

r-2,c-6-Diphenylpiperidine

V. Maheshwaran,^a S. Abdul Basheer,^b A. Akila,^b
S. Ponnuswamy^b and M. N. Ponnuswamy^{a*}

^aCentre of Advanced Study in Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai 600 025, India, and ^bDepartment of Chemistry, Government Arts College (Autonomous), Coimbatore 641 018, India
Correspondence e-mail: mnpsy2004@yahoo.com

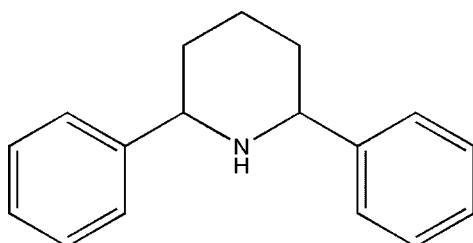
Received 1 July 2013; accepted 23 July 2013

Key indicators: single-crystal X-ray study; $T = 293\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$; R factor = 0.043; wR factor = 0.125; data-to-parameter ratio = 16.8.

In the title compound, $\text{C}_{17}\text{H}_{19}\text{N}$, the piperidine ring adopts a chair conformation. The phenyl rings substituted at the 2- and 6-positions of the piperidine ring subtend dihedral angles of $81.04(7)$ and $81.10(7)^\circ$ with the best plane of the piperidine ring. The crystal packing features $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For the biological activity of piperidine derivatives, see: Aridoss *et al.* (2009); Boehringer & Söhne GmbH (1961); Jain *et al.* (2005); Kubota *et al.* (1998); Mobio *et al.* (1989); Rubiralta *et al.* (1991). For the synthesis of the title compound, see: Ponnuswamy *et al.* (2002). For puckering parameters, see: Cremer & Pople (1975). For asymmetry parameters, see: Nardelli (1983).

**Experimental***Crystal data*

$\text{C}_{17}\text{H}_{19}\text{N}$	$\gamma = 81.466(9)^\circ$
$M_r = 237.33$	$V = 693.53(18)\text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 5.6450(9)\text{ \AA}$	$\text{Mo } K\alpha$ radiation
$b = 11.2255(17)\text{ \AA}$	$\mu = 0.07\text{ mm}^{-1}$
$c = 11.5281(17)\text{ \AA}$	$T = 293\text{ K}$
$\alpha = 73.911(9)^\circ$	$0.21 \times 0.19 \times 0.18\text{ mm}$
$\beta = 89.898(9)^\circ$	

Data collection

Bruker SMART APEXII CCD diffractometer	9781 measured reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2008)	2813 independent reflections
$T_{\min} = 0.986$, $T_{\max} = 0.988$	2113 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.032$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.043$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.125$	$\Delta\rho_{\max} = 0.11\text{ e \AA}^{-3}$
$S = 1.06$	$\Delta\rho_{\min} = -0.20\text{ e \AA}^{-3}$
2813 reflections	
167 parameters	

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$ and $Cg2$ are the centroids of the C13–C18 and C7–C12 rings, respectively.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}3-\text{H}3\cdots Cg1^i$	0.97	3.00	3.719 (2)	132
$\text{C}10-\text{H}10\cdots Cg1^{ii}$	0.93	3.01	3.760 (2)	139
$\text{C}16-\text{H}16\cdots Cg2^{iii}$	0.93	3.03	3.799 (2)	141

Symmetry codes: (i) $-x + 2, -y + 1, -z + 2$; (ii) $-x + 1, -y + 2, -z + 1$; (iii) $x, y - 1, z$.

Data collection: *APEX2* (Bruker, 2008); cell refinement: *SAINT* (Bruker, 2008); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012); software used to prepare material for publication: *SHELXL97* and *PLATON* (Spek, 2009).

SP thanks the UGC, New Delhi, for financial assistance in the form of a Major Research Project.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT6919).

