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# Methyl 2-[(3,5-di-tert-butyl-4-hydroxybenzyl)sulfanyl]pyridine-3-carboxylate *n*-hexane hemisolvate

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Key indicators: single-crystal X-ray study; T = 100 K; mean  $\sigma$ (C–C) = 0.002 Å; R factor = 0.047; wR factor = 0.113; data-to-parameter ratio = 20.7.

The title solvated ester, C<sub>22</sub>H<sub>29</sub>NO<sub>3</sub>S·0.5C<sub>6</sub>H<sub>14</sub>, crystallizes with two independent molecules along with a hexane molecule in the asymmetric unit. The two aromatic rings are separated by an  $-CH_2-S-$  linkage; the rings are aligned at 83.27 (4)° in one molecule and 47.66  $(7)^{\circ}$  in the other. The hydroxy group of one molecule forms an  $O-H \cdots O$  hydrogen bond to the other molecule.

#### **Related literature**

For the synthesis of carboxylic acid, see: Mansor et al. (2008).



**Experimental** 

Crystal data C22H29NO3S-0.5C6H14  $M_r = 430.61$ 

Monoclinic,  $P2_1/c$ a = 15.0665 (5) Å

b = 9.4818 (3) Å c = 34.6700 (13) Å $\beta = 90.796 \ (3)^{\circ}$ V = 4952.4 (3) Å<sup>3</sup> Z = 8

#### Data collection

Agilent SuperNova Dual
diffractometer with an Atlas
detector
Absorption correction: multi-scan
(CrysAlis PRO; Agilent, 2011)
$T_{\min} = 0.941, \ T_{\max} = 0.955$

#### Refinement

2

$R[F^2 > 2\sigma(F^2)] = 0.047$	H atoms treated by a mixture of
$wR(F^2) = 0.113$	independent and constrained
S = 1.00	refinement
11426 reflections	$\Delta \rho_{\rm max} = 0.33 \text{ e } \text{\AA}^{-3}$
551 parameters	$\Delta \rho_{\rm min} = -0.30 \text{ e } \text{\AA}^{-3}$
2 restraints	

#### Table 1 Hydrogen-bond geometry (Å, °).

Cg is the centroid of the C1-C6 ring.

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$	
O1−H1···O6	0.83 (1)	2.22 (2)	2.767 (2)	124 (2)	
Symmetry code: (i) $-x + 1$ , $y - \frac{1}{2}$ , $-z + \frac{1}{2}$ .					

Data collection: CrysAlis PRO (Agilent, 2011); cell refinement: CrysAlis PRO; data reduction: CrysAlis PRO; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5894).

#### References

Agilent (2011). CrysAlis PRO. Agilent Technologies, Yarnton, England. Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191. Mansor, S., Yehye, W. A., Ariffin, A., Rahman, N. A. & Ng, S. W. (2008). Acta Cryst. E64, 01778.

Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.

Westrip, S. P. (2010). J. Appl. Cryst. 43, 920-925.

Mo  $K\alpha$  radiation

 $0.40 \times 0.35 \times 0.30$  mm

34553 measured reflections 11425 independent reflections

8907 reflections with  $I > 2\sigma(I)$ 

 $\mu = 0.16 \text{ mm}^{-1}$ 

T = 100 K

 $R_{\rm int} = 0.041$ 

# supporting information

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# Methyl 2-[(3,5-di-*tert*-butyl-4-hydroxybenzyl)sulfanyl]pyridine-3-carboxylate *n*-hexane hemisolvate

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## S1. Comment

The study continues from an earlier study on 2-([(3,5–3,5-di-*tert*-butyl-4-hydroxphenyl)methyl]sulfanyl)pyridine-3-carboxylic acid Mansor *et al.*, 2008). The compound is readily esterified with methanol medium when the reaction is catalyzed with *p*-toulenesulfonic acid. The ester crystallizes with a hexane molecule of crystallization (Scheme I, Fig. 1). The two aromatic rings that are separated by an  $-CH_2$ -S- linkage are aligned at 83.27 (4) ° in one molecule and 47.66 (7) ° in the other. The hydroxyl group of one molecule forms a O—H···O hydrogen bond, whereas the other one shows a O— H··· $\pi$  contact (Table 1).

## S2. Experimental

2-((3,5-Di-*tert*-butyl-4-hydroxybenzyl)thio)nicotinic acid (1.36 g m, 3.97 mmol,) was dissolved in methanol (50 ml). A small quantity of *p*-toluenesulfonic acid was added as catalyst. The solution was heated for 6 h. The solvent was removed and sodium bicarbonate in water was added until the solution was approximately netural. The organic compound was extracted by ethyl acetate; the organic phase was dried with magnesium sulfate. The ethyl acetate was evaporated and the product recrystallized from *n*-hexane to give light yellow crystals. The methy ester crystallizes as two independent molecules along with a solvent (Scheme I, Fig. 1). The two aromatic rings that are separated by the  $-CH_2-S-$  linkage are aligned at 83.27 (4) ° in one molecule and 47.66 (7) ° in the other. The molecule having the larger twist uses its hydroxy H atom to form a hydrogen bond to the carbonyl O atom of the molecule having the smaller twist (Table 1).

## S3. Refinement

Carbon-bound H-atoms were placed in calculated positions [C–H 0.95 to 0.99 Å,  $U_{iso}$ (H) 1.2 to 1.5 $U_{eq}$ (C)] and were included in the refinement in the riding model approximation.

The hydroxy H-atoms were located in a difference Fourier map, and were refined isotropically with a distance restraint of O–H  $0.84\pm0.01$  Å.



