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5-Methyl-3,3-bis(4-methylpiperazin-1yl)-1-[2-(4-methylpiperazin-1-yl)ethyl]indolin-2-one

Hui-Hui Lin, Xiao-Lin Zheng and Sheng-Li Cao*

Department of Chemistry, Capital Normal University, Beijing 100048, People's Republic of China Correspondence e-mail: slcao@cnu.edu.cn

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Key indicators: single-crystal X-ray study; T = 296 K; mean σ (C–C) = 0.003 Å; R factor = 0.047; wR factor = 0.129; data-to-parameter ratio = 15.6.

In the title compound, C₂₆H₄₃N₇O, each piperazine ring adopts a chair conformation. Two 1-methylpiperazine rings bond to one Csp^3 of the pyrrole ring via the piperazine N atoms, while the third one links to the N atom of the indolin-2one unit through a flexible ethyl group with an almost syn conformation. In the crystal, molecules are connected through methylene-carbonyl C-H···O interactions into an infinite chain along the c axis. The almost parallel arrays are stacked, forming a three-dimensional framework.

Related literature

For the background to indoline-2,3-dione and its derivatives, see Chiyanzu et al. (2005); Solomon et al. (2009); Sriram et al. (2004). For a related structure, see: Lin et al. (2012).



17378 measured reflections

 $R_{\rm int} = 0.036$

4785 independent reflections

3332 reflections with $I > 2\sigma(I)$

Experimental

Crystal data

| $C_{26}H_{43}N_7O$ | V = 2725.7 (2) Å ³ |
|---------------------------------|---|
| $M_r = 469.67$ | Z = 4 |
| Monoclinic, $P2_1/c$ | Mo $K\alpha$ radiation |
| a = 15.9433 (8) Å | $\mu = 0.07 \text{ mm}^{-1}$ |
| b = 14.4097 (6) Å | T = 296 K |
| c = 12.5310 (6) Å | $0.33 \times 0.21 \times 0.20 \text{ mm}$ |
| $\beta = 108.774 \ (3)^{\circ}$ | |

Data collection

Bruker APEXII CCD area-detector diffractometer Absorption correction: multi-scan (SADABS; Bruker, 2007) $T_{\min} = 0.613, \ T_{\max} = 0.746$

Refinement

| $R[F^2 > 2\sigma(F^2)] = 0.047$ | 307 parameters |
|---------------------------------|--|
| $wR(F^2) = 0.129$ | H-atom parameters constrained |
| S = 1.03 | $\Delta \rho_{\rm max} = 0.23 \text{ e } \text{\AA}^{-3}$ |
| 4785 reflections | $\Delta \rho_{\rm min} = -0.20 \text{ e } \text{\AA}^{-3}$ |

Table 1

Hydrogen-bond geometry (Å, °).

 $D - H \cdot \cdot \cdot A$ D-H $H \cdots A$ $D \cdot \cdot \cdot A$ $D - H \cdot \cdot \cdot A$ $C19-H19B\cdots O1^{i}$ 0.97 2.64 3.388 (2) 135

Symmetry code: (i) $x, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2007); cell refinement: APEX2 and SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL and PLATON (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZJ2076).

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5-Methyl-3,3-bis(4-methylpiperazin-1-yl)-1-[2-(4-methylpiperazin-1-yl)ethyl]indolin-2-one

Hui-Hui Lin, Xiao-Lin Zheng and Sheng-Li Cao

S1. Comment

Mannich base derivatives of indoline-2,3-dione (isatin) exhibit antibacterial (Chiyanzu *et al.*, 2005), anti-HIV (Sriram *et al.*, 2004) and anticancer activity (Solomon *et al.*, 2009).Recently, we obtained an isatin derivative with an flexible ethylene linker between the N atom of the isatin moiety and the amine group of the morpholine through the reaction of 1-(2bromoethyl)-5-methylindoline-2,3-dione with excess of morpholine, namely 5-methyl-3,3-bis(morpholin-4yl)-1-[2(morpholin-4-yl)ethyl] -2,3-dihydro-1H-indol-2-one. As a continue research on the synthesis of indole-2,3dione(isatin) derivatives, herein we report the synthesis and crystal structure of a new compound based on indoline-2,3dione through the reaction of 1-(2-bromoethyl)-5-methylindoline-2,3-dione with 1-methylpiperazine, namely 5methyl-3,3-bis(4-methylpiperazin-1-yl)-1-(2-(4-methylpiperazin-1-yl)ethyl) indolin-2-one.

