

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

9-(3,4-Dimethoxyphenyl)-3,3,6,6-tetramethyl-1,2,3,4,5,6,7,8,9,10-decahydroacridine-1,8-dione

Rajni Kant,^a* Vivek K. Gupta,^a Kamini Kapoor,^a D. R. Patil,^b D. R. Chandam^b and Madhukar B. Deshmukh^b

^aX-ray Crystallography Laboratory, Post-Graduate Department of Physics & Electronics, University of Jammu, Jammu Tawi 180 006, India, and ^bDepartment of Chemistry, Shivaji University, Kolhapur, 416 004 (MS), India Correspondence e-mail: rkvk.paper11@gmail.com

Received 8 January 2013; accepted 23 January 2013

Key indicators: single-crystal X-ray study; T = 293 K; mean σ (C–C) = 0.006 Å; disorder in main residue; R factor = 0.059; wR factor = 0.127; data-to-parameter ratio = 10.6.

The asymmetric unit of the title compound, $C_{25}H_{31}NO_4$, contains two independent molecules. In one molecule, the benzene ring and an attached methoxy group were refined as disordered over two sets of sites in a 0.65 (4): 0.35 (4) ratio. In both molecules, the central ring of the acridinedione system adopts a flattened boat conformation. The four essentially planar atoms of this ring [maximum deviations = 0.006 (5) Å in both molecules] forms dihedral angles of 86.8 (2) and 87.6 (2)°, respectively, with the major and minor components in the disordered benzene ring and 87.3 (2)° with the benzene ring in the fully ordered molecule. The two outer rings of the acridinedione system adopt sofa conformations in both molecules. In the crystal, $N-H\cdots O$ hydrogen bonds form two independent chains along [100]. $C-H\cdots O$ hydrogen bonds link the chains, forming a three-dimensional network.

Related literature

For applications of acridines, see: Murugan *et al.* (1998); Leon *et al.* (2008); Josephrajan *et al.* (2005); Srividya *et al.* (1998, 1996). For related structures, see: Balamurugan *et al.* (2009); Zhao & Teng (2008); Kant *et al.* (2013*a*,*b*). For ring conformations, see: Duax & Norton (1975).



Experimental

Crystal data

 $C_{25}H_{31}NO_4$ $M_r = 409.51$ Orthorhombic, $Pca2_1$ a = 14.1607 (6) Å b = 15.3126 (10) Å c = 21.1196 (14) Å

Data collection

Oxford Diffraction Xcalibur Sapphire3 diffractometer Absorption correction: multi-scan (*CrysAlis PRO*; Oxford Diffraction, 2010) $T_{min} = 0.672, T_{max} = 1.000$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.059$ $wR(F^2) = 0.127$ S = 1.016386 reflections 604 parameters $V = 4579.5 (5) Å^{3}$ Z = 8 Mo K\alpha radiation \mu = 0.08 mm^{-1} T = 293 K 0.3 \times 0.2 \times 0.2 mm

14639 measured reflections 6386 independent reflections 4075 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.055$

 $\begin{array}{l} 53 \text{ restraints} \\ \text{H-atom parameters constrained} \\ \Delta \rho_{max} = 0.17 \text{ e } \text{ Å}^{-3} \\ \Delta \rho_{min} = -0.17 \text{ e } \text{ Å}^{-3} \end{array}$

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N12A - H12A \cdots O1A^{i}$	0.86	2.07	2.846 (4)	150
$N12B - H12B \cdot \cdot \cdot O8B^{ii}$	0.86	2.05	2.885 (4)	165
$C2A - H2AA \cdots O1B^{ii}$	0.97	2.45	3.400 (6)	166
$C16A - H16D \cdots O25A^{iii}$	0.96	2.61	3.477 (12)	150
$C7A - H7AB \cdots O27B^{iv}$	0.97	2.47	3.432 (6)	169
$C7B - H7BB \cdots O8A^{i}$	0.97	2.57	3.507 (5)	162

Symmetry codes: (i) $x - \frac{1}{2}, -y + 1, z$; (ii) $x + \frac{1}{2}, -y, z$; (iii) $-x + \frac{1}{2}, y, z - \frac{1}{2}$; (iv) $-x, -y + 1, z + \frac{1}{2}$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

RK acknowledges the Department of Science & Technology for access to the single-crystal X-ray diffractometer sanctioned as a National Facility under project No. SR/S2/ CMP-47/2003.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH5574).

References

- Balamurugan, P., Jagan, R., Thiagarajan, V. M., Yamin, B. & Sivakumar, K. (2009). Acta Cryst. E65, o271.
- Duax, W. L. & Norton, D. A. (1975). Atlas of Steroid Structures, Vol. 1. New York: Plenum Press.
- Farrugia, L. J. (2012). J. Appl. Cryst. 45, 849-854.
- Josephrajan, T., Ramakrishnan, V. T., Kathiravan, G. & Muthumary, J. (2005). *ARKIVOC*, pp. 124–136.

- Kant, R., Gupta, V. K., Kapoor, K., Patil, D. R., Patil, P. P. & Deshmukh, M. B. (2013a). Acta Cryst. E69, o100.
- Kant, R., Gupta, V. K., Kapoor, K., Patil, D. R., Jagadale, S. D. & Deshmukh, M. B. (2013b). Acta Cryst. E69, o101.
- Leon, R., Rios, C., Contelles, J. M., Lopez, G. M., Garcia, A. G. & Villarroya, M. (2008). Eur. J. Med. Chem. 43, 668–674.
- Murugan, P., Shanmugasundaram, P., Ramakrishnan, V. T., Venkatachalapathy, B., Srividya, N., Ramamurthy, P., Gunasekaran, K. & Velmurugan, D. (1998). J. Chem. Soc. Perkin Trans. 2, pp. 999–1003.
- Oxford Diffraction (2010). CrysAlis PRO. Oxford Diffraction Ltd, Yarnton, England.
- Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.
- Spek, A. L. (2009). Acta Cryst. D65, 148-155.
- Srividya, N., Ramamurthy, P. & Ramakrishnan, V. T. (1998). Spectrochim. Acta Part A, 54, 245–253.
- Srividya, N., Ramamurthy, P., Shanmugasundaram, P. & Ramakrishnan, V. T. (1996). J. Org. Chem. 61, 5083–5089.
- Zhao, L.-L. & Teng, D. (2008). Acta Cryst. E64, o1772-o1773.

supporting information

Acta Cryst. (2013). E69, o297-o298 [doi:10.1107/S1600536813002250]

9-(3,4-Dimethoxyphenyl)-3,3,6,6-tetramethyl-1,2,3,4,5,6,7,8,9,10-decahydro-acridine-1,8-dione

Rajni Kant, Vivek K. Gupta, Kamini Kapoor, D. R. Patil, D. R. Chandam and Madhukar B. Deshmukh

S1. Comment

The 1,4- dihydropyridine (DHP) nucleus act as a versatile intermediate for the synthesis of several pharmaceuticals together with those of cardiovascular drugs and as a calcium channel modulators, laser dyes and photo initiators (Leon *et al.*, 2008). Acridines, the earliest known antibiotics and are toxic towards bacteria. Some acridinedione derivatives show good inhibition against the pathogen Vibrio isolate-I (Josephrajan *et al.*, 2005). Certain acridine-1,8-diones exhibit fluorescence activities (Murugan *et al.*, 1998) and a few acridinedione derivatives also show photophysical (Srividya *et al.*, 1998) and electrochemical properties (Srividya *et al.*, 1996). Thus, the accurate description of crystal structures of substituted acridinediones are expected to provide useful information on the role of substituents in influencing molecular conformation which has a direct relationship to biological activity. This paper deals with the crystal structure of a 3,4-dimethoxyphenyl substituted tetramethyl acridinedione, (I).

