

Redetermination of $\text{Nd}_2\text{Ti}_2\text{O}_7$: a non-centrosymmetric structure with perovskite-type slabs

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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{Ti}–\text{O}) = 0.004$ Å; R factor = 0.024; wR factor = 0.027; data-to-parameter ratio = 49.7.

Single crystals of dineodymium(III) dititanium(IV) heptoxide, $\text{Nd}_2\text{Ti}_2\text{O}_7$, were synthesized by the flux method and found to belong to the family of compounds with perovskite-type structural motifs. The asymmetric unit contains four Nd, four Ti and 14 O-atom sites. The perovskite-type slabs are stacked parallel to (010) with a thickness corresponding to four corner-sharing TiO_6 octahedra. The Nd and Ti ions are displaced from the geometrical centres of respective coordination polyhedra so that the net polarization occurs along the c axis. The investigated crystals were all twinned and have a halved monoclinic unit cell in comparison with the first structure determination of this compound [Scheunemann & Müller-Buschbaum (1975). *J. Inorg. Nucl. Chem.* **37**, 2261–2263].

Related literature

For previous determinations of $\text{Nd}_2\text{Ti}_2\text{O}_7$, see: Scheunemann & Müller-Buschbaum (1975); Harvey *et al.* (2005). For related compounds, see: Gasperin (1975); Ishizawa *et al.* (1980); Schmalle *et al.* (1993). For the extinction method, see: Becker & Coppens (1974).

Experimental

Crystal data

$\text{Nd}_2\text{Ti}_2\text{O}_7$

$M_r = 496.2$

Monoclinic, $P112_1$
 $a = 7.6747$ (1) Å
 $b = 13.0025$ (2) Å
 $c = 5.4640$ (1) Å
 $\gamma = 98.5165$ (5)°
 $V = 539.24$ (2) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 21.77$ mm⁻¹
 $T = 293$ K
 $0.11 \times 0.08 \times 0.07$ mm

Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (*TWINABS*; Bruker, 2008)
 $T_{\min} = 0.134$, $T_{\max} = 0.206$

25187 measured reflections
6608 independent reflections
6200 reflections with $I > 3\sigma(I)$
 $R_{\text{int}} = 0.049$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.024$
 $wR(F^2) = 0.027$
 $S = 1.37$
6608 reflections
133 parameters

$\Delta\rho_{\max} = 2.44$ e Å⁻³
 $\Delta\rho_{\min} = -1.57$ e Å⁻³
Absolute structure: Flack (1983),
3019 Friedel pairs
Flack parameter: 0.220 (12)

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT*; program(s) used to solve structure: *SUPERFLIP* (Palatinus & Chapuis, 2007); program(s) used to refine structure: *JANA2006* (Petříček *et al.*, 2006); molecular graphics: *ATOMS* (Dowty, 2006) and *DIAMOND* (Brandenburg & Putz, 2005); software used to prepare material for publication: *PLATON* (Spek, 2009) and *publCIF* (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WM2720).

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supporting information

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S1. Comment

The structure of $\text{Nd}_2\text{Ti}_2\text{O}_7$ contains perovskite-type slabs consisting of TiO_6 octahedra and Nd ions (Fig. 1 and Fig. 2(a)). The slabs are stacked along **b** with interconnecting Nd—O bonds. The thickness of the slabs corresponds to four corner-sharing TiO_6 octahedra. The TiO_6 octahedra have two tilt systems about **c** and **b*** which are closely related to the displacements of Nd atoms. The mode of the Nd atom displacements along **b*** is schematically illustrated by arrows in Fig. 2(a) with respect to the dashed lines drawn parallel to **a**. The Nd and Ti ions are displaced from the geometrical centres of respective coordination polyhedra so that a net polarization occurs in the crystal. The magnitude of the spontaneous polarization was estimated to be approximately $18 \mu\text{C cm}^{-2}$ along the polar *c* axis assuming formal charges for constituent atoms.

