

Crystal structure of 2-(4-methylphenyl)-4*H*-1,3-benzothiazine

N. C. Sandhya,^a Chandra,^b G. P. Suresha,^a
N. K. Lokanath^b and M. Mahendra^{b*}

^aDepartment of Studies in Chemistry, Manasagangotri, University of Mysore, Mysore 570 006, India, and ^bDepartment of Studies in Physics, Manasagangotri, University of Mysore, Mysore 570 006, India. *Correspondence e-mail: mahendra@physics.uni-mysore.ac.in

Received 29 November 2014; accepted 12 December 2014

Edited by H. Ishida, Okayama University, Japan

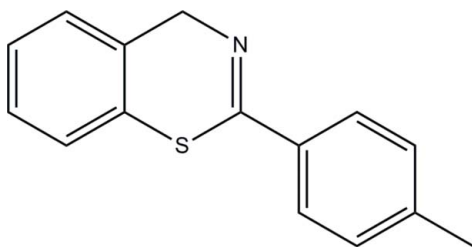
In the title compound, C₁₅H₁₃NS, the thiazine ring adopts a boat conformation. The dihedral angle between the planes of the benzene ring of the benzothiazine unit and the tolyl ring is 19.52 (9)°. In the crystal, molecules are linked by weak C—H···π interactions into a tape structure along the *b*-axis direction.

Keywords: crystal structure; benzothiazine derivative; biological properties; C—H···π interactions.

CCDC reference: 1039090

1. Related literature

For the biological importance of benzothiazine derivatives, see: Ahmad *et al.* (2010); Gupta *et al.* (2002); Lazzeri *et al.* (2001); Parveen *et al.* (2014); Zia-ur-Rehman *et al.* (2006).



2. Experimental

2.1. Crystal data

C₁₅H₁₃NS

M_r = 239.33

Monoclinic, *P*2₁/*c*
a = 15.1241 (9) Å
b = 6.0111 (4) Å
c = 14.3212 (9) Å
β = 110.268 (2)°
V = 1221.36 (13) Å³

Z = 4
Cu *Kα* radiation
μ = 2.13 mm⁻¹
T = 293 K
0.30 × 0.25 × 0.20 mm

2.2. Data collection

Bruker X8 Proteum diffractometer
10379 measured reflections
1988 independent reflections

1890 reflections with *I* > 2σ(*I*)
*R*_{int} = 0.041

2.3. Refinement

R[*F*² > 2σ(*F*²)] = 0.036
wR(*F*²) = 0.102
S = 1.05
1988 reflections

156 parameters
H-atom parameters constrained
Δρ_{max} = 0.18 e Å⁻³
Δρ_{min} = -0.20 e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

*C*_g is the centroid of the C3/C2/C7–C10 ring.

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C7—H7··· <i>C</i> _g ⁱ	0.93	2.75	3.485 (2)	136

Symmetry code: (i) $-x, y + \frac{1}{2}, -z - \frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

Acknowledgements

MM would like to thank the UGC, New Delhi, Government of India, for the award of a project under the head F. No. 41–920/2012(SR) dated: 25-07-2012.

Supporting information for this paper is available from the IUCr electronic archives (Reference: IS5385).

References

- Ahmad, M., Siddiqui, H. L., Zia-ur-Rehman, M. & Parvez, M. (2010). *Eur. J. Med. Chem.* **45**, 698–704.
- Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Gupta, S. K., Bansal, P., Bhardwaj, R. K., Jaiswal, J. & Velpandian, T. (2002). *Skin Pharmacol. Appl. Skin Physiol.* **15**, 105–111.
- Lazzeri, N., Belvisi, M. G., Patel, H. J., Yacoub, M. H., Chung, K. F. & Mitchell, J. A. (2001). *Am. J. Respir. Cell Mol. Biol.* **24**, 44–48.
- Parveen, S., Hussain, S., Zhu, S., Qin, X., Hao, X., Zhang, S., Lu, J. & Zhu, C. (2014). *RSC Adv.* **4**, 21134–21140.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Zia-ur-Rehman, M., Choudary, J. A., Ahmad, S. & Siddiqui, H. L. (2006). *Chem. Pharm. Bull.* **54**, 1175–1178.

supporting information

Acta Cryst. (2015). E71, o74 [doi:10.1107/S205698901402725X]

Crystal structure of 2-(4-methylphenyl)-4*H*-1,3-benzothiazine

N. C. Sandhya, Chandra, G. P. Suresha, N. K. Lokanath and M. Mahendra

S1. Comment

Benzothiazines have been found to possess versatile biological activities, such as analgesic (Gupta *et al.*, 2002), anti-bacterial (Zia-ur-Rehman *et al.*, 2006) and antioxidant activities (Ahmad *et al.*, 2010). Also, benzothiazine derivatives have shown activities for the treatment of asthmatic therapy (Lazzeri *et al.*, 2001). Recently, 1,2-benzothiazine-1,1-dioxide and its derivatives were reported as aldose reductase inhibitors (Parveen *et al.*, 2014). With this potential background of benzothiazine derivatives, we have synthesized the title compound to study its crystal structure.

