*-diisopropylethylamine (DIEA) (2 ml) into the mixture for 15 min, the BF₃—OEt₂ (2.0 ml) was added into the reaction mixture and stirring was continued for another 30 min. The resulting mixture was evaporated, and the residue was chromatographed on a silica gel column using CH₂Cl₂ as eluent. Repeated chromatography followed by recrystallization from CH₂Cl₂ and MeOH gave the target compound as red crystals. Yield: 130 mg, 28.9%. Anal. for C₁₉H₁₈BF₂IN₂: Calc. C, 50.70; H, 4.03; N, 6.22; Found: C, 50.42; H, 4.17; N, 6.31%. The No. of CCDC: 863227.

S3. Refinement

All H atoms were placed in geometrically idealized positions and treated as riding on their parent atoms, with C—H = 0.93 - 0.96 Å and $U_{\text{iso}}(\text{H}) > 1.2\text{--}1.5 U_{\text{eq}}(\text{C})$.

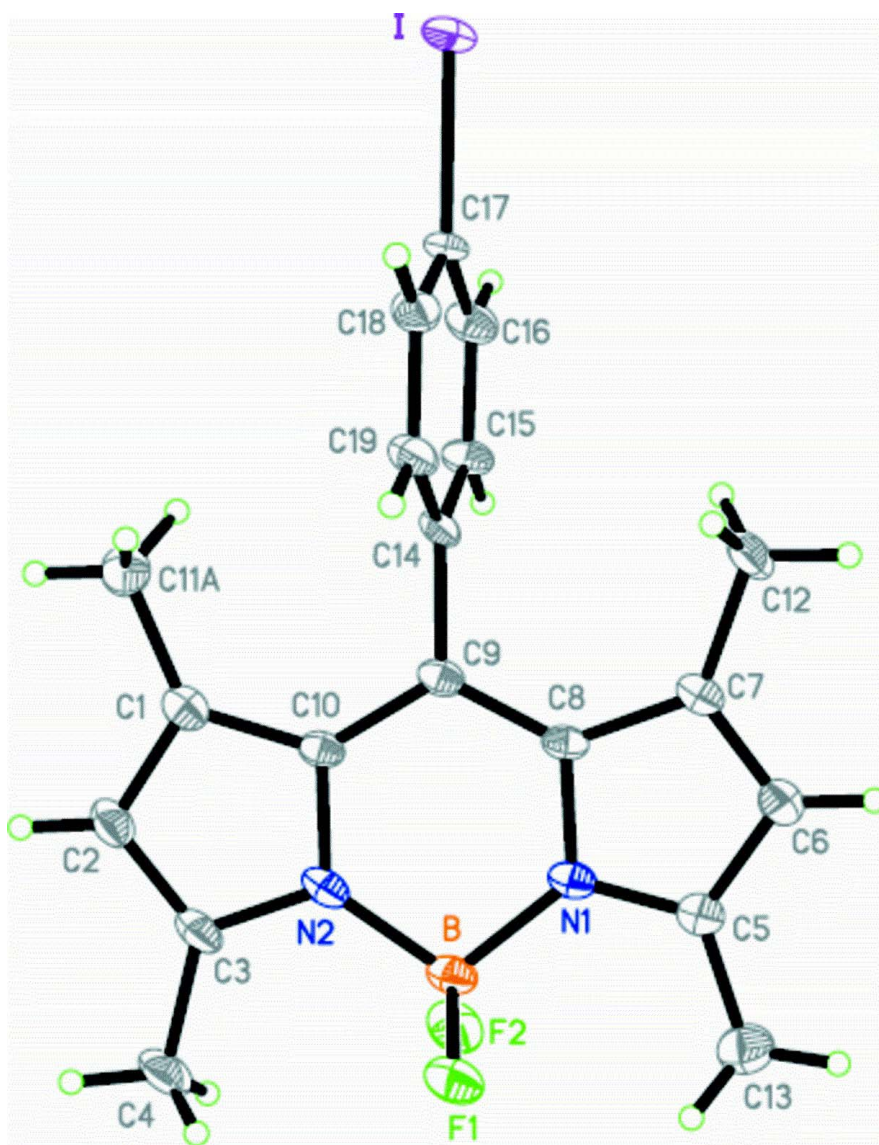


Figure 1

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 30% probability level.