References

- Aridoss, G., Parthiban, P., Ramachandran, R., Prakash, M., Kabilan, S. & Jeong, Y. T. (2009). *Eur. J. Med. Chem.* **44**, 577–592.
- Boehringer & Söhne GmbH (1961). *Chem. Abstr.* **55**, 24796g.
- Bruker (2008). *APEX2*, *SADABS* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Farrugia, L. J. (2012). *J. Appl. Cryst.* **45**, 849–854.
- Jain, R., Chen, D., White, R. J., Patel, D. V. & Yuan, Z. (2005). *Curr. Med. Chem.* **12**, 1607–1627.
- Kubota, H., Fujii, M., Ikeda, K., Takeuchi, M., Shibanuma, T. & Isomura, Y. (1998). *Chem. Pharm. Bull.* **46**, 351–354.
- Mobio, I. G., Soldatenkov, A. T., Federov, V. O., Ageev, E. A., Sargeeva, N. D., Lin, S., Stashenko, E. E., Prostakov, N. S. & Andreeva, E. I. (1989). *Khim. Farm. Zh.* **23**, 421–427.
- Nardelli, M. (1983). *Acta Cryst. C39*, 1141–1142.
- Ponnuswamy, S., Venkatraj, M., Jeyaraman, R., Suresh Kumar, M., Kumaran, D. & Ponnuswamy, M. N. (2002). *Indian J. Chem. Sect. B*, **41**, 614–627.
- Rubiralta, M., Giralt, E. & Diez, A. (1991). *Piperidine: Structure, Preparation, Reactivity, and Synthetic Applications of Piperidine and its Derivatives*, pp. 225–312. Amsterdam: Elsevier.
- Sheldrick, G. M. (2008). *Acta Cryst. A64*, 112–122.
- Spek, A. L. (2009). *Acta Cryst. D65*, 148–155.

Experimental*Crystal data*

$\text{C}_{17}\text{H}_{19}\text{N}$	$\gamma = 81.466(9)^\circ$
$M_r = 237.33$	$V = 693.53(18)\text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 5.6450(9)\text{ \AA}$	$\text{Mo } K\alpha$ radiation
$b = 11.2255(17)\text{ \AA}$	$\mu = 0.07\text{ mm}^{-1}$
$c = 11.5281(17)\text{ \AA}$	$T = 293\text{ K}$
$\alpha = 73.911(9)^\circ$	$0.21 \times 0.19 \times 0.18\text{ mm}$
$\beta = 89.898(9)^\circ$	

supporting information

Acta Cryst. (2013). E69, o1371 [doi:10.1107/S1600536813020382]

r-2,c-6-Diphenylpiperidine

V. Maheshwaran, S. Abdul Basheer, A. Akila, S. Ponnuswamy and M. N. Ponnuswamy

S1. Comment

Piperidines are valuable heterocyclic compounds found in natural substances and pharmaceutical products (Rubiralta *et al.*, 1991; Jain *et al.*, 2005; Kubota *et al.*, 1998). Several 2,6-substituted piperidine derivatives were found to be useful as tranquilisers (Boehringer & Söhne GmbH, 1961) and possess a wide range of biological activities such as antiviral, antimarial, antibacterial and antifungal activities (Aridoss *et al.*, 2009, Mobio *et al.*, 1989). In view of the above importance, the crystallographic study of the title compound has been carried out to establish its molecular structure.

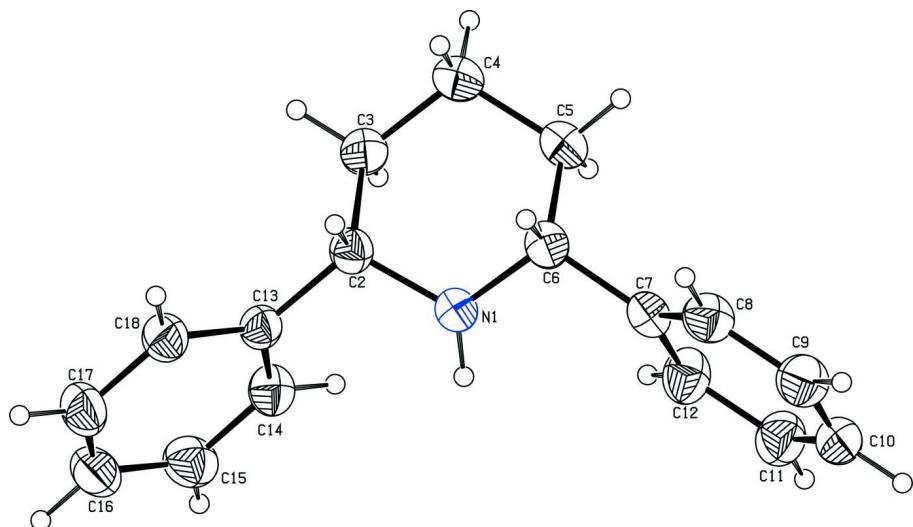
The *ORTEP* plot of the molecule is shown in Fig. 1. The piperidine ring adopts a chair conformation with puckering parameters (Cremer & Pople, 1975) and asymmetry parameters (Nardelli, 1983): $q_2=0.0420$ (15) Å, $q_3 = -0.5799$ (15) Å, $\varphi_2 = 190$ (2)° and Δ_s (N1& C4)= 0.75 (12)°. The phenyl rings at 2,6-positions of the piperidine ring occupy equatorial positions. The corresponding torsion angles are [C13—C2—C3—C4] -178.62 (11)° & [C4—C5—C6—C7] 177.89 (12)°, respectively. The dihedral angle between the two phenyl rings is 60.0 (7)°. The phenyl rings [C7—C12 & C13—C18] are twisted away from the best plane of the piperidine moiety by 81.04 (7)° & 81.10 (7)°, respectively. The molecules in the unit cell are connected by C—H $\cdots \pi$ interactions (Fig. 2 & Table. 1; Cg1 is the centroid of the ring C13 to C18 and Cg2 is the centroid of the ring C7 to C12).