In the title compound, each piperazine ring adopts a chair conformation. Two 1-methylpiperazine rings bond to one $C8(sp^3)$ of the pyrrole ring via the piperazine N atoms (N2, N6), while the third one links to the N1 atom of indolin-2-one moiety through a flexible ethyl group with an almost *syn* conformation (N1-C15-C16-N5 torsion angle of 58.0 (3)°, as shown in Fig. 1). This steric configuration is similar to that of 5-methyl-3,3-bis(morpholin-4-yl)-1-[2(morpholin-4-yl)ethyl] -2,3-dihydro-1H-indol-2-one (59.7 (3)°) reported by us recently (Lin *et al.*, 2012). Through C19—H19b(methyl-ene)···O2ⁱ(carbonyl) interactions(see Table 1), the molecules are interconnected and arranged into an array along the *c* direction (i x, -y + 0.5, z +0.5), as shown in Fig. 2. The almost parallel arrays are stacked to form a three-dimensional framework (Fig. 3)

S2. Experimental

To a solution of 1-(2-bromoethyl)-5-methylindoline-2,3-dione (0.27 g, 1 mmol) in N,N-dimethylformamide (5 ml) was added dropwise 1-methylpiperazine (0.60 g, 6 mmol). The mixture was stirred at 80–90°C for 3 h. The colorless crystals of the title compound were deposited by evaporation of the resulting solution in room temperature for one day (m.p. 401.2–403.0 K; yield 45%).

S3. Refinement

All H atoms were discernible in the difference electron density maps. Nevertheless, the hydrogen atoms were placed into idealized positions and allowed to ride on their respective carrier atoms, with C—H = 0.93 and 0.97 Å for aryl and methylene hydrogens, respectively. $U_{iso}(H) = 1.2U_{eq}(C)_{aryl/methylene}$.



Figure 1

The title molecule with the atomic numbering scheme. The displacement ellipsoids of the non-hydrogen atoms are shown at the 30% probability level.



Figure 2

The array of the title compound along the *c* direction. The red-dashed lines indicate C19—H19b···O2ⁱ interactions (i x, -y + 0.5, z + 0.5).



Figure 3

View down the c direction of the stacking structure of the title compound. The stacking structures are shown with different color for clarity.

5-Methyl-3,3-bis(4-methylpiperazin-1-yl)-1-[2-(4-methylpiperazin-1-yl)ethyl]indolin-2-one

| Crystal data | |
|---|---|
| $C_{26}H_{43}N_7O$ $M_r = 469.67$ Monoclinic, $P2_1/c$ Hall symbol: -P 2ybc $a = 15.9433$ (8) Å $b = 14.4097$ (6) Å $c = 12.5310$ (6) Å $\beta = 108.774$ (3)° $V = 2725.7$ (2) Å ³ $Z = 4$ | F(000) = 1024 $D_x = 1.145 \text{ Mg m}^{-3}$ Mo Ka radiation, $\lambda = 0.71073 \text{ Å}$ Cell parameters from 233 reflections $\theta = 2.2-27.0^{\circ}$ $\mu = 0.07 \text{ mm}^{-1}$ T = 296 K Block, colorless $0.33 \times 0.21 \times 0.20 \text{ mm}$ |
| Data collection | |
| Bruker APEXII CCD area-detector diffractometer Radiation source: fine-focus sealed tube Graphite monochromator ω scans Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2007) $T_{min} = 0.613, T_{max} = 0.746$ | 17378 measured reflections 4785 independent reflections 3332 reflections with $I > 2\sigma(I)$ $R_{int} = 0.036$ $\theta_{max} = 25.0^{\circ}, \theta_{min} = 2.2^{\circ}$ $h = -18 \rightarrow 12$ $k = -15 \rightarrow 17$ $l = -14 \rightarrow 14$ |
| Refinement | |
| Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.047$ $wR(F^2) = 0.129$ S = 1.03 4785 reflections 307 parameters 0 restraints Primary atom site location: structure-invariant direct methods | Secondary atom site location: difference Fourier map Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0638P)^2 + 0.4418P] P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.23$ e Å ⁻³ $\Delta\rho_{min} = -0.20$ e Å ⁻³ |

