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## Structure Reports

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# 1-Acetyl-5-(4-fluorophenyl)-2-sulfanylideneimidazolidin-4-one

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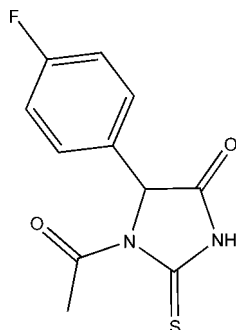
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 Key indicators: single-crystal X-ray study;  $T = 123$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.031;  $wR$  factor = 0.082; data-to-parameter ratio = 16.4.

In the title compound,  $\text{C}_{11}\text{H}_9\text{FN}_2\text{O}_2\text{S}$ , the 2-sulfanylideneimidazolidin-4-one moiety is essentially planar, with a maximum deviation of 0.0183 (14) Å. The mean plane of this moiety is approximately coplanar with the attached acetyl group and perpendicular to the benzene ring, making dihedral angles of 9.70 (14) and 86.70 (6)°, respectively. In the crystal, molecules are linked by  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds between the amide NH and acetyl  $\text{C}=\text{O}$  groups, forming a  $C(6)$  chain along the  $a$ -axis direction.

## Related literature

For applications and the biological activity of 2-sulfanylideneimidazolidin-4-ones, see: Marton *et al.* (1993). For the crystal structures of related compounds, see: Casas *et al.* (1998); Sulbaran *et al.* (2007); Taniguchi *et al.* (2009). For a description of the Cambridge Structural Database, see: Allen (2002). For hydrogen-bond motifs, see: Etter (1990). For the synthetic procedure, see: Schlack & Kumpf (1926).



## Experimental

### Crystal data

 $\text{C}_{11}\text{H}_9\text{FN}_2\text{O}_2\text{S}$ 
 $M_r = 252.27$ 

 Monoclinic,  $P2_1/n$   
 $a = 7.1327$  (9) Å  
 $b = 23.852$  (3) Å  
 $c = 7.3437$  (10) Å  
 $\beta = 113.541$  (3)°  
 $V = 1145.4$  (3) Å<sup>3</sup>
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.29$  mm<sup>-1</sup>  
 $T = 123$  K  
 $0.30 \times 0.10 \times 0.08$  mm

### Data collection

 Rigaku/MSC Mercury CCD diffractometer  
 Absorption correction: multi-scan (REQAB; Rigaku, 1998)  
 $T_{\min} = 0.829$ ,  $T_{\max} = 0.977$ 

 12234 measured reflections  
 2612 independent reflections  
 2418 reflections with  $F^2 > 2\sigma(F^2)$   
 $R_{\text{int}} = 0.024$ 

### Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.031$   
 $wR(F^2) = 0.082$   
 $S = 1.06$   
 2612 reflections  
 159 parameters

 H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.32$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.23$  e Å<sup>-3</sup>
**Table 1**

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H2}\cdots\text{O2}^i$	0.84 (2)	1.96 (2)	2.7836 (16)	167 (2)

 Symmetry code: (i)  $x + 1, y, z$ .

Data collection: *CrystalClear* (Rigaku, 2006); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SIR2008* in *Il Milione* (Burla *et al.*, 2007); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 2012) and *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *CrystalStructure* (Rigaku, 2010).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IS5315).

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## supporting information

*Acta Cryst.* (2013). E69, o1699 [doi:10.1107/S1600536813028560]

## 1-Acetyl-5-(4-fluorophenyl)-2-sulfanylideneimidazolidin-4-one

Soh-ichi Kitoh, Yijing Feng, Shuhei Fujinami, Masaki Ichitani, Mitsunori Honda and Ko-Ki Kunimoto