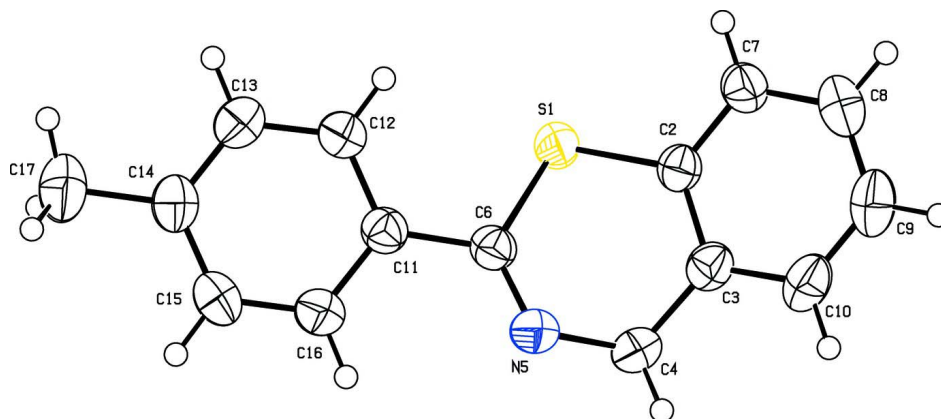
In the title compound (Fig. 1), the mean plane of the benzothiazine moiety (S1/C2/C7–C10/C3/C4/N5/C6) makes a dihedral angle of 19.52 (9)° with the benzene ring (C11–C16). The central thiazine ring adopts a boat conformation with puckering parameter $Q = 0.5848$ (16) Å and $\varphi = 183.41$ (17)°, and the maximum deviation found on the puckered atom at C6 is -0.170 (2) Å. There are no classic hydrogen bonds. Instead, a weak C—H $\cdots\pi$ interaction is observed (C7—H7 \cdots Cgⁱ; Cg: C3/C2/C7–C10; Table 1). The molecular packing exhibits layered stacking when viewed down the *b* axis as shown in Fig. 2.

S2. Experimental

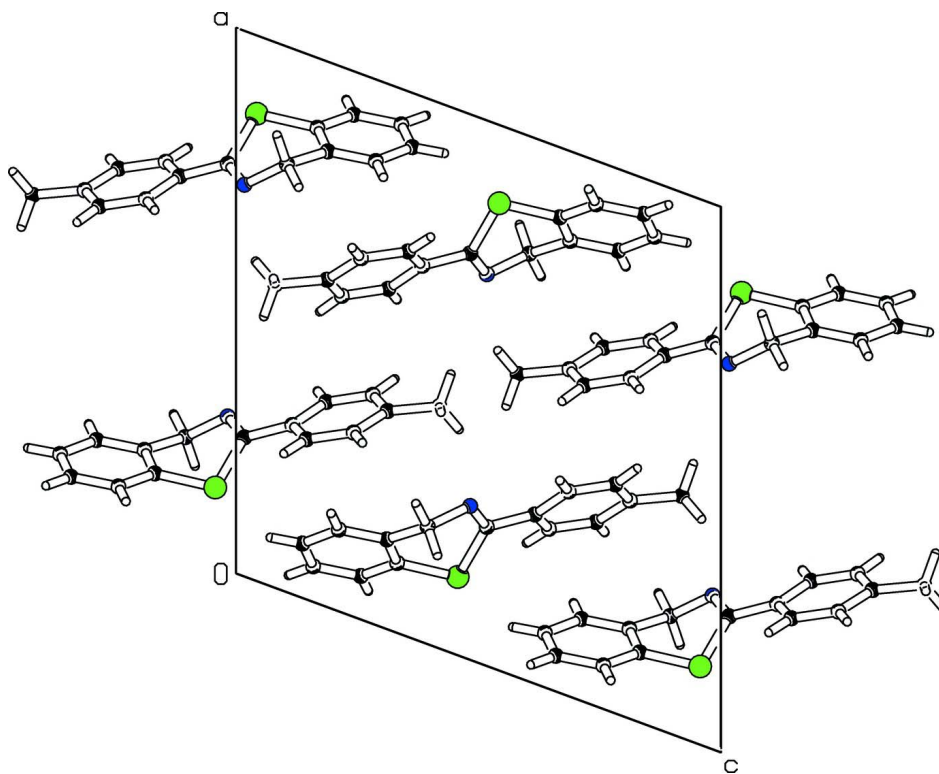
4-Methyl-*N*-[(phenylthio)methyl]benzamide was heated with POCl₃ (10 ml) on an oil bath for 1 h. The reaction mixture was cooled by treated with ice, neutralized with Na₂CO₃, and extracted with dichloromethane. The combined extracts were dried (Na₂SO₄) and solvent was evaporated off. The residue was recrystallized from hot ethanol to get crystals of the title compound.

