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Crystal structure of Hg₂SO₄ – a redetermination

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The crystal structure of mercury(I) sulfate (or mercurous sulfate), Hg₂SO₄, was re-determined based on modern CCD data. In comparison with the previous determination from Weissenberg film data [Dorm (1969). *Acta Chem. Scand.* **23**, 1607–1615], all atoms were refined with anisotropic displacement parameters, leading to higher precision in terms of bond lengths and angles [*e.g.* Hg–Hg = 2.5031 (7) compared to 2.500 (3) Å]. The structure consists of alternating rows along [001] of Hg₂²⁺ dumbbells (generated by inversion symmetry) and SO₄²⁻ tetrahedra (symmetry 2). The dumbbells are linked *via* short O–Hg–Hg–O bonds to the sulfate tetrahedra into chains extending parallel to [20 $\bar{1}$]. More remote O–Hg–Hg–O bonds connect these chains into a three-dimensional framework.

Keywords: crystal structure; redetermination; mercurous; Hg/S/O system.

CCDC reference: 1004277

1. Related literature

Structural data of the previous refinement of Hg₂SO₄ (Dorm, 1969) were deposited with the ICSD (2014), but contain an error in the *z* coordinate of the sulfur atom. Other phases in the system Hg/S/O were structurally characterized by Aurivillius & Stålhandske (1980) [HgSO₄], Weil (2001) [Hg₃(SO₄)O₂] and Logemann & Wickleder (2013) [Hg(S₂O₇)]. For a review on Hg–Hg bond lengths in mercurous compounds, see: Weil *et al.* (2005).

2. Experimental

2.1. Crystal data

Hg₂O₄S
M_r = 497.24
 Monoclinic, *P2*/*c*
 $a = 6.2771$ (8) Å
 $b = 4.4290$ (6) Å
 $c = 8.3596$ (10) Å

$\beta = 91.695$ (4)°
 $V = 232.31$ (5) Å³
 $Z = 2$
 Mo *K*α radiation

$\mu = 66.35$ mm⁻¹
 $T = 295$ K
 $0.18 \times 0.08 \times 0.04$ mm

2.2. Data collection

Bruker SMART CCD diffractometer
 Absorption correction: numerical (*HABITUS*; Herrendorf, 1997)
 $T_{\min} = 0.012$, $T_{\max} = 0.119$
 1737 measured reflections
 701 independent reflections
 629 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.060$

2.3. Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.087$
 $S = 1.06$
 701 reflections
 34 parameters
 $\Delta\rho_{\max} = 3.57$ e Å⁻³
 $\Delta\rho_{\min} = -3.18$ e Å⁻³

Table 1

Selected geometric parameters (Å, °).

| | | | |
|------------------------------------|-------------|--------------------|------------|
| Hg–O ⁱ | 2.193 (6) | Hg–Hg ^v | 2.5031 (7) |
| Hg–O ⁱⁱ | 2.518 (6) | S–O1 | 1.450 (7) |
| Hg–O1 ⁱⁱⁱ | 2.725 (6) | S–O2 | 1.509 (6) |
| Hg–O1 ^{iv} | 2.898 (7) | | |
| O ⁱ –Hg–Hg ^v | 164.47 (14) | | |

Symmetry codes: (i) $-x + 1, -y + 1, -z$; (ii) $x, y - 1, z$; (iii) $-x + 1, y, -z + \frac{1}{2}$; (iv) $-x + 1, -y, -z$; (v) $-x, -y, -z$.

Data collection: *SMART* (Bruker, 2005); cell refinement: *SAINTE* (Bruker, 2005); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ATOMS for Windows* (Dowty, 2006); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

Acknowledgements

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Supporting information for this paper is available from the IUCr electronic archives (Reference: HB0012).