#### Figure 1

Anisotropic displacement ellipsoid plot (Barbour, 2001) of at the two independent molecules and the solvent molecule at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

#### Methyl 2-[(3,5-di-tert-butyl-4-hydroxybenzyl)sulfanyl]pyridine-3-carboxylate n-hexane hemisolvate

Crystal data

$C_{22}H_{29}NO_{3}S{\cdot}0.5C_{6}H_{14}$
$M_r = 430.61$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
<i>a</i> = 15.0665 (5) Å
<i>b</i> = 9.4818 (3) Å
<i>c</i> = 34.6700 (13) Å
$\beta = 90.796 \ (3)^{\circ}$
$V = 4952.4 (3) \text{ Å}^3$
Z = 8

#### Data collection

Agilent SuperNova Dual	$T_{\rm min} = 0.941, \ T_{\rm max} = 0.941$
diffractometer with an Atlas detector	34553 measured reflect
Radiation source: SuperNova (Mo) X-ray	11425 independent ret
Source	8907 reflections with
Mirror monochromator	$R_{\rm int} = 0.041$
Detector resolution: 10.4041 pixels mm <sup>-1</sup>	$\theta_{\rm max} = 27.6^{\circ}, \ \theta_{\rm min} = 2.4$
$\omega$ scan	$h = -19 \rightarrow 16$
Absorption correction: multi-scan	$k = -12 \rightarrow 11$
(CrysAlis PRO; Agilent, 2011)	$l = -45 \rightarrow 41$

F(000) = 1864 $D_{\rm x} = 1.155 {\rm Mg} {\rm m}^{-3}$ Mo K $\alpha$  radiation,  $\lambda = 0.71073$  Å Cell parameters from 9294 reflections  $\theta = 2.4 - 27.5^{\circ}$  $\mu = 0.16 \text{ mm}^{-1}$ T = 100 KPrism, light yellow  $0.40 \times 0.35 \times 0.30$  mm

955 ctions flections  $I > 2\sigma(I)$ ۱°

Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.047$	Hydrogen site location: inferred from
$wR(F^2) = 0.113$	neighbouring sites
S = 1.00	H atoms treated by a mixture of independent
11426 reflections	and constrained refinement
551 parameters	$w = 1/[\sigma^2(F_o^2) + (0.0395P)^2 + 3.1619P]$
2 restraints	where $P = (F_o^2 + 2F_c^2)/3$
Primary atom site location: structure-invariant	$(\Delta/\sigma)_{\rm max} = 0.001$
direct methods	$\Delta \rho_{\rm max} = 0.33 \text{ e} \text{ Å}^{-3}$
	$\Delta \rho_{\min} = -0.30 \text{ e} \text{ Å}^{-3}$

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
<b>S</b> 1	0.72068 (3)	0.35500 (4)	0.438613 (12)	0.01646 (10)	
01	0.82843 (9)	0.76073 (14)	0.29857 (4)	0.0244 (3)	
H1	0.8214 (15)	0.728 (2)	0.2764 (4)	0.036 (7)*	
O2	0.83997 (9)	0.17218 (14)	0.47471 (4)	0.0259 (3)	
03	0.81798 (8)	0.06381 (13)	0.53138 (4)	0.0206 (3)	
N1	0.59707 (9)	0.39135 (15)	0.49119 (4)	0.0185 (3)	
C1	0.78031 (11)	0.69231 (17)	0.32609 (5)	0.0148 (3)	
C2	0.71649 (10)	0.58834 (17)	0.31702 (5)	0.0134 (3)	
C3	0.67382 (10)	0.52510 (16)	0.34816 (5)	0.0134 (3)	
Н3	0.6304	0.4545	0.3431	0.016*	
C4	0.69244 (10)	0.56147 (17)	0.38616 (5)	0.0131 (3)	
C5	0.75521 (10)	0.66599 (17)	0.39378 (5)	0.0141 (3)	
H5	0.7677	0.6913	0.4198	0.017*	
C6	0.80023 (10)	0.73445 (17)	0.36441 (5)	0.0142 (3)	
C7	0.86849 (11)	0.85133 (18)	0.37354 (5)	0.0187 (4)	
C8	0.84068 (12)	0.99148 (18)	0.35410 (6)	0.0226 (4)	
H8A	0.7853	1.0251	0.3654	0.034*	
H8B	0.8317	0.9762	0.3264	0.034*	
H8C	0.8874	1.0620	0.3582	0.034*	
C9	0.96164 (12)	0.8069 (2)	0.35996 (6)	0.0264 (4)	
H9A	0.9811	0.7231	0.3743	0.040*	
H9B	1.0037	0.8841	0.3646	0.040*	
H9C	0.9592	0.7853	0.3323	0.040*	
C10	0.87606 (13)	0.8805 (2)	0.41700 (6)	0.0265 (4)	
H10A	0.8959	0.7948	0.4304	0.040*	
H10B	0.8180	0.9089	0.4267	0.040*	
H10C	0.9191	0.9563	0.4216	0.040*	
C11	0.69576 (11)	0.54066 (18)	0.27526 (5)	0.0168 (3)	
C12	0.77809 (12)	0.46918 (19)	0.25784 (5)	0.0229 (4)	
H12A	0.7953	0.3880	0.2737	0.034*	
H12B	0.8272	0.5368	0.2572	0.034*	
H12C	0.7640	0.4374	0.2316	0.034*	
C13	0.66440 (13)	0.66520 (19)	0.24972 (5)	0.0241 (4)	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters  $(\mathring{A}^2)$ 