Scheunemann & Müller-Buschbaum (1975) (hereafter abbreviated as SMB) have previously determined the structure of $\text{Nd}_2\text{Ti}_2\text{O}_7$ based on single-crystal X-ray diffraction data and reported monoclinic symmetry and space group $P2_1$ with unit-cell dimensions $a = 7.677 \text{ \AA}$, $b = 26.013 \text{ \AA}$, $c = 5.465 \text{ \AA}$, and $\gamma = 98.4^\circ$ (the original cell setting in $P12_11$ has been converted to the current setting in $P112_1$ with the *c* axis unique). The unit cell of the SMB structure (Fig. 2(b)) is doubled along **b** compared with that of the present study. The SMB structure contains two kinds of perovskite-type slabs with octahedra coloured in green and orange, respectively. The green-coloured slab is essentially the same as that in the present study, which can be understood in terms of the octahedral tilt about **b*** and the mode of the Nd atom displacements along **b*** as indicated by arrows. In contrast, the orange-coloured slab in the SMB structure has very small octahedral tilts about **b*** in combination with negligible Nd atom displacements along **b***. The existence of different types of slabs stacked alternately along **b** justifies the doubled unit cell in the SMB structure. The presence of two modifications (Fig. 2(a) and (b)) at room temperature, having different unit cells with the same space group, leaves room for further investigations on the crystal chemistry of $\text{Nd}_2\text{Ti}_2\text{O}_7$.

Harvey *et al.* (2005) (hereafter abbreviated as HWLSR) studied a $\text{Nd}_2(\text{Zr}_{1-x}\text{Ti}_x)\text{O}_7$ solid solution and reported a monoclinic modification for the end member $\text{Nd}_2\text{Ti}_2\text{O}_7$ based on neutron powder data and analyzed by the Rietveld method. The HWLSR structure is plotted in Fig. 2(c). The monoclinic unit cell of the HWLSR structure has similar dimensions to that of SMB. Although no reference is given in their paper to the initial structure model for the Rietveld refinement, it appears possible that the starting SMB model, if used, could easily converge to the HWLSR structure because the symmetry is the same and the parameter shifts are rather small. It is important, however, that the SMB and HWLSR structures are different in that the latter consists of essentially the same green-type slabs. The simulated powder diffraction diagrams of the SMB and HWLSR structures are shown in Fig. 3, indicating a difference in intensities of reflections at low angles with $k = \text{odd}$. The structure-checking program *PLATON* (Spek, 2009) actually found additional twofold screw symmetries lying at the boundary of the slabs in the HWLSR structure with atom shifts less than 0.22 \AA ,

hence suggesting a halved monoclinic unit cell as shown in Fig. 2(a).

Diffraction data in the present study were taken using a conventional sealed X-ray tube. Within the limitations of the experiment, we could not find any superstructure reflections that would double the unit cell volume. It should be noted that twinning of the investigated crystal does emerge as extra reflections with $h = \text{odd}$ (Fig. 4), and that the reciprocal pattern of the components can be indexed on basis of doubled monoclinic unit cells like those of HWLSR or SMB. However, it is not difficult to distinguish the twinning because the extra extinction condition appears for reflections with $h = \text{even}$ and $k = \text{odd}$ (Fig. 4). A trial integration was carried out for the present crystal, assuming a doubled monoclinic cell similar like that of SMB or HWLSR, but the mode of Nd atom displacements in the doubled cell was essentially the same as those in Fig. 2(c), and no structure similar to Fig. 2(b) was obtained.