The asymmetric unit of the title compound comprises of two crystallographically independent molecules, A and B (Fig.1). In molecule A, the benzene ring and one attached methoxy group is disordered over two sets of sites in a 0.65 (4):0.35 (4) ratio. Bond lengths and angles are normal and correspond to those observed in related structures (Balamurugan *et al.*, 2009; Zhao & Teng 2008; Kant *et al.* (2013*a,b*)). The central ring (C9/C10/C11/N12/C13/C14) of the acridinedione moiety adopts a flattened boat conformation (Δ Cs(C9A) = 1.32 & Δ Cs (C13A—C14A) = 10.43; Δ Cs(C9B) = 1.34 & Δ Cs (C13B—C14B) = 12.58) and the four essentially planar atoms (C10/C11/C13/C14) of this ring (maximum deviation-0.006 (5) Å for C11A and -0.006 (5) Å for C13B) forms a dihedral angle of 86.8 (2)° and 87.6 (2)° with the major and minor components of disorder benzene ring in molecule A and 87.3 (2)° with the benzene ring in molecule B. Both the outer rings adopt sofa conformations (Δ Cs (C3A) = 7.99; Δ Cs (C6A) = 4.61; Δ Cs (C3B) = 8.34; Δ Cs (C6B) = 8.93) (Duax & Norton, 1975). In the crystal N—H···O hydrogen bonds (Table 1) form two independent one-dimensional chains parallel to [100]. In addition weak C—H···O hydrogen bonds link chains to form a three-dimensional network (Fig. 2).

S2. Experimental

In a 50 ml round bottom flask, a mixture of dimedone (2 mmole), 3,4 - dimethoxy benzaldehyde 2(1 mmole) and ammonium acetate (1.2 mmole) in aqu. ethanol (7 ml) was stirred at room temperature for 5 min. To this [CMIM][HSO4] (3-carboxymethyl-1-methylimidazolium bisulfate)(20 mol %) was added and the reaction mixture heated at 348–351 K for 1.5 hrs. The progress of reaction was monitored by TLC. After completion of reaction, the mixture was gradually cooled to RT and poured on ice water under stirring. A precipitate was formed which was filtered and dried. The crude product was recrystallized from ethanol.

M.P.: 559–563 K, Yield:78%. IR(KBr): 3273, 3200, 3071, 2954, 1642, 1610 cm-1. 1H NMR(300 MHz, DMSO-d6): δ = 7.9 (s, 1H, NH); 7.2–6.2 (m, 3H, Ar—H); 5.0 (s, 1H, CH); 3.8 (s, 3H, OCH3); 3.7 (s, 3H, OCH3); 2.3–2.1 (m, 8H, CH2); 1.0 (s, 6H, CH3); 0.9 (s,6*H*, CH3).

S3. Refinement

All H atoms were positioned geometrically and were treated as riding on their parent C atoms, with N—H distances of 0.86 Å, C—H distances of 0.93–0.98 Å and with $U_{iso}(H) = 1.2U_{eq}(C/N)$ or $1.5U_{eq}(methyl C)$.



Figure 1

The asymmetric unit of the title compound with ellipsoids drawn at the 40% probability level. H atoms are shown as small spheres of arbitrary radii. Atoms labeled with 'C' are the minor component of disorder.



Figure 2

The packing arrangement of molecules viewed along the *b* axis. The dashed lines show intermolecular C—H \cdots O and N—H \cdots O hydrogen bonds. Only H atoms involved in hydrogen bonds are shown. The disorder is not shown.

9-(3,4-Dimethoxyphenyl)-3,3,6,6-tetramethyl-1,2,3,4,5,6,7,8,9,10-decahydroacridine-1,8-dione

Crystal data	
$C_{25}H_{31}NO_4$	F(000) = 1760
$M_r = 409.51$	$D_{\rm x} = 1.188 {\rm Mg} {\rm m}^{-3}$
Orthorhombic, $Pca2_1$	Mo K α radiation, $\lambda = 0.71073$ Å
Hall symbol: P 2c -2ac	Cell parameters from 6069 reflections
a = 14.1607 (6) Å	$\theta = 3.5 - 29.0^{\circ}$
b = 15.3126(10) Å	$\mu = 0.08 \text{ mm}^{-1}$
c = 21.1196(14) Å	T = 293 K
V = 4579.5 (5) Å ³	Block, yellow
Z = 8	$0.3 \times 0.2 \times 0.2 \text{ mm}$
Data collection	
Oxford Diffraction Xcalibur Sapphire3	Absorption correction: multi-scan
diffractometer	(CrysAlis PRO; Oxford Diffraction,
Radiation source: fine-focus sealed tube	$T_{\min} = 0.672, \ T_{\max} = 1.000$
Graphite monochromator	14639 measured reflections
Detector resolution: 16.1049 pixels mm ⁻¹	6386 independent reflections
ω scans	4075 reflections with $I > 2\sigma(I)$
	$R_{\rm int} = 0.055$

2010)

$\theta_{\text{max}} = 25.0^{\circ}, \ \theta_{\text{min}} = 3.5^{\circ}$	$k = -18 \rightarrow 13$
$h = -16 \rightarrow 16$	$l = -25 \rightarrow 21$
Refinement	
Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.059$	Hydrogen site location: inferred from
$wR(F^2) = 0.127$	neighbouring sites
<i>S</i> = 1.01	H-atom parameters constrained
6386 reflections	$w = 1/[\sigma^2(F_o^2) + (0.0377P)^2]$
604 parameters	where $P = (F_0^2 + 2F_c^2)/3$
53 restraints	$(\Delta/\sigma)_{\rm max} = 0.002$
Primary atom site location: structure-invariant	$\Delta \rho_{\rm max} = 0.17 \text{ e } \text{\AA}^{-3}$
direct methods	$\Delta \rho_{\rm min} = -0.17 \text{ e } \text{\AA}^{-3}$

Special details

Experimental. *CrysAlis PRO*, Oxford Diffraction Ltd., Version 1.171.34.40 (release 27–08-2010 CrysAlis171. NET) (compiled Aug 27 2010,11:50:40) Empirical absorption correction using spherical harmonics, implemented in SCALE3 ABSPACK scaling algorithm.

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	Occ. (<1)
O1A	0.41575 (18)	0.5230 (2)	0.62071 (19)	0.0548 (9)	
08A	0.26645 (19)	0.7813 (2)	0.73491 (19)	0.0666 (10)	
O27A	0.4145 (2)	0.4704 (2)	0.92959 (18)	0.0681 (10)	
N12A	0.0923 (2)	0.5381 (2)	0.6676 (2)	0.0474 (10)	
H12A	0.0388	0.5115	0.6674	0.057*	
C1A	0.3348 (3)	0.4965 (3)	0.6129 (3)	0.0428 (12)	
C2A	0.3141 (3)	0.4280 (3)	0.5643 (3)	0.0550 (13)	
H2AA	0.3702	0.3927	0.5582	0.066*	
H2AB	0.3000	0.4565	0.5244	0.066*	
C3A	0.2315 (3)	0.3681 (3)	0.5818 (2)	0.0458 (11)	
C4A	0.1465 (2)	0.4260 (3)	0.5952 (2)	0.0448 (11)	
H4AA	0.1228	0.4491	0.5555	0.054*	
H4AB	0.0969	0.3909	0.6140	0.054*	
C5A	0.0075 (3)	0.6611 (3)	0.7126 (2)	0.0445 (11)	
H5AA	-0.0369	0.6173	0.7272	0.053*	
H5AB	-0.0182	0.6874	0.6745	0.053*	
C6A	0.0183 (3)	0.7312 (3)	0.7634 (2)	0.0412 (10)	
C7A	0.0994 (3)	0.7905 (3)	0.7444 (2)	0.0465 (12)	
H7AA	0.0810	0.8232	0.7071	0.056*	
H7AB	0.1104	0.8322	0.7782	0.056*	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