H13A	0.6111	0.7074	0.2608	0.036*
H13B	0.6507	0.6309	0.2236	0.036*
H13C	0.7115	0.7363	0.2486	0.036*
C14	0.62038 (12)	0.43170 (19)	0.27365 (5)	0.0232 (4)
H14A	0.5667	0.4726	0.2848	0.035*
H14B	0.6379	0.3478	0.2884	0.035*
H14C	0.6084	0.4052	0.2468	0.035*
C15	0.64611 (11)	0.49028 (17)	0.41924 (5)	0.0149 (3)
H15A	0.5899	0.4468	0.4101	0.018*
H15B	0.6321	0.5601	0.4395	0.018*
C16	0.67199 (11)	0.32072 (17)	0.48357 (5)	0.0148 (3)
C17	0.55904 (11)	0.3716 (2)	0.52556 (5)	0.0221 (4)
H17	0.5066	0.4231	0.5309	0.027*
C18	0.59154 (12)	0.2812 (2)	0.55354 (5)	0.0222 (4)
H18	0.5622	0.2703	0.5774	0.027*
C19	0.66795 (11)	0.20683 (19)	0.54570 (5)	0.0187 (4)
H19	0.6915	0.1424	0.5642	0.022*
C20	0.71060 (11)	0.22649 (17)	0.51058 (5)	0.0149 (3)
C21	0.79532 (11)	0.15355 (17)	0.50291 (5)	0.0168 (3)
C22	0.90344 (12)	-0.0032 (2)	0.52704 (6)	0.0245 (4)
H22A	0.9144	-0.0673	0.5488	0.037*
H22B	0.9500	0.0690	0.5266	0.037*
H22C	0.9039	-0.0566	0.5029	0.037*
S2	0.84000 (3)	0.64682 (4)	0.164523 (12)	0.01465 (10)
O4	0.50710 (8)	0.22359 (13)	0.16869 (4)	0.0251 (3)
H4	0.4579 (9)	0.260 (2)	0.1644 (7)	0.045 (7)*
05	0.98462 (9)	0.99237 (14)	0.22179 (4)	0.0313 (3)
O6	0.89539 (8)	0.80484 (13)	0.22566 (4)	0.0220 (3)
N2	0.95527 (9)	0.70873 (14)	0.10963 (4)	0.0157 (3)
C23	0.57637 (11)	0.30871 (17)	0.15767 (5)	0.0154 (3)
C24	0.66138 (11)	0.24573 (17)	0.15966 (5)	0.0157 (3)
C25	0.73273 (11)	0.32794 (17)	0.14796 (5)	0.0150 (3)
H25	0.7907	0.2885	0.1489	0.018*
C26	0.72205 (10)	0.46567 (17)	0.13502 (5)	0.0129 (3)
C27	0.63775 (10)	0.52384 (17)	0.13415 (5)	0.0140 (3)
H27	0.6308	0.6187	0.1258	0.017*
C28	0.56262 (11)	0.44839 (17)	0.14504 (5)	0.0142 (3)
C29	0.46915 (11)	0.51433 (18)	0.14173 (5)	0.0186 (4)
C30	0.47325 (12)	0.66895 (19)	0.12899 (7)	0.0293 (5)
H30A	0.5024	0.6752	0.1039	0.044*
H30B	0.5071	0.7236	0.1481	0.044*
H30C	0.4129	0.7069	0.1267	0.044*
C31	0.42186 (13)	0.5140(2)	0.18096 (6)	0.0312 (5)
H31A	0.3649	0.5632	0.1784	0.047*
H31B	0.4591	0.5622	0.2002	0.047*
H31C	0.4116	0.4165	0.1891	0.047*
C32	0.41416 (12)	0.4369 (2)	0.11052 (6)	0.0282 (4)
H32A	0.4460	0.4391	0.0861	0.042*
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H32B	0.3565	0.4837	0.1072	0.042*
H32C	0.4050	0.3388	0.1184	0.042*
C33	0.67511 (12)	0.09333 (17)	0.17365 (6)	0.0209 (4)
C34	0.64672 (15)	0.0793 (2)	0.21592 (6)	0.0327 (5)
H34A	0.6838	0.1407	0.2322	0.049*
H34B	0.6539	-0.0188	0.2243	0.049*
H34C	0.5844	0.1071	0.2182	0.049*
C35	0.62306 (13)	-0.01154 (19)	0.14800 (6)	0.0292(5)
H35A	0.6364	-0.1081	0.1563	0.044*
H35B	0.6404	0.0007	0.1211	0.044*
H35C	0 5593	0.0061	0 1504	0.044*
C36	0 77303 (13)	0.0501(2)	0 17171 (7)	0.0328(5)
H36A	0.8087	0.1124	0.1883	0.049*
H36B	0.7932	0.0576	0.1450	0.049*
H36C	0.7797	-0.0476	0.1806	0.049*
C37	0.7797 0.80198 (11)	0.54897 (17)	0.12235 (5)	0.045
H37A	0.7855	0.54697 (17)	0.12235 (3)	0.0108
1137A 1137B	0.7855	0.0143	0.1012	0.019
1137B C28	0.0491 0.02120(10)	0.4049	0.1152	$0.013^{\circ}$
C38	1.02407(11)	0.74019(10) 0.77768(18)	0.14380(3)	0.0130(3)
C39	1.02497 (11)	0.77708 (18)	0.09300 (3)	0.0194 (4)
П39	1.0420	0.7347	0.0090	0.025
C40	1.0/252 (11)	0.88001 (19)	0.114/1 (0)	0.0213 (4)
H40	1.1213	0.9263	0.1031	0.026*
C41	1.04/32 (11)	0.91313 (18)	0.15174 (5)	0.0191 (4)
H41	1.0786	0.9833	0.1660	0.023*
C42	0.97578 (10)	0.84316 (17)	0.16818 (5)	0.0150 (3)
C43	0.94730 (11)	0.87594 (17)	0.20780 (5)	0.0172 (4)
C44	0.95887 (17)	1.0294 (3)	0.26088 (6)	0.0443 (6)
H44A	0.9950	1.1087	0.2701	0.066*
H44B	0.9682	0.9481	0.2779	0.066*
H44C	0.8960	1.0561	0.2609	0.066*
C45	0.90495 (18)	0.4113 (3)	0.56068 (9)	0.0575 (8)
H45A	0.9585	0.3687	0.5499	0.086*
H45B	0.8611	0.3375	0.5657	0.086*
H45C	0.9202	0.4596	0.5849	0.086*
C46	0.86628 (15)	0.5170 (2)	0.53206 (8)	0.0415 (6)
H46A	0.9113	0.5904	0.5268	0.050*
H46B	0.8527	0.4680	0.5075	0.050*
C47	0.78269 (14)	0.5873 (2)	0.54639 (6)	0.0331 (5)
H47A	0.7971	0.6402	0.5703	0.040*
H47B	0.7390	0.5135	0.5530	0.040*
C48	0.74063 (15)	0.6876 (2)	0.51729 (7)	0.0344 (5)
H48A	0.7845	0.7615	0.5109	0.041*
H48B	0.7273	0.6345	0.4933	0.041*
C49	0.65646 (18)	0.7586 (3)	0.53037 (8)	0.0523 (7)
H49A	0.6696	0.8135	0.5541	0.063*
H49B	0.6124	0.6852	0.5370	0.063*
C50	0.6163 (2)	0.8559 (3)	0.50025 (10)	0.0761 (11)
				(-)