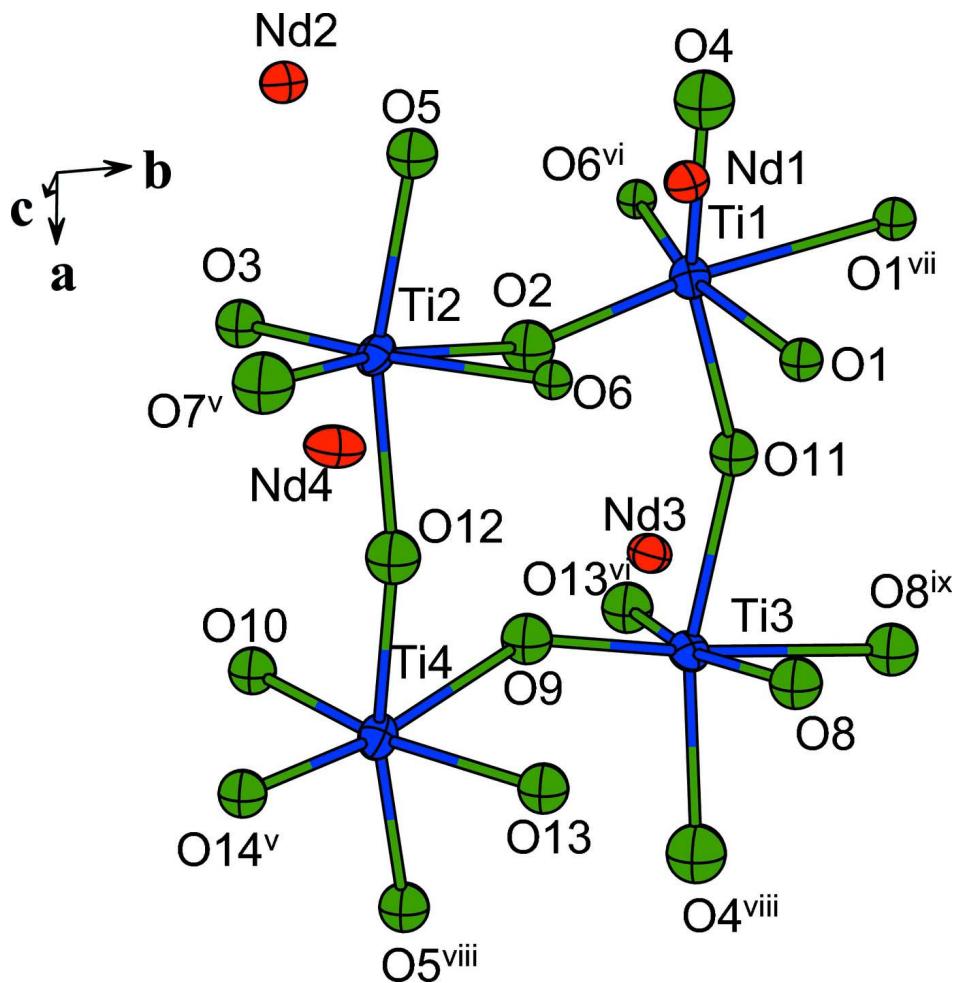
As a conclusion, the HWLSR structure model is supposedly the same as the present one. On the other hand, it seems difficult to discard the possibility of two monoclinic modifications for $\text{Nd}_2\text{Ti}_2\text{O}_7$ at room temperature as far as the SMB structure exists.

S2. Experimental

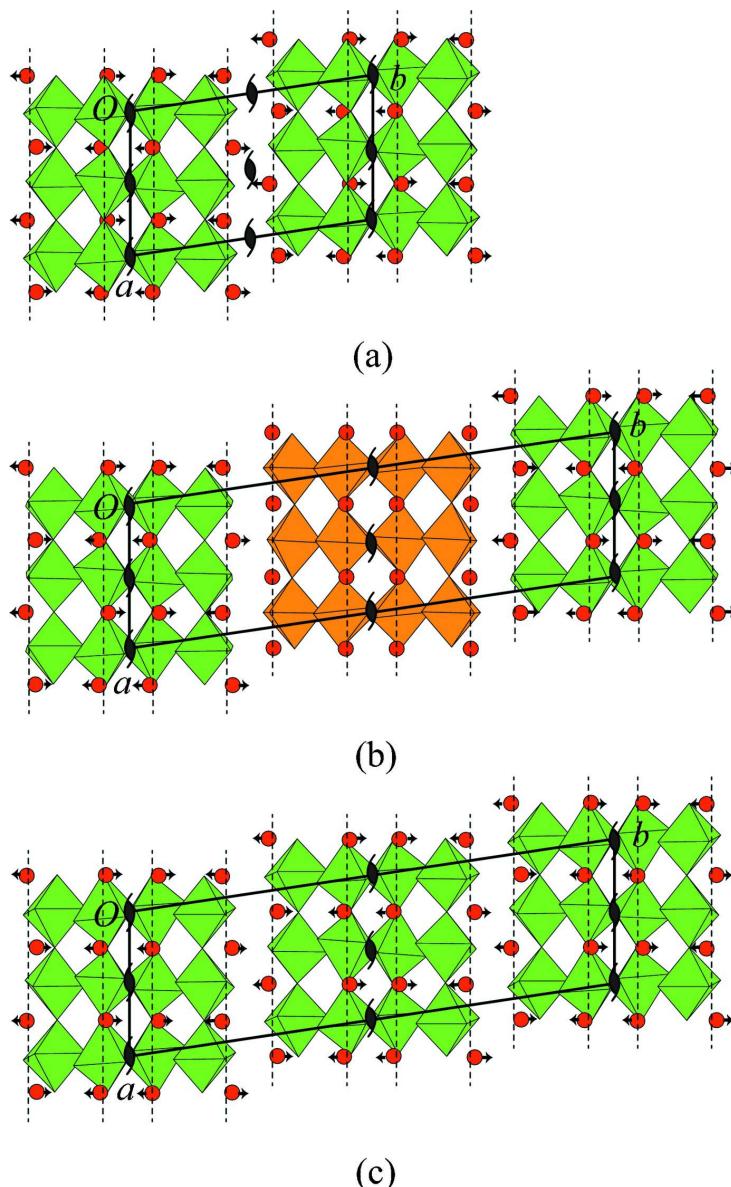
Crystals were grown by the flux method using Nd_2O_3 (Wako Co., 99.9%) and TiO_2 (Wako, 99%) as starting materials, and $\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$ as the flux. A preliminary heat treatment at 1273 K for 12 h was applied to Nd_2O_3 before weighing. A 15 g mixture containing a 20 mol% solute corresponding to the $\text{Nd}_2\text{Ti}_2\text{O}_7$ composition was put in a platinum crucible. The platinum crucible was then placed in an alumina crucible with alumina powder and heated in an electric furnace under air atmosphere. The sample was kept at 1373 K for 10 h, and then cooled to 1173 K at the rate of 5 K h^{-1} , with subsequent furnace-cooling by turning off the power. Purple and transparent crystals with approximate dimensions of $200 \times 300 \times 50 \mu\text{m}^3$ were obtained after washing the flux component with warm water. Neither the flux component nor other impurities were detected beyond the limit of detection using energy dispersive spectroscopic analysis with a JED-2300 instrument (Jeol Ltd.).