S3. Refinement

All H atoms were positioned geometrically and allowed to ride on their parent atom, with C—H = 0.93–0.97 Å, and with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$.

**Figure 1**

The molecular structure of the title compound with 50% probability displacement ellipsoids. H atoms are drawn as spheres of arbitrary radii.

**Figure 2**

A packing diagram of the title compound viewed along the *b* axis.

2-(4-Methylphenyl)-4H-1,3-benzothiazine

Crystal data

$C_{15}H_{13}NS$

$M_r = 239.33$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 15.1241(9)\ \text{\AA}$

$b = 6.0111(4)\ \text{\AA}$

$c = 14.3212(9)\ \text{\AA}$

$\beta = 110.268(2)^\circ$

$V = 1221.36(13)\ \text{\AA}^3$

$Z = 4$

$F(000) = 504$
 $D_x = 1.302 \text{ Mg m}^{-3}$
 Cu $K\alpha$ radiation, $\lambda = 1.54178 \text{ \AA}$
 Cell parameters from 1988 reflections
 $\theta = 3.1\text{--}64.6^\circ$

$\mu = 2.13 \text{ mm}^{-1}$
 $T = 293 \text{ K}$
 Block, light yellow
 $0.30 \times 0.25 \times 0.20 \text{ mm}$

Data collection

Bruker X8 Proteum
 diffractometer
 Radiation source: Bruker MicroStar microfocus
 rotating anode
 Helios multilayer optics monochromator
 Detector resolution: $10.7 \text{ pixels mm}^{-1}$
 φ and ω scans
 10379 measured reflections

1988 independent reflections
 1890 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.041$
 $\theta_{\text{max}} = 64.6^\circ$, $\theta_{\text{min}} = 3.1^\circ$
 $h = -17 \rightarrow 16$
 $k = -3 \rightarrow 6$
 $l = -16 \rightarrow 16$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.036$
 $wR(F^2) = 0.102$
 $S = 1.05$
 1988 reflections
 156 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods
 Secondary atom site location: difference Fourier
 map

Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0608P)^2 + 0.3174P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.18 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$
 Extinction correction: *SHELXL97* (Sheldrick,
 2008), $\text{FC}^* = \text{KFC}[1 + 0.001\text{XFC}^2\Lambda^3/\text{SIN}(2\Theta)]^{-1/4}$
 Extinction coefficient: 0.0063 (7)

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.85647 (3)	0.07017 (7)	0.54278 (3)	0.0440 (2)
N5	0.71804 (10)	-0.2361 (2)	0.51727 (10)	0.0482 (5)
C2	0.87171 (11)	0.0058 (3)	0.66754 (11)	0.0383 (5)
C3	0.83561 (11)	-0.1929 (3)	0.68916 (12)	0.0422 (5)
C4	0.78453 (13)	-0.3459 (3)	0.60484 (13)	0.0519 (6)
C6	0.74506 (11)	-0.0644 (2)	0.48288 (12)	0.0394 (5)
C7	0.92024 (12)	0.1532 (3)	0.74273 (12)	0.0461 (5)
C8	0.93470 (13)	0.0983 (3)	0.84074 (13)	0.0554 (6)
C9	0.90031 (14)	-0.0989 (4)	0.86320 (14)	0.0595 (7)
C10	0.85010 (12)	-0.2426 (3)	0.78801 (13)	0.0534 (6)