# supporting information

H50A	0.5629	0.9004	0.5105	0.114*
H50B	0.6005	0.8014	0.4771	0.114*
H50C	0.6595	0.9289	0.4936	0.114*

Atomic displacement parameters  $(Å^2)$ 

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
<b>S</b> 1	0.0185 (2)	0.0192 (2)	0.0119 (2)	0.00352 (17)	0.00481 (16)	0.00417 (16)
01	0.0316 (7)	0.0283 (7)	0.0134 (7)	-0.0131 (6)	0.0065 (6)	0.0007 (6)
O2	0.0283 (7)	0.0288 (7)	0.0210 (7)	0.0090 (6)	0.0107 (6)	0.0077 (6)
O3	0.0176 (6)	0.0261 (7)	0.0181 (7)	0.0058 (5)	0.0027 (5)	0.0067 (5)
N1	0.0160 (7)	0.0247 (8)	0.0150 (8)	0.0017 (6)	0.0025 (6)	0.0043 (6)
C1	0.0161 (8)	0.0151 (8)	0.0133 (8)	0.0010 (7)	0.0033 (6)	0.0033 (6)
C2	0.0143 (8)	0.0138 (8)	0.0120 (8)	0.0025 (6)	-0.0004 (6)	0.0003 (6)
C3	0.0132 (7)	0.0120 (7)	0.0150 (8)	0.0009 (6)	-0.0001 (6)	0.0009 (6)
C4	0.0122 (7)	0.0139 (8)	0.0133 (8)	0.0041 (6)	0.0022 (6)	0.0033 (6)
C5	0.0145 (8)	0.0153 (8)	0.0125 (8)	0.0040 (6)	-0.0013 (6)	-0.0005 (6)
C6	0.0132 (7)	0.0137 (8)	0.0156 (8)	0.0009 (6)	0.0001 (6)	0.0008 (6)
C7	0.0169 (8)	0.0189 (8)	0.0203 (9)	-0.0041 (7)	-0.0024 (7)	-0.0001 (7)
C8	0.0208 (9)	0.0199 (9)	0.0270 (10)	-0.0051 (7)	0.0001 (8)	0.0016 (8)
C9	0.0174 (9)	0.0276 (10)	0.0343 (12)	-0.0043 (8)	-0.0016 (8)	0.0021 (8)
C10	0.0286 (10)	0.0266 (10)	0.0240 (10)	-0.0092 (8)	-0.0056 (8)	-0.0021 (8)
C11	0.0213 (8)	0.0175 (8)	0.0116 (8)	-0.0016 (7)	0.0005 (7)	-0.0005 (6)
C12	0.0284 (10)	0.0236 (9)	0.0168 (9)	-0.0014 (8)	0.0040 (8)	-0.0031 (7)
C13	0.0316 (10)	0.0245 (9)	0.0162 (9)	-0.0008 (8)	-0.0040 (8)	0.0032 (7)
C14	0.0290 (10)	0.0246 (9)	0.0159 (9)	-0.0063 (8)	-0.0024 (8)	-0.0044 (7)
C15	0.0154 (8)	0.0176 (8)	0.0118 (8)	0.0015 (7)	0.0022 (6)	0.0030 (6)
C16	0.0165 (8)	0.0170 (8)	0.0111 (8)	-0.0040 (7)	0.0028 (6)	0.0006 (6)
C17	0.0156 (8)	0.0314 (10)	0.0195 (9)	0.0038 (8)	0.0058 (7)	0.0052 (8)
C18	0.0195 (9)	0.0324 (10)	0.0146 (9)	-0.0002 (8)	0.0051 (7)	0.0067 (8)
C19	0.0190 (8)	0.0240 (9)	0.0130 (9)	-0.0009 (7)	0.0001 (7)	0.0067 (7)
C20	0.0152 (8)	0.0162 (8)	0.0135 (8)	-0.0012 (7)	0.0007 (6)	0.0006 (6)
C21	0.0186 (8)	0.0167 (8)	0.0150 (9)	-0.0012 (7)	0.0003 (7)	0.0004 (7)
C22	0.0189 (9)	0.0319 (10)	0.0229 (10)	0.0086 (8)	0.0023 (7)	0.0047 (8)
S2	0.01407 (19)	0.0163 (2)	0.0137 (2)	-0.00558 (16)	0.00258 (15)	-0.00077 (16)
O4	0.0151 (6)	0.0191 (6)	0.0412 (9)	-0.0031 (5)	0.0048 (6)	0.0072 (6)
05	0.0383 (8)	0.0320 (7)	0.0236 (8)	-0.0210 (6)	0.0053 (6)	-0.0104 (6)
06	0.0276 (7)	0.0219 (6)	0.0166 (7)	-0.0074 (5)	0.0033 (5)	-0.0007 (5)
N2	0.0152 (7)	0.0160 (7)	0.0160 (7)	0.0006 (6)	0.0032 (6)	0.0013 (6)
C23	0.0142 (8)	0.0162 (8)	0.0156 (9)	-0.0043 (7)	0.0016 (6)	-0.0012 (7)
C24	0.0181 (8)	0.0133 (8)	0.0155 (9)	-0.0017 (7)	-0.0021 (7)	-0.0016 (6)
C25	0.0139 (8)	0.0165 (8)	0.0146 (8)	0.0003 (7)	-0.0022 (6)	-0.0033 (7)
C26	0.0130 (7)	0.0152 (8)	0.0103 (8)	-0.0036 (6)	-0.0002 (6)	-0.0027 (6)
C27	0.0172 (8)	0.0128 (8)	0.0119 (8)	-0.0012 (6)	-0.0016 (6)	-0.0002 (6)
C28	0.0141 (8)	0.0156 (8)	0.0129 (8)	-0.0011 (6)	-0.0006 (6)	-0.0030 (6)
C29	0.0140 (8)	0.0174 (8)	0.0245 (10)	0.0003 (7)	0.0008 (7)	-0.0009 (7)
C30	0.0168 (9)	0.0207 (9)	0.0503 (14)	0.0036 (8)	-0.0038 (9)	0.0031 (9)
C31	0.0248 (10)	0.0353 (11)	0.0338 (12)	0.0097 (9)	0.0087 (9)	-0.0012 (9)