S3. Refinement

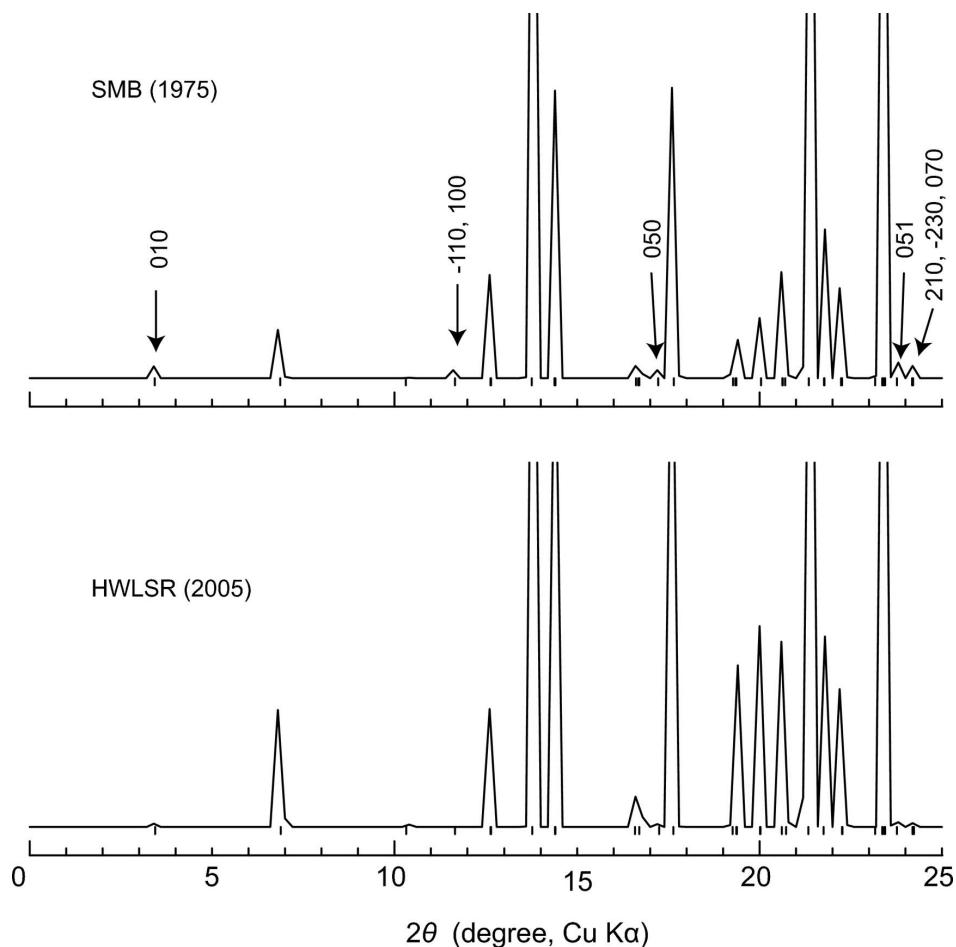
The charge-flipping structure solution program *SUPERFLIP* (Palatinus & Chapuis, 2007) indicated that the present $\text{Nd}_2\text{Ti}_2\text{O}_7$ crystals are isostructural with the monoclinic modifications of $\text{La}_2\text{Ti}_2\text{O}_7$ (Gasperin, 1975) and $\text{Ca}_2\text{Nb}_2\text{O}_7$ (Ishizawa *et al.*, 1980). Further refinements were thus carried out using the atom positions of monoclinic $\text{Ca}_2\text{Nb}_2\text{O}_7$ as starting parameters. All the crystals showed twinning where one twin component (m_2) is obtained by rotating the other (m_1) by 180° about \mathbf{b}^* . The orientation relationships of the two twin components are illustrated in Fig. 4. The twin scheme is essentially the same as that described for $\text{La}_2\text{Ti}_2\text{O}_7$ (Schmalle *et al.*, 1993). Since the monoclinic lattice is metrically the sublattice of the pseudo-orthorhombic one in reciprocal space (Fig. 4), all the reflections of the twin components with $h = \text{even}$ are almost perfectly overlapped whereas those with $h = \text{odd}$ are not overlapped at all. The integration of peak intensities were carried out separately for the lattices of the m_1 and m_2 components, and an absorption correction was processed by *TWINABS* (Bruker, 2008). In addition to the above, crystals undergo twinning by merohedry where the spontaneous polarization vectors of the ferroelectric domains are aligned oppositely along \mathbf{c} . The refinement was thus carried out assuming four twin components, m_1 , m_2 and their inverted ones, resulting in roughly similar volume fractions of 23 (3)%, 28 (2)%, 22 (1)% and 27 (1)%, respectively. The lowest angle reflections, 010 and its equivalents at $\theta = 1.58^\circ$, were removed from the refinement because their intensities were seriously affected by the beamstop. Anisotropic and isotropic atomic displacement parameters were used for the two types of metal and the O atoms, respectively.