### S1. Comment

2-Sulfanylideneimidazolidin-4-one (2-thiohydantoin) derivatives are useful synthetic intermediates in a wide range of applications, such as therapeutics, fungicides and herbicides (Marton *et al.*, 1993). We have been studying crystal structures and hydrogen-bonding patterns of the polymorphic forms of 2-thiohydantoin derivatives. The Cambridge Structural Database survey (Ver. 5.34; Allen, 2002) indicates that 1-acetyl-2-thiohydantoin with an unsubstituted N atom show three types of N—H $\cdots$ O hydrogen-bonding patterns: (i) the amide NH and the acetyl C=O groups form a chain with a *C*(6) graph-set motif (Etter *et al.*, 1990) [triclinic polymorph of 1-acetyl-2-thiohydantoin (NIFHIT01) (Taniguchi *et al.*, 2009) and two other derivatives (KABRIQ and KOMGUO)]; (ii) the amide NH and the amide C=O groups form a chain with *C*(4) [monoclinic polymorph of 1-acetyl-2-thiohydantoin (NIFHIT) (Casas *et al.*, 1998) and one other derivative (DOKXUX)]; (iii) the amide NH and the amide C=O groups form a ring with *R*<sup>2</sup><sub>2</sub>(8) [1-acetyl-5-methyl-2-thiohydantoin (DIKWAW) (Sulbaran *et al.*, 2007)]. As an extension of our research, we report on the crystal structure of the title compound, C<sub>11</sub>H<sub>9</sub>FN<sub>2</sub>O<sub>2</sub>S.

In the title molecule (Fig. 1), the bond lengths and angles are normal and comparable to those observed in the reported 1-acetyl-2-thiohydantoin with an unsubstituted N atom. The 2-thiohydantoin moiety (N1/C1/S1/N2/C2/O1/C3) is essentially planar, with maximum deviations of 0.0183 (14) Å for C3 atom and -0.0138 (13) Å for N1 atom. The acetyl group (C4/O2/C5) is almost coplanar with the 2-thiohydantoin moiety, and the dihedral angle between the acetyl group and the 2-thiohydantoin moiety is 9.70 (14)°.

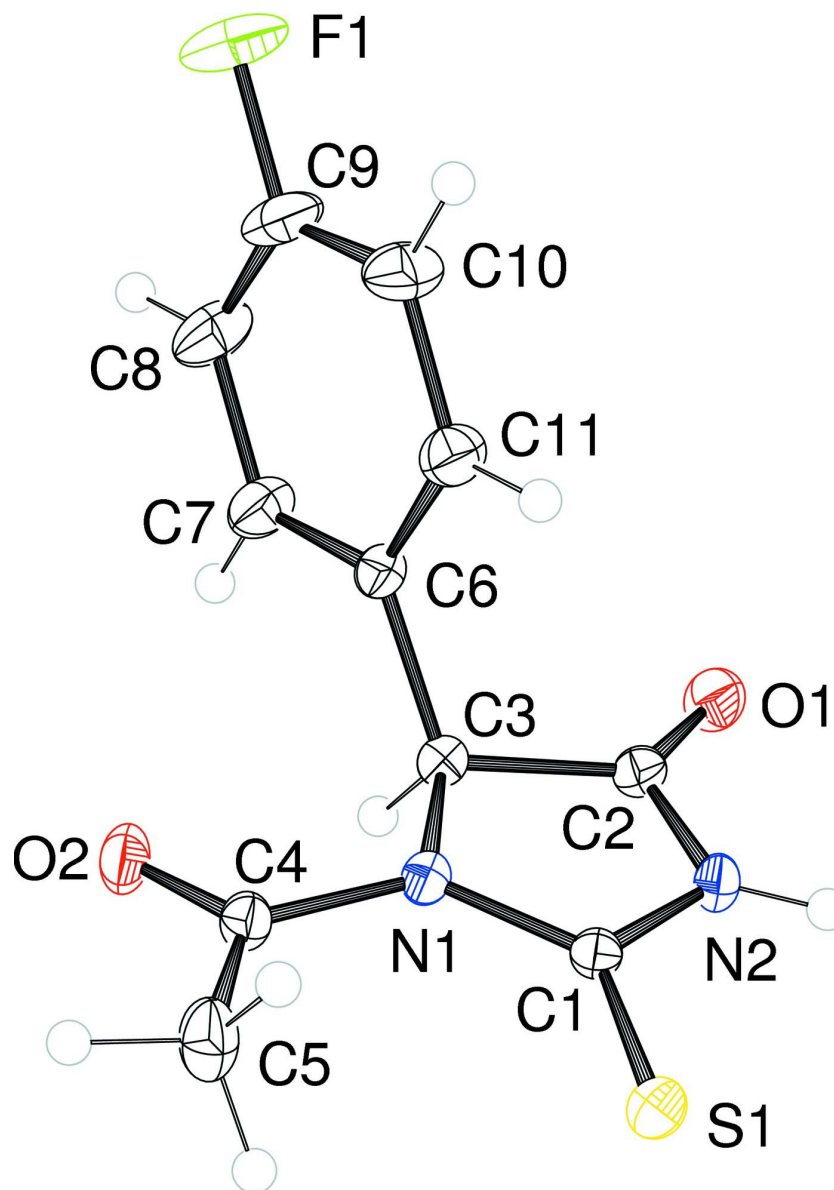
In the crystal structure (Fig. 2), the molecules are linked by an N—H $\cdots$ O hydrogen bond between the amide NH and acetyl C=O groups, forming an infinite one-dimensional chain along the *a* axis, with a *C*(6) graph-set motif (Table 1).