C8A	0.1907 (3)	0.7431 (3)	0.7306 (2)	0.0439 (11)	
C9A	0.2752 (2)	0.6035 (3)	0.6965 (2)	0.0407 (11)	
H9AA	0.3221	0.6436	0.6786	0.049*	
C10A	0.2565 (2)	0.5319 (3)	0.6482 (2)	0.0369 (10)	
C11A	0.1688 (3)	0.5000 (3)	0.6383 (2)	0.0351 (11)	
C13A	0.0997 (2)	0.6177 (3)	0.6969 (2)	0.0382 (10)	
C14A	0.1847 (2)	0.6543 (3)	0.7080 (2)	0.0365 (10)	
C15A	0.2574 (3)	0.3134 (3)	0.6400 (3)	0.0685 (16)	
H15D	0.2033	0.2805	0.6534	0.103*	
H15E	0.2773	0.3514	0.6736	0.103*	
H15F	0.3078	0.2742	0.6294	0.103*	
C16A	0.2084 (3)	0.3074 (3)	0.5267 (3)	0.0715 (16)	
H16D	0.1920	0.3416	0.4902	0.107*	
H16E	0.1562	0.2706	0.5380	0.107*	
H16F	0.2625	0.2719	0.5172	0.107*	
C17A	0.0379 (3)	0.6895 (3)	0.8280(2)	0.0603 (13)	
H17D	0.0908	0.6506	0.8245	0.090*	
H17E	-0.0167	0.6574	0.8415	0.090*	
H17F	0.0519	0.7344	0.8583	0.090*	
C18A	-0.0729(3)	0.7854 (3)	0.7663 (3)	0.0587 (13)	
H18D	-0.0866	0.8084	0.7250	0.088*	
H18E	-0.0649	0.8327	0.7956	0.088*	
H18F	-0.1241	0.7490	0.7799	0.088*	
C19A	0.3146 (3)	0.5673 (3)	0.7580(2)	0.0402 (10)	
C22A	0.3848 (3)	0.5054 (3)	0.8740 (2)	0.0494 (12)	
O25A	0.2757 (9)	0.3913 (8)	0.8741 (5)	0.074 (4)	0.65 (4)
C20A	0.2766 (6)	0.4936 (5)	0.7872 (3)	0.043 (3)	0.65 (4)
H20B	0.2269	0.4644	0.7676	0.052*	0.65 (4)
C21A	0.3099 (6)	0.4630 (5)	0.8431 (3)	0.045 (2)	0.65 (4)
C23A	0.4224 (6)	0.5783 (6)	0.8451 (4)	0.065 (4)	0.65 (4)
H23B	0.4721	0.6075	0.8647	0.078*	0.65 (4)
C24A	0.3881 (6)	0.6096 (6)	0.7876 (4)	0.055 (3)	0.65 (4)
H24B	0.4148	0.6589	0.7692	0.066*	0.65 (4)
C26A	0.2016 (14)	0.3452 (12)	0.8480(7)	0.078 (6)	0.65 (4)
H26D	0.1777	0.3044	0.8786	0.117*	0.65 (4)
H26E	0.2231	0.3141	0.8113	0.117*	0.65 (4)
H26F	0.1523	0.3849	0.8361	0.117*	0.65 (4)
025C	0.2505 (17)	0.420 (3)	0.8889(11)	0.092 (9)	0.35 (4)
C20C	0.2630 (12)	0.5092 (18)	0.7952 (8)	0.043 (4)	0.35 (4)
H20C	0.2046	0.4898	0.7808	0.052*	0.35 (4)
C21C	0.2953 (12)	0.4801 (19)	0.8519 (8)	0.056(5)	0.35 (4)
C23C	0.4336(12)	0.5662 (16)	0.8386 (9)	0.062(5)	0.35 (4)
H23C	0 4891	0 5898	0.8549	0.075*	0.35(4)
C24C	0.4021(14)	0.593 (2)	0.7793(10)	0.055(3)	0.35(4)
H24C	0.4401	0.6286	0.7541	0.066*	0.35(1)
C26C	0.184(3)	0.365 (3)	0.8620(15)	0.081 (14)	0.35(4)
H26G	0.1587	0.3272	0.8939	0.122*	0.35(1)
H26H	0.2137	0.3307	0.8295	0.122*	0 35 (4)
	······ · · /	0.0001	0.04/0	···	0.00 (1)

H26I	0.1344	0.3993	0.8438	0.122*	0.35 (4)
C28A	0.5051 (4)	0.4963 (4)	0.9522 (3)	0.0720 (18)	
H28D	0.5200	0.4642	0.9899	0.108*	
H28E	0.5045	0.5577	0.9615	0.108*	
H28F	0.5518	0.4846	0.9204	0.108*	
O1B	-0.0121(2)	-0.2853(2)	0.5241 (2)	0.0810(12)	
O8B	-0.16622(19)	-0.0344(2)	0.6456 (2)	0.0734 (12)	
O25B	-0.0012 (2)	0.1058 (2)	0.40468 (18)	0.0766 (11)	
O27B	-0.1489(2)	0.0429 (2)	0.34795 (17)	0.0604 (9)	
N12B	0.1604 (2)	-0.0524(2)	0.61100 (19)	0.0458 (10)	
H12B	0.2141	-0.0265	0.6140	0.055*	
C1B	0.0648 (3)	-0.2511(3)	0.5358 (3)	0.0541 (13)	
C2B	0.1562(3)	-0.2996(3)	0.5243(3)	0.0589(14)	
H2BA	0.1472	-0 3398	0.4893	0.071*	
H2BR	0.1710	-0.3340	0.5616	0.071*	
C3B	0.2397(3)	-0.2402(3)	0.5010	0.071	
C4B	0.2397(3) 0.2483(3)	-0.1713(3)	0.5617(2)	0.0500(12) 0.0502(12)	
	0.2483 (3)	-0.1083	0.5005	0.060*	
	0.2742	-0.1263	0.5995	0.060*	
C5D	0.2920	0.1203	0.5481	0.000°	
	0.1020 (5)	0.0332 (3)	0.0808 (5)	0.0551 (15)	
	0.1330	0.0892	0.0714	0.004	
	0.1228	0.0238 0.1122 (2)	0.7238 0.7014 (2)	0.004°	
	0.0174(3)	0.1125(3)	0.7014(3)	0.0333(13)	
	-0.0673(3)	0.0519(3)	0.7119 (3)	0.0622 (14)	
H/BA	-0.0574	0.0193	0.7507	0.075*	
H/BB	-0.1233	0.0875	0.7177	0.075*	
C8B	-0.0852(3)	-0.0110 (3)	0.6595 (3)	0.0491 (14)	
C9B	-0.0208 (2)	-0.1138 (3)	0.5758 (2)	0.0424 (11)	
H9BA	-0.0683	-0.1550	0.5913	0.051*	
C10B	0.0698 (3)	-0.1646 (3)	0.5638 (2)	0.0401 (11)	
C11B	0.1546 (3)	-0.1297 (3)	0.5778 (2)	0.0399 (10)	
C13B	0.0824 (3)	-0.0154 (3)	0.6397 (3)	0.0407 (11)	
C14B	-0.0046(2)	-0.0469 (3)	0.6264 (2)	0.0406 (11)	
C15B	0.2225 (4)	-0.1969 (4)	0.4448 (3)	0.0794 (17)	
H15A	0.2190	-0.2411	0.4127	0.119*	
H15B	0.2734	-0.1577	0.4355	0.119*	
H15C	0.1641	-0.1650	0.4460	0.119*	
C16B	0.3307 (3)	-0.2941 (4)	0.5059 (3)	0.0820 (19)	
H16A	0.3447	-0.3180	0.5468	0.123*	
H16B	0.3818	-0.2573	0.4925	0.123*	
H16C	0.3226	-0.3408	0.4760	0.123*	
C17B	-0.0042 (3)	0.1733 (3)	0.6459 (3)	0.0672 (15)	
H17A	-0.0608	0.2055	0.6547	0.101*	
H17B	-0.0127	0.1394	0.6081	0.101*	
H17C	0.0475	0.2131	0.6401	0.101*	
C18B	0.0376 (4)	0.1652 (4)	0.7612 (3)	0.0847 (19)	
H18A	-0.0162	0.2009	0.7711	0.127*	
H18B	0.0917	0.2018	0.7544	0.127*	