**Figure 1**

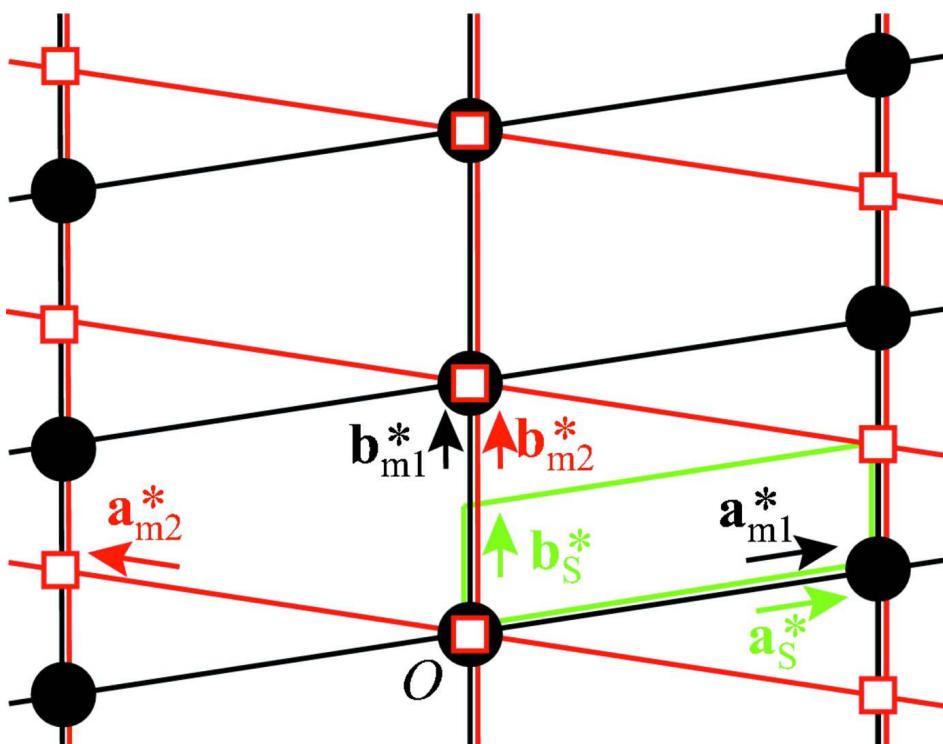
View of a part of the Nd₂Ti₂O₇ structure with displacement ellipsoids drawn at the 95% probability level. Symmetry codes are given in the geometric parameters Table.