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2$ sigma(F^2) is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

 $U_{\rm iso} * / U_{\rm eq}$ Ζ x v 01 0.0505 (4) 0.61856 (8) 0.42163 (9) 0.51428 (11) N1 0.63105 (9) 0.48564 (10) 0.68622 (13) 0.0417(4)N2 0.81475 (9) 0.38396 (9) 0.65785 (12) 0.0382(4)N3 0.84094 (11) 0.18704 (11) 0.0508 (4) 0.66137 (14) N4 0.56630(11) 0.10611 (11) 0.64539 (14) 0.0509(4)N5 0.56530(10) 0.29754 (10) 0.71078 (12) 0.0445(4)0.52537 (10) N6 0.78671 (10) 0.55620 (12) 0.0426 (4) N7 0.83904(12)0.65751 (12) 0.42139 (14) 0.0594(5)C1 0.93336 (16) 0.62601 (16) 1.02535 (18) 0.0685(7)H1A 0.9181 0.6508 0.103* 1.0878 H1B 0.9600 0.6738 0.9936 0.103* H1C 0.9745 0.5758 1.0509 0.103* C2 0.85101 (14) 0.59104 (13) 0.93714 (16) 0.0508(5)C3 0.76872 (15) 0.59731 (14) 0.95242 (18) 0.0567 (6) H3A 0.068* 0.7651 0.6244 1.0182 C4 0.69174 (14) 0.56483 (13) 0.87348 (17) 0.0516(5) H4A 0.5694 0.062* 0.6373 0.8856 C5 0.69861 (12) 0.52560(12) 0.77654 (15) 0.0411(4)C6 0.77938 (12) 0.51941 (11) 0.0385 (4) 0.75634 (15) 0.85514 (13) C7 0.55156 (12) 0.83746 (15) 0.0457(5)0.055* H7A 0.9095 0.5467 0.8253 0.47217 (12) C8 0.76536(11) 0.64355 (14) 0.0382(4)C9 0.66284 (12) 0.45605 (12) 0.60356 (15) 0.0395 (4) C10 0.89282 (18) 0.10229 (16) 0.6758(2)0.0817(8)H10A 0.8837 0.0659 0.7352 0.123* H10B 0.9545 0.1177 0.6947 0.123* H10C 0.8746 0.6069 0.123* 0.0674 C14 0.80539(13) 0.32577 (12) 0.74919 (16) 0.0457(5)H14A 0.055* 0.7441 0.3069 0.7319 H14B 0.8220 0.3611 0.8188 0.055* C12 0.79338 (14) 0.32871 (14) 0.55410 (16) 0.0529(5)0.4938 0.064* H12A 0.8012 0.3663 H12B 0.7319 0.3092 0.5320 0.064* C13 0.86357 (14) 0.24095 (14) 0.76392 (16) 0.0529(5)H13A 0.9251 0.2600 0.7847 0.063*