H18C	0.0499	0.1262	0.7959	0.127*
C19B	-0.0573 (3)	-0.0748 (3)	0.5145 (2)	0.0448 (12)
C20B	-0.0120 (3)	-0.0035 (3)	0.4882 (2)	0.0424 (12)
H20A	0.0406	0.0195	0.5087	0.051*
C21B	-0.0421 (3)	0.0343 (3)	0.4328 (2)	0.0466 (11)
C22B	-0.1219 (3)	0.0023 (3)	0.4027 (3)	0.0522 (14)
C23B	-0.1665 (3)	-0.0674 (4)	0.4282 (3)	0.083 (2)
H23A	-0.2191	-0.0906	0.4079	0.100*
C24B	-0.1351 (3)	-0.1051 (4)	0.4845 (3)	0.0768 (18)
H24A	-0.1682	-0.1518	0.5017	0.092*
C26B	0.0708 (3)	0.1487 (4)	0.4394 (3)	0.0827 (19)
H26A	0.0927	0.1984	0.4159	0.124*
H26B	0.1224	0.1091	0.4463	0.124*
H26C	0.0461	0.1676	0.4794	0.124*
C28B	-0.2427 (3)	0.0244 (3)	0.3247 (3)	0.0665 (15)
H28A	-0.2557	0.0605	0.2887	0.100*
H28B	-0.2880	0.0362	0.3574	0.100*
H28C	-0.2466	-0.0360	0.3127	0.100*

Atomic displacement parameters (\mathring{A}^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
OIA	0.0224 (15)	0.062 (2)	0.080 (3)	0.0021 (15)	0.0027 (15)	0.0113 (19)
O8A	0.0491 (19)	0.047 (2)	0.103 (3)	-0.0204 (15)	0.0055 (18)	-0.010 (2)
O27A	0.056 (2)	0.093 (3)	0.056 (2)	-0.0114 (19)	-0.0208 (18)	0.008 (2)
N12A	0.0218 (16)	0.045 (2)	0.076 (3)	-0.0030 (15)	0.0028 (17)	-0.020 (2)
C1A	0.035 (3)	0.038 (3)	0.055 (3)	0.0047 (18)	0.001 (2)	0.008 (2)
C2A	0.040 (3)	0.055 (3)	0.071 (4)	0.005 (2)	0.012 (2)	-0.005 (3)
C3A	0.042 (2)	0.043 (2)	0.052 (3)	0.010 (2)	0.003 (2)	-0.005 (2)
C4A	0.030 (2)	0.046 (3)	0.059 (3)	0.0029 (19)	-0.005 (2)	-0.013 (2)
C5A	0.036 (2)	0.047 (3)	0.050 (3)	0.0037 (19)	-0.002 (2)	-0.015 (2)
C6A	0.041 (2)	0.039 (3)	0.043 (3)	0.0016 (19)	-0.001 (2)	-0.003 (2)
C7A	0.049 (3)	0.037 (2)	0.054 (3)	-0.001 (2)	-0.002 (2)	0.004 (2)
C8A	0.046 (3)	0.042 (3)	0.043 (3)	-0.008(2)	-0.003 (2)	0.004 (2)
C9A	0.024 (2)	0.037 (2)	0.061 (3)	-0.0044 (17)	-0.0048 (19)	0.003 (2)
C10A	0.023 (2)	0.034 (2)	0.054 (3)	0.0009 (18)	-0.0026 (18)	0.002 (2)
C11A	0.032 (2)	0.040(2)	0.034 (3)	0.0054 (16)	-0.0045 (19)	-0.005 (2)
C13A	0.028 (2)	0.041 (2)	0.046 (3)	-0.0004 (17)	0.0014 (18)	-0.005 (2)
C14A	0.028 (2)	0.036 (2)	0.045 (3)	-0.0037 (17)	-0.0025 (18)	-0.008(2)
C15A	0.056 (3)	0.053 (3)	0.096 (4)	-0.001 (3)	-0.010 (3)	0.021 (3)
C16A	0.072 (4)	0.058 (4)	0.084 (4)	0.017 (3)	0.003 (3)	-0.021 (3)
C17A	0.068 (3)	0.060 (3)	0.052 (3)	0.005 (3)	0.002 (2)	0.004 (3)
C18A	0.055 (3)	0.052 (3)	0.070 (3)	0.015 (2)	0.003 (3)	-0.017 (3)
C19A	0.029 (2)	0.039 (2)	0.053 (3)	-0.0024 (16)	-0.0120 (18)	-0.004 (2)
C22A	0.037 (2)	0.062 (3)	0.050 (3)	-0.0014 (18)	-0.011 (2)	0.000 (2)
O25A	0.068 (5)	0.093 (6)	0.060 (5)	-0.036 (5)	-0.011 (4)	0.020 (4)
C20A	0.033 (4)	0.051 (6)	0.046 (5)	-0.016 (4)	-0.004 (3)	-0.010 (3)
C21A	0.033 (4)	0.056 (5)	0.046 (4)	-0.006 (3)	0.007 (3)	0.001 (3)

