

Structural characterization of three norfloxacin salts with N-heterocyclic mono- and dicarboxylic acids

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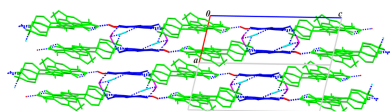
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Cocrystallization of norfloxacin (NFX) with three heterocyclic cofomers, namely, pyridine-3,5-dicarboxylic acid, pyridazine-3-carboxylic acid and pyrimidine-5-carboxylic acid, yielded three molecular salts: norfloxacinium [4-(3-carboxy-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinolin-7-yl)piperazin-1-ium] 5-carboxypyridine-3-carboxylate trihydrate, $2C_{16}H_{19}FN_3O_3^+ \cdot 2C_7H_4NO_4^- \cdot 3H_2O$, (**1**), norfloxacinium pyridazine-3-carboxylate, $C_{16}H_{19}FN_3O_3^+ \cdot C_5H_3N_2O_2^-$, (**2**), and norfloxacinium pyrimidine-5-carboxylate monohydrate, $C_{16}H_{19}FN_3O_3^+ \cdot C_5H_3N_2O_2^- \cdot H_2O$, (**3**). In all structures, the quinolone skeleton of the norfloxacin cation is essentially planar, while the piperazine ring adopts a chair conformation. The crystal packing is governed primarily by $N-H \cdots O$ and $N-H \cdots N$ hydrogen bonds, which generate distinct mono-periodic and di-periodic motifs depending on the cofomer. These assemblies are further linked by weak $C-H \cdots O$ and $C-H \cdots N$ interactions into tri-periodic supramolecular networks. The crystal structures are additionally stabilized by aromatic $\pi-\pi$ interactions, which differ in their stacking arrangements among the three salts. Hirshfeld surface analysis of the norfloxacin cations shows that $H \cdots H$ and $O \cdots H/H \cdots O$ contacts dominate, whereas variations in the contributions from $C \cdots C$, $C \cdots H/H \cdots C$ and $N \cdots H/H \cdots N$ contacts reflect differences in the supramolecular packing.

1. Chemical context

Pharmaceutical multicomponent crystal forms, including cocrystals and molecular salts, have become an important strategy in crystal engineering for improving the physicochemical properties of active pharmaceutical ingredients (APIs), such as solubility, dissolution behaviour, stability and bioavailability, without altering their pharmacological activity. The design of such crystalline materials relies on the rational manipulation of intermolecular interactions and supramolecular synthons to produce solids with predictable structural features and tailored properties (Desiraju, 1989, 1995; Bolla *et al.*, 2022).

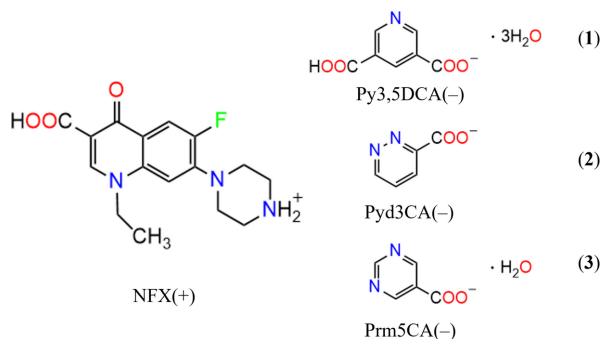
Norfloxacin, 1-ethyl-6-fluoro-1,4-dihydro-6-fluoro-4-oxo-7-(piperazin-1-yl)quinoline-3-carboxylic acid (NFX), a second-generation fluoroquinolone discovered in the late 1970s, is a suitable candidate for multicomponent crystal design because of its poor aqueous solubility and low permeability. Consequently, it is classified as a Biopharmaceutics Classification System (BCS) class IV drug (Gunnam & Nangia, 2023). Norfloxacin is active against Gram-positive and Gram-negative bacteria (Zeng *et al.*, 2024) and is used in the treatment of bacterial infections of the prostate and the urinary and biliary



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tracts (Holmes *et al.*, 1985; Tabeta *et al.*, 1987; Lee & Wong, 1988).



