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catena-Poly[[silver(I)- μ -2-[(pyrazin-2-yl)- κ^2 N¹:N⁴]aminomethyl]phenol] nitrate]

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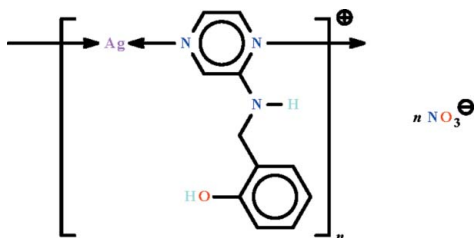
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Key indicators: single-crystal X-ray study; $T = 294$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.075; wR factor = 0.224; data-to-parameter ratio = 16.1.

The Ag^I atom in the polycationic salt, {[Ag(C₁₁H₁₁N₃O)]NO₃]_{*n*}, shows a linear coordination [N—Ag—N = 175.0 (2)°]; the polymeric nature arises from bridging by the pyrazine portion of the ligand, resulting in chains extending parallel to [100]. The NO₃[−] counter-ions surround the polymeric chain and interact only weakly with it [Ag···O = 2.701 (4) and 2.810 (5) Å]. Adjacent chains are linked into a three-dimensional network by O—H···O and N—H···O hydrogen bonds.

Related literature

For the structure of 2-[(pyrazin-2-yl)amino]methyl]phenol, see: Gao & Ng (2012).



Experimental

Crystal data

 [Ag(C₁₁H₁₁N₃O)]NO₃
 $M_r = 371.11$

 Monoclinic, $P2_1/c$
 $a = 7.1265$ (9) Å

 $b = 9.5249$ (14) Å

 $c = 18.654$ (2) Å

 $\beta = 97.240$ (4)°

 $V = 1256.1$ (3) Å³
 $Z = 4$

 Mo $K\alpha$ radiation

 $\mu = 1.63$ mm^{−1}
 $T = 293$ K

 $0.27 \times 0.18 \times 0.13$ mm

Data collection

Rigaku R-Axis RAPID IP diffractometer

Absorption correction: multi-scan (ABSCOR; Higashi, 1995)

 $T_{\min} = 0.668$, $T_{\max} = 0.817$

11969 measured reflections

2864 independent reflections

 2038 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.125$
 $S = 1.02$

2864 reflections

182 parameters

H-atom parameters constrained

 $\Delta\rho_{\max} = 0.64$ e Å^{−3}
 $\Delta\rho_{\min} = -0.77$ e Å^{−3}

Table 1

Selected bond lengths (Å).

Ag1—N1	2.172 (4)	Ag1—N2 ⁱ	2.195 (4)
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 Symmetry code: (i) $x - 1, y, z$.

Table 2

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
O1—H1···O3 ⁱⁱ	0.84	1.96	2.795 (6)	171
N3—H3···O4 ⁱⁱⁱ	0.88	2.22	2.982 (6)	145

 Symmetry codes: (ii) $x, -y + \frac{3}{2}, z + \frac{1}{2}$; (iii) $-x, -y + 1, -z + 1$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalClear* (Rigaku/MSK, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5602).

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catena-Poly[[silver(I)- μ -2-[(pyrazin-2-yl- κ^2 N¹:N⁴)aminomethyl]phenol] nitrate]