C11	0.68996 (11)	0.0295 (3)	0.38404 (12)	0.0399 (5)
C12	0.70649 (13)	0.2415 (3)	0.35473 (13)	0.0494 (6)
C13	0.66238 (14)	0.3119 (3)	0.25780 (14)	0.0548 (6)
C14	0.59980 (13)	0.1777 (3)	0.18745 (13)	0.0520 (6)
C15	0.58033 (14)	-0.0299 (3)	0.21806 (14)	0.0577 (6)
C16	0.62461 (13)	-0.1037 (3)	0.31386 (13)	0.0501 (6)
C17	0.55616 (18)	0.2545 (4)	0.08089 (15)	0.0766 (8)
H4A	0.83080	-0.42520	0.58460	0.0620*
H4B	0.75040	-0.45520	0.62890	0.0620*
H7	0.94270	0.28720	0.72730	0.0550*
H8	0.96770	0.19490	0.89160	0.0660*
H9	0.91090	-0.13580	0.92930	0.0710*
H10	0.82580	-0.37380	0.80390	0.0640*
H12	0.74760	0.33660	0.40080	0.0590*
H13	0.67520	0.45350	0.23960	0.0660*
H15	0.53630	-0.12110	0.17280	0.0690*
H16	0.61070	-0.24440	0.33200	0.0600*
H17A	0.56570	0.41170	0.07730	0.1150*
H17B	0.48980	0.22310	0.05730	0.1150*
H17C	0.58520	0.17760	0.04030	0.1150*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0437 (3)	0.0489 (3)	0.0373 (3)	-0.0101 (2)	0.0113 (2)	0.0032 (2)
N5	0.0492 (8)	0.0407 (8)	0.0485 (8)	-0.0079 (6)	0.0092 (6)	0.0025 (6)
C2	0.0358 (8)	0.0401 (8)	0.0380 (8)	0.0031 (7)	0.0117 (6)	0.0024 (7)
C3	0.0395 (8)	0.0430 (9)	0.0442 (9)	0.0027 (7)	0.0145 (7)	0.0064 (7)
C4	0.0598 (11)	0.0388 (10)	0.0518 (10)	-0.0056 (8)	0.0127 (8)	0.0080 (8)
C6	0.0402 (9)	0.0373 (9)	0.0396 (8)	-0.0010 (6)	0.0126 (7)	-0.0030 (6)
C7	0.0434 (9)	0.0466 (10)	0.0428 (9)	0.0001 (7)	0.0081 (7)	-0.0009 (7)
C8	0.0515 (11)	0.0685 (12)	0.0395 (9)	0.0058 (9)	0.0074 (8)	-0.0055 (8)
C9	0.0585 (11)	0.0820 (14)	0.0385 (9)	0.0096 (10)	0.0174 (9)	0.0103 (9)
C10	0.0514 (10)	0.0612 (12)	0.0497 (10)	0.0034 (9)	0.0203 (8)	0.0177 (9)
C11	0.0392 (8)	0.0393 (9)	0.0405 (9)	0.0006 (7)	0.0128 (7)	-0.0015 (7)
C12	0.0546 (10)	0.0414 (10)	0.0449 (9)	-0.0046 (8)	0.0078 (8)	-0.0011 (7)
C13	0.0610 (11)	0.0452 (10)	0.0533 (10)	0.0016 (8)	0.0135 (9)	0.0084 (8)
C14	0.0480 (10)	0.0587 (11)	0.0440 (9)	0.0060 (8)	0.0091 (8)	0.0053 (8)
C15	0.0520 (11)	0.0627 (12)	0.0461 (10)	-0.0094 (9)	0.0014 (8)	-0.0031 (9)
C16	0.0487 (10)	0.0453 (10)	0.0500 (10)	-0.0075 (8)	0.0092 (8)	0.0003 (8)
C17	0.0780 (15)	0.0877 (16)	0.0494 (11)	0.0016 (12)	0.0035 (10)	0.0156 (11)

Geometric parameters (Å, °)

S1—C2	1.7635 (16)	C14—C15	1.388 (3)
S1—C6	1.7967 (17)	C14—C17	1.510 (3)
N5—C4	1.465 (2)	C15—C16	1.375 (3)
N5—C6	1.2705 (19)	C4—H4A	0.9700