supporting information

C23A	0.050 (7)	0.063 (7)	0.081 (7)	-0.012 (5)	-0.036 (6)	0.008 (6)
C24A	0.041 (3)	0.046 (5)	0.078 (4)	-0.012 (3)	-0.027 (3)	0.011 (4)
C26A	0.055 (9)	0.071 (8)	0.108 (13)	-0.016 (6)	0.005 (7)	-0.005 (8)
C19C	0.029 (2)	0.039 (2)	0.053 (3)	-0.0024 (16)	-0.0120 (18)	-0.004 (2)
C22C	0.037 (2)	0.062 (3)	0.050 (3)	-0.0014 (18)	-0.011 (2)	0.000 (2)
O25C	0.093 (11)	0.127 (17)	0.056 (9)	-0.057 (12)	-0.011 (9)	0.028 (11)
C20C	0.027 (7)	0.046 (10)	0.056 (9)	0.004 (6)	-0.010 (5)	0.006 (7)
C21C	0.048 (8)	0.080 (13)	0.039 (9)	-0.022 (8)	-0.010 (6)	0.004 (7)
C23C	0.029 (9)	0.077 (15)	0.081 (12)	-0.003 (7)	-0.022 (7)	0.020 (10)
C24C	0.041 (3)	0.046 (5)	0.078 (4)	-0.012 (3)	-0.027 (3)	0.011 (4)
C26C	0.054 (15)	0.11 (3)	0.083 (17)	-0.05 (2)	0.001 (15)	0.02 (2)
C28A	0.048 (3)	0.103 (5)	0.066 (4)	0.004 (3)	-0.030 (3)	-0.005 (3)
O1B	0.063 (2)	0.050 (2)	0.130 (4)	-0.0236 (19)	-0.002 (2)	-0.003 (2)
O8B	0.0238 (16)	0.066 (2)	0.131 (4)	0.0006 (16)	0.0040 (18)	0.018 (2)
O25B	0.080 (2)	0.080 (3)	0.070 (3)	-0.046 (2)	-0.0272 (19)	0.031 (2)
O27B	0.0521 (19)	0.074 (2)	0.055 (2)	-0.0110 (17)	-0.0196 (16)	0.0152 (19)
N12B	0.0212 (17)	0.054 (2)	0.062 (3)	-0.0031 (16)	-0.0006 (16)	-0.004 (2)
C1B	0.060 (3)	0.043 (3)	0.059 (3)	-0.013 (2)	0.002 (2)	0.007 (3)
C2B	0.060 (3)	0.044 (3)	0.072 (4)	0.008 (2)	0.008 (3)	0.007 (3)
C3B	0.047 (3)	0.052 (3)	0.053 (3)	-0.001 (2)	0.009 (2)	0.005 (3)
C4B	0.037 (2)	0.055 (3)	0.058 (3)	0.004 (2)	0.010 (2)	0.005 (3)
C5B	0.039 (2)	0.062 (3)	0.058 (3)	0.010 (2)	-0.006 (2)	-0.006 (3)
C6B	0.036 (2)	0.058 (3)	0.066 (4)	0.010 (2)	0.004 (2)	-0.001 (3)
C7B	0.049 (3)	0.055 (3)	0.082 (4)	0.013 (2)	0.017 (3)	0.012 (3)
C8B	0.028 (3)	0.050 (3)	0.069 (4)	0.006 (2)	0.004 (2)	0.021 (3)
C9B	0.025 (2)	0.038 (2)	0.064 (3)	-0.0089 (18)	-0.001 (2)	0.017 (2)
C10B	0.032 (2)	0.042 (3)	0.047 (3)	-0.0053 (18)	0.0007 (19)	0.013 (2)
C11B	0.035 (2)	0.040 (2)	0.044 (3)	-0.0010 (19)	0.0000 (19)	0.000 (2)
C13B	0.026 (2)	0.043 (3)	0.054 (3)	0.0071 (18)	0.001 (2)	0.005 (2)
C14B	0.024 (2)	0.042 (3)	0.056 (3)	0.0056 (18)	-0.001 (2)	0.014 (2)
C15B	0.098 (4)	0.079 (4)	0.061 (4)	-0.008 (3)	0.018 (3)	0.012 (3)
C16B	0.063 (3)	0.084 (5)	0.100 (5)	0.020 (3)	0.021 (3)	-0.008 (4)
C17B	0.052 (3)	0.048 (3)	0.102 (4)	-0.002 (2)	0.001 (3)	0.018 (3)
C18B	0.083 (4)	0.082 (5)	0.089 (4)	0.019 (3)	0.002 (3)	-0.031 (4)
C19B	0.030 (2)	0.042 (3)	0.062 (3)	-0.0089 (19)	-0.005 (2)	0.012 (2)
C20B	0.033 (2)	0.047 (3)	0.048 (3)	-0.0096 (18)	-0.012 (2)	0.012 (2)
C21B	0.041 (2)	0.049 (3)	0.050 (3)	-0.011 (2)	-0.006 (2)	0.010 (3)
C22B	0.050 (3)	0.056 (3)	0.051 (3)	-0.010 (2)	-0.019 (3)	0.009 (3)
C23B	0.065 (3)	0.088 (4)	0.097 (5)	-0.039 (3)	-0.048 (3)	0.036 (4)
C24B	0.066 (3)	0.072 (4)	0.093 (4)	-0.039 (3)	-0.037 (3)	0.044 (3)
C26B	0.070 (3)	0.087 (4)	0.091 (4)	-0.054 (3)	-0.029 (3)	0.035 (4)
C28B	0.052 (3)	0.086 (4)	0.062 (4)	-0.004 (3)	-0.021 (3)	0.006 (3)

Geometric parameters (Å, °)

01A—C1A	1.227 (5)	С26С—Н26Н	0.9600
O8A—C8A	1.225 (4)	C26C—H26I	0.9600
O27A—C22A	1.357 (6)	C28A—H28D	0.9600

O27A—C28A	1.425 (5)	C28A—H28E	0.9600
N12A—C13A	1.372 (5)	C28A—H28F	0.9600
N12A—C11A	1.377 (5)	O1B—C1B	1.233 (5)
N12A—H12A	0.8600	O8B—C8B	1.238 (5)
C1A—C10A	1.441 (6)	O25B—C21B	1.373 (5)
C1A—C2A	1.497 (7)	025B—C26B	1.417 (5)
C2A - C3A	1.532 (6)	027B—C22B	1.368 (6)
C2A—H2AA	0 9700	O27B-C28B	1443(5)
C2A—H2AB	0.9700	N12B-C11B	1.718(5) 1.378(5)
C3A - C4A	1 523 (5)	N12B—C13B	1.370(5)
C_{3A} $-C_{16A}$	1.525 (6)	N12B—H12B	0.8600
C_{3A} $-C_{15A}$	1.525 (6)	C1B - C10B	1 453 (6)
	1.331(0) 1.487(6)	CIR C2R	1.433(0)
	0.0700	C1B - C2B C2B - C3B	1.511(0) 1.525(6)
	0.9700	$C_{2}D_{-}C_{3$	1.323(0)
C_{4A} H_{4AB} C_{5A} C_{12A}	0.9700	C2D—II2DA C2D—II2DD	0.9700
CSA—CISA	1.501 (5)	$C2D - \Pi 2DD$	0.9700
CSA—C6A	1.527 (5)	C3B—C4B	1.532 (6)
С5А—Н5АА	0.9700	C3B-C16B	1.532 (6)
С5А—Н5АВ	0.9/00	C3B—C15B	1.537 (6)
C6A—C/A	1.518 (6)	C4B—C11B	1.510 (5)
C6A—C1/A	1.530 (6)	C4B—H4BA	0.9700
C6A—C18A	1.536 (5)	C4B—H4BB	0.9700
C7A—C8A	1.512 (6)	C5B—C13B	1.474 (6)
С7А—Н7АА	0.9700	C5B—C6B	1.533 (5)
C7A—H7AB	0.9700	C5B—H5BA	0.9700
C8A—C14A	1.444 (5)	C5B—H5BB	0.9700
C9A—C19A	1.518 (5)	C6B—C18B	1.528 (7)
C9A—C14A	1.519 (5)	C6B—C17B	1.530 (6)
C9A—C10A	1.520 (6)	C6B—C7B	1.531 (6)
С9А—Н9АА	0.9800	C7B—C8B	1.489 (7)
C10A—C11A	1.351 (5)	C7B—H7BA	0.9700
C13A—C14A	1.348 (5)	C7B—H7BB	0.9700
C15A—H15D	0.9600	C8B—C14B	1.446 (6)
C15A—H15E	0.9600	C9B—C14B	1.499 (6)
C15A—H15F	0.9600	C9B—C19B	1.516 (6)
C16A—H16D	0.9600	C9B—C10B	1.520 (5)
C16A—H16E	0.9600	С9В—Н9ВА	0.9800
C16A—H16F	0.9600	C10B—C11B	1.348 (5)
C17A—H17D	0.9600	C13B—C14B	1.352 (5)
C17A—H17E	0.9600	C15B—H15A	0.9600
C17A—H17F	0.9600	C15B—H15B	0.9600
C18A - H18D	0.9600	C15B—H15C	0.9600
C18A—H18F	0.9600	C16B—H16A	0.9600
C18A—H18F	0.9600	C16B—H16B	0.9600
	1 376 (5)	CI6B HI6C	0.000
$C_{12A} = C_{24A}$	1.370(3) 1 304 (7)	C17B H17A	0.2000
$C_{12A} = C_{20A}$	1.374(7) 1.270(7)	C17D = H17D	0.2000
$C_{22A} = C_{23A}$	1.3/9(7)	$C_{1/D}$ $-\Pi_{1/D}$	0.9000
UZZA-UZIA	1.404 (/)	$U / D = \Pi I / U$	0.9000