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S1. Comment

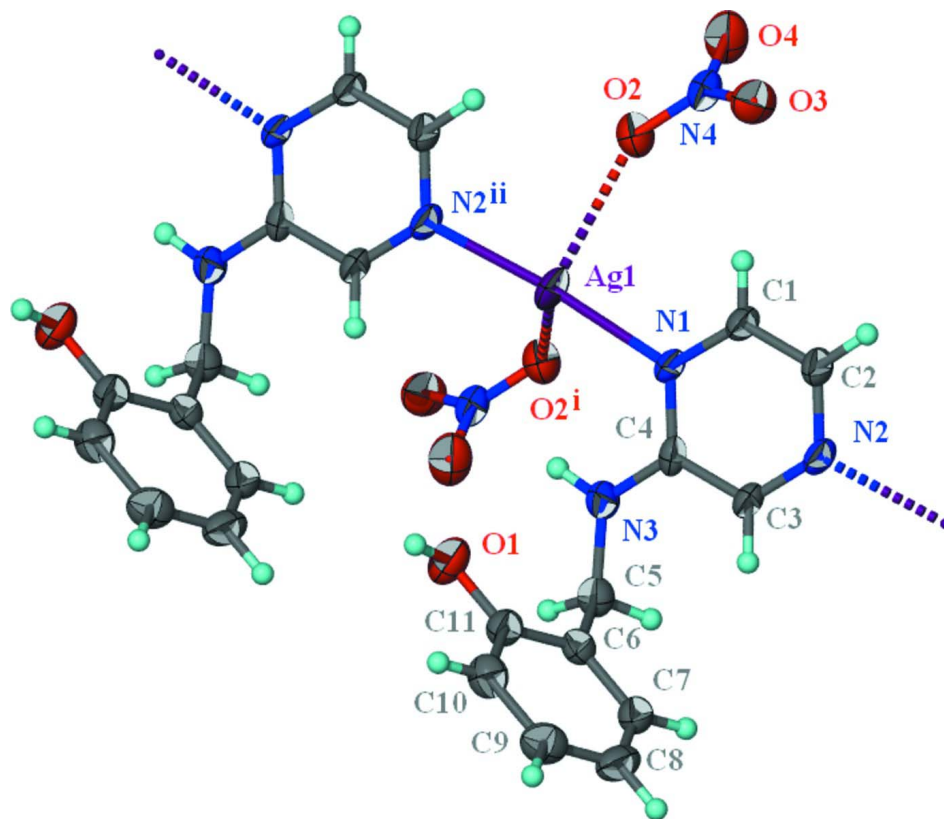
A recent study reports 2-[(pyrazin-2-ylamino)methyl]phenol, a reduced Schiff-base that possesses an acidic phenolic group (Gao & Ng, 2012). The reaction with silver nitrate yields polycationic [Ag(C₁₁H₁₁N₃O)]_n nNO₃ (Scheme I). The polymeric nature arises from bridging by the pyrazine portion of the ligand. The counterions surround the chain and interact only weakly with it [Ag \cdots O 2.701 (4), 2.810 (5) Å] (Fig. 1). Adjacent chains are linked into a three-dimensional network by O–H \cdots O and N–H \cdots O hydrogen bonds (Table 1).

S2. Experimental

An acetonitrile solution (10 ml) of silver nitrate (1 mmol) was added to a methanol solution (5 ml) of 2-[(pyrazin-2-yl-amino)methyl]phenol (1 mmol). The solution was filtered and then side aside, away from light, for the growth of crystals. Colorless crystals were obtained after several days.

S3. Refinement

Hydrogen atoms were placed in calculated positions (C–H 0.93–0.97, N–H 0.88, O–H 0.84 Å) and were included in the refinement in the riding model approximation, with $U(\text{H})$ set to 1.2–1.5 $U(\text{C}, \text{N}, \text{O})$.

**Figure 1**

Thermal ellipsoid plot (Barbour, 2001) of the polymeric chain structure of $[\text{Ag}(\text{C}_{11}\text{H}_{11}\text{N}_3\text{O})]_n\text{NO}_3$ at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

catena-Poly[[silver(I)- μ -2-[(pyrazin-2-yl- $\kappa^2\text{N}^1:\text{N}^4)$ aminomethyl]phenol] nitrate]

Crystal data

$[\text{Ag}(\text{C}_{11}\text{H}_{11}\text{N}_3\text{O})]\text{NO}_3$

$M_r = 371.11$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 7.1265$ (9) Å

$b = 9.5249$ (14) Å

$c = 18.654$ (2) Å

$\beta = 97.240$ (4)°

$V = 1256.1$ (3) Å³

$Z = 4$

$F(000) = 736$

$D_x = 1.962$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 5867 reflections