### S2. Experimental

The title compound was synthesized using a slight modification of a reported method (Schlack & Kumpf, 1926). 4-Fluorophenylglycine (0.507 g, 3.00 mmol) was allowed to react with a mixture of ammonium thiocyanate (0.234 g, 3.07 mmol), acetic anhydride (10 ml), and acetic acid (2 ml) at 100 °C for 1 h. A white precipitate was obtained by adding 25 ml distilled water and subsequent cooling the solution in a refrigerator. The crude product was purified by recrystallization from an ethanol solution (yield: 47%). Single crystals suitable for X-ray diffraction were obtained from the ethanol solution.

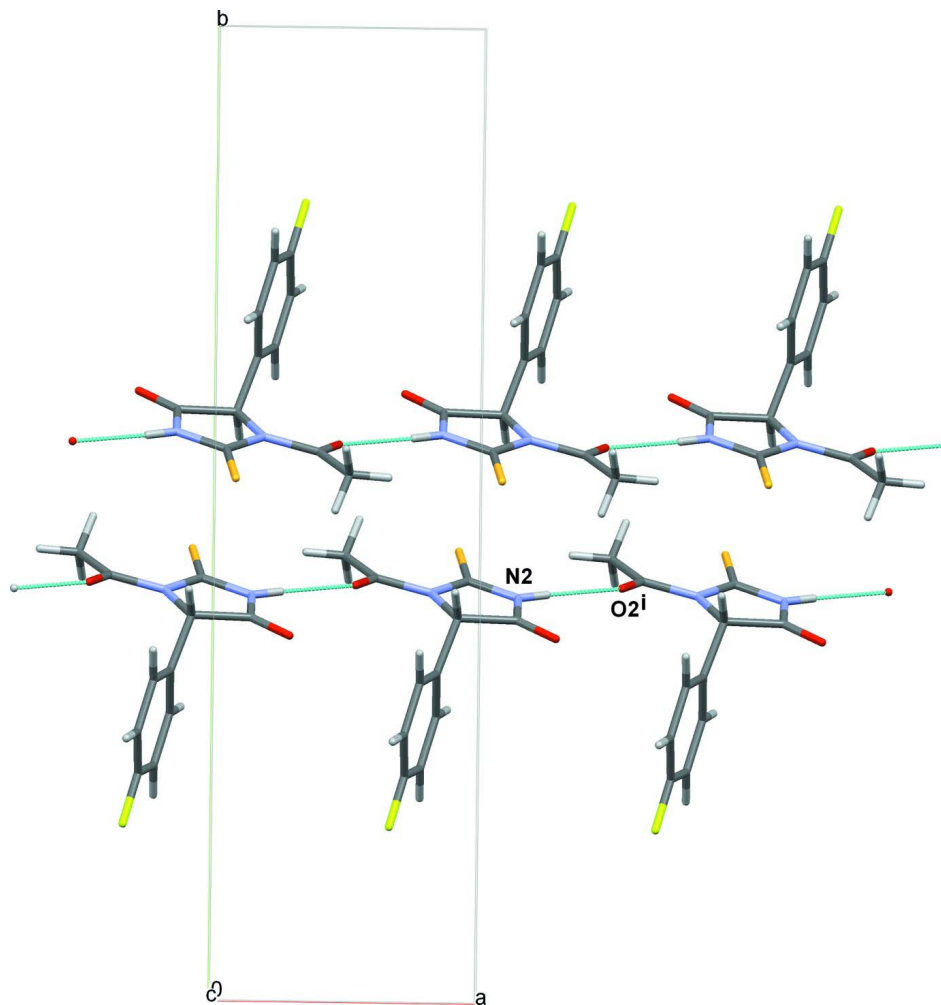
### S3. Refinement

The N-bound H atom was located in a difference map and refined freely [N2—H2 = 0.84 (2) Å]. The remaining H atoms were positioned geometrically (C—H = 0.95, 0.98 or 1.00 Å) and refined using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ . A rotating group model was applied to the methyl group.



**Figure 1**

The molecular structure of the title compound, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.



**Figure 2**

A partial packing diagram of the title compound, viewed down the *c* axis. Hydrogen bonds are shown as dashed cyan lines (see Table 1 for details).

### 1-Acetyl-5-(4-fluorophenyl)-2-sulfanylideneimidazolidin-4-one

#### Crystal data

$C_{11}H_9FN_2O_2S$

$M_r = 252.27$

Monoclinic,  $P2_1/n$

Hall symbol:  $-P 2_1n$

$a = 7.1327$  (9) Å

$b = 23.852$  (3) Å

$c = 7.3437$  (10) Å

$\beta = 113.541$  (3)°

$V = 1145.4$  (3) Å<sup>3</sup>

$Z = 4$

$F(000) = 520$

$D_x = 1.463$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71070$  Å

Cell parameters from 4829 reflections

$\theta = 3.0$ – $27.5$ °

$\mu = 0.29$  mm<sup>-1</sup>

$T = 123$  K

Prism, colorless

$0.30 \times 0.10 \times 0.08$  mm

*Data collection*

Rigaku/MSM Mercury CCD  
diffractometer  
Detector resolution: 7.314 pixels mm<sup>-1</sup>  
 $\omega$  scans  
Absorption correction: multi-scan  
(*REQAB*; Rigaku, 1998)  
 $T_{\min} = 0.829$ ,  $T_{\max} = 0.977$   