C32	0.0174 (9)	0.0286 (10)	0.0384 (12)	0.0016 (8)	-0.0074 (8)	-0.0085 (9)
C33	0.0201 (9)	0.0130 (8)	0.0296 (11)	-0.0007 (7)	-0.0011 (8)	0.0032 (7)
C34	0.0455 (12)	0.0218 (10)	0.0308 (12)	-0.0004 (9)	-0.0006 (10)	0.0110 (8)
C35	0.0318 (10)	0.0140 (9)	0.0416 (13)	-0.0016 (8)	-0.0028 (9)	-0.0027 (8)
C36	0.0259 (10)	0.0162 (9)	0.0562 (15)	0.0034 (8)	-0.0051 (10)	0.0060 (9)
C37	0.0157 (8)	0.0173 (8)	0.0140 (8)	-0.0038 (7)	0.0012 (6)	-0.0014 (7)
C38	0.0109 (7)	0.0120 (7)	0.0179 (9)	0.0008 (6)	-0.0002 (6)	0.0030 (6)
C39	0.0153 (8)	0.0210 (9)	0.0219 (10)	0.0011 (7)	0.0075 (7)	0.0025 (7)
C40	0.0151 (8)	0.0220 (9)	0.0271 (10)	-0.0029 (7)	0.0066 (7)	0.0052 (8)
C41	0.0144 (8)	0.0178 (8)	0.0249 (10)	-0.0032 (7)	-0.0024 (7)	0.0028 (7)
C42	0.0135 (8)	0.0151 (8)	0.0162 (9)	0.0008 (7)	-0.0017 (6)	0.0030 (7)
C43	0.0165 (8)	0.0169 (8)	0.0181 (9)	-0.0020 (7)	-0.0036 (7)	0.0008 (7)
C44	0.0615 (15)	0.0463 (13)	0.0254 (12)	-0.0297 (12)	0.0110 (11)	-0.0175 (10)
C45	0.0509 (15)	0.0401 (14)	0.081 (2)	0.0019 (12)	-0.0356 (15)	-0.0080 (14)
C46	0.0361 (12)	0.0357 (12)	0.0522 (16)	-0.0046 (10)	-0.0122 (11)	-0.0077 (11)
C47	0.0400 (12)	0.0284 (10)	0.0306 (12)	-0.0051 (9)	-0.0111 (9)	-0.0040 (9)
C48	0.0422 (12)	0.0262 (10)	0.0343 (12)	-0.0072 (9)	-0.0154 (10)	-0.0012 (9)
C49	0.0567 (15)	0.0454 (14)	0.0541 (17)	0.0146 (12)	-0.0279 (13)	-0.0226 (12)
C50	0.076 (2)	0.0335 (14)	0.117 (3)	0.0065 (14)	-0.062 (2)	-0.0126 (16)

# Geometric parameters (Å, °)

S1—C16	1.7621 (17)	C23—C24	1.414 (2)
S1—C15	1.8271 (17)	C24—C25	1.393 (2)
O1—C1	1.3696 (19)	C24—C33	1.537 (2)
O1—H1	0.832 (9)	C25—C26	1.389 (2)
O2—C21	1.208 (2)	C25—H25	0.9500
O3—C21	1.344 (2)	C26—C27	1.385 (2)
O3—C22	1.445 (2)	C26—C37	1.511 (2)
N1—C16	1.342 (2)	C27—C28	1.396 (2)
N1—C17	1.343 (2)	C27—H27	0.9500
C1—C2	1.410 (2)	C28—C29	1.544 (2)
C1—C6	1.415 (2)	C29—C30	1.533 (2)
C2—C3	1.399 (2)	C29—C32	1.540 (3)
C2—C11	1.545 (2)	C29—C31	1.544 (3)
C3—C4	1.387 (2)	C30—H30A	0.9800
С3—Н3	0.9500	C30—H30B	0.9800
C4—C5	1.393 (2)	C30—H30C	0.9800
C4—C15	1.510(2)	C31—H31A	0.9800
C5—C6	1.392 (2)	C31—H31B	0.9800
С5—Н5	0.9500	C31—H31C	0.9800
C6—C7	1.542 (2)	C32—H32A	0.9800
C7—C10	1.535 (3)	С32—Н32В	0.9800
С7—С9	1.545 (2)	C32—H32C	0.9800
С7—С8	1.545 (2)	C33—C36	1.534 (3)
C8—H8A	0.9800	C33—C34	1.538 (3)
C8—H8B	0.9800	C33—C35	1.541 (3)
C8—H8C	0.9800	C34—H34A	0.9800