**Figure 2**

Comparison of the $\text{Nd}_2\text{Ti}_2\text{O}_7$ structures, (a) present study, (b) Scheunemann & Müller-Buschbaum (1975), (c) Harvey *et al.* (2005). Unit cells and atomic parameters are shifted for comparison. Two kinds of perovskite-type slabs are illustrated using green- and orange-coloured TiO_6 octahedra, respectively. The small black arrows indicate the displacements of Nd atoms (red spheres) along \mathbf{b}^* with respect to the dashed lines drawn parallel to \mathbf{a} .

**Figure 3**

Powder diffraction diagrams of $\text{Nd}_2\text{Ti}_2\text{O}_7$ calculated from the atomic coordinates given by Scheunemann & Müller-Buschbaum (1975) (top) and Harvey *et al.* (2005) (bottom). Indices are given for several peaks which characterise the difference between the two structures.

**Figure 4**

Schematic representation of the $hk0$ reciprocal section of $\text{Nd}_2\text{Ti}_2\text{O}_7$. The monoclinic twin components, $m1$ and $m2$, are represented by black and red colour, respectively. The doubled monoclinic unit cell, S , determined by Scheunemann & Müller-Buschbaum (1975), and Harvey *et al.* (2005) is given in green colour. The c^* axes of all modifications are perpendicular to the paper.

Dineodymium(III) dititanium(IV) heptaoxide

Crystal data

$\text{Nd}_2\text{Ti}_2\text{O}_7$
 $M_r = 496.2$
Monoclinic, $P112_1$
Hall symbol: P 2zc
 $a = 7.6747 (1)$ Å
 $b = 13.0025 (2)$ Å
 $c = 5.4640 (1)$ Å
 $\beta = 90^\circ$
 $V = 539.24 (2)$ Å³
 $Z = 4$

$F(000) = 880$
 $D_x = 6.110 \text{ Mg m}^{-3}$
Mo $K\alpha$ radiation, $\lambda = 0.71069$ Å
Cell parameters from 12720 reflections
 $\theta = 3.2\text{--}40.2^\circ$
 $\mu = 21.77 \text{ mm}^{-1}$
 $T = 293$ K
Block, purple
 $0.11 \times 0.08 \times 0.07$ mm

Data collection

Bruker APEXII CCD
diffractometer
Radiation source: X-ray tube
 φ and ω scans
Absorption correction: multi-scan
(*TWINABS*; Bruker, 2008)
 $T_{\min} = 0.134$, $T_{\max} = 0.206$
25187 measured reflections

6608 independent reflections
6200 reflections with $I > 3\sigma(I)$
 $R_{\text{int}} = 0.049$
 $\theta_{\max} = 40.3^\circ$, $\theta_{\min} = 1.6^\circ$
 $h = -13 \rightarrow 13$
 $k = -23 \rightarrow 23$
 $l = -9 \rightarrow 9$

*Refinement*Refinement on F

$$R[F^2 > 2\sigma(F^2)] = 0.024$$

$$wR(F^2) = 0.027$$

$$S = 1.37$$

6608 reflections

133 parameters

0 restraints

1 constraint

Weighting scheme based on measured s.u.'s $w = 1/(\sigma^2(F) + 0.0001F^2)$

$$(\Delta/\sigma)_{\max} = 0.016$$

$$\Delta\rho_{\max} = 2.44 \text{ e } \text{\AA}^{-3}$$

$$\Delta\rho_{\min} = -1.57 \text{ e } \text{\AA}^{-3}$$

Extinction correction: B-C type 1 Gaussian isotropic (Becker & Coppens, 1974)

Extinction coefficient: 349 (12)