O25A—C21A	1.366 (7)	C18B—H18A	0.9600
O25A—C26A	1.380 (7)	C18B—H18B	0.9600
C20A—C21A	1.356 (8)	C18B—H18C	0.9600
C20A—H20B	0.9300	C19B—C24B	1.353 (5)
C23A—C24A	1.394 (7)	C19B—C20B	1.383 (6)
C23A—H23B	0.9300	C20B—C21B	1.373 (6)
C24A—H24B	0.9300	C20B—H20A	0.9300
C26A—H26D	0.9600	C21B—C22B	1.387 (6)
С26А—Н26Е	0.9600	C22B—C23B	1.353 (6)
C26A—H26F	0.9600	C23B—C24B	1.393 (6)
O25C—C21C	1.366 (7)	C23B—H23A	0.9300
O25C—C26C	1.380 (7)	C24B—H24A	0.9300
C20C—C21C	1.356 (8)	C26B—H26A	0.9600
C20C—H20C	0.9300	C26B—H26B	0.9600
C23C—C24C	1.394 (7)	C26B—H26C	0.9600
C23C—H23C	0.9300	C28B—H28A	0.9600
C24C—H24C	0.9300	C28B—H28B	0.9600
C26C—H26G	0.9600	C28B—H28C	0.9600
C22A—O27A—C28A	117.3 (4)	H28D—C28A—H28F	109.5
C13A—N12A—C11A	121.3 (3)	H28E—C28A—H28F	109.5
C13A—N12A—H12A	119.4	C21B—O25B—C26B	116.7 (4)
C11A—N12A—H12A	119.4	C22B—O27B—C28B	117.0 (4)
O1A—C1A—C10A	121.7 (4)	C11B—N12B—C13B	121.8 (3)
O1A—C1A—C2A	120.4 (4)	C11B—N12B—H12B	119.1
C10A—C1A—C2A	117.9 (4)	C13B—N12B—H12B	119.1
C1A—C2A—C3A	113.8 (4)	O1B—C1B—C10B	120.7 (4)
C1A—C2A—H2AA	108.8	O1B—C1B—C2B	121.0 (5)
C3A—C2A—H2AA	108.8	C10B—C1B—C2B	118.2 (4)
C1A—C2A—H2AB	108.8	C1B—C2B—C3B	113.8 (4)
C3A—C2A—H2AB	108.8	C1B—C2B—H2BA	108.8
H2AA—C2A—H2AB	107.7	C3B—C2B—H2BA	108.8
C4A—C3A—C16A	109.1 (3)	C1B—C2B—H2BB	108.8
C4A—C3A—C15A	111.0 (4)	C3B—C2B—H2BB	108.8
C16A—C3A—C15A	109.4 (4)	H2BA—C2B—H2BB	107.7
C4A—C3A—C2A	107.4 (4)	C2B—C3B—C4B	109.0 (4)
C16A—C3A—C2A	110.2 (4)	C2B—C3B—C16B	110.0 (4)
C15A—C3A—C2A	109.8 (4)	C4B—C3B—C16B	109.9 (4)
C11A—C4A—C3A	112.9 (3)	C2B—C3B—C15B	108.5 (4)
C11A—C4A—H4AA	109.0	C4B—C3B—C15B	110.8 (4)
СЗА—С4А—Н4АА	109.0	C16B—C3B—C15B	108.7 (4)
C11A—C4A—H4AB	109.0	C11B—C4B—C3B	112.5 (4)
СЗА—С4А—Н4АВ	109.0	C11B—C4B—H4BA	109.1
Н4АА—С4А—Н4АВ	107.8	C3B—C4B—H4BA	109.1
C13A—C5A—C6A	112.3 (3)	C11B—C4B—H4BB	109.1
С13А—С5А—Н5АА	109.1	C3B—C4B—H4BB	109.1
С6А—С5А—Н5АА	109.1	H4BA—C4B—H4BB	107.8
С13А—С5А—Н5АВ	109.1	C13B—C5B—C6B	114.1 (4)

C6A—C5A—H5AB	109.1	С13В—С5В—Н5ВА	108.7
Н5АА—С5А—Н5АВ	107.9	C6B—C5B—H5BA	108.7
C7A—C6A—C5A	108.0 (4)	C13B—C5B—H5BB	108.7
C7A—C6A—C17A	110.4 (3)	C6B—C5B—H5BB	108.7
C5A—C6A—C17A	110.5 (4)	H5BA—C5B—H5BB	107.6
C7A—C6A—C18A	108.9 (4)	C18B—C6B—C17B	110.3 (4)
C5A—C6A—C18A	108.9 (3)	C18B—C6B—C7B	110.3 (4)
C17A—C6A—C18A	110.1 (4)	C17B—C6B—C7B	108.8 (4)
C8A—C7A—C6A	114.3 (3)	C18B—C6B—C5B	109.5 (4)
C8A—C7A—H7AA	108 7	C17B - C6B - C5B	111 2 (4)
C6A - C7A - H7AA	108.7	C7B-C6B-C5B	106.6(4)
C8A - C7A - H7AB	108.7	C8B-C7B-C6B	114.6(4)
C6A - C7A - H7AB	108.7	C8B - C7B - H7BA	108.6
H7AA - C7A - H7AB	107.6	C6B-C7B-H7BA	108.6
084 - C84 - C144	121.7(4)	C8B-C7B-H7BB	108.6
084 - C84 - C74	121.7(4) 120.4(4)	C6B-C7B-H7BB	108.6
$C_{14A} = C_{8A} = C_{7A}$	120.4(4) 117.7(3)		107.6
$C_{14A} = C_{0A} = C_{14A}$	117.7(3)	$\Pi/BA - C/B - \Pi/BB$	107.0 120.5(5)
$C_{19A} = C_{9A} = C_{10A}$	111.1(3) 1120(3)	$O_{8}^{R} C_{8}^{R} C_{7}^{R}$	120.3(3) 121.4(4)
$C_{14A} = C_{9A} = C_{10A}$	112.0(3) 100.2(3)	$C_{AB} = C_{AB} = C_{AB}$	121.4(4)
$C_{14A} = C_{9A} = C_{10A}$	109.2 (5)	$C_{14}D = C_{8}D = C_{7}D$	110.1(4) 112.1(4)
C14A = C9A = H9AA	100.1	$C_{14}D = C_{9}D = C_{19}D$	113.1(4) 100.0(2)
C10A = C0A = H0AA	108.1	$C_{14} = C_{9} = C_{10} = C_$	109.9(3)
C10A - C10A - C10A	108.1	C14D = C0D = U0DA	107.9
CIIA—CIOA—CIA	119.4 (4)	C14B - C9B - H9BA	107.8
CIIA = CI0A = C0A	121.7 (4)	C19B - C9B - H9BA	107.8
CIA-CIUA-CYA	118.9 (3)	CIUB-CUB-H9BA	107.8
CIOA—CIIA—NI2A	120.0 (4)	CIIB—CI0B—CIB	119.6 (4)
CIOA—CIIA—C4A	124.5 (4)	CIIB—CI0B—C9B	120.8 (4)
NI2A—CIIA—C4A	115.5 (3)	CIB-CI0B-C9B	119.6 (4)
CI4A—CI3A—NI2A	121.0 (3)	CI0B—CIIB—NI2B	120.4 (4)
C14A—C13A—C5A	123.7 (4)	Clob—Clib—C4B	124.5 (4)
N12A—C13A—C5A	115.2 (3)	N12B—C11B—C4B	115.1 (3)
C13A—C14A—C8A	120.1 (3)	C14B—C13B—N12B	119.5 (4)
C13A—C14A—C9A	120.9 (4)	C14B—C13B—C5B	124.5 (4)
C8A—C14A—C9A	119.0 (3)	N12B—C13B—C5B	115.9 (3)
C3A—C15A—H15D	109.5	C13B—C14B—C8B	118.9 (4)
C3A—C15A—H15E	109.5	C13B—C14B—C9B	122.0 (4)
H15D—C15A—H15E	109.5	C8B—C14B—C9B	119.0 (4)
C3A—C15A—H15F	109.5	C3B—C15B—H15A	109.5
H15D—C15A—H15F	109.5	C3B—C15B—H15B	109.5
H15E—C15A—H15F	109.5	H15A—C15B—H15B	109.5
C3A—C16A—H16D	109.5	C3B—C15B—H15C	109.5
C3A—C16A—H16E	109.5	H15A—C15B—H15C	109.5
H16D—C16A—H16E	109.5	H15B—C15B—H15C	109.5
C3A—C16A—H16F	109.5	C3B—C16B—H16A	109.5
H16D—C16A—H16F	109.5	C3B—C16B—H16B	109.5
H16E—C16A—H16F	109.5	H16A—C16B—H16B	109.5
C6A—C17A—H17D	109.5	C3B—C16B—H16C	109.5