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

| H13B | 0.8566 | 0.2029 | 0.8244 | 0.063* |
|------|--------------|--------------|--------------|------------|
| C11 | 0.85262 (15) | 0.24454 (14) | 0.57235 (17) | 0.0575 (6) |
| H11A | 0.8384 | 0.2089 | 0.5031 | 0.069* |
| H11B | 0.9140 | 0.2641 | 0.5924 | 0.069* |
| C21 | 0.61197 (17) | 0.03109 (15) | 0.6078 (2) | 0.0715 (7) |
| H21A | 0.6241 | -0.0186 | 0.6616 | 0.107* |
| H21B | 0.6666 | 0.0540 | 0.6012 | 0.107* |
| H21C | 0.5751 | 0.0087 | 0.5358 | 0.107* |
| C19 | 0.62220 (14) | 0.13991 (14) | 0.75484 (17) | 0.0523 (5) |
| H19A | 0.6796 | 0.1574 | 0.7500 | 0.063* |
| H19B | 0.6312 | 0.0905 | 0.8100 | 0.063* |
| C20 | 0.58116 (14) | 0.22236 (14) | 0.79314 (16) | 0.0536 (5) |
| H20A | 0.5256 | 0.2040 | 0.8030 | 0.064* |
| H20B | 0.6204 | 0.2439 | 0.8653 | 0.064* |
| C18 | 0.54742 (15) | 0.18231 (14) | 0.56412 (17) | 0.0555 (5) |
| H18A | 0.5075 | 0.1606 | 0.4925 | 0.067* |
| H18B | 0.6021 | 0.2019 | 0.5525 | 0.067* |
| C17 | 0.50618 (13) | 0.26400 (14) | 0.60349 (16) | 0.0526 (5) |
| H17A | 0.4948 | 0.3133 | 0.5479 | 0.063* |
| H17B | 0.4501 | 0.2456 | 0.6118 | 0.063* |
| C16 | 0.53429 (13) | 0.38064 (13) | 0.75112 (18) | 0.0516 (5) |
| H16A | 0.5688 | 0.3899 | 0.8297 | 0.062* |
| H16B | 0.4730 | 0.3717 | 0.7470 | 0.062* |
| C15 | 0.54066 (12) | 0.46790 (13) | 0.68498 (17) | 0.0498 (5) |
| H15A | 0.5025 | 0.4610 | 0.6076 | 0.060* |
| H15B | 0.5198 | 0.5207 | 0.7172 | 0.060* |
| C22 | 0.73461 (14) | 0.60954 (14) | 0.51764 (18) | 0.0550 (5) |
| H22A | 0.7543 | 0.6581 | 0.5738 | 0.066* |
| H22B | 0.6725 | 0.5973 | 0.5061 | 0.066* |
| C25 | 0.87969 (13) | 0.54758 (15) | 0.57798 (18) | 0.0558 (5) |
| H25A | 0.9160 | 0.4938 | 0.6086 | 0.067* |
| H25B | 0.8973 | 0.5975 | 0.6325 | 0.067* |
| C26 | 0.8496 (2) | 0.68403 (19) | 0.3141 (2) | 0.0871 (8) |
| H26A | 0.8137 | 0.7375 | 0.2846 | 0.131* |
| H26B | 0.8315 | 0.6336 | 0.2617 | 0.131* |
| H26C | 0.9107 | 0.6985 | 0.3254 | 0.131* |
| C24 | 0.89265 (16) | 0.57653 (16) | 0.4690 (2) | 0.0669 (6) |
| H24A | 0.9547 | 0.5908 | 0.4823 | 0.080* |
| H24B | 0.8764 | 0.5256 | 0.4156 | 0.080* |
| C23 | 0.74686 (15) | 0.63994 (15) | 0.40848 (18) | 0.0606 (6) |
| H23A | 0.7244 | 0.5921 | 0.3520 | 0.073* |
| H23B | 0.7128 | 0.6960 | 0.3821 | 0.073* |
| | | | | |

Atomic displacement parameters $(Å^2)$

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|------------|------------|------------|-------------|------------|-------------|
| 01 | 0.0457 (8) | 0.0621 (8) | 0.0380 (8) | -0.0015 (6) | 0.0053 (6) | -0.0015 (7) |
| N1 | 0.0393 (8) | 0.0425 (8) | 0.0440 (9) | 0.0020 (7) | 0.0145 (7) | -0.0015 (7) |