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supporting information

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Crystal structure of Hg_2SO_4 – a redetermination

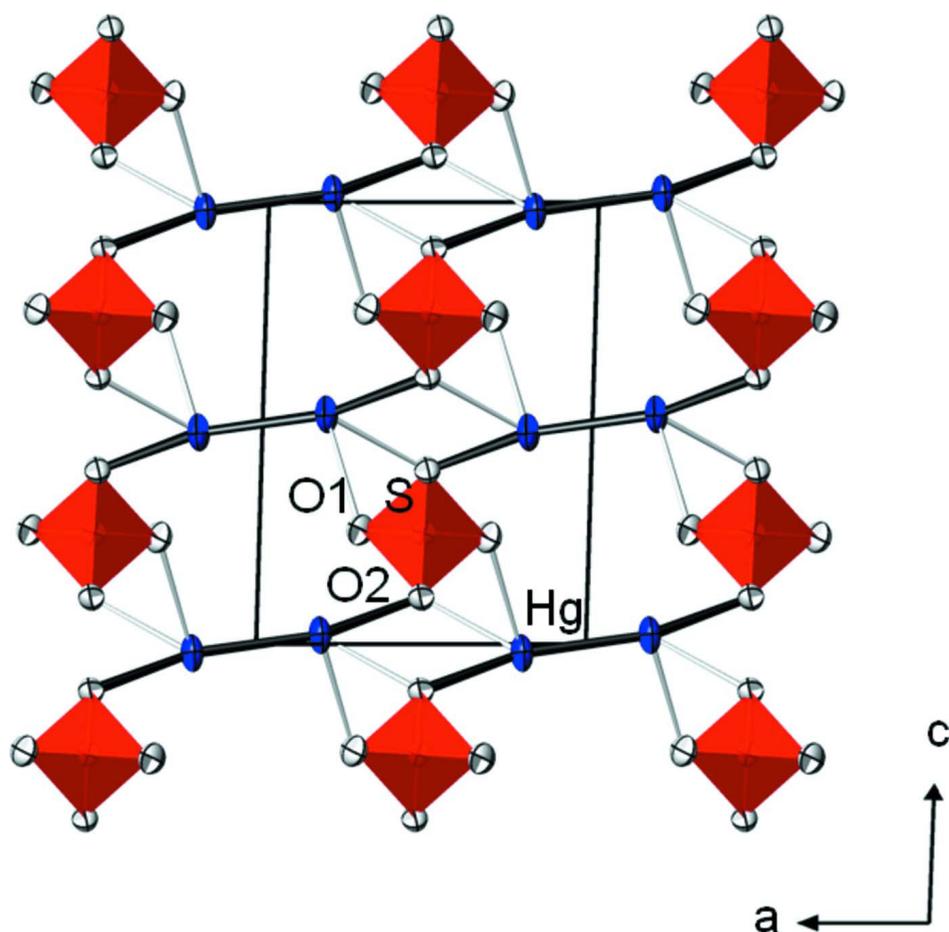
Matthias Weil

S1. Experimental

1 g HgO was suspended in 20 ml water. 4 ml sulfuric acid (96%_{w/w}) and 2 drops CS_2 were added to the mixture, transferred into a 50 ml polypropylene beaker that was sealed and heated for 12 h at 393 K. Besides a polycrystalline dirty-white solid with an unknown diffraction pattern, few colourless and transparent single crystals of the title compound were present in the reaction mixture.

S2. Refinement

The coordinates of the previous refinement (Dorm, 1969) were used as starting parameters. The highest and lowest remaining electron density is 0.84 Å and 1.25 Å, respectively, from the Hg atom. It should be noted that in the current version (01/2014) of the Inorganic Structure Data Base (ICSD, 2014), the deposited structure data of the previous refinement by Dorm (1969) contain an error: The z parameter of the sulfur atom must be 1/4, not 3/4.

**Figure 1**

The crystal structure of Hg_2SO_4 in a projection along $[010]$. Displacement ellipsoids are drawn at the 74% probability level; short Hg—O bonds are displayed with closed black lines, longer Hg—O bonds with open lines.