C6A—C17A—H17E	109.5	H16A—C16B—H16C	109.5
H17D—C17A—H17E	109.5	H16B—C16B—H16C	109.5
C6A—C17A—H17F	109.5	C6B—C17B—H17A	109.5
H17D—C17A—H17F	109.5	C6B—C17B—H17B	109.5
H17E—C17A—H17F	109.5	H17A—C17B—H17B	109.5
C6A—C18A—H18D	109.5	C6B—C17B—H17C	109.5
C6A—C18A—H18E	109.5	H17A—C17B—H17C	109.5
H18D—C18A—H18E	109.5	H17B—C17B—H17C	109.5
C6A—C18A—H18F	109.5	C6B—C18B—H18A	109.5
H18D—C18A—H18F	109.5	C6B—C18B—H18B	109.5
H18E—C18A—H18F	109.5	H18A—C18B—H18B	109.5
C_{24A} — C_{19A} — C_{20A}	118.2 (4)	C6B-C18B-H18C	109.5
C24A - C19A - C9A	119.6 (4)	H18A - C18B - H18C	109.5
C_{20A} C_{19A} C_{9A}	122.1 (4)	H18B-C18B-H18C	109.5
O27A - C22A - C23A	125.6(5)	C_{24B} C_{19B} C_{20B}	107.3 1174(4)
O27A - C22A - C21A	123.0(5)	$C_{24B} = C_{19B} = C_{20B}$	122.9(4)
C_{23A} C_{22A} C_{21A}	117.0(5) 117.4(5)	$C_{20B} = C_{19B} = C_{9B}$	122.9(1) 1197(4)
$C_{23} = C_{23} = C$	119.4 (6)	$C_{20B} = C_{20B} = C_{19B}$	117.7(4) 122.1(4)
$C_{21A} = C_{20A} = C_{20A}$	119.4(0) 122.0(5)	$C_{21B} = C_{20B} = C_{19B}$	122.1 (4)
$C_{21A} = C_{20A} = C_{19A}$	122.0 (5)	$C_{21D} = C_{20D} = H_{20A}$	119.0
$C_{21A} = C_{20A} = H_{20B}$	119.0	$C_{19}^{19} = C_{20}^{19} = C_{20}^{11} = $	119.0 125.0(4)
C_{10}^{20} C_{20}^{21} C_{20}^{20}	124.8 (5)	$C_{20B} = C_{21B} = C_{23B}$	123.0(4) 110.7(4)
$C_{20A} = C_{21A} = O_{23A}$	124.8(5)	$C_{20} = C_{21} = C_{22} = C$	115.7(4)
$C_{20}A = C_{21}A = C_{22}A$	120.0(5)	$C_{23}D = C_{21}D = C_{22}D$	113.2(4) 124.5(4)
$C_{22A} = C_{21A} = C_{22A}$	114.0(5)	$C_{23}D = C_{22}D = C_{21}D$	124.3(4)
$C_{22}A = C_{23}A = C_{24}A$	122.0 (3)	$C_{23}D = C_{22}D = C_{21}D$	118.3(3) 117.1(4)
С22А—С23А—Н23В	119.0	$O_2/B = C_{22}B = C_{21}B$	11/.1(4) 121.2(4)
$C_{24}A - C_{23}A - H_{23}B$	119.0	$C_{22}B = C_{23}B = C_{24}B$	121.2 (4)
C19A - C24A - C23A	119.8 (5)	C22B—C23B—H23A	119.4
C19A—C24A—H24B	120.1	$C_{24}B - C_{23}B - H_{23}A$	119.4
C23A—C24A—H24B	120.1	C19B - C24B - C23B	121.2 (4)
$C_{21}C_{}O_{25}C_{}C_{26}C_{}C_{2$	119.3 (6)	C19B - C24B - H24A	119.4
C21C—C20C—H20C	119.0	C23B—C24B—H24A	119.4
$C_{20}C_{-}C_{21}C_{-}O_{25}C_{$	124.7 (5)	025B—C26B—H26A	109.5
C24C - C23C - H23C	119.1	025B—C26B—H26B	109.5
C23C—C24C—H24C	120.1	H26A—C26B—H26B	109.5
025C—C26C—H26G	109.5	025B—C26B—H26C	109.5
025С—С26С—Н26Н	109.5	H26A—C26B—H26C	109.5
H26G—C26C—H26H	109.5	H26B—C26B—H26C	109.5
O25C—C26C—H26I	109.5	027B—C28B—H28A	109.5
H26G—C26C—H26I	109.5	O27B—C28B—H28B	109.5
H26H—C26C—H26I	109.5	H28A—C28B—H28B	109.5
O27A—C28A—H28D	109.5	O27B—C28B—H28C	109.5
027A—C28A—H28E	109.5	H28A—C28B—H28C	109.5
H28D—C28A—H28E	109.5	H28B—C28B—H28C	109.5
O27A—C28A—H28F	109.5		
O1A $C1A$ $C2A$ $C2A$	1/18 7 (A)	OIR CIR C2P C2P	151 5 (5)
$C_{10A} = C_{1A} = C_{2A} = C_{2A}$	$-23 \Lambda (6)$	C10P C1P C2P C3P	$-31 \Lambda (6)$
UIUA - UIA - UZA - UJA	55.4 (0)	CIUD-CID-C2D-C3D	-31.4 (0)