Absolute structure: Flack (1983), 3019 Friedel pairs

Absolute structure parameter: 0.220 (12)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Nd1	0.22817 (5)	0.906931 (15)	0.75507 (4)	0.00599 (5)
Nd2	0.14522 (4)	0.574954 (15)	0.34489 (4)	0.00563 (5)
Nd3	0.71953 (5)	0.881350 (15)	0.74678 (4)	0.00533 (5)
Nd4	0.64866 (4)	0.612953 (17)	0.28408 (4)	0.00796 (5)
Ti1	0.47020 (19)	0.87935 (5)	0.25800 (16)	0.00463 (15)
Ti2	0.41480 (19)	0.67491 (5)	0.78904 (13)	0.00476 (16)
Ti3	0.96662 (19)	0.88160 (5)	0.25934 (16)	0.00437 (15)
Ti4	0.92426 (19)	0.67911 (5)	0.78238 (14)	0.00521 (16)
O1	0.5306 (4)	0.9810 (2)	0.5246 (6)	0.0050 (6)*
O2	0.5001 (5)	0.7695 (3)	0.4494 (7)	0.0081 (6)*
O3	0.4073 (5)	0.5586 (2)	0.5773 (6)	0.0064 (5)*
O4	0.2248 (6)	0.8923 (2)	0.3081 (6)	0.0099 (5)*
O5	0.1746 (6)	0.6954 (2)	0.6923 (6)	0.0072 (5)*
O6	0.4338 (5)	0.8188 (2)	0.9387 (6)	0.0044 (5)*
O7	0.3756 (5)	0.6078 (3)	0.0680 (7)	0.0114 (7)*
O8	0.9591 (5)	0.9796 (3)	0.5265 (6)	0.0079 (6)*
O9	0.8867 (5)	0.7715 (3)	0.4553 (7)	0.0073 (6)*
O10	0.8731 (5)	0.5702 (2)	0.5633 (6)	0.0065 (5)*
O11	0.7275 (6)	0.9089 (2)	0.1730 (6)	0.0062 (5)*
O12	0.6740 (6)	0.6933 (2)	0.8421 (6)	0.0084 (5)*
O13	0.9747 (5)	0.8155 (2)	0.9432 (6)	0.0072 (6)*
O14	0.9226 (5)	0.5940 (2)	0.0518 (6)	0.0063 (6)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Nd1	0.00488 (8)	0.00601 (7)	0.00701 (8)	0.00062 (9)	-0.00008 (13)	0.00075 (7)
Nd2	0.00502 (8)	0.00630 (7)	0.00558 (8)	0.00087 (10)	-0.00014 (11)	-0.00051 (6)
Nd3	0.00446 (8)	0.00581 (7)	0.00580 (8)	0.00100 (9)	-0.00004 (12)	0.00079 (6)
Nd4	0.00543 (9)	0.01077 (8)	0.00800 (9)	0.00231 (10)	-0.00042 (12)	-0.00119 (6)
Ti1	0.0050 (3)	0.0047 (2)	0.0043 (2)	0.0010 (4)	0.0009 (7)	0.0001 (2)
Ti2	0.0042 (3)	0.0046 (2)	0.0052 (3)	-0.0002 (3)	0.0013 (6)	-0.00020 (18)
Ti3	0.0041 (3)	0.0049 (2)	0.0044 (2)	0.0014 (3)	0.0011 (7)	0.0000 (2)
Ti4	0.0055 (3)	0.0045 (2)	0.0056 (3)	0.0003 (4)	0.0013 (7)	-0.0005 (2)

Geometric parameters (\AA , \circ)

