



Crystal structure of *N*-[2-(benzo[*d*][1,3]-dioxol-5-yl)ethyl]-4-methylbenzenesulfonamide

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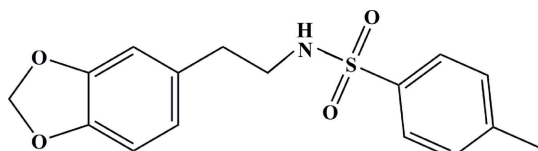
In the title compound, C₁₆H₁₇NO₄S, the heterocyclic ring is almost planar (r.m.s. deviation = 0.007 Å) and the dihedral angle between the benzene rings is 28.18 (10)°. The N–C–C–C torsion angle for the central chain is 62.4 (3)°; overall, the molecule has a Z-shape. In the crystal, inversion dimers linked by pairs of N–H···O hydrogen bonds generate R₂²(8) loops.

Keywords: crystal structure; methylbenzenesulfonamide derivatives; hydrogen bonding.

CCDC reference: 1401539

1. Related literature

For background to methylbenzenesulfonamide derivatives, see: Barn *et al.* (2001); Ghorai *et al.* (2010).



2. Experimental

2.1. Crystal data

C₁₆H₁₇NO₄S

M_r = 319.37

Monoclinic, *P*2₁/*n*
a = 12.3265 (2) Å
b = 9.96026 (16) Å
c = 12.7021 (3) Å
 β = 100.5980 (18)°
V = 1532.90 (5) Å³

Z = 4
 Mo *K*α radiation
 μ = 0.23 mm⁻¹
T = 293 K
 0.40 × 0.20 × 0.12 mm

2.2. Data collection

Agilent SuperNova (single source at offset), Eos diffractometer
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2011)
T_{min} = 0.868, *T_{max}* = 1.000

12219 measured reflections
 3136 independent reflections
 2587 reflections with *I* > 2σ(*I*)
R_{int} = 0.029

2.3. Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.043
 $wR(F^2)$ = 0.115
S = 1.05
 3136 reflections
 204 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max}$ = 0.19 e Å⁻³
 $\Delta\rho_{\min}$ = -0.39 e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
N1–H1···O4 ⁱ	0.84 (2)	2.19 (2)	3.026 (2)	172 (2)

Symmetry code: (i) $-x, -y, -z + 1$.

Data collection: *CrysAlis PRO* (Agilent, 2011); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *OLEX* (Dolomanov *et al.*, 2009); software used to prepare material for publication: *OLEX*.

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Supporting information for this paper is available from the IUCr electronic archives (Reference: HB7436).

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supporting information

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Crystal structure of *N*-[2-(benzo[*d*][1,3]dioxol-5-yl)ethyl]-4-methylbenzenesulfonamide

Ke-Bin Huang and Gui-Jie Zhang

S1. Experimental

A solution of sulfonylchloride (10 mmol) in dichloromethane (15 mL) was slowly added to a cooled solution of methylenedioxyphenethylamine (15 mmol) in dichloromethane (10 ml) and triethylamine (15 mmol). Yellow blocks of the title compound were obtained by slow evaporation of a solution in methanol.

S2. Refinement

Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

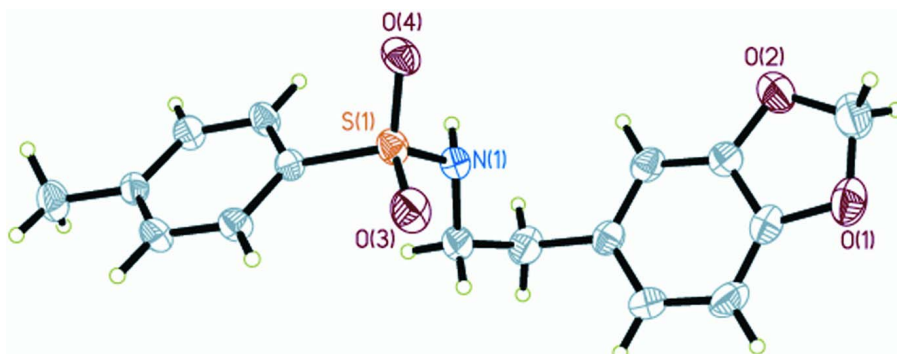


Figure 1

The molecular structure of title compound, with atom labelling. Displacement ellipsoids are drawn at 30% probability level.