12234 measured reflections

2612 independent reflections  
2418 reflections with  $F^2 > 2\sigma(F^2)$   
 $R_{\text{int}} = 0.024$   
 $\theta_{\max} = 27.5^\circ$ ,  $\theta_{\min} = 3.4^\circ$   
 $h = -9 \rightarrow 9$   
 $k = -30 \rightarrow 30$   
 $l = -9 \rightarrow 8$

*Refinement*

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.031$   
 $wR(F^2) = 0.082$   
 $S = 1.06$   
2612 reflections  
159 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites  
H atoms treated by a mixture of independent  
and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0441P)^2 + 0.4044P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.32 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.23 \text{ e } \text{Å}^{-3}$

*Special details*

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement was performed using all reflections. The weighted  $R$ -factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ .  $R$ -factor (gt) are based on  $F$ . The threshold expression of  $F^2 > 2.0 \sigma(F^2)$  is used only for calculating  $R$ -factor (gt).

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{Å}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.92257 (4)	0.464054 (12)	0.18414 (4)	0.02040 (10)
F1	0.66717 (16)	0.18005 (4)	0.78913 (15)	0.0472 (3)
O1	1.28538 (13)	0.37499 (4)	0.83947 (13)	0.0254 (2)
O2	0.53419 (13)	0.42791 (4)	0.53528 (13)	0.0240 (2)
N1	0.81733 (14)	0.42098 (4)	0.47872 (14)	0.0166 (2)
N2	1.14408 (15)	0.41667 (4)	0.53118 (15)	0.0180 (2)
C1	0.95711 (17)	0.43386 (4)	0.39663 (17)	0.0164 (3)
C2	1.14023 (17)	0.39352 (5)	0.70131 (17)	0.0186 (3)
C3	0.91869 (16)	0.39534 (5)	0.67800 (16)	0.0166 (3)
C4	0.61024 (17)	0.43576 (5)	0.41600 (18)	0.0187 (3)
C5	0.49412 (19)	0.45857 (6)	0.21277 (19)	0.0265 (3)
C6	0.84028 (17)	0.33792 (5)	0.69937 (17)	0.0184 (3)
C7	0.7772 (2)	0.32809 (5)	0.8518 (2)	0.0264 (3)
C8	0.7187 (3)	0.27457 (6)	0.8832 (2)	0.0338 (4)
C9	0.7236 (3)	0.23252 (6)	0.7578 (3)	0.0314 (3)
C10	0.7823 (3)	0.24057 (6)	0.6034 (2)	0.0304 (3)
C11	0.8411 (2)	0.29430 (5)	0.57419 (19)	0.0249 (3)