Nd1—O1	2.691 (3)	Nd4—O9	2.710 (3)
Nd1—O1 ⁱ	2.631 (3)	Nd4—O10	2.426 (4)
Nd1—O4	2.450 (3)	Nd4—O12 ^{vi}	2.627 (3)
Nd1—O5	2.742 (3)	Nd4—O14	2.499 (4)
Nd1—O6	2.312 (4)	Ti1—O1	1.975 (3)
Nd1—O8 ⁱⁱ	2.701 (4)	Ti1—O1 ^{vii}	2.220 (3)
Nd1—O8 ⁱ	2.659 (4)	Ti1—O2	1.812 (4)
Nd1—O11 ⁱ	2.411 (3)	Ti1—O4	1.935 (5)
Nd1—O13 ⁱⁱ	2.360 (4)	Ti1—O6 ^{vi}	1.917 (3)
Nd2—O3	2.415 (4)	Ti1—O11	2.010 (4)
Nd2—O5	2.450 (3)	Ti2—O2	2.269 (4)
Nd2—O7	2.318 (4)	Ti2—O3	1.898 (3)
Nd2—O10 ⁱⁱ	2.398 (4)	Ti2—O5	1.973 (4)
Nd2—O10 ⁱⁱⁱ	2.423 (3)	Ti2—O6	2.028 (3)
Nd2—O14 ⁱⁱ	2.381 (4)	Ti2—O7 ^v	1.760 (4)
Nd2—O14 ^{iv}	2.457 (3)	Ti2—O12	1.989 (5)
Nd3—O1	2.410 (3)	Ti3—O4 ^{viii}	1.984 (5)
Nd3—O2	2.621 (3)	Ti3—O8	1.944 (4)
Nd3—O4 ⁱ	2.931 (3)	Ti3—O8 ^{ix}	2.213 (3)
Nd3—O6	2.458 (4)	Ti3—O9	1.820 (4)
Nd3—O8	2.401 (3)	Ti3—O11	1.977 (4)
Nd3—O9	2.601 (4)	Ti3—O13 ^{vi}	1.935 (3)
Nd3—O11 ^v	2.355 (3)	Ti4—O5 ^{viii}	1.964 (4)
Nd3—O12	2.474 (3)	Ti4—O9	2.196 (4)
Nd3—O13	2.494 (4)	Ti4—O10	1.851 (3)
Nd4—O2	2.636 (4)	Ti4—O12	1.984 (5)
Nd4—O3	2.472 (4)	Ti4—O13	1.965 (3)
Nd4—O3 ⁱⁱⁱ	2.481 (3)	Ti4—O14 ^v	1.841 (3)
Nd4—O7	2.398 (4)		
O1—Ti1—O1 ^{vii}	84.56 (13)	O4 ^{viii} —Ti3—O8	88.84 (15)
O1—Ti1—O2	93.27 (15)	O4 ^{viii} —Ti3—O8 ^{ix}	83.46 (13)
O1—Ti1—O4	88.45 (14)	O4 ^{viii} —Ti3—O9	101.10 (15)
O1—Ti1—O6 ^{vi}	162.01 (13)	O4 ^{viii} —Ti3—O11	164.50 (12)
O1—Ti1—O11	85.18 (13)	O4 ^{viii} —Ti3—O13 ^{vi}	93.06 (16)
O1 ^{vii} —Ti1—O2	172.96 (16)	O8—Ti3—O8 ^{ix}	85.77 (14)
O1 ^{vii} —Ti1—O4	83.65 (13)	O8—Ti3—O9	91.99 (16)
O1 ^{vii} —Ti1—O6 ^{vi}	78.18 (12)	O8—Ti3—O11	86.75 (14)
O1 ^{vii} —Ti1—O11	80.38 (13)	O8—Ti3—O13 ^{vi}	165.40 (14)
O2—Ti1—O4	103.01 (15)	O8 ^{ix} —Ti3—O9	174.89 (17)
O2—Ti1—O6 ^{vi}	103.26 (15)	O8 ^{ix} —Ti3—O11	81.40 (13)
O2—Ti1—O11	92.78 (15)	O8 ^{ix} —Ti3—O13 ^{vi}	80.08 (13)
O4—Ti1—O6 ^{vi}	94.59 (16)	O9—Ti3—O11	93.89 (15)
O4—Ti1—O11	163.29 (13)	O9—Ti3—O13 ^{vi}	101.82 (15)
O6 ^{vi} —Ti1—O11	86.97 (15)	O11—Ti3—O13 ^{vi}	87.61 (16)
O2—Ti2—O3	84.58 (13)	O5 ^{viii} —Ti4—O9	86.72 (14)

O2—Ti2—O5	84.71 (13)	O5 ^{viii} —Ti4—O10	90.78 (16)
O2—Ti2—O6	81.60 (13)	O5 ^{viii} —Ti4—O12	167.53 (13)
O2—Ti2—O7 ^v	172.61 (17)	O5 ^{viii} —Ti4—O13	87.47 (15)
O2—Ti2—O12	81.44 (14)	O5 ^{viii} —Ti4—O14 ^v	100.64 (16)
O3—Ti2—O5	91.46 (16)	O9—Ti4—O10	82.06 (14)
O3—Ti2—O6	166.09 (13)	O9—Ti4—O12	82.86 (14)
O3—Ti2—O7 ^v	98.63 (15)	O9—Ti4—O13	83.96 (14)
O3—Ti2—O12	95.54 (16)	O9—Ti4—O14 ^v	171.88 (16)
O5—Ti2—O6	85.68 (14)	O10—Ti4—O12	94.53 (16)
O5—Ti2—O7 ^v	101.79 (17)	O10—Ti4—O13	165.99 (15)
O5—Ti2—O12	163.81 (12)	O10—Ti4—O14 ^v	94.32 (14)
O6—Ti2—O7 ^v	95.28 (15)	O12—Ti4—O13	84.64 (15)
O6—Ti2—O12	84.04 (14)	O12—Ti4—O14 ^v	90.22 (16)
O7 ^v —Ti2—O12	91.59 (17)	O13—Ti4—O14 ^v	99.66 (14)

Symmetry codes: (i) $-x+1, -y+2, z+1/2$; (ii) $x-1, y, z$; (iii) $-x+1, -y+1, z-1/2$; (iv) $-x+1, -y+1, z+1/2$; (v) $x, y, z+1$; (vi) $x, y, z-1$; (vii) $-x+1, -y+2, z-1/2$; (viii) $x+1, y, z$; (ix) $-x+2, -y+2, z-1/2$.