S2. Experimental

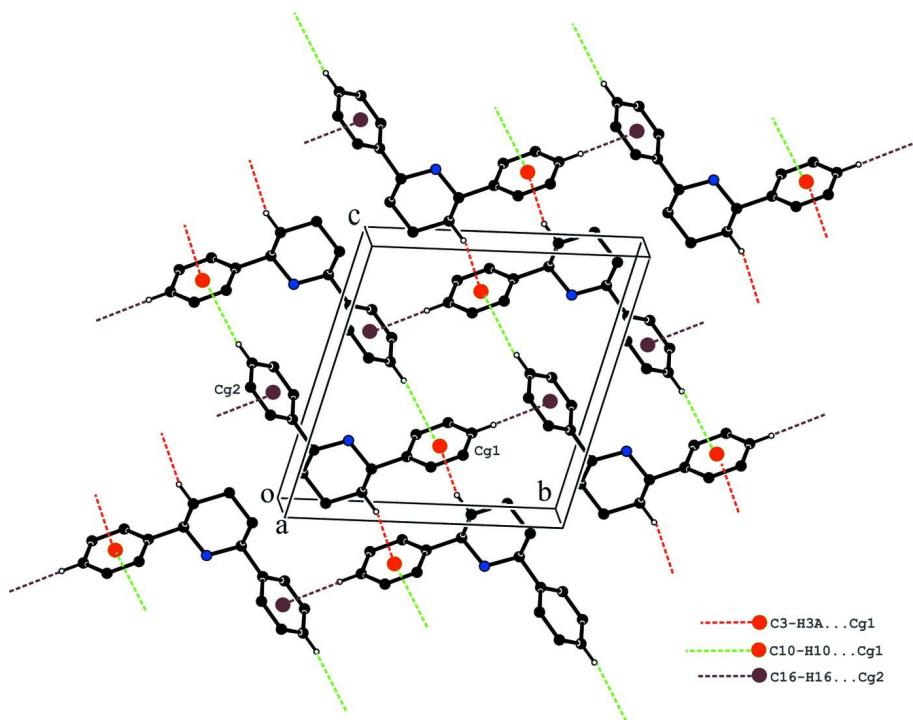
A mixture of piperidin-4-one (10 mM) and 80% hydrazine hydrate (3.1 ml) in diethylene glycol (100 ml) was heated on a steam bath for 2 hrs (Ponnuswamy *et al.*, 2002). Potassium hydroxide pellets (2.8 g) were added to the mixture and the contents were allowed to reflux vigorously on a heating mantle for another 2 hrs and the reaction mixture was cooled. The product formed was filtered and recrystallized from ethanol.

S3. Refinement

All H atoms were found in a difference map. Nevertheless, those bonded to C were positioned geometrically (C—H = 0.93–0.98 Å) and allowed to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. The H atom bonded to N was freely refined.

**Figure 1**

The molecular structure of the title compound, showing the atomic numbering and displacement ellipsoids drawn at 30% probability level.