С9—Н9А	0.9800	C34—H34B	0.9800
С9—Н9В	0.9800	С34—Н34С	0.9800
С9—Н9С	0.9800	С35—Н35А	0.9800
C10—H10A	0.9800	С35—Н35В	0.9800
C10—H10B	0.9800	С35—Н35С	0.9800
C10—H10C	0.9800	С36—Н36А	0.9800
C11—C14	1.536 (2)	C36—H36B	0.9800
C11—C12	1.544 (2)	C36—H36C	0.9800
C11—C13	1 546 (2)	C37—H37A	0.9900
C12—H12A	0.9800	C37—H37B	0.9900
C12—H12B	0.9800	C38—C42	1410(2)
C12—H12C	0.9800	$C_{39}$ $C_{40}$	1.110(2) 1.381(3)
C13—H13A	0.9800	C39—H39	0.9500
C13—H13B	0.9800	C40—C41	1380(3)
C13—H13C	0.9800	C40 - H40	0.9500
C14—H14A	0.9800	C41-C42	1.394(2)
C14—H14B	0.9800	C41 - H41	0.9500
C14 H14C	0.9800	C42-C43	1.478(2)
C15—H15A	0.9900	C44—H44A	0.9800
C15—H15B	0.9900	C44—H44B	0.9800
$C_{10} = 1100$	1.414(2)	C44 - H44C	0.9800
$C_{10} = C_{20}$	1.414(2) 1 380(3)	C45-C46	1.521(3)
C17 H17	0.9500	$C_{45} = C_{40}$	0.0800
C18 $C19$	1.380(2)	$C_{45}$ H45R	0.9800
C18—H18	0.9500	C45 - H45C	0.9800
C19-C20	1 397 (2)	C46-C47	1.515(3)
C19 H19	0.9500	$C_{46} = H_{46A}$	0.0000
$C_{10}$ $C_{21}$	1.479(2)	$C_{40}$ H46B	0.9900
$C_{20} = C_{21}$	1.479(2)	C40 $C47$ $C48$	1.510(3)
C22—1122A C22 H22B	0.9800	C47 = C48	0.0000
C22—H22C	0.9800	C47 = H47P	0.9900
S2 C38	0.9800	$C_{4} = C_{4}$	0.9900
S2-C37	1.7030(10) 1.8174(17)	$C_{40}$ $U_{49}$	0.0000
52037	1.01/4(1/) 1.278(2)	$C_{40}$ $H_{40}$ $H_{40}$	0.9900
04 114	1.370(2)	$C_{40} = C_{50}$	0.9900
04 - H4	0.830(9) 1.228(2)	C49 - C30	1.314(4)
05 C44	1.320(2)	C49—R49A	0.9900
05-044	1.430(2)	C50 U50A	0.9900
V0	1.209 (2)	C50_H50A	0.9800
N2 C29	1.342(2) 1.242(2)	C50_H50C	0.9800
$\begin{array}{c} N2 - C38 \\ C22 - C28 \\ \end{array}$	1.343(2)	С30—Н30С	0.9800
C23—C28	1.409 (2)		
C16—S1—C15	101.14 (8)	С28—С27—Н27	118.8
C1—O1—H1	113.9 (16)	C27—C28—C23	116.71 (14)
C21—O3—C22	114.81 (13)	C27—C28—C29	120.98 (14)
C16—N1—C17	118.41 (15)	C23—C28—C29	122.26 (14)
O1—C1—C2	122.82 (15)	C30—C29—C32	106.08 (16)
O1—C1—C6	114.38 (14)	C30—C29—C28	111.62 (14)

$C^{2}$ $C^{1}$ $C^{2}$	122.90(15)	$C_{22}$ $C_{20}$ $C_{29}$	100.01.(1.4)
$C_2 = C_1 = C_0$	122.80 (15)	$C_{32} - C_{29} - C_{28}$	109.91 (14)
	116.55 (15)	$C_{30} - C_{29} - C_{31}$	106.05 (15)
C3—C2—C11	120.59 (14)	C32—C29—C31	111.57 (15)
C1C2C11	122.83 (14)	C28—C29—C31	111.43 (15)
C4—C3—C2	122.51 (15)	С29—С30—Н30А	109.5
С4—С3—Н3	118.7	С29—С30—Н30В	109.5
С2—С3—Н3	118.7	H30A—C30—H30B	109.5
C3—C4—C5	119.02 (15)	С29—С30—Н30С	109.5
C3—C4—C15	121.38 (15)	H30A—C30—H30C	109.5
C5—C4—C15	119.59 (15)	H30B-C30-H30C	109.5
C6—C5—C4	121.97 (16)	C29—C31—H31A	109.5
С6—С5—Н5	119.0	C29—C31—H31B	109.5
С4—С5—Н5	119.0	H31A—C31—H31B	109.5
C5—C6—C1	117.13 (15)	C29—C31—H31C	109.5
C5—C6—C7	121.02 (15)	H31A—C31—H31C	109.5
C1 - C6 - C7	121.02(15) 121.85(15)	H31B_C31_H31C	109.5
$C_{10}$ $C_{7}$ $C_{6}$	121.00(13) 111.80(14)	$C_{20}$ $C_{32}$ $H_{32A}$	109.5
$C_{10} = C_{7} = C_{0}$	107.03(15)	$C_{29} = C_{32} = H_{32R}$	109.5
$C_{10} = C_{10} = C_{10}$	107.03(13) 110.22(14)	H22A C22 H22D	109.5
$C_0 - C_7 - C_9$	110.55(14) 10(.92(15))	H32A - C32 - H32B	109.5
$C_{10} - C_{7} - C_{8}$	100.82(15)	С29—С32—H32С	109.5
	110.64 (14)	H32A—C32—H32C	109.5
C9—C7—C8	110.11 (14)	H32B—C32—H32C	109.5
С7—С8—Н8А	109.5	C36—C33—C24	111.28 (14)
С7—С8—Н8В	109.5	C36—C33—C34	107.41 (17)
H8A—C8—H8B	109.5	C24—C33—C34	110.11 (15)
С7—С8—Н8С	109.5	C36—C33—C35	106.51 (16)
H8A—C8—H8C	109.5	C24—C33—C35	111.05 (15)
H8B—C8—H8C	109.5	C34—C33—C35	110.37 (16)
С7—С9—Н9А	109.5	С33—С34—Н34А	109.5
С7—С9—Н9В	109.5	C33—C34—H34B	109.5
H9A—C9—H9B	109.5	H34A—C34—H34B	109.5
С7—С9—Н9С	109.5	C33—C34—H34C	109.5
Н9А—С9—Н9С	109.5	H34A—C34—H34C	109.5
H9B—C9—H9C	109.5	H34B—C34—H34C	109.5
C7-C10-H10A	109.5	C33—C35—H35A	109.5
C7-C10-H10B	109.5	C33—C35—H35B	109.5
$H_{10A}$ $C_{10}$ $H_{10B}$	109.5	H354_C35_H35B	109.5
C7 $C10$ $H10C$	109.5	1135A - C35 - 1155B	109.5
	109.5		109.5
H10A - C10 - H10C	109.5	нээд—Сээ—нээС	109.5
	109.5	H35B-C35-H35C	109.5
	106./5 (14)	C33—C36—H36A	109.5
C14—C11—C2	111.75 (14)	С33—С36—Н36В	109.5
C12—C11—C2	110.00 (14)	H36A—C36—H36B	109.5
C14—C11—C13	105.85 (14)	C33—C36—H36C	109.5
C12—C11—C13	110.63 (14)	H36A—C36—H36C	109.5
C2-C11-C13	111.68 (14)	H36B—C36—H36C	109.5
C11—C12—H12A	109.5	C26—C37—S2	106.06 (11)
C11—C12—H12B	109.5	С26—С37—Н37А	110.5