$\theta = 3.1\text{--}27.5^\circ$

$\mu = 1.63$ mm⁻¹

$T = 293$ K

Prism, colorless

$0.27 \times 0.18 \times 0.13$ mm

Data collection

Rigaku R-AXIS RAPID IP

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω scan

Absorption correction: multi-scan

(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.668$, $T_{\max} = 0.817$

11969 measured reflections

2864 independent reflections

2038 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.042$

$\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 3.1^\circ$

$h = -9 \rightarrow 8$

$k = -12 \rightarrow 12$

$l = -24 \rightarrow 24$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.125$

$S = 1.02$

2864 reflections

182 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0508P)^2 + 4.1394P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.64 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.77 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Ag1	0.06666 (5)	0.59659 (5)	0.58246 (2)	0.05110 (18)
O1	0.3023 (5)	0.4649 (5)	0.8136 (2)	0.0571 (10)
H1	0.2582	0.5291	0.8375	0.086*
O2	-0.0170 (6)	0.6917 (5)	0.4458 (2)	0.0575 (11)
O3	0.1837 (5)	0.8322 (5)	0.4064 (2)	0.0574 (11)
O4	-0.0257 (6)	0.7272 (6)	0.3321 (2)	0.0658 (12)
N1	0.3716 (5)	0.5848 (4)	0.5842 (2)	0.0316 (8)
N2	0.7616 (5)	0.5983 (4)	0.5896 (2)	0.0361 (9)
N3	0.4029 (6)	0.4097 (5)	0.6707 (2)	0.0434 (10)
H3	0.2785	0.4081	0.6661	0.052*
N4	0.0450 (6)	0.7509 (5)	0.3943 (2)	0.0414 (10)
C1	0.4556 (6)	0.6685 (5)	0.5406 (3)	0.0352 (10)
H1A	0.3804	0.7245	0.5078	0.042*
C2	0.6473 (7)	0.6755 (5)	0.5421 (3)	0.0397 (11)
H2	0.6993	0.7341	0.5100	0.048*
C3	0.6828 (6)	0.5159 (6)	0.6338 (3)	0.0368 (11)
H3A	0.7599	0.4648	0.6683	0.044*
C4	0.4845 (6)	0.5027 (5)	0.6304 (2)	0.0328 (10)
C5	0.5036 (8)	0.3108 (6)	0.7217 (3)	0.0442 (12)
H5A	0.6001	0.2642	0.6980	0.053*
H5B	0.4151	0.2395	0.7333	0.053*
C6	0.5967 (7)	0.3745 (5)	0.7914 (3)	0.0371 (11)
C7	0.7896 (7)	0.3554 (6)	0.8134 (3)	0.0451 (12)
H7	0.8625	0.3062	0.7839	0.054*
C8	0.8733 (8)	0.4086 (6)	0.8783 (3)	0.0510 (14)
H8	1.0016	0.3941	0.8926	0.061*
C9	0.7677 (8)	0.4832 (7)	0.9221 (3)	0.0529 (14)
H9	0.8247	0.5192	0.9658	0.063*
C10	0.5791 (8)	0.5042 (6)	0.9013 (3)	0.0472 (13)
H10	0.5082	0.5552	0.9308	0.057*
C11	0.4926 (7)	0.4498 (6)	0.8363 (3)	0.0404 (11)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Ag1	0.0201 (2)	0.0684 (3)	0.0657 (3)	0.00250 (18)	0.00874 (18)	0.0059 (2)
O1	0.039 (2)	0.071 (3)	0.062 (3)	0.008 (2)	0.0080 (19)	-0.008 (2)
O2	0.050 (2)	0.066 (3)	0.059 (2)	0.000 (2)	0.018 (2)	0.015 (2)
O3	0.039 (2)	0.064 (3)	0.067 (3)	-0.0142 (19)	-0.0030 (19)	0.011 (2)
O4	0.048 (2)	0.102 (4)	0.047 (2)	-0.010 (2)	0.0027 (19)	-0.012 (2)
N1	0.0186 (16)	0.038 (2)	0.038 (2)	0.0005 (15)	0.0023 (15)	0.0005 (17)
N2	0.0183 (16)	0.044 (2)	0.046 (2)	-0.0003 (16)	0.0028 (16)	0.0007 (19)
N3	0.028 (2)	0.059 (3)	0.043 (2)	-0.006 (2)	0.0052 (18)	0.010 (2)
N4	0.028 (2)	0.046 (3)	0.050 (3)	-0.0013 (18)	0.0086 (19)	-0.002 (2)
C1	0.024 (2)	0.035 (3)	0.047 (3)	0.0019 (19)	0.005 (2)	0.004 (2)
C2	0.029 (2)	0.041 (3)	0.050 (3)	-0.002 (2)	0.012 (2)	0.008 (2)
C3	0.022 (2)	0.048 (3)	0.041 (3)	0.005 (2)	0.0058 (19)	0.002 (2)
C4	0.025 (2)	0.041 (3)	0.034 (2)	-0.0041 (19)	0.0101 (18)	-0.007 (2)
C5	0.042 (3)	0.046 (3)	0.045 (3)	0.002 (2)	0.010 (2)	0.003 (2)
C6	0.038 (2)	0.036 (3)	0.039 (3)	-0.001 (2)	0.009 (2)	0.007 (2)
C7	0.038 (3)	0.047 (3)	0.051 (3)	0.005 (2)	0.013 (2)	0.014 (3)
C8	0.035 (3)	0.049 (3)	0.067 (4)	0.002 (2)	-0.002 (3)	0.014 (3)
C9	0.052 (3)	0.053 (4)	0.051 (3)	-0.008 (3)	-0.005 (3)	0.003 (3)
C10	0.052 (3)	0.046 (3)	0.044 (3)	-0.002 (3)	0.010 (3)	-0.006 (2)
C11	0.035 (2)	0.040 (3)	0.047 (3)	0.001 (2)	0.007 (2)	0.005 (2)