Mercury(I) sulfate

Crystal data

$\text{Hg}_2\text{O}_4\text{S}$

$M_r = 497.24$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 6.2771\ (8)\ \text{\AA}$

$b = 4.4290\ (6)\ \text{\AA}$

$c = 8.3596\ (10)\ \text{\AA}$

$\beta = 91.695\ (4)^\circ$

$V = 232.31\ (5)\ \text{\AA}^3$

$Z = 2$

$F(000) = 416$

$D_x = 7.109\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 1233 reflections

$\theta = 3.3\text{--}30.4^\circ$

$\mu = 66.35\ \text{mm}^{-1}$

$T = 295\ \text{K}$

Fragment, colourless

$0.18 \times 0.08 \times 0.04\ \text{mm}$

Data collection

Bruker SMART CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω -scans

Absorption correction: numerical
(*HABITUS*; Herrendorf, 1997)

$T_{\min} = 0.012$, $T_{\max} = 0.119$

1737 measured reflections

701 independent reflections

629 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.060$
 $\theta_{\text{max}} = 30.4^\circ$, $\theta_{\text{min}} = 4.6^\circ$

$h = -8 \rightarrow 8$
 $k = -6 \rightarrow 6$
 $l = -11 \rightarrow 8$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.035$
 $wR(F^2) = 0.087$
 $S = 1.06$
 701 reflections
 34 parameters
 0 restraints

Primary atom site location: isomorphous
 structure methods
 $w = 1/[\sigma^2(F_o^2) + (0.052P)^2 + 1.7115P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 3.57 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -3.18 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL*,
 $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.0118 (12)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|----|-------------|-------------|--------------|----------------------------------|
| Hg | 0.19318 (5) | 0.05289 (9) | -0.02034 (4) | 0.0275 (2) |
| S | 0.5000 | 0.5674 (5) | 0.2500 | 0.0134 (5) |
| O1 | 0.6943 (11) | 0.3901 (16) | 0.2586 (8) | 0.0224 (12) |
| O2 | 0.5038 (9) | 0.7720 (13) | 0.1058 (6) | 0.0172 (10) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|----|-------------|-------------|-------------|---------------|--------------|--------------|
| Hg | 0.0119 (2) | 0.0392 (3) | 0.0315 (3) | -0.00369 (11) | 0.00211 (13) | 0.00072 (13) |
| S | 0.0125 (12) | 0.0141 (11) | 0.0136 (11) | 0.000 | 0.0000 (8) | 0.000 |
| O1 | 0.018 (3) | 0.027 (3) | 0.022 (3) | 0.009 (2) | -0.001 (2) | 0.004 (2) |
| O2 | 0.015 (2) | 0.020 (2) | 0.016 (2) | 0.001 (2) | 0.0010 (18) | 0.004 (2) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|----------------------|------------|----------------------|-------------|
| Hg—O2 ⁱ | 2.193 (6) | S—Hg ⁱⁱⁱ | 3.7082 (15) |
| Hg—O2 ⁱⁱ | 2.518 (6) | S—Hg ^{ix} | 3.8936 (17) |
| Hg—O1 ⁱⁱⁱ | 2.725 (6) | S—Hg ^{iv} | 3.8936 (17) |
| Hg—O1 ^{iv} | 2.898 (7) | O1—Hg ⁱⁱⁱ | 2.725 (6) |
| Hg—Hg ^v | 2.5031 (7) | O1—Hg ^{iv} | 2.898 (7) |
| S—O1 ⁱⁱⁱ | 1.450 (7) | O1—Hg ⁱ | 3.261 (7) |
| S—O1 | 1.450 (7) | O1—Hg ^{vii} | 3.716 (7) |