| N2 | 0.0452 (9) | 0.0371 (8) | 0.0313 (8) | 0.0021 (6) | 0.0110 (7) | 0.0003 (7) |
|-----|-------------|-------------|-------------|--------------|-------------|--------------|
| N3 | 0.0609 (10) | 0.0404 (9) | 0.0488 (10) | 0.0074 (8) | 0.0145 (8) | -0.0025 (8) |
| N4 | 0.0605 (10) | 0.0466 (9) | 0.0479 (10) | -0.0058 (8) | 0.0206 (8) | -0.0001 (8) |
| N5 | 0.0497 (9) | 0.0447 (9) | 0.0377 (9) | -0.0012 (7) | 0.0122 (7) | 0.0013 (8) |
| N6 | 0.0429 (9) | 0.0427 (8) | 0.0428 (9) | 0.0008 (7) | 0.0145 (7) | 0.0079 (7) |
| N7 | 0.0748 (12) | 0.0513 (10) | 0.0556 (11) | -0.0103 (9) | 0.0261 (10) | 0.0090 (9) |
| C1 | 0.0831 (16) | 0.0589 (13) | 0.0493 (13) | -0.0106 (12) | 0.0016 (12) | -0.0094 (11) |
| C2 | 0.0689 (14) | 0.0371 (10) | 0.0406 (11) | -0.0035 (9) | 0.0097 (10) | -0.0022 (9) |
| C3 | 0.0830 (16) | 0.0438 (11) | 0.0445 (12) | -0.0027 (11) | 0.0225 (12) | -0.0111 (10) |
| C4 | 0.0633 (13) | 0.0445 (11) | 0.0530 (12) | 0.0023 (10) | 0.0270 (11) | -0.0069 (10) |
| C5 | 0.0499 (11) | 0.0331 (9) | 0.0404 (11) | 0.0029 (8) | 0.0146 (9) | -0.0004 (8) |
| C6 | 0.0443 (10) | 0.0337 (9) | 0.0366 (10) | 0.0010 (8) | 0.0118 (8) | 0.0000 (8) |
| C7 | 0.0499 (11) | 0.0401 (10) | 0.0442 (11) | -0.0021 (8) | 0.0111 (9) | -0.0003 (9) |
| C8 | 0.0401 (10) | 0.0388 (9) | 0.0353 (10) | 0.0005 (8) | 0.0118 (8) | -0.0003 (8) |
| C9 | 0.0428 (10) | 0.0382 (9) | 0.0358 (10) | 0.0028 (8) | 0.0102 (9) | 0.0049 (9) |
| C10 | 0.109 (2) | 0.0567 (14) | 0.0793 (18) | 0.0317 (14) | 0.0299 (16) | 0.0022 (13) |
| C14 | 0.0561 (12) | 0.0420 (10) | 0.0409 (11) | 0.0034 (9) | 0.0182 (9) | 0.0005 (9) |
| C12 | 0.0704 (14) | 0.0486 (11) | 0.0365 (11) | 0.0082 (10) | 0.0125 (10) | -0.0037 (10) |
| C13 | 0.0628 (13) | 0.0489 (11) | 0.0428 (11) | 0.0087 (10) | 0.0113 (10) | 0.0053 (10) |
| C11 | 0.0729 (14) | 0.0542 (12) | 0.0475 (12) | 0.0109 (11) | 0.0224 (11) | -0.0067 (11) |
| C21 | 0.0855 (17) | 0.0564 (13) | 0.0788 (17) | -0.0015 (12) | 0.0349 (14) | -0.0081 (13) |
| C19 | 0.0576 (12) | 0.0496 (11) | 0.0488 (12) | -0.0002 (10) | 0.0160 (10) | 0.0078 (10) |
| C20 | 0.0658 (13) | 0.0564 (12) | 0.0389 (11) | -0.0026 (10) | 0.0175 (10) | 0.0048 (10) |
| C18 | 0.0689 (14) | 0.0559 (12) | 0.0400 (11) | -0.0113 (11) | 0.0152 (10) | -0.0015 (10) |
| C17 | 0.0530 (12) | 0.0550 (12) | 0.0439 (11) | -0.0058 (10) | 0.0072 (10) | 0.0039 (10) |
| C16 | 0.0478 (11) | 0.0562 (12) | 0.0568 (13) | -0.0028 (9) | 0.0251 (10) | -0.0066 (11) |
| C15 | 0.0408 (11) | 0.0527 (12) | 0.0568 (12) | 0.0077 (9) | 0.0169 (9) | 0.0004 (10) |
| C22 | 0.0590 (13) | 0.0463 (11) | 0.0610 (14) | 0.0050 (10) | 0.0210 (11) | 0.0112 (11) |
| C25 | 0.0480 (12) | 0.0617 (13) | 0.0589 (13) | -0.0020 (10) | 0.0189 (10) | 0.0096 (11) |
| C26 | 0.121 (2) | 0.0796 (18) | 0.0699 (17) | -0.0179 (16) | 0.0443 (16) | 0.0146 (14) |
| C24 | 0.0666 (14) | 0.0676 (14) | 0.0772 (16) | -0.0018 (12) | 0.0379 (13) | 0.0128 (13) |
| C23 | 0.0716 (15) | 0.0481 (12) | 0.0570 (14) | -0.0014 (11) | 0.0136 (11) | 0.0137 (11) |
| | | | | | | |