Geometric parameters (\AA , $^\circ$)

Ag1—N1	2.172 (4)	C1—H1A	0.9300
Ag1—N2 ⁱ	2.195 (4)	C2—H2	0.9300
Ag1—O2	2.701 (4)	C3—C4	1.412 (6)
Ag1—O2 ⁱⁱ	2.810 (5)	C3—H3A	0.9300
O1—C11	1.376 (6)	C5—C6	1.511 (7)
O1—H1	0.8400	C5—H5A	0.9700
O2—N4	1.242 (6)	C5—H5B	0.9700
O3—N4	1.254 (6)	C6—C7	1.396 (7)
O4—N4	1.225 (6)	C6—C11	1.386 (7)
N1—C1	1.333 (6)	C7—C8	1.377 (8)
N1—C4	1.352 (6)	C7—H7	0.9300
N2—C3	1.314 (6)	C8—C9	1.376 (9)
N2—C2	1.344 (6)	C8—H8	0.9300
N2—Ag1 ⁱⁱⁱ	2.195 (4)	C9—C10	1.366 (8)
N3—C4	1.342 (6)	C9—H9	0.9300
N3—C5	1.461 (7)	C10—C11	1.390 (8)
N3—H3	0.8800	C10—H10	0.9300
C1—C2	1.364 (6)		
N1—Ag1—N2 ⁱ	175.01 (15)	C4—C3—H3A	119.0
N1—Ag1—O2	97.63 (13)	N1—C4—N3	118.3 (4)
N2 ⁱ —Ag1—O2	87.27 (14)	N1—C4—C3	119.3 (4)