**Figure 2**

The crystal packing of the molecules viewed down a -axis.

r-2,c-6-Diphenylpiperidine

Crystal data

$C_{17}H_{19}N$
 $M_r = 237.33$

Triclinic, $P\bar{1}$
Hall symbol: -P 1

$a = 5.6450 (9)$ Å
 $b = 11.2255 (17)$ Å
 $c = 11.5281 (17)$ Å
 $\alpha = 73.911 (9)^\circ$
 $\beta = 89.898 (9)^\circ$
 $\gamma = 81.466 (9)^\circ$
 $V = 693.53 (18)$ Å³
 $Z = 2$
 $F(000) = 256$

$D_x = 1.136$ Mg m⁻³
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 2113 reflections
 $\theta = 1.8\text{--}26.6^\circ$
 $\mu = 0.07$ mm⁻¹
 $T = 293$ K
Block, white
 $0.21 \times 0.19 \times 0.18$ mm

Data collection

Bruker SMART APEXII CCD
diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
 ω and φ scans
Absorption correction: multi-scan
(SADABS; Bruker, 2008)
 $T_{\min} = 0.986$, $T_{\max} = 0.988$

9781 measured reflections
2813 independent reflections
2113 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.032$
 $\theta_{\max} = 26.6^\circ$, $\theta_{\min} = 1.8^\circ$
 $h = -7 \rightarrow 7$
 $k = -13 \rightarrow 14$
 $l = -14 \rightarrow 14$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.125$
 $S = 1.06$
2813 reflections
167 parameters
0 restraints
Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map
Hydrogen site location: inferred from
neighbouring sites
H atoms treated by a mixture of independent
and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0582P)^2 + 0.0755P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.11$ e Å⁻³
 $\Delta\rho_{\min} = -0.20$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C2	0.3110 (2)	0.28559 (11)	0.14071 (11)	0.0526 (3)
H2	0.4718	0.2820	0.1084	0.063*
C3	0.1318 (3)	0.28661 (13)	0.04187 (13)	0.0649 (4)
H3A	-0.0288	0.2927	0.0720	0.078*
H3B	0.1375	0.3595	-0.0264	0.078*
C4	0.1868 (3)	0.16823 (14)	0.00071 (13)	0.0762 (4)
H4A	0.3376	0.1680	-0.0398	0.091*

H4B	0.0619	0.1672	-0.0565	0.091*
C5	0.2026 (3)	0.05172 (13)	0.10787 (13)	0.0680 (4)
H5A	0.2539	-0.0222	0.0809	0.082*
H5B	0.0452	0.0454	0.1407	0.082*
C6	0.3782 (2)	0.05592 (11)	0.20619 (12)	0.0547 (3)
H6	0.5379	0.0580	0.1730	0.066*
C7	0.3915 (2)	-0.05597 (11)	0.31576 (12)	0.0525 (3)
C8	0.5902 (3)	-0.14804 (13)	0.34202 (13)	0.0643 (4)
H8	0.7197	-0.1406	0.2919	0.077*
C9	0.5997 (3)	-0.25123 (14)	0.44165 (15)	0.0737 (4)
H9	0.7353	-0.3123	0.4580	0.088*
C10	0.4108 (3)	-0.26438 (14)	0.51663 (13)	0.0699 (4)
H10	0.4176	-0.3341	0.5835	0.084*
C11	0.2118 (3)	-0.17362 (14)	0.49196 (15)	0.0747 (4)
H11	0.0829	-0.1816	0.5425	0.090*
C12	0.2023 (3)	-0.07085 (13)	0.39271 (14)	0.0685 (4)
H12	0.0662	-0.0101	0.3769	0.082*
C13	0.2607 (2)	0.39961 (11)	0.18678 (11)	0.0501 (3)
C14	0.0632 (3)	0.41659 (13)	0.25525 (13)	0.0618 (4)
H14	-0.0411	0.3577	0.2714	0.074*
C15	0.0190 (3)	0.51912 (14)	0.29970 (15)	0.0735 (4)
H15	-0.1140	0.5289	0.3459	0.088*
C16	0.1715 (3)	0.60761 (14)	0.27597 (15)	0.0754 (4)
H16	0.1422	0.6768	0.3063	0.090*
C17	0.3659 (3)	0.59283 (14)	0.20757 (15)	0.0734 (4)
H17	0.4680	0.6527	0.1908	0.088*
C18	0.4119 (3)	0.48965 (12)	0.16320 (12)	0.0604 (4)
H18	0.5453	0.4804	0.1172	0.072*
N1	0.30501 (19)	0.17179 (9)	0.24060 (10)	0.0527 (3)
H1	0.397 (3)	0.1732 (13)	0.3027 (14)	0.066 (4)*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C2	0.0521 (7)	0.0509 (7)	0.0556 (7)	-0.0122 (5)	0.0034 (6)	-0.0143 (6)
C3	0.0785 (9)	0.0594 (8)	0.0559 (8)	-0.0142 (7)	-0.0073 (7)	-0.0128 (6)
C4	0.1052 (12)	0.0700 (9)	0.0583 (8)	-0.0204 (8)	-0.0091 (8)	-0.0225 (7)
C5	0.0859 (10)	0.0597 (8)	0.0655 (9)	-0.0183 (7)	-0.0016 (7)	-0.0255 (7)
C6	0.0543 (7)	0.0504 (7)	0.0614 (8)	-0.0093 (5)	0.0064 (6)	-0.0183 (6)
C7	0.0547 (7)	0.0476 (7)	0.0595 (7)	-0.0083 (5)	0.0012 (6)	-0.0217 (6)
C8	0.0604 (8)	0.0652 (9)	0.0672 (9)	-0.0004 (7)	0.0016 (7)	-0.0231 (7)
C9	0.0791 (10)	0.0627 (9)	0.0734 (10)	0.0087 (7)	-0.0137 (8)	-0.0192 (8)
C10	0.0935 (11)	0.0570 (8)	0.0580 (8)	-0.0140 (8)	-0.0090 (8)	-0.0130 (7)
C11	0.0819 (11)	0.0673 (9)	0.0728 (10)	-0.0147 (8)	0.0162 (8)	-0.0145 (8)
C12	0.0646 (9)	0.0564 (8)	0.0788 (10)	-0.0017 (6)	0.0114 (7)	-0.0134 (7)
C13	0.0523 (7)	0.0469 (7)	0.0489 (7)	-0.0089 (5)	-0.0042 (5)	-0.0093 (5)
C14	0.0610 (8)	0.0571 (8)	0.0674 (8)	-0.0119 (6)	0.0053 (7)	-0.0160 (7)
C15	0.0747 (10)	0.0723 (10)	0.0744 (10)	-0.0002 (8)	0.0043 (8)	-0.0277 (8)