Geometric parameters (Å, °)

| 01—С9 | 1.220 (2) | C14—H14A | 0.9700 |
|--------|-----------|----------|-----------|
| N1-C9 | 1.360 (2) | C14—H14B | 0.9700 |
| N1—C5 | 1.410(2) | C12—C11 | 1.509 (3) |
| N1—C15 | 1.459 (2) | C12—H12A | 0.9700 |
| N2-C14 | 1.465 (2) | C12—H12B | 0.9700 |
| N2—C12 | 1.468 (2) | C13—H13A | 0.9700 |
| N2—C8 | 1.476 (2) | C13—H13B | 0.9700 |
| N3—C13 | 1.445 (2) | C11—H11A | 0.9700 |
| N3—C11 | 1.449 (3) | C11—H11B | 0.9700 |
| N3—C10 | 1.453 (3) | C21—H21A | 0.9600 |
| N4—C19 | 1.458 (3) | C21—H21B | 0.9600 |
| N4—C18 | 1.461 (2) | C21—H21C | 0.9600 |
| N4—C21 | 1.464 (3) | C19—C20 | 1.507 (3) |
| | | | |

| N5-C16 | 1447(2) | C19—H19A | 0 9700 |
|---------------------------|----------------------|---------------|-------------|
| N5-C17 | 1 454 (2) | C19—H19B | 0.9700 |
| N5-C20 | 1 461 (2) | C20—H20A | 0.9700 |
| N6-C25 | 1.454(2) | C_{20} H20R | 0.9700 |
| N6-C22 | 1.462 (2) | C18 - C17 | 1 507 (3) |
| N6-C8 | 1.102(2) 1 464(2) | C18 - H18A | 0.9700 |
| N7C23 | 1.404(2) 1.448(3) | C18—H18B | 0.9700 |
| N7C24 | 1.456 (3) | C17_H17A | 0.9700 |
| N7 C26 | 1.450 (3) | C17 H17R | 0.9700 |
| 11 - 220 | 1.459 (3) | C16 C15 | 1.527(3) |
| $C_1 = H_1 A$ | 0.9600 | C16 H16A | 1.327(3) |
| | 0.9000 | | 0.9700 |
| | 0.9000 | | 0.9700 |
| | 0.9000 | CI5—HI5A | 0.9700 |
| $C_2 = C_3$ | 1.388 (3) | C15—H15B | 0.9700 |
| | 1.393 (3) | C22—C23 | 1.508 (3) |
| | 1.387 (3) | C22—H22A | 0.9700 |
| C3—H3A | 0.9300 | C22—H22B | 0.9700 |
| C4—C5 | 1.376 (3) | C25—C24 | 1.505 (3) |
| C4—H4A | 0.9300 | C25—H25A | 0.9700 |
| C5—C6 | 1.393 (2) | C25—H25B | 0.9700 |
| C6—C7 | 1.384 (2) | C26—H26A | 0.9600 |
| C6—C8 | 1.519 (2) | C26—H26B | 0.9600 |
| С7—Н7А | 0.9300 | C26—H26C | 0.9600 |
| C8—C9 | 1.565 (2) | C24—H24A | 0.9700 |
| C10—H10A | 0.9600 | C24—H24B | 0.9700 |
| C10—H10B | 0.9600 | C23—H23A | 0.9700 |
| C10—H10C | 0.9600 | C23—H23B | 0.9700 |
| C14—C13 | 1.509 (3) | | |
| | | | |
| C9—N1—C5 | 111.34 (14) | N3—C11—C12 | 110.32 (17) |
| C9—N1—C15 | 123.09 (16) | N3—C11—H11A | 109.6 |
| C5—N1—C15 | 125.23 (15) | C12—C11—H11A | 109.6 |
| C14—N2—C12 | 109.19 (14) | N3—C11—H11B | 109.6 |
| C14—N2—C8 | 113.78 (13) | C12—C11—H11B | 109.6 |
| C12—N2—C8 | 113.39 (13) | H11A—C11—H11B | 108.1 |
| C13—N3—C11 | 108.67 (15) | N4—C21—H21A | 109.5 |
| C13—N3—C10 | 111.86 (17) | N4—C21—H21B | 109.5 |
| C11—N3—C10 | 111.54 (17) | H21A—C21—H21B | 109.5 |
| C19—N4—C18 | 109.32 (15) | N4—C21—H21C | 109.5 |
| C19—N4—C21 | 109.44 (17) | H21A—C21—H21C | 109.5 |
| C18—N4—C21 | 110.04 (16) | H21B—C21—H21C | 109.5 |
| C16—N5—C17 | 113.91 (16) | N4—C19—C20 | 111.48 (17) |
| C16—N5—C20 | 111.57 (15) | N4—C19—H19A | 109.3 |
| C17—N5—C20 | 108.67 (15) | C20—C19—H19A | 109.3 |
| $C_{25} = N_{6} = C_{22}$ | 108.43 (15) | N4—C19—H19B | 109.3 |
| $C_{25} = N_{6} = C_{8}$ | 116 32 (14) | C20-C19-H19B | 109.3 |
| $C_{22} = N_{6} = C_{8}$ | 115.95 (14) | H19A—C19—H19B | 108.0 |
| C_{23} N7 C_{24} | 110.28 (16) | N5-C20-C19 | 110 71 (16) |
| 023 117 024 | 110.20 (10) | 113 020 017 | 110.71 (10) |