N-[2-(Benzo[*d*][1,3]dioxol-5-yl)ethyl]-4-methylbenzenesulfonamide

Crystal data

$C_{16}H_{17}NO_4S$

$M_r = 319.37$

Monoclinic, $P2_1/n$

$a = 12.3265$ (2) Å

$b = 9.96026$ (16) Å

$c = 12.7021$ (3) Å

$\beta = 100.5980$ (18)°

$V = 1532.90$ (5) Å³

$Z = 4$

$F(000) = 672$

$D_x = 1.384$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.7107$ Å

Cell parameters from 5283 reflections

$\theta = 2.9$ – 28.7 °

$\mu = 0.23 \text{ mm}^{-1}$
 $T = 293 \text{ K}$

Block, yellow
 $0.40 \times 0.20 \times 0.12 \text{ mm}$

Data collection

Agilent SuperNova (single source at offset), Eos diffractometer
 Radiation source: SuperNova (Mo) X-ray Source
 Mirror monochromator
 Detector resolution: 16.1623 pixels mm^{-1}
 ω scans
 Absorption correction: multi-scan (CrysAlis PRO; Agilent, 2011)

$T_{\min} = 0.868$, $T_{\max} = 1.000$
 12219 measured reflections
 3136 independent reflections
 2587 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.029$
 $\theta_{\max} = 26.4^\circ$, $\theta_{\min} = 2.9^\circ$
 $h = -15 \rightarrow 15$
 $k = -12 \rightarrow 12$
 $l = -15 \rightarrow 15$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.115$
 $S = 1.05$
 3136 reflections
 204 parameters
 0 restraints
 Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
 Hydrogen site location: inferred from neighbouring sites
 H atoms treated by a mixture of independent and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0499P)^2 + 0.4898P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.19 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.39 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.01040 (4)	0.17100 (5)	0.62290 (4)	0.04836 (17)
O1	0.10241 (14)	-0.39697 (19)	0.99233 (14)	0.0834 (5)
O2	0.07337 (17)	-0.43875 (18)	0.81134 (15)	0.0899 (6)
O3	0.04945 (13)	0.23231 (16)	0.72422 (12)	0.0689 (4)
O4	0.08943 (10)	0.12122 (14)	0.56207 (12)	0.0602 (4)
N1	-0.06387 (14)	0.04327 (17)	0.64128 (15)	0.0521 (4)
H1	-0.0777 (17)	-0.003 (2)	0.5849 (18)	0.058 (7)*
C1	0.1333 (2)	-0.4786 (3)	0.9109 (2)	0.0844 (8)
H1A	0.2117	-0.4692	0.9115	0.101*
H1B	0.1181	-0.5721	0.9239	0.101*
C2	0.02598 (17)	-0.3080 (2)	0.93953 (17)	0.0564 (5)
C3	-0.0287 (2)	-0.2082 (3)	0.98013 (18)	0.0725 (7)