C16	0.0932 (12)	0.0606 (9)	0.0760 (10)	-0.0017 (8)	-0.0151 (9)	-0.0300 (8)
C17	0.0845 (11)	0.0602 (9)	0.0810 (10)	-0.0244 (8)	-0.0089 (9)	-0.0215 (8)
C18	0.0621 (8)	0.0595 (8)	0.0618 (8)	-0.0179 (6)	-0.0014 (6)	-0.0162 (6)
N1	0.0583 (6)	0.0468 (6)	0.0540 (6)	-0.0091 (5)	-0.0049 (5)	-0.0154 (5)

Geometric parameters (\AA , $^{\circ}$)

C2—N1	1.4685 (16)	C9—C10	1.370 (2)
C2—C13	1.5064 (17)	C9—H9	0.9300
C2—C3	1.5217 (19)	C10—C11	1.372 (2)
C2—H2	0.9800	C10—H10	0.9300
C3—C4	1.5211 (19)	C11—C12	1.376 (2)
C3—H3A	0.9700	C11—H11	0.9300
C3—H3B	0.9700	C12—H12	0.9300
C4—C5	1.521 (2)	C13—C14	1.3857 (18)
C4—H4A	0.9700	C13—C18	1.3881 (17)
C4—H4B	0.9700	C14—C15	1.375 (2)
C5—C6	1.522 (2)	C14—H14	0.9300
C5—H5A	0.9700	C15—C16	1.380 (2)
C5—H5B	0.9700	C15—H15	0.9300
C6—N1	1.4648 (16)	C16—C17	1.367 (2)
C6—C7	1.5072 (18)	C16—H16	0.9300
C6—H6	0.9800	C17—C18	1.382 (2)
C7—C8	1.3789 (19)	C17—H17	0.9300
C7—C12	1.3867 (19)	C18—H18	0.9300
C8—C9	1.381 (2)	N1—H1	0.891 (16)
C8—H8	0.9300		
N1—C2—C13	109.73 (10)	C9—C8—H8	119.5
N1—C2—C3	108.14 (10)	C10—C9—C8	120.55 (14)
C13—C2—C3	113.09 (11)	C10—C9—H9	119.7
N1—C2—H2	108.6	C8—C9—H9	119.7
C13—C2—H2	108.6	C9—C10—C11	119.28 (14)
C3—C2—H2	108.6	C9—C10—H10	120.4
C4—C3—C2	111.00 (12)	C11—C10—H10	120.4
C4—C3—H3A	109.4	C10—C11—C12	120.17 (15)
C2—C3—H3A	109.4	C10—C11—H11	119.9
C4—C3—H3B	109.4	C12—C11—H11	119.9
C2—C3—H3B	109.4	C11—C12—C7	121.37 (14)
H3A—C3—H3B	108.0	C11—C12—H12	119.3
C3—C4—C5	110.73 (12)	C7—C12—H12	119.3
C3—C4—H4A	109.5	C14—C13—C18	118.18 (12)
C5—C4—H4A	109.5	C14—C13—C2	120.72 (11)
C3—C4—H4B	109.5	C18—C13—C2	121.09 (12)
C5—C4—H4B	109.5	C15—C14—C13	121.03 (13)
H4A—C4—H4B	108.1	C15—C14—H14	119.5
C4—C5—C6	111.35 (11)	C13—C14—H14	119.5
C4—C5—H5A	109.4	C14—C15—C16	120.15 (15)