N1—Ag1—O2 ⁱⁱ	93.12 (13)	N3—C4—C3	122.4 (5)
N2 ⁱ —Ag1—O2 ⁱⁱ	85.21 (14)	N3—C5—C6	115.3 (4)
O2—Ag1—O2 ⁱⁱ	98.24 (12)	N3—C5—H5A	108.5
C11—O1—H1	109.5	C6—C5—H5A	108.5
N4—O2—Ag1	145.6 (3)	N3—C5—H5B	108.5
C1—N1—C4	117.2 (4)	C6—C5—H5B	108.5
C1—N1—Ag1	119.1 (3)	H5A—C5—H5B	107.5
C4—N1—Ag1	123.6 (3)	C7—C6—C11	118.1 (5)
C3—N2—C2	117.9 (4)	C7—C6—C5	120.7 (5)
C3—N2—Ag1 ⁱⁱⁱ	122.6 (3)	C11—C6—C5	121.2 (5)
C2—N2—Ag1 ⁱⁱⁱ	119.1 (3)	C6—C7—C8	120.8 (5)
C4—N3—C5	125.3 (4)	C6—C7—H7	119.6
C4—N3—H3	117.3	C8—C7—H7	119.6
C5—N3—H3	117.3	C9—C8—C7	120.2 (5)
O4—N4—O2	120.3 (5)	C9—C8—H8	119.9
O4—N4—O3	120.3 (5)	C7—C8—H8	119.9
O2—N4—O3	119.4 (5)	C8—C9—C10	120.0 (5)
N1—C1—C2	122.9 (5)	C8—C9—H9	120.0
N1—C1—H1A	118.6	C10—C9—H9	120.0
C2—C1—H1A	118.6	C9—C10—C11	120.3 (5)
C1—C2—N2	120.6 (4)	C9—C10—H10	119.8
C1—C2—H2	119.7	C11—C10—H10	119.8
N2—C2—H2	119.7	C6—C11—C10	120.5 (5)
N2—C3—C4	121.9 (4)	C6—C11—O1	116.7 (5)
N2—C3—H3A	119.0	C10—C11—O1	122.7 (5)
N1—Ag1—O2—N4	-19.9 (6)	C5—N3—C4—N1	178.2 (5)
N2 ⁱ —Ag1—O2—N4	161.0 (6)	C5—N3—C4—C3	-0.7 (8)
O2 ⁱⁱ —Ag1—O2—N4	-114.2 (6)	N2—C3—C4—N1	-5.1 (7)
O2—Ag1—N1—C1	26.1 (4)	N2—C3—C4—N3	173.8 (5)
O2 ⁱⁱ —Ag1—N1—C1	124.9 (4)	C4—N3—C5—C6	73.9 (6)
O2 ⁱⁱ —Ag1—N1—C4	-59.3 (4)	N3—C5—C6—C7	-124.6 (5)
Ag1—O2—N4—O4	152.0 (5)	N3—C5—C6—C11	57.3 (6)
Ag1—O2—N4—O3	-26.1 (9)	C11—C6—C7—C8	0.7 (8)
C4—N1—C1—C2	-1.2 (7)	C5—C6—C7—C8	-177.4 (5)
Ag1—N1—C1—C2	174.9 (4)	C6—C7—C8—C9	-0.8 (8)
N1—C1—C2—N2	-1.3 (8)	C7—C8—C9—C10	0.2 (9)
C3—N2—C2—C1	0.5 (7)	C8—C9—C10—C11	0.5 (9)
Ag1 ⁱⁱⁱ —N2—C2—C1	173.7 (4)	C7—C6—C11—C10	-0.1 (8)
C2—N2—C3—C4	2.6 (7)	C5—C6—C11—C10	178.1 (5)
Ag1 ⁱⁱⁱ —N2—C3—C4	-170.3 (4)	C7—C6—C11—O1	-179.0 (5)
C1—N1—C4—N3	-174.8 (4)	C5—C6—C11—O1	-0.9 (7)
Ag1—N1—C4—N3	9.2 (6)	C9—C10—C11—C6	-0.6 (8)
C1—N1—C4—C3	4.2 (7)	C9—C10—C11—O1	178.4 (5)
Ag1—N1—C4—C3	-171.8 (3)		

Symmetry codes: (i) $x-1, y, z$; (ii) $-x, -y+1, -z+1$; (iii) $x+1, y, z$.

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O1—H1 \cdots O3 ^{iv}	0.84	1.96	2.795 (6)	171
N3—H3 \cdots O4 ⁱⁱ	0.88	2.22	2.982 (6)	145

Symmetry codes: (ii) $-x, -y+1, -z+1$; (iv) $x, -y+3/2, z+1/2$.