In this work, three new molecular salts of norfloxacin with N-heterocyclic mono- and dicarboxylic acids have been prepared and structurally characterized. The selected cofomers are pyridine-3,5-dicarboxylic acid (Py3,5DCA) pyridazine-3-carboxylic acid (Pyd3CA) and pyrimidine-5-carboxylic acid (Prm5CA). The crystal structures of three NFX multi-

component forms (1)–(3) are reported and discussed with emphasis on the intermolecular interactions governing crystal packing.

2. Structural commentary

All three investigated multicomponent crystals of norfloxacin are molecular salts. The formal positive charge is located on the nitrogen atom (N3/N6) of the piperazine ring, while the negative charge is delocalized over the carboxylate group of cofomer monoanions.

All three molecular salts crystallize in the triclinic space group $P\bar{1}$. The asymmetric units (Fig. 1) of $2\text{NFX}^+ \cdot 2\text{Py3,5DCA}^- \cdot 3\text{H}_2\text{O}$ (1) and $\text{NFX}^+ \cdot \text{Pyd3CA}^-$ (2) each comprise two norfloxacin cations and two cofomer anions; in addition, the asymmetric unit of (1) contains three water molecules. In contrast, the asymmetric unit of $\text{NFX}^+ \cdot \text{Prm5CA}^- \cdot \text{H}_2\text{O}$ (3) consists of one norfloxacin cation, one pyrimidine-5-carboxylate anion and one water molecule.