2,2-difluoro-8-(4-iodophenyl)-4,6,10,12-tetramethyl-1 λ^5 ,3-diaza-2 λ^4 - boratricyclo[7.3.0.0^(3,7)]dodeca-1(12),4,6,8,10-pentaen-1-ylum

Crystal data

$C_{19}H_{18}BF_2IN_2$

$M_r = 450.06$

Monoclinic, $P2_1/c$

$a = 12.1004 (3) \text{ \AA}$

$b = 8.1992 (2) \text{ \AA}$

$c = 18.0607 (4) \text{ \AA}$

$\beta = 90.577 (3)^\circ$

$V = 1791.77 (8) \text{ \AA}^3$

$Z = 4$

$F(000) = 888$

$D_x = 1.668 \text{ Mg m}^{-3}$

Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$

Cell parameters from 3440 reflections

$\mu = 14.24 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Block, red

$0.20 \times 0.18 \times 0.16 \text{ mm}$

Data collection

Bruker SMART 1000 CCD area-detector
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 phi and ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2007)
 $T_{\min} = 0.102$, $T_{\max} = 0.110$

6829 measured reflections
 3348 independent reflections
 2770 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.044$
 $\theta_{\max} = 70.8^\circ$, $\theta_{\min} = 4.9^\circ$
 $h = -9 \rightarrow 14$
 $k = -10 \rightarrow 8$
 $l = -21 \rightarrow 21$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.062$
 $wR(F^2) = 0.166$
 $S = 1.05$
 3348 reflections
 230 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0917P)^2 + 0.3111P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 1.33 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -1.36 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
I	0.46758 (4)	−0.22865 (6)	0.54374 (2)	0.0515 (2)
F1	1.0026 (3)	0.3904 (5)	0.8798 (2)	0.0540 (10)
N2	0.8395 (4)	0.2246 (6)	0.8934 (3)	0.0334 (11)
F2	0.8499 (4)	0.5070 (5)	0.9279 (2)	0.0569 (11)
N1	0.8479 (3)	0.4411 (6)	0.7983 (2)	0.0305 (10)
C15	0.7292 (4)	0.0148 (7)	0.6658 (3)	0.0345 (13)
H6	0.8056	0.0233	0.6619	0.041*
C2	0.8031 (5)	−0.0117 (8)	0.9496 (3)	0.0401 (14)
H7	0.8028	−0.0933	0.9853	0.048*
C9	0.7415 (4)	0.1937 (7)	0.7768 (3)	0.0278 (11)
C1	0.7510 (4)	−0.0199 (7)	0.8812 (3)	0.0332 (12)
C14	0.6760 (4)	0.0933 (7)	0.7238 (3)	0.0280 (11)
C10	0.7747 (4)	0.1305 (7)	0.8456 (3)	0.0272 (11)
C8	0.7754 (4)	0.3485 (7)	0.7541 (3)	0.0273 (11)
C7	0.7538 (4)	0.4399 (7)	0.6886 (3)	0.0325 (12)
C13	0.9491 (6)	0.7053 (9)	0.7908 (4)	0.0510 (17)