C1A—C2A—C3A—C4A	54.9 (5)	C1B—C2B—C3B—C4B	53.3 (6)
C1A—C2A—C3A—C16A	173.6 (4)	C1B—C2B—C3B—C16B	173.8 (4)
C1A—C2A—C3A—C15A	-65.9 (5)	C1B-C2B-C3B-C15B	-67.4 (6)
C16A—C3A—C4A—C11A	-167.4 (4)	C2B-C3B-C4B-C11B	-46.6 (5)
C15A—C3A—C4A—C11A	72.0 (5)	C16B—C3B—C4B—C11B	-167.2 (4)
C2A—C3A—C4A—C11A	-48.1 (5)	C15B—C3B—C4B—C11B	72.6 (5)
C13A—C5A—C6A—C7A	49.7 (5)	C13B—C5B—C6B—C18B	166.2 (5)
C13A—C5A—C6A—C17A	-71.2 (4)	C13B—C5B—C6B—C17B	-71.7 (5)
C13A—C5A—C6A—C18A	167.8 (4)	C13B—C5B—C6B—C7B	46.8 (6)
C5A—C6A—C7A—C8A	-53.4(5)	C18B—C6B—C7B—C8B	-172.3(4)
C17A - C6A - C7A - C8A	67 5 (5)	C17B-C6B-C7B-C8B	66 5 (5)
$C_{18A} - C_{6A} - C_{7A} - C_{8A}$	-1715(4)	C5B-C6B-C7B-C8B	-53.5(5)
C6A - C7A - C8A - O8A	-1563(4)	C6B - C7B - C8B - O8B	-1469(4)
C6A - C7A - C8A - C14A	28.0 (6)	C6B - C7B - C8B - C14B	35.2 (6)
O1A - C1A - C10A - C11A	-1794(4)	O1B-C1B-C10B-C11B	1782(0)
C_{2} C_{1} C_{1	27(6)	$C^{2}B$ $C^{1}B$ $C^{1}0B$ $C^{1}1B$	170.2(4)
$O_{1A} = C_{1A} = C_{10A} = C_{10A}$	-0.3(7)	$O_{1B} = C_{1B} = C_{10B} = C_{1B}$	-32(7)
$C_{2A} = C_{1A} = C_{10A} = C_{9A}$	-178.2(4)	C_{1D} C_{1D} C_{10D} C_{2D} C_{10D} C_{2D}	3.2(7)
$C_{2A} = C_{1A} = C_{10A} = C_{9A}$	170.2(4)	$C_{2}D - C_{1}D - C_{1}D - C_{2}D$	-24.2(5)
C14A = C9A = C10A = C11A	100.3(4)	$C_{14} = C_{2} = C_{10} = C_{10} = C_{11} = C_{10} = C_$	24.2(3)
C10A = C0A = C10A = C1A	-23.3(0)	C14D $C0D$ $C10D$ $C1D$	101.1(3) 157.2(4)
C14A = C0A = C10A = C1A	-76.6(3)	C14B - C9B - C10B - C1B	137.2(4)
C14A - C9A - C10A - C1A	137.7(4)	C19B - C9B - C10B - C1B	-77.3(3)
CIA = CI0A = CI1A = NI2A	-1/4.4(4)	CIB—CI0B—CIIB—NI2B	-1/2.4(4)
CIA CIDA CITA NIZA	0.0 (6) 2.5 (7)	C9B—C10B—C11B—N12B	9.0 (6)
CIA—CI0A—CIIA—C4A	3.5 (7)	CIB—CI0B—CIIB—C4B	5.0 (7)
C9A—C10A—C11A—C4A	-175.6(4)	C9B—C10B—C11B—C4B	-173.7 (4)
C13A—N12A—C11A—C10A	12.7 (7)	C13B—N12B—C11B—C10B	10.4 (7)
C13A—N12A—C11A—C4A	-165.4 (4)	C13B—N12B—C11B—C4B	-167.2 (4)
C3A—C4A—C11A—C10A	21.3 (6)	C3B—C4B—C11B—C10B	19.3 (6)
C3A—C4A—C11A—N12A	-160.8 (4)	C3B—C4B—C11B—N12B	-163.2 (4)
C11A—N12A—C13A—C14A	-11.7 (7)	C11B—N12B—C13B—C14B	-11.4 (7)
C11A—N12A—C13A—C5A	165.7 (4)	C11B—N12B—C13B—C5B	164.6 (4)
C6A—C5A—C13A—C14A	-22.1 (6)	C6B—C5B—C13B—C14B	-23.1 (7)
C6A—C5A—C13A—N12A	160.6 (4)	C6B—C5B—C13B—N12B	161.2 (4)
N12A—C13A—C14A—C8A	171.2 (4)	N12B—C13B—C14B—C8B	176.4 (4)
C5A—C13A—C14A—C8A	-6.0 (6)	C5B—C13B—C14B—C8B	0.8 (7)
N12A—C13A—C14A—C9A	-8.4 (6)	N12B—C13B—C14B—C9B	-7.3 (6)
C5A—C13A—C14A—C9A	174.4 (4)	C5B—C13B—C14B—C9B	177.0 (4)
O8A—C8A—C14A—C13A	-172.4 (4)	O8B—C8B—C14B—C13B	175.3 (4)
C7A—C8A—C14A—C13A	3.2 (6)	C7B—C8B—C14B—C13B	-6.7 (6)
O8A—C8A—C14A—C9A	7.2 (6)	O8B—C8B—C14B—C9B	-1.0 (6)
C7A—C8A—C14A—C9A	-177.2 (4)	C7B-C8B-C14B-C9B	176.9 (4)
C19A—C9A—C14A—C13A	-100.0 (4)	C19B—C9B—C14B—C13B	-100.3 (4)
C10A—C9A—C14A—C13A	24.1 (5)	C10B—C9B—C14B—C13B	23.5 (6)
C19A—C9A—C14A—C8A	80.4 (5)	C19B—C9B—C14B—C8B	75.9 (5)
C10A—C9A—C14A—C8A	-155.5 (4)	C10B—C9B—C14B—C8B	-160.3 (4)
C14A—C9A—C19A—C24A	-100.4 (6)	C14B—C9B—C19B—C24B	-128.4 (5)
C10A—C9A—C19A—C24A	137.1 (6)	C10B—C9B—C19B—C24B	108.1 (5)

77.3 (6)	C14B—C9B—C19B—C20B	49.7 (5)
-45.1 (6)	C10B—C9B—C19B—C20B	-73.9 (5)
-16.7 (7)	C24B—C19B—C20B—C21B	-1.9 (7)
162.8 (6)	C9B—C19B—C20B—C21B	180.0 (4)
0.00 (10)	C19B—C20B—C21B—O25B	179.1 (5)
-177.8 (3)	C19B—C20B—C21B—C22B	1.8 (8)
180.00 (6)	C26B—O25B—C21B—C20B	-7.2 (7)
0.00 (11)	C26B—O25B—C21B—C22B	170.1 (5)
-0.4 (17)	C28B—O27B—C22B—C23B	15.7 (8)
179.6 (17)	C28B—O27B—C22B—C21B	-165.6 (5)
-179.5 (4)	C20B—C21B—C22B—C23B	-1.7 (8)
0.00 (9)	O25B—C21B—C22B—C23B	-179.2 (5)
0.5 (4)	C20B—C21B—C22B—O27B	179.5 (4)
180.00 (5)	O25B—C21B—C22B—O27B	2.0 (7)
179.5 (4)	O27B—C22B—C23B—C24B	-179.5 (5)
0.00 (9)	C21B—C22B—C23B—C24B	1.7 (9)
0.00 (9)	C20B—C19B—C24B—C23B	1.9 (8)
177.8 (3)	C9B—C19B—C24B—C23B	180.0 (5)
0.00 (11)	C22B—C23B—C24B—C19B	-1.9 (10)
20 (4)		
	$\begin{array}{c} 77.3 \ (6) \\ -45.1 \ (6) \\ -16.7 \ (7) \\ 162.8 \ (6) \\ 0.00 \ (10) \\ -177.8 \ (3) \\ 180.00 \ (6) \\ 0.00 \ (11) \\ -0.4 \ (17) \\ 179.6 \ (17) \\ -179.5 \ (4) \\ 0.00 \ (9) \\ 0.5 \ (4) \\ 180.00 \ (5) \\ 179.5 \ (4) \\ 0.00 \ (9) \\ 0.00 \ (9) \\ 177.8 \ (3) \\ 0.00 \ (11) \\ 20 \ (4) \end{array}$	$\begin{array}{llllllllllllllllllllllllllllllllllll$

Hydrogen-bond geometry (Å, °)

D—H···A	D—H	$H \cdots A$	D··· A	D—H··· A
N12A—H12A…O1A ⁱ	0.86	2.07	2.846 (4)	150
N12 <i>B</i> —H12 <i>B</i> ···O8 <i>B</i> ⁱⁱ	0.86	2.05	2.885 (4)	165
$C2A$ — $H2AA$ ···O1 B^{ii}	0.97	2.45	3.400 (6)	166
C16A—H16D…O25A ⁱⁱⁱ	0.96	2.61	3.477 (12)	150
$C7A$ — $H7AB$ ···O27 B^{iv}	0.97	2.47	3.432 (6)	169
$C7B$ — $H7BB$ ···· $O8A^{i}$	0.97	2.57	3.507 (5)	162

Symmetry codes: (i) *x*-1/2, -*y*+1, *z*; (ii) *x*+1/2, -*y*, *z*; (iii) -*x*+1/2, *y*, *z*-1/2; (iv) -*x*, -*y*+1, *z*+1/2.