H3	-0.0176	-0.1910	1.0533	0.087*
C4	-0.10220 (19)	-0.1330 (2)	0.90745 (18)	0.0657 (6)
H4	-0.1404	-0.0633	0.9331	0.079*
C5	-0.12075 (16)	-0.1574 (2)	0.79891 (16)	0.0515 (5)
C6	-0.06264 (17)	-0.2603 (2)	0.76031 (16)	0.0556 (5)
H6	-0.0725	-0.2786	0.6874	0.067*
C7	0.00883 (17)	-0.3334 (2)	0.83196 (17)	0.0538 (5)
C8	-0.20189 (18)	-0.0740 (2)	0.7228 (2)	0.0676 (6)
H8A	-0.2671	-0.0597	0.7541	0.081*
H8B	-0.2244	-0.1242	0.6569	0.081*
C9	-0.15798 (18)	0.0611 (2)	0.6958 (2)	0.0654 (6)
H9A	-0.2159	0.1108	0.6499	0.078*
H9B	-0.1351	0.1124	0.7610	0.078*
C10	-0.07539 (13)	0.28694 (17)	0.54277 (14)	0.0397 (4)
C11	-0.08585 (15)	0.41631 (18)	0.57810 (15)	0.0462 (4)
H11	-0.0486	0.4434	0.6451	0.055*
C12	-0.15261 (15)	0.50503 (18)	0.51218 (16)	0.0491 (4)
H12	-0.1592	0.5927	0.5354	0.059*
C13	-0.20962 (14)	0.46749 (17)	0.41312 (15)	0.0437 (4)
C14	-0.28241 (19)	0.5655 (2)	0.34252 (19)	0.0665 (6)
H14A	-0.2651	0.5636	0.2719	0.100*
H14B	-0.2702	0.6543	0.3717	0.100*
H14C	-0.3583	0.5411	0.3389	0.100*
C15	-0.19811 (16)	0.33719 (18)	0.37970 (15)	0.0505 (5)
H15	-0.2365	0.3099	0.3132	0.061*
C16	-0.13115 (16)	0.24717 (18)	0.44277 (15)	0.0490 (5)
H16	-0.1233	0.1602	0.4186	0.059*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0424 (3)	0.0496 (3)	0.0495 (3)	0.00838 (19)	-0.0009 (2)	0.0022 (2)
O1	0.0771 (11)	0.0969 (13)	0.0694 (11)	0.0077 (10)	-0.0045 (9)	0.0205 (10)
O2	0.1130 (14)	0.0765 (11)	0.0788 (12)	0.0330 (10)	0.0134 (11)	-0.0013 (10)
O3	0.0695 (9)	0.0724 (10)	0.0540 (9)	0.0114 (8)	-0.0171 (7)	-0.0045 (8)
O4	0.0426 (7)	0.0613 (8)	0.0773 (10)	0.0105 (6)	0.0124 (7)	0.0019 (7)
N1	0.0576 (10)	0.0477 (9)	0.0524 (10)	0.0109 (7)	0.0135 (8)	0.0075 (8)
C1	0.0721 (16)	0.0828 (18)	0.100 (2)	0.0123 (14)	0.0206 (15)	0.0238 (16)
C2	0.0544 (11)	0.0642 (13)	0.0483 (11)	-0.0089 (10)	0.0031 (9)	0.0088 (10)
C3	0.0907 (18)	0.0882 (17)	0.0389 (11)	-0.0001 (14)	0.0124 (11)	-0.0021 (12)
C4	0.0749 (14)	0.0733 (14)	0.0537 (13)	0.0090 (12)	0.0244 (11)	0.0001 (11)
C5	0.0477 (10)	0.0564 (11)	0.0514 (11)	-0.0090 (8)	0.0115 (9)	0.0062 (9)
C6	0.0674 (13)	0.0554 (12)	0.0416 (11)	-0.0091 (10)	0.0037 (9)	-0.0052 (9)
C7	0.0590 (12)	0.0484 (11)	0.0537 (12)	-0.0069 (9)	0.0100 (9)	-0.0037 (9)
C8	0.0518 (12)	0.0843 (16)	0.0671 (14)	0.0040 (11)	0.0118 (10)	0.0182 (12)
C9	0.0657 (13)	0.0674 (14)	0.0676 (14)	0.0216 (11)	0.0244 (11)	0.0168 (11)
C10	0.0367 (8)	0.0398 (9)	0.0414 (9)	0.0008 (7)	0.0039 (7)	0.0008 (7)
C11	0.0465 (10)	0.0457 (10)	0.0441 (10)	-0.0004 (8)	0.0020 (8)	-0.0077 (8)

C12	0.0532 (10)	0.0364 (9)	0.0562 (12)	0.0027 (8)	0.0059 (9)	-0.0071 (8)
C13	0.0414 (9)	0.0421 (9)	0.0475 (10)	0.0016 (7)	0.0079 (8)	0.0052 (8)
C14	0.0713 (14)	0.0559 (12)	0.0669 (14)	0.0134 (10)	-0.0019 (11)	0.0107 (11)
C15	0.0574 (11)	0.0472 (10)	0.0417 (10)	0.0022 (8)	-0.0043 (8)	-0.0042 (8)
C16	0.0574 (11)	0.0382 (9)	0.0476 (11)	0.0040 (8)	-0.0005 (9)	-0.0057 (8)

Geometric parameters (Å, °)

S1—O3	1.4259 (15)	C6—C7	1.356 (3)
S1—O4	1.4383 (14)	C8—H8A	0.9700
S1—N1	1.6092 (18)	C8—H8B	0.9700
S1—C10	1.7584 (17)	C8—C9	1.513 (3)
O1—C1	1.421 (3)	C9—H9A	0.9700
O1—C2	1.375 (3)	C9—H9B	0.9700
O2—C1	1.400 (3)	C10—C11	1.378 (2)
O2—C7	1.371 (3)	C10—C16	1.386 (2)
N1—H1	0.84 (2)	C11—H11	0.9300
N1—C9	1.467 (3)	C11—C12	1.380 (3)
C1—H1A	0.9700	C12—H12	0.9300
C1—H1B	0.9700	C12—C13	1.375 (3)
C2—C3	1.355 (3)	C13—C14	1.505 (3)
C2—C7	1.367 (3)	C13—C15	1.381 (3)
C3—H3	0.9300	C14—H14A	0.9600
C3—C4	1.388 (3)	C14—H14B	0.9600
C4—H4	0.9300	C14—H14C	0.9600
C4—C5	1.377 (3)	C15—H15	0.9300
C5—C6	1.390 (3)	C15—C16	1.372 (3)
C5—C8	1.507 (3)	C16—H16	0.9300
C6—H6	0.9300		
O3—S1—O4	118.88 (9)	C5—C8—H8B	108.7
O3—S1—N1	108.32 (10)	C5—C8—C9	114.39 (18)
O3—S1—C10	107.96 (9)	H8A—C8—H8B	107.6
O4—S1—N1	105.37 (9)	C9—C8—H8A	108.7
O4—S1—C10	108.07 (9)	C9—C8—H8B	108.7
N1—S1—C10	107.79 (8)	N1—C9—C8	110.25 (17)
C2—O1—C1	105.37 (18)	N1—C9—H9A	109.6
C7—O2—C1	105.8 (2)	N1—C9—H9B	109.6
S1—N1—H1	109.7 (15)	C8—C9—H9A	109.6
C9—N1—S1	119.55 (15)	C8—C9—H9B	109.6
C9—N1—H1	114.4 (15)	H9A—C9—H9B	108.1
O1—C1—H1A	109.8	C11—C10—S1	120.52 (13)
O1—C1—H1B	109.8	C11—C10—C16	120.34 (16)
O2—C1—O1	109.2 (2)	C16—C10—S1	119.13 (13)
O2—C1—H1A	109.8	C10—C11—H11	120.6
O2—C1—H1B	109.8	C10—C11—C12	118.82 (17)
H1A—C1—H1B	108.3	C12—C11—H11	120.6
C3—C2—O1	129.2 (2)	C11—C12—H12	119.1