H15B	1.0239	0.6742	0.7807	0.077*
H15C	0.9397	0.7166	0.8432	0.077*
H15A	0.9331	0.8074	0.7669	0.077*
C6	0.8157 (5)	0.5794 (8)	0.6943 (3)	0.0390 (13)
H16	0.8192	0.6614	0.6588	0.047*
C4	0.9226 (6)	0.2003 (10)	1.0198 (4)	0.0522 (18)
H17C	0.8915	0.3008	1.0371	0.078*
H17A	0.9972	0.2187	1.0043	0.078*
H17B	0.9223	0.1214	1.0591	0.078*
C17	0.5569 (5)	−0.0896 (7)	0.6206 (3)	0.0357 (13)
C5	0.8727 (5)	0.5783 (8)	0.7620 (3)	0.0354 (13)
C18	0.5021 (5)	−0.0124 (8)	0.6784 (3)	0.0370 (13)
H20	0.4259	−0.0218	0.6826	0.044*
C16	0.6705 (5)	−0.0756 (8)	0.6139 (3)	0.0389 (14)
H21	0.7069	−0.1262	0.5750	0.047*
C19	0.5624 (5)	0.0783 (8)	0.7293 (3)	0.0362 (13)
H22	0.5260	0.1299	0.7679	0.043*
C3	0.8559 (5)	0.1382 (8)	0.9563 (3)	0.0384 (14)
B	0.8873 (5)	0.3956 (9)	0.8772 (4)	0.0346 (14)
C12	0.6745 (5)	0.4019 (9)	0.6266 (3)	0.0465 (16)
H25A	0.6999	0.3077	0.6002	0.070*
H25C	0.6703	0.4933	0.5935	0.070*
H25B	0.6027	0.3804	0.6465	0.070*
C11A	0.6824 (5)	−0.1606 (8)	0.8552 (4)	0.0419 (14)
H1AB	0.6056	−0.1314	0.8567	0.063*
H1AC	0.6955	−0.2529	0.8868	0.063*
H1AA	0.7020	−0.1878	0.8053	0.063*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
I	0.0601 (4)	0.0609 (4)	0.0331 (3)	−0.0223 (2)	−0.0201 (2)	−0.00197 (17)
F1	0.0302 (17)	0.081 (3)	0.050 (2)	−0.0144 (19)	−0.0154 (16)	0.009 (2)
N2	0.031 (2)	0.048 (3)	0.021 (2)	0.001 (2)	−0.0043 (19)	−0.003 (2)
F2	0.082 (3)	0.057 (2)	0.0314 (18)	−0.001 (2)	−0.0013 (19)	−0.0118 (17)
N1	0.025 (2)	0.039 (3)	0.027 (2)	−0.001 (2)	−0.0030 (17)	−0.006 (2)
C15	0.023 (2)	0.047 (3)	0.034 (3)	0.000 (2)	−0.006 (2)	−0.009 (3)
C2	0.043 (3)	0.048 (4)	0.029 (3)	0.003 (3)	−0.009 (2)	0.008 (3)
C9	0.021 (2)	0.037 (3)	0.026 (3)	0.002 (2)	−0.002 (2)	−0.006 (2)
C1	0.028 (3)	0.042 (3)	0.030 (3)	0.004 (3)	0.000 (2)	−0.001 (2)
C14	0.027 (2)	0.040 (3)	0.017 (2)	−0.003 (2)	−0.0031 (19)	0.001 (2)
C10	0.020 (2)	0.035 (3)	0.027 (2)	0.001 (2)	−0.0028 (19)	−0.005 (2)
C8	0.021 (2)	0.033 (3)	0.028 (2)	0.000 (2)	−0.0051 (19)	−0.007 (2)
C7	0.029 (3)	0.043 (3)	0.026 (2)	0.002 (3)	−0.003 (2)	−0.003 (2)
C13	0.047 (4)	0.051 (4)	0.055 (5)	−0.007 (3)	−0.001 (3)	−0.005 (3)
C6	0.046 (3)	0.042 (3)	0.029 (3)	−0.002 (3)	0.000 (2)	0.001 (2)
C4	0.053 (4)	0.075 (5)	0.029 (3)	−0.011 (4)	−0.016 (3)	0.001 (3)
C17	0.044 (3)	0.037 (3)	0.026 (3)	−0.012 (3)	−0.017 (2)	−0.002 (2)

C5	0.034 (3)	0.039 (3)	0.033 (3)	−0.001 (3)	−0.002 (2)	−0.003 (2)
C18	0.027 (3)	0.048 (4)	0.036 (3)	−0.009 (3)	−0.004 (2)	0.004 (3)
C16	0.034 (3)	0.054 (4)	0.028 (3)	0.001 (3)	0.002 (2)	−0.008 (3)
C19	0.035 (3)	0.049 (3)	0.025 (3)	−0.004 (3)	−0.004 (2)	−0.002 (2)
C3	0.038 (3)	0.055 (4)	0.023 (3)	−0.001 (3)	−0.008 (2)	0.002 (3)
B	0.028 (3)	0.047 (4)	0.028 (3)	−0.006 (3)	−0.002 (2)	−0.007 (3)
C12	0.047 (3)	0.061 (4)	0.032 (3)	−0.007 (3)	−0.015 (3)	0.012 (3)
C11A	0.038 (3)	0.044 (4)	0.044 (3)	0.000 (3)	−0.001 (3)	0.003 (3)

Geometric parameters (Å, °)