H12A—C12—H12B	109.5	S2—C37—H37A	110.5
C11—C12—H12C	109.5	С26—С37—Н37В	110.5
H12A—C12—H12C	109.5	S2—C37—H37B	110.5
H12B—C12—H12C	109.5	Н37А—С37—Н37В	108.7
C11—C13—H13A	109.5	N2-C38-C42	122.44 (14)
C11—C13—H13B	109.5	N2—C38—S2	116.86 (12)
H13A—C13—H13B	109.5	C42—C38—S2	120.71 (13)
C11—C13—H13C	109.5	N2-C39-C40	123.92 (17)
H13A—C13—H13C	109.5	N2—C39—H39	118.0
H13B—C13—H13C	109.5	С40—С39—Н39	118.0
C11—C14—H14A	109.5	C41—C40—C39	118.20 (16)
C11—C14—H14B	109.5	C41—C40—H40	120.9
H14A—C14—H14B	109.5	C39—C40—H40	120.9
C11—C14—H14C	109.5	C40—C41—C42	119.75 (16)
H14A—C14—H14C	109.5	C40—C41—H41	120.1
H14B—C14—H14C	109.5	C42—C41—H41	120.1
C4—C15—S1	107.74 (11)	C41—C42—C38	117.91 (16)
C4—C15—H15A	110.2	C41—C42—C43	121.06 (15)
S1—C15—H15A	110.2	C38—C42—C43	121.03 (14)
C4—C15—H15B	110.2	O6—C43—O5	123.38 (16)
S1—C15—H15B	110.2	O6—C43—C42	123.82 (15)
H15A—C15—H15B	108.5	O5—C43—C42	112.79 (14)
N1—C16—C20	121.64 (15)	O5—C44—H44A	109.5
N1—C16—S1	116.34 (12)	O5—C44—H44B	109.5
C20—C16—S1	122.00 (12)	H44A—C44—H44B	109.5
N1—C17—C18	123.98 (16)	O5—C44—H44C	109.5
N1—C17—H17	118.0	H44A—C44—H44C	109.5
C18—C17—H17	118.0	H44B—C44—H44C	109.5
C17—C18—C19	117.90 (16)	C46—C45—H45A	109.5
C17—C18—H18	121.0	C46—C45—H45B	109.5
C19—C18—H18	121.0	H45A—C45—H45B	109.5
C18—C19—C20	119.90 (16)	C46—C45—H45C	109.5
C18—C19—H19	120.1	H45A—C45—H45C	109.5
С20—С19—Н19	120.1	H45B—C45—H45C	109.5
C19—C20—C16	118.14 (15)	C47—C46—C45	112.9 (2)
C19—C20—C21	120.18 (15)	C47—C46—H46A	109.0
C16—C20—C21	121.65 (15)	C45—C46—H46A	109.0
O2—C21—O3	123.27 (15)	C47—C46—H46B	109.0
O2—C21—C20	124.68 (16)	C45—C46—H46B	109.0
O3—C21—C20	112.05 (14)	H46A—C46—H46B	107.8
O3—C22—H22A	109.5	C46—C47—C48	113.50 (19)
O3—C22—H22B	109.5	С46—С47—Н47А	108.9
H22A—C22—H22B	109.5	С48—С47—Н47А	108.9
O3—C22—H22C	109.5	C46—C47—H47B	108.9
H22A—C22—H22C	109.5	C48—C47—H47B	108.9
H22B—C22—H22C	109.5	H47A—C47—H47B	107.7
C38—S2—C37	101.46 (8)	C49—C48—C47	115.1 (2)
C23—O4—H4	112.5 (17)	C49—C48—H48A	108.5
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C43—O5—C44	115.01 (14)	C47—C48—H48A	108.5
C39—N2—C38	117.77 (15)	C49—C48—H48B	108.5
O4—C23—C28	121.88 (14)	C47—C48—H48B	108.5
O4—C23—C24	115.35 (14)	H48A—C48—H48B	107.5
C28—C23—C24	122.77 (15)	C48—C49—C50	113.2 (3)
C25—C24—C23	116.85 (15)	C48—C49—H49A	108.9
C25—C24—C33	121.13 (15)	C50—C49—H49A	108.9
C23—C24—C33	122.02 (14)	C48—C49—H49B	108.9
C26—C25—C24	122.25 (15)	С50—С49—Н49В	108.9
С26—С25—Н25	118.9	H49A—C49—H49B	107.8
С24—С25—Н25	118.9	C49—C50—H50A	109.5
C27—C26—C25	118.92 (14)	C49—C50—H50B	109.5
C27—C26—C37	121.34 (14)	H50A—C50—H50B	109.5
C25—C26—C37	119.74 (14)	С49—С50—Н50С	109.5
C26—C27—C28	122.49 (15)	H50A-C50-H50C	109.5
С26—С27—Н27	118.8	H50B-C50-H50C	109.5
Q1—C1—C2—C3	178.61 (14)	Q4—C23—C24—C33	-0.7(2)
C6-C1-C2-C3	-1.2(2)	$C_{28}$ $C_{23}$ $C_{24}$ $C_{33}$	179.78 (16)
01 - C1 - C2 - C11	0.8(2)	$C_{23}$ $C_{24}$ $C_{25}$ $C_{26}$	0.3 (2)
C6-C1-C2-C11	-179.02(15)	$C_{33}$ $C_{24}$ $C_{25}$ $C_{26}$	179.69 (16)
C1 - C2 - C3 - C4	0.0(2)	$C_{24}$ $C_{25}$ $C_{26}$ $C_{27}$	0.8 (2)
$C_{11} = C_{2} = C_{3} = C_{4}$	177 91 (15)	$C_{24}$ $C_{25}$ $C_{26}$ $C_{27}$ $C_{27}$	-17940(15)
$C_2 - C_3 - C_4 - C_5$	0.7(2)	$C_{25}$ $C_{25}$ $C_{25}$ $C_{27}$ $C_{28}$	-13(2)
$C_2 = C_3 = C_4 = C_1^{-1}$	-17929(14)	$C_{23}^{37}$ $C_{26}^{37}$ $C_{27}^{37}$ $C_{28}^{37}$	178 84 (15)
$C_{2}^{-}$ $C_{3}^{-}$ $C_{4}^{-}$ $C_{5}^{-}$ $C_{6}^{-}$	-0.4(2)	$C_{26}^{26} - C_{27}^{27} - C_{28}^{28} - C_{23}^{23}$	0.8(2)
$C_{15} = C_{4} = C_{5} = C_{6}$	170.4(2)	$C_{20} = C_{27} = C_{28} = C_{29}$	-176.89(15)
$C_{13} - C_{4} - C_{5} - C_{6} - C_{1}$	-0.7(2)	$C_{20} = C_{27} = C_{28} = C_{27}$	-170.39(15)
C4 - C5 - C6 - C7	-0.7(2)	$C_{24} = C_{23} = C_{26} = C_{27}$	-1/9.21(10)
C4 - C5 - C0 - C7	1/9.13(13) 179.21(14)	$C_{24} = C_{23} = C_{26} = C_{27}$	0.5(2)
01 - C1 - C0 - C3	-1/8.31(14)	04 - 023 - 028 - 029	-1.3(3)
$C_2 - C_1 - C_6 - C_3$	1.5(2)	$C_{24} = C_{23} = C_{28} = C_{29}$	1//.95 (10)
01 - C1 - C6 - C7	1.9 (2)	$C_2/-C_2 = C_2 = C_3 =$	-4.8(2)
C2-C1-C6-C/	-1/8.29 (15)	$C_{23}$ $C_{28}$ $C_{29}$ $C_{30}$	1/7.59 (16)
C5—C6—C7—C10	-1.5 (2)	$C_{27} - C_{28} - C_{29} - C_{32}$	112.57 (18)
C1 - C6 - C7 - C10	1/8.2/(15)	$C_{23}$ — $C_{28}$ — $C_{29}$ — $C_{32}$	-65.0 (2)
C5—C6—C7—C9	117.45 (17)	C27—C28—C29—C31	-123.21 (18)
C1—C6—C7—C9	-62.7(2)	C23—C28—C29—C31	59.2 (2)
C5—C6—C7—C8	-120.46 (17)	C25—C24—C33—C36	-1.3 (2)
C1—C6—C7—C8	59.3 (2)	C23—C24—C33—C36	178.15 (17)
C3—C2—C11—C14	5.1 (2)	C25—C24—C33—C34	117.73 (18)
C1—C2—C11—C14	-177.15 (15)	C23—C24—C33—C34	-62.9 (2)
C3—C2—C11—C12	-113.29 (16)	C25—C24—C33—C35	-119.72 (18)
C1—C2—C11—C12	64.5 (2)	C23—C24—C33—C35	59.7 (2)
C3—C2—C11—C13	123.47 (16)	C27—C26—C37—S2	86.45 (17)
C1—C2—C11—C13	-58.8 (2)	C25—C26—C37—S2	-93.38 (16)
C3—C4—C15—S1	99.30 (16)	C38—S2—C37—C26	-179.22 (11)
C5—C4—C15—S1	-80.75 (16)	C39—N2—C38—C42	1.0 (2)
C16—S1—C15—C4	163.90 (12)	C39—N2—C38—S2	-179.48 (12)