C3—C2—C7	121.3 (2)	C13—C12—C11	121.88 (17)
C7—C2—O1	109.5 (2)	C13—C12—H12	119.1
C2—C3—H3	121.5	C12—C13—C14	121.13 (17)
C2—C3—C4	116.9 (2)	C12—C13—C15	118.21 (16)
C4—C3—H3	121.5	C15—C13—C14	120.67 (18)
C3—C4—H4	118.7	C13—C14—H14A	109.5
C5—C4—C3	122.5 (2)	C13—C14—H14B	109.5
C5—C4—H4	118.7	C13—C14—H14C	109.5
C4—C5—C6	118.8 (2)	H14A—C14—H14B	109.5
C4—C5—C8	120.9 (2)	H14A—C14—H14C	109.5
C6—C5—C8	120.3 (2)	H14B—C14—H14C	109.5
C5—C6—H6	120.9	C13—C15—H15	119.4
C7—C6—C5	118.18 (19)	C16—C15—C13	121.29 (17)
C7—C6—H6	120.9	C16—C15—H15	119.4
C2—C7—O2	110.14 (19)	C10—C16—H16	120.3
C6—C7—O2	127.6 (2)	C15—C16—C10	119.46 (16)
C6—C7—C2	122.2 (2)	C15—C16—H16	120.3
C5—C8—H8A	108.7		
S1—N1—C9—C8	-169.30 (16)	C3—C2—C7—C6	0.5 (3)
S1—C10—C11—C12	178.98 (14)	C3—C4—C5—C6	-0.9 (3)
S1—C10—C16—C15	-179.88 (15)	C3—C4—C5—C8	179.5 (2)
O1—C2—C3—C4	-179.9 (2)	C4—C5—C6—C7	0.9 (3)
O1—C2—C7—O2	-0.2 (2)	C4—C5—C8—C9	79.5 (3)
O1—C2—C7—C6	-179.87 (19)	C5—C6—C7—O2	179.6 (2)
O3—S1—N1—C9	55.48 (18)	C5—C6—C7—C2	-0.8 (3)
O3—S1—C10—C11	6.21 (18)	C5—C8—C9—N1	62.4 (3)
O3—S1—C10—C16	-174.78 (15)	C6—C5—C8—C9	-100.1 (2)
O4—S1—N1—C9	-176.30 (16)	C7—O2—C1—O1	1.0 (3)
O4—S1—C10—C11	-123.57 (16)	C7—C2—C3—C4	-0.4 (3)
O4—S1—C10—C16	55.44 (17)	C8—C5—C6—C7	-179.48 (18)
N1—S1—C10—C11	123.02 (16)	C10—S1—N1—C9	-61.09 (18)
N1—S1—C10—C16	-57.97 (17)	C10—C11—C12—C13	0.7 (3)
C1—O1—C2—C3	-179.6 (2)	C11—C10—C16—C15	-0.9 (3)
C1—O1—C2—C7	0.8 (2)	C11—C12—C13—C14	179.75 (19)
C1—O2—C7—C2	-0.5 (3)	C11—C12—C13—C15	-0.5 (3)
C1—O2—C7—C6	179.1 (2)	C12—C13—C15—C16	-0.5 (3)
C2—O1—C1—O2	-1.1 (3)	C13—C15—C16—C10	1.1 (3)
C2—C3—C4—C5	0.6 (4)	C14—C13—C15—C16	179.33 (19)
C3—C2—C7—O2	-179.8 (2)	C16—C10—C11—C12	0.0 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O4 ⁱ	0.84 (2)	2.19 (2)	3.026 (2)	172 (2)

Symmetry code: (i) $-x, -y, -z+1$.