I—C17	2.089 (5)	C7—C12	1.500 (7)
F1—B	1.397 (7)	C13—C5	1.483 (9)
N2—C3	1.352 (8)	C13—H15B	0.9600
N2—C10	1.393 (7)	C13—H15C	0.9600
N2—B	1.546 (9)	C13—H15A	0.9600
F2—B	1.373 (8)	C6—C5	1.399 (8)
N1—C5	1.337 (8)	C6—H16	0.9300
N1—C8	1.403 (6)	C4—C3	1.486 (8)
N1—B	1.544 (8)	C4—H17C	0.9600
C15—C16	1.387 (8)	C4—H17A	0.9600
C15—C14	1.392 (7)	C4—H17B	0.9600
C15—H6	0.9300	C17—C16	1.386 (8)
C2—C1	1.382 (8)	C17—C18	1.394 (8)
C2—C3	1.390 (9)	C18—C19	1.386 (8)
C2—H7	0.9300	C18—H20	0.9300
C9—C8	1.396 (8)	C16—H21	0.9300
C9—C10	1.402 (7)	C19—H22	0.9300
C9—C14	1.485 (7)	C12—H25A	0.9600
C1—C10	1.422 (8)	C12—H25C	0.9600
C1—C11A	1.494 (8)	C12—H25B	0.9600
C14—C19	1.385 (8)	C11A—H1AB	0.9600
C8—C7	1.422 (8)	C11A—H1AC	0.9600
C7—C6	1.370 (9)	C11A—H1AA	0.9600
C3—N2—C10	107.9 (5)	H17C—C4—H17A	109.5
C3—N2—B	125.6 (5)	C3—C4—H17B	109.5
C10—N2—B	126.4 (5)	H17C—C4—H17B	109.5
C5—N1—C8	108.6 (4)	H17A—C4—H17B	109.5
C5—N1—B	125.9 (5)	C16—C17—C18	120.5 (5)
C8—N1—B	125.4 (5)	C16—C17—I	119.6 (4)
C16—C15—C14	121.2 (5)	C18—C17—I	119.9 (4)
C16—C15—H6	119.4	N1—C5—C6	108.8 (5)
C14—C15—H6	119.4	N1—C5—C13	124.2 (5)
C1—C2—C3	109.1 (5)	C6—C5—C13	127.0 (6)
C1—C2—H7	125.5	C19—C18—C17	119.4 (5)
C3—C2—H7	125.5	C19—C18—H20	120.3
C8—C9—C10	120.9 (5)	C17—C18—H20	120.3

C8—C9—C14	118.1 (5)	C17—C16—C15	119.2 (5)
C10—C9—C14	120.8 (5)	C17—C16—H21	120.4
C2—C1—C10	105.7 (5)	C15—C16—H21	120.4
C2—C1—C11A	124.5 (6)	C14—C19—C18	121.0 (5)
C10—C1—C11A	129.9 (5)	C14—C19—H22	119.5
C19—C14—C15	118.7 (5)	C18—C19—H22	119.5
C19—C14—C9	121.7 (5)	N2—C3—C2	109.0 (5)
C15—C14—C9	119.6 (4)	N2—C3—C4	122.9 (6)
N2—C10—C9	120.0 (5)	C2—C3—C4	128.0 (6)
N2—C10—C1	108.3 (5)	F2—B—F1	109.5 (5)
C9—C10—C1	131.7 (5)	F2—B—N1	110.8 (5)
C9—C8—N1	120.6 (5)	F1—B—N1	109.7 (5)
C9—C8—C7	132.1 (5)	F2—B—N2	110.6 (5)
N1—C8—C7	107.2 (5)	F1—B—N2	109.9 (5)
C6—C7—C8	106.4 (5)	N1—B—N2	106.4 (5)
C6—C7—C12	125.0 (6)	C7—C12—H25A	109.5
C8—C7—C12	128.5 (5)	C7—C12—H25C	109.5
C5—C13—H15B	109.5	H25A—C12—H25C	109.5
C5—C13—H15C	109.5	C7—C12—H25B	109.5
H15B—C13—H15C	109.5	H25A—C12—H25B	109.5
C5—C13—H15A	109.5	H25C—C12—H25B	109.5
H15B—C13—H15A	109.5	C1—C11A—H1AB	109.5
H15C—C13—H15A	109.5	C1—C11A—H1AC	109.5
C7—C6—C5	108.9 (5)	H1AB—C11A—H1AC	109.5
C7—C6—H16	125.5	C1—C11A—H1AA	109.5
C5—C6—H16	125.5	H1AB—C11A—H1AA	109.5
C3—C4—H17C	109.5	H1AC—C11A—H1AA	109.5
C3—C4—H17A	109.5		