| C23—N7—C26 | 110.92 (19) | N5—C20—H20A | 109.5 |
|--|--------------------------|--|---------------------|
| C24—N7—C26 | 111.1 (2) | С19—С20—Н20А | 109.5 |
| C2—C1—H1A | 109.5 | N5—C20—H20B | 109.5 |
| C2—C1—H1B | 109.5 | С19—С20—Н20В | 109.5 |
| H1A—C1—H1B | 109.5 | H20A—C20—H20B | 108.1 |
| C2—C1—H1C | 109.5 | N4—C18—C17 | 111.87 (16) |
| H1A—C1—H1C | 109.5 | N4—C18—H18A | 109.2 |
| H1B—C1—H1C | 109.5 | C17—C18—H18A | 109.2 |
| C3—C2—C7 | 118.01 (18) | N4—C18—H18B | 109.2 |
| C3—C2—C1 | 121.10 (19) | C17—C18—H18B | 109.2 |
| C7—C2—C1 | 120.9 (2) | H18A—C18—H18B | 107.9 |
| C4—C3—C2 | 122.57 (19) | N5—C17—C18 | 109.94 (16) |
| C4—C3—H3A | 118.7 | N5—C17—H17A | 109.7 |
| C2—C3—H3A | 118.7 | С18—С17—Н17А | 109.7 |
| C5-C4-C3 | 117.70 (19) | N5-C17-H17B | 109.7 |
| C5-C4-H4A | 121.1 | C18—C17—H17B | 109.7 |
| C3-C4-H4A | 121.1 | H17A—C17—H17B | 108.2 |
| C4-C5-C6 | 121.85 (18) | N5-C16-C15 | 113.69 (16) |
| C4—C5—N1 | 128.26 (17) | N5-C16-H16A | 108.8 |
| C6-C5-N1 | 109 88 (15) | C15—C16—H16A | 108.8 |
| C7—C6—C5 | 118.97 (17) | N5-C16-H16B | 108.8 |
| C7—C6—C8 | 131.72 (17) | C15—C16—H16B | 108.8 |
| C_{5} C_{6} C_{8} | 109.27(15) | H16A—C16—H16B | 107.7 |
| C6-C7-C2 | 120 87 (18) | N1-C15-C16 | 112 12 (15) |
| C6-C7-H7A | 119.6 | N1-C15-H15A | 109.2 |
| $C_2 - C_7 - H_7 A$ | 119.6 | C16— $C15$ — $H15A$ | 109.2 |
| N6-C8-N2 | 107 37 (13) | N1-C15-H15B | 109.2 |
| N6-C8-C6 | 117 46 (14) | C16-C15-H15B | 109.2 |
| $N_2 - C_8 - C_6$ | 117.40(14) 111.11(14) | H15A - C15 - H15B | 107.9 |
| N6-C8-C9 | 107.98 (13) | N6-C22-C23 | 107.9 108.37(17) |
| N2 - C8 - C9 | 111 87 (14) | N6_C22_H22A | 110.0 |
| C6 - C8 - C9 | 100.97(14) | C_{23} C_{22} H_{22A} | 110.0 |
| 01 - C9 - N1 | 100.97(14) 125.35(17) | N6_C22_H22B | 110.0 |
| 01 - 09 - 08 | 125.35(17) 126.21(16) | $C_{22} = H_{22}$ | 110.0 |
| N1 C9 C8 | 120.21(10) 108.44(15) | H22A C22 H22B | 108.4 |
| $N_1 = C_2 = C_3$ $N_3 = C_{10} = H_{10A}$ | 100.5 | N6 C25 C24 | 108.4 108.67(17) |
| N3_C10_H10B | 109.5 | N6-C25-H25A | 110.0 |
| $H_{10A} = C_{10} = H_{10B}$ | 109.5 | $C_{23} = C_{23} = H_{23}$ | 110.0 |
| $N_2 C_{10} H_{10}C$ | 109.5 | $N_{12} = 0.25 = 0.25 R$ | 110.0 |
| $H_{10A} = C_{10} = H_{10C}$ | 109.5 | $C_{23} = C_{23} = C$ | 110.0 |
| H10R C10 H10C | 109.5 | 1254 - 225 - 1125B | 108.3 |
| $\frac{1100}{10} - \frac{10}{110}$ | 107.5 | N7 C26 H26A | 108.5 |
| $N_2 = C_{14} = C_{15}$ | 100.6 | $N/-C_{20}$ -H26R | 109.5 |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 109.0 | $H_{20} = H_{20}$ | 109.5 |
| $\mathbf{N}_{1}^{T} = \mathbf{C}_{14}^{T} = \mathbf{\Pi}_{14}^{T} \mathbf{H}_{14}^{T} \mathbf{H}_{14}^{$ | 109.0 | $\mathbf{N7} \mathbf{C26} \mathbf{H26C}$ | 107.5 |
| $\begin{array}{cccccccccccccccccccccccccccccccccccc$ | 109.0 | $H_{26A} = C_{26} = H_{26C}$ | 109.5 |
| $U_{1J} = U_{14} = \Pi_{14} \square \square$ | 109.0 | $H_{20} = C_{20} = H_{20} = H$ | 109.3 |
| $\frac{1114A}{12} = 0.14 = 0.14$ | 100.1 | $\mathbf{N7} \mathbf{C24} \mathbf{C25}$ | 107.3 |
| NZ-UIZ-UII | 110.23 (16) | N/-C24-C23 | 111.09 (19) |