H2	1.252 (3)	0.4205 (7)	0.513 (3)	0.027 (4)*
H3	0.9063	0.4211	0.7800	0.0199*
H5A	0.3491	0.4617	0.1885	0.0318*
H5B	0.5095	0.4333	0.1143	0.0318*
H5C	0.5474	0.4957	0.2019	0.0318*
H7	0.7738	0.3581	0.9356	0.0317*
H8	0.6765	0.2673	0.9884	0.0406*
H10	0.7828	0.2104	0.5189	0.0364*
H11	0.8822	0.3012	0.4681	0.0299*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.02194 (17)	0.02105 (16)	0.02123 (17)	0.00019 (10)	0.01181 (13)	0.00233 (10)
F1	0.0632 (7)	0.0242 (5)	0.0522 (6)	-0.0180 (4)	0.0208 (5)	0.0062 (4)
O1	0.0181 (5)	0.0278 (5)	0.0267 (5)	0.0001 (4)	0.0052 (4)	0.0056 (4)
O2	0.0175 (4)	0.0300 (5)	0.0283 (5)	0.0008 (4)	0.0131 (4)	0.0028 (4)
N1	0.0148 (5)	0.0173 (5)	0.0188 (5)	-0.0003 (4)	0.0078 (4)	0.0014 (4)
N2	0.0141 (5)	0.0196 (5)	0.0223 (5)	-0.0004 (4)	0.0095 (4)	-0.0004 (4)
C1	0.0170 (5)	0.0122 (5)	0.0220 (6)	-0.0019 (4)	0.0099 (5)	-0.0034 (4)
C2	0.0183 (6)	0.0156 (5)	0.0231 (6)	-0.0024 (4)	0.0094 (5)	-0.0016 (4)
C3	0.0162 (6)	0.0163 (6)	0.0179 (6)	-0.0008 (4)	0.0075 (5)	0.0006 (4)
C4	0.0153 (6)	0.0183 (6)	0.0235 (6)	-0.0011 (5)	0.0087 (5)	-0.0018 (5)
C5	0.0174 (6)	0.0386 (8)	0.0229 (6)	0.0029 (5)	0.0072 (5)	0.0036 (5)
C6	0.0151 (5)	0.0180 (6)	0.0218 (6)	-0.0016 (4)	0.0072 (5)	0.0006 (5)
C7	0.0316 (7)	0.0243 (7)	0.0279 (7)	-0.0064 (5)	0.0168 (6)	-0.0020 (5)
C8	0.0424 (8)	0.0321 (8)	0.0323 (7)	-0.0116 (6)	0.0205 (7)	0.0034 (6)
C9	0.0326 (7)	0.0207 (6)	0.0370 (8)	-0.0097 (6)	0.0098 (6)	0.0061 (6)
C10	0.0363 (8)	0.0187 (6)	0.0353 (8)	-0.0058 (6)	0.0135 (6)	-0.0051 (5)
C11	0.0284 (7)	0.0219 (6)	0.0278 (7)	-0.0035 (5)	0.0146 (6)	-0.0022 (5)

*Geometric parameters (Å, °)*

S1—C1	1.6454 (13)	C7—C8	1.391 (2)
F1—C9	1.3621 (19)	C8—C9	1.372 (3)
O1—C2	1.2073 (13)	C9—C10	1.370 (3)
O2—C4	1.2150 (19)	C10—C11	1.392 (2)
N1—C1	1.3899 (19)	N2—H2	0.84 (2)
N1—C3	1.4810 (14)	C3—H3	1.000
N1—C4	1.4053 (16)	C5—H5A	0.980
N2—C1	1.3676 (14)	C5—H5B	0.980
N2—C2	1.3762 (18)	C5—H5C	0.980
C2—C3	1.5206 (18)	C7—H7	0.950
C3—C6	1.5111 (18)	C8—H8	0.950
C4—C5	1.4906 (17)	C10—H10	0.950
C6—C7	1.383 (3)	C11—H11	0.950
C6—C11	1.3900 (19)		