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	0.5 (2) -178.10 (14) 2.45 (15) -176.15 (14) -1.0 (3) 0.2 (3) 1.1 (3) -1.6 (3) 1.76.41 (16) 0.8 (3) 1.79.29 (13) -177.18 (15) 1.3 (2) 3.8 (2) -175.61 (14) -174.27 (18) 3.6 (3) 5.1 (2) -176.95 (15) 78.72 (15)	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} -6.51 (14) \\ 173.05 (13) \\ -0.9 (3) \\ 0.3 (3) \\ 0.3 (3) \\ -0.2 (2) \\ 179.85 (16) \\ -0.4 (2) \\ -179.96 (12) \\ 179.52 (15) \\ 0.0 (2) \\ 0.3 (3) \\ -179.87 (17) \\ -168.10 (16) \\ 12.0 (3) \\ 12.1 (2) \\ -167.86 (15) \\ -177.04 (18) \\ 179.34 (18) \\ -179.21 (19) \end{array}$
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# Hydrogen-bond geometry (Å, °)

Cg is the centroid of the C1–C6 ring.

D—H···A	D—H	H···A	D···A	D—H…A
01—H1…O6	0.83 (1)	2.22 (2)	2.767 (2)	124 (2)
$O4$ — $H4$ ···· $Cg^{i}$	0.83 (1)	3.25	?	?

Symmetry code: (i) -x+1, y-1/2, -z+1/2.