C6—C5—H5A	109.4	C14—C15—H15	119.9
C4—C5—H5B	109.4	C16—C15—H15	119.9
C6—C5—H5B	109.4	C17—C16—C15	119.52 (14)
H5A—C5—H5B	108.0	C17—C16—H16	120.2
N1—C6—C7	110.02 (10)	C15—C16—H16	120.2
N1—C6—C5	108.40 (11)	C16—C17—C18	120.58 (14)
C7—C6—C5	112.75 (10)	C16—C17—H17	119.7
N1—C6—H6	108.5	C18—C17—H17	119.7
C7—C6—H6	108.5	C17—C18—C13	120.54 (14)
C5—C6—H6	108.5	C17—C18—H18	119.7
C8—C7—C12	117.66 (13)	C13—C18—H18	119.7
C8—C7—C6	121.30 (12)	C6—N1—C2	113.14 (10)
C12—C7—C6	121.04 (11)	C6—N1—H1	110.1 (9)
C7—C8—C9	120.97 (14)	C2—N1—H1	109.7 (9)
C7—C8—H8	119.5		
N1—C2—C3—C4	-56.90 (15)	C6—C7—C12—C11	179.08 (13)
C13—C2—C3—C4	-178.62 (11)	N1—C2—C13—C14	-50.66 (15)
C2—C3—C4—C5	53.73 (18)	C3—C2—C13—C14	70.16 (15)
C3—C4—C5—C6	-53.20 (18)	N1—C2—C13—C18	128.55 (12)
C4—C5—C6—N1	55.84 (15)	C3—C2—C13—C18	-110.63 (14)
C4—C5—C6—C7	177.89 (12)	C18—C13—C14—C15	-0.6 (2)
N1—C6—C7—C8	-130.46 (12)	C2—C13—C14—C15	178.63 (13)
C5—C6—C7—C8	108.40 (14)	C13—C14—C15—C16	0.3 (2)
N1—C6—C7—C12	50.58 (15)	C14—C15—C16—C17	0.3 (2)
C5—C6—C7—C12	-70.56 (16)	C15—C16—C17—C18	-0.6 (2)
C12—C7—C8—C9	-0.1 (2)	C16—C17—C18—C13	0.4 (2)
C6—C7—C8—C9	-179.06 (12)	C14—C13—C18—C17	0.25 (19)
C7—C8—C9—C10	0.1 (2)	C2—C13—C18—C17	-178.98 (12)
C8—C9—C10—C11	-0.2 (2)	C7—C6—N1—C2	174.32 (10)
C9—C10—C11—C12	0.2 (2)	C5—C6—N1—C2	-61.97 (13)
C10—C11—C12—C7	-0.2 (2)	C13—C2—N1—C6	-173.66 (10)
C8—C7—C12—C11	0.1 (2)	C3—C2—N1—C6	62.57 (14)

Hydrogen-bond geometry (Å, °)

Cg1 and Cg2 are the centroids of the C13—C18 and C7—C12 rings, respectively.

D—H···A	D—H	H···A	D···A	D—H···A
C3—H3A···Cg1 ⁱ	0.97	3.00	3.719 (2)	132
C10—H10···Cg1 ⁱⁱ	0.93	3.01	3.760 (2)	139
C16—H16···Cg2 ⁱⁱⁱ	0.93	3.03	3.799 (2)	141

Symmetry codes: (i) -x+2, -y+1, -z+2; (ii) -x+1, -y+2, -z+1; (iii) x, y-1, z.