C1—N1—C3	111.61 (9)	C8—C9—C10	123.48 (15)
C1—N1—C4	130.13 (10)	C9—C10—C11	117.85 (14)
C3—N1—C4	117.51 (11)	C6—C11—C10	120.41 (15)
C1—N2—C2	114.14 (12)	C1—N2—H2	123.1 (10)
S1—C1—N1	130.27 (8)	C2—N2—H2	122.7 (10)
S1—C1—N2	123.36 (11)	N1—C3—H3	109.392
N1—C1—N2	106.37 (11)	C2—C3—H3	109.399
O1—C2—N2	126.05 (13)	C6—C3—H3	109.408
O1—C2—C3	127.51 (13)	C4—C5—H5A	109.473
N2—C2—C3	106.44 (9)	C4—C5—H5B	109.477
N1—C3—C2	101.43 (11)	C4—C5—H5C	109.470
N1—C3—C6	114.97 (9)	H5A—C5—H5B	109.474
C2—C3—C6	111.93 (10)	H5A—C5—H5C	109.461
O2—C4—N1	116.01 (10)	H5B—C5—H5C	109.473
O2—C4—C5	123.27 (12)	C6—C7—H7	119.804
N1—C4—C5	120.71 (13)	C8—C7—H7	119.807
C3—C6—C7	119.42 (11)	C7—C8—H8	120.984
C3—C6—C11	120.67 (13)	C9—C8—H8	120.972
C7—C6—C11	119.82 (12)	C9—C10—H10	121.071
C6—C7—C8	120.39 (14)	C11—C10—H10	121.082
C7—C8—C9	118.04 (17)	C6—C11—H11	119.797
F1—C9—C8	118.03 (16)	C10—C11—H11	119.789
F1—C9—C10	118.50 (14)		
C1—N1—C3—C2	-1.25 (11)	O1—C2—C3—C6	-55.41 (16)
C1—N1—C3—C6	-122.21 (10)	N2—C2—C3—N1	0.71 (11)
C3—N1—C1—S1	-178.11 (9)	N2—C2—C3—C6	123.78 (9)
C3—N1—C1—N2	1.32 (11)	N1—C3—C6—C7	-126.90 (11)
C1—N1—C4—O2	-166.54 (10)	N1—C3—C6—C11	56.50 (14)
C1—N1—C4—C5	14.60 (17)	C2—C3—C6—C7	118.06 (11)
C4—N1—C1—S1	-8.49 (18)	C2—C3—C6—C11	-58.54 (12)
C4—N1—C1—N2	170.93 (10)	C3—C6—C7—C8	-175.27 (9)
C3—N1—C4—O2	2.57 (15)	C3—C6—C11—C10	175.46 (9)
C3—N1—C4—C5	-176.29 (9)	C7—C6—C11—C10	-1.12 (16)
C4—N1—C3—C2	-172.31 (9)	C11—C6—C7—C8	1.35 (17)
C4—N1—C3—C6	66.73 (13)	C6—C7—C8—C9	-0.67 (18)
C1—N2—C2—O1	179.24 (10)	C7—C8—C9—F1	179.71 (11)
C1—N2—C2—C3	0.04 (12)	C7—C8—C9—C10	-0.3 (2)
C2—N2—C1—S1	178.64 (9)	F1—C9—C10—C11	-179.49 (10)
C2—N2—C1—N1	-0.83 (12)	C8—C9—C10—C11	0.5 (2)
O1—C2—C3—N1	-178.48 (12)	C9—C10—C11—C6	0.21 (18)

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N2—H2 $\cdots$ O2 <sup>i</sup>	0.84 (2)	1.96 (2)	2.7836 (16)	167 (2)

Symmetry code: (i)  $x+1, y, z$ .