C2—C3	1.392 (3)	C4—H4B	0.9700
C2—C7	1.391 (2)	C7—H7	0.9300
C3—C4	1.502 (2)	C8—H8	0.9300
C3—C10	1.388 (2)	C9—H9	0.9300
C6—C11	1.483 (2)	C10—H10	0.9300
C7—C8	1.384 (2)	C12—H12	0.9300
C8—C9	1.377 (3)	C13—H13	0.9300
C9—C10	1.384 (3)	C15—H15	0.9300
C11—C12	1.391 (3)	C16—H16	0.9300
C11—C16	1.393 (3)	C17—H17A	0.9600
C12—C13	1.382 (3)	C17—H17B	0.9600
C13—C14	1.379 (3)	C17—H17C	0.9600
C2—S1—C6	99.02 (8)	N5—C4—H4B	109.00
C4—N5—C6	118.70 (15)	C3—C4—H4A	109.00
S1—C2—C3	119.34 (12)	C3—C4—H4B	109.00
S1—C2—C7	119.55 (13)	H4A—C4—H4B	108.00
C3—C2—C7	121.11 (15)	C2—C7—H7	120.00
C2—C3—C4	118.61 (14)	C8—C7—H7	120.00
C2—C3—C10	118.43 (15)	C7—C8—H8	120.00
C4—C3—C10	122.95 (16)	C9—C8—H8	120.00
N5—C4—C3	114.95 (14)	C8—C9—H9	120.00
S1—C6—N5	123.71 (13)	C10—C9—H9	120.00
S1—C6—C11	114.10 (11)	C3—C10—H10	120.00
N5—C6—C11	121.96 (15)	C9—C10—H10	120.00
C2—C7—C8	119.27 (17)	C11—C12—H12	120.00
C7—C8—C9	120.18 (17)	C13—C12—H12	120.00
C8—C9—C10	120.36 (17)	C12—C13—H13	119.00
C3—C10—C9	120.62 (17)	C14—C13—H13	119.00
C6—C11—C12	122.43 (15)	C14—C15—H15	119.00
C6—C11—C16	119.58 (15)	C16—C15—H15	119.00
C12—C11—C16	117.75 (16)	C11—C16—H16	120.00
C11—C12—C13	120.59 (17)	C15—C16—H16	120.00
C12—C13—C14	121.68 (17)	C14—C17—H17A	110.00
C13—C14—C15	117.54 (17)	C14—C17—H17B	109.00
C13—C14—C17	120.58 (18)	C14—C17—H17C	109.00
C15—C14—C17	121.87 (18)	H17A—C17—H17B	109.00
C14—C15—C16	121.47 (18)	H17A—C17—H17C	109.00
C11—C16—C15	120.87 (17)	H17B—C17—H17C	110.00
N5—C4—H4A	109.00		
C6—S1—C2—C3	31.75 (16)	S1—C6—C11—C12	-20.0 (2)
C6—S1—C2—C7	-148.81 (15)	S1—C6—C11—C16	154.33 (14)
C2—S1—C6—N5	-30.02 (16)	N5—C6—C11—C12	165.34 (17)
C2—S1—C6—C11	155.37 (12)	N5—C6—C11—C16	-20.4 (3)
C6—N5—C4—C3	47.9 (2)	C2—C7—C8—C9	-0.8 (3)
C4—N5—C6—S1	-6.7 (2)	C7—C8—C9—C10	-0.8 (3)
C4—N5—C6—C11	167.49 (15)	C8—C9—C10—C3	1.6 (3)

S1—C2—C3—C4	-0.1 (2)	C6—C11—C12—C13	171.49 (18)
S1—C2—C3—C10	178.66 (14)	C16—C11—C12—C13	-2.9 (3)
C7—C2—C3—C4	-179.54 (17)	C6—C11—C16—C15	-172.56 (18)
C3—C2—C7—C8	1.6 (3)	C12—C11—C16—C15	2.0 (3)
C7—C2—C3—C10	-0.8 (3)	C11—C12—C13—C14	1.0 (3)
S1—C2—C7—C8	-177.85 (15)	C12—C13—C14—C15	1.8 (3)
C2—C3—C4—N5	-43.9 (2)	C12—C13—C14—C17	-177.1 (2)
C10—C3—C4—N5	137.37 (18)	C13—C14—C15—C16	-2.7 (3)
C2—C3—C10—C9	-0.8 (3)	C17—C14—C15—C16	176.1 (2)
C4—C3—C10—C9	177.91 (19)	C14—C15—C16—C11	0.8 (3)

Hydrogen-bond geometry (Å, °)

Cg is the centroid of the C3/C2/C7—C10 ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C7—H7... <i>Cg</i> ⁱ	0.93	2.75	3.485 (2)	136

Symmetry code: (i) $-x, y+1/2, -z-1/2$.