| N2—C12—H12A | 109.6 | N7—C24—H24A | 109.4 |
|---------------|-------------|---------------|-------------|
| C11—C12—H12A | 109.6 | C25—C24—H24A | 109.4 |
| N2—C12—H12B | 109.6 | N7—C24—H24B | 109.4 |
| C11—C12—H12B | 109.6 | C25—C24—H24B | 109.4 |
| H12A—C12—H12B | 108.1 | H24A—C24—H24B | 108.0 |
| N3—C13—C14 | 111.04 (16) | N7—C23—C22 | 111.87 (18) |
| N3—C13—H13A | 109.4 | N7—C23—H23A | 109.2 |
| C14—C13—H13A | 109.4 | С22—С23—Н23А | 109.2 |
| N3—C13—H13B | 109.4 | N7—C23—H23B | 109.2 |
| C14—C13—H13B | 109.4 | С22—С23—Н23В | 109.2 |
| H13A—C13—H13B | 108.0 | H23A—C23—H23B | 107.9 |
| | | | |

Hydrogen-bond geometry (Å, °)

| D—H···A | D—H | H···A | D····A | <i>D</i> —H··· <i>A</i> |
|-----------------------------------|------|-------|-----------|-------------------------|
| C19—H19 <i>B</i> …O1 ⁱ | 0.97 | 2.64 | 3.388 (2) | 135 |

Symmetry code: (i) x, -y+1/2, z+1/2.