| | | | |
|---|-------------|--|-------------|
| S—O2 ⁱⁱⁱ | 1.509 (6) | O1—Hg ^x | 4.090 (6) |
| S—O2 | 1.509 (6) | O2—Hg ⁱ | 2.193 (6) |
| S—Hg ^{vi} | 3.2315 (13) | O2—Hg ^{viii} | 2.518 (6) |
| S—Hg ⁱ | 3.2315 (13) | O2—Hg ^{vi} | 3.812 (5) |
| S—Hg ^{vii} | 3.6309 (15) | O2—Hg ^{vii} | 4.097 (6) |
| S—Hg ^{viii} | 3.6309 (15) | | |
| O2 ⁱ —Hg—Hg ^v | 164.47 (14) | S—O1—Hg ^x | 156.0 (3) |
| O2 ⁱ —Hg—O2 ⁱⁱ | 69.1 (2) | Hg ⁱⁱⁱ —O1—Hg ^x | 36.63 (8) |
| Hg ^v —Hg—O2 ⁱⁱ | 126.30 (13) | Hg ^{iv} —O1—Hg ^x | 77.69 (14) |
| O2 ⁱ —Hg—O1 ⁱⁱⁱ | 82.0 (2) | Hg ⁱ —O1—Hg ^x | 117.52 (19) |
| Hg ^v —Hg—O1 ⁱⁱⁱ | 102.86 (14) | Hg ^{vii} —O1—Hg ^x | 89.04 (14) |
| O2 ⁱⁱ —Hg—O1 ⁱⁱⁱ | 75.8 (2) | S—O1—Hg | 62.8 (3) |
| O2 ⁱ —Hg—O1 ^{iv} | 77.5 (2) | Hg ⁱⁱⁱ —O1—Hg | 115.4 (2) |
| Hg ^v —Hg—O1 ^{iv} | 102.91 (14) | Hg ^{iv} —O1—Hg | 64.22 (13) |
| O2 ⁱⁱ —Hg—O1 ^{iv} | 75.64 (18) | Hg ⁱ —O1—Hg | 95.95 (15) |
| O1 ⁱⁱⁱ —Hg—O1 ^{iv} | 149.3 (3) | Hg ^{vii} —O1—Hg | 138.14 (18) |
| O1 ⁱⁱⁱ —S—O1 | 114.5 (6) | Hg ^x —O1—Hg | 129.82 (18) |
| O1 ⁱⁱⁱ —S—O2 ⁱⁱⁱ | 109.4 (3) | S—O2—Hg ⁱ | 120.5 (3) |
| O1—S—O2 ⁱⁱⁱ | 108.6 (4) | S—O2—Hg ^{viii} | 126.9 (3) |
| O1 ⁱⁱⁱ —S—O2 | 108.6 (4) | Hg ⁱ —O2—Hg ^{viii} | 110.9 (2) |
| O1—S—O2 | 109.4 (3) | S—O2—Hg ^{vi} | 56.41 (19) |
| O2 ⁱⁱⁱ —S—O2 | 106.2 (5) | Hg ⁱ —O2—Hg ^{vi} | 131.7 (2) |
| S—O1—Hg ⁱⁱⁱ | 122.3 (4) | Hg ^{viii} —O2—Hg ^{vi} | 80.47 (14) |
| S—O1—Hg ^{iv} | 123.6 (4) | S—O2—Hg | 72.7 (2) |
| Hg ⁱⁱⁱ —O1—Hg ^{iv} | 96.8 (2) | Hg ⁱ —O2—Hg | 129.6 (2) |
| S—O1—Hg ⁱ | 76.0 (3) | Hg ^{viii} —O2—Hg | 85.11 (15) |
| Hg ⁱⁱⁱ —O1—Hg ⁱ | 148.1 (3) | Hg ^{vi} —O2—Hg | 97.20 (13) |
| Hg ^{iv} —O1—Hg ⁱ | 91.77 (18) | S—O2—Hg ^{vii} | 61.6 (2) |
| S—O1—Hg ^{vii} | 75.3 (3) | Hg ⁱ —O2—Hg ^{vii} | 77.46 (15) |
| Hg ⁱⁱⁱ —O1—Hg ^{vii} | 85.41 (17) | Hg ^{viii} —O2—Hg ^{vii} | 122.59 (19) |
| Hg ^{iv} —O1—Hg ^{vii} | 153.3 (2) | Hg ^{vi} —O2—Hg ^{vii} | 58.72 (8) |
| Hg ⁱ —O1—Hg ^{vii} | 73.82 (14) | Hg—O2—Hg ^{vii} | 134.35 (14) |

Symmetry codes: (i) $-x+1, -y+1, -z$; (ii) $x, y-1, z$; (iii) $-x+1, y, -z+1/2$; (iv) $-x+1, -y, -z$; (v) $-x, -y, -z$; (vi) $x, -y+1, z+1/2$; (vii) $-x+1, y+1, -z+1/2$; (viii) $x, y+1, z$; (ix) $x, -y, z+1/2$; (x) $x+1, -y, z+1/2$.