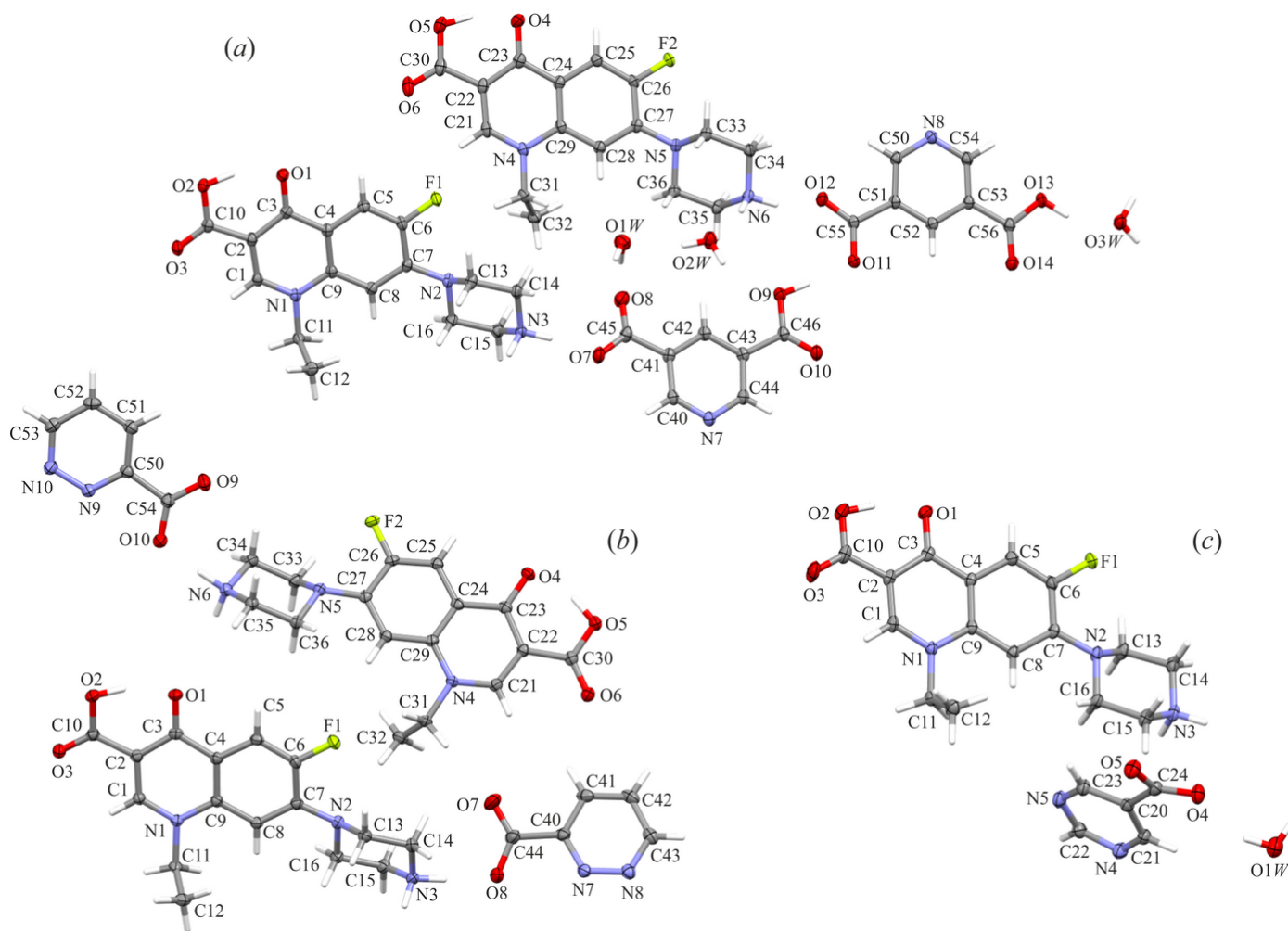


Figure 1
The molecular structures of (a) $2\text{NFX}^+ \cdot 2\text{Py3,5DCA}^- \cdot 3\text{H}_2\text{O}$ (1), (b) $\text{NFX}^+ \cdot \text{Pyd3CA}^-$ (2) and (c) $\text{NFX}^+ \cdot \text{Prm5CA}^- \cdot \text{H}_2\text{O}$ (3) with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.

Table 1

Geometric parameters describing the molecular conformation of norfloxacin in molecular salts **(1)**–**(3)**.

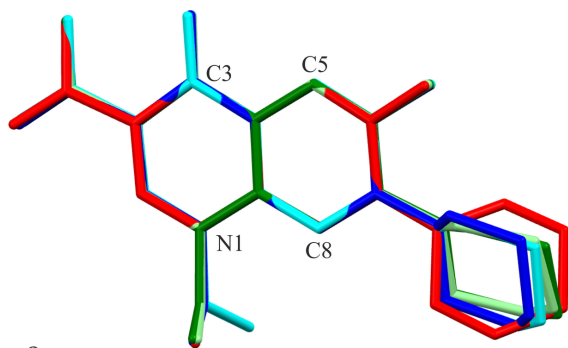
C1–N1–C11–C12 or C21–N4–C31–C32 torsion angle ($^{\circ}$) defining the orientation of the ethyl group to heterocyclic ring of the norfloxacin molecule; Cremer & Pople puckering parameters defining the chair conformation of the piperazine ring: Q ($\text{\AA}$) – the puckering amplitude; the angular parameters θ ($^{\circ}$) and φ ($^{\circ}$); Analysis of ring substituent in the piperazine ring: the angle ($^{\circ}$) between the C7–N2 or C27–N5 bond and the normal to Cremer & Pople plane; dihedral angle ($^{\circ}$) between the best planes of heterocyclic and benzene rings, forming the ten-membered quinolone system; dihedral angle ($^{\circ}$) between the best planes of the ten-membered quinolone system and the piperazine ring.

Structure	Torsion angle	Q	θ	φ	Angle C7–N2 bond/CP plane	Dihedral angle heterocyclic/benzene	Dihedral angle quinolone/piperazine
(1)	–99.62 (13) 105.88 (13)	0.5782 (14) 0.5727 (13)	180.00 (13) 178.13 (14)	159 (21) 96 (4)	79.72 (8) 83.11 (8)	1.67 (6) 2.97 (6)	40.55 (5) 37.45 (5)
(2)	–104.20 (11) –104.23 (11)	0.5705 (12) 0.5730 (12)	175.26 (11) 171.82 (11)	167.5 (15) 183.4 (9)	85.91 (7) 88.23 (7)	3.64 (5) 2.55 (5)	43.77 (5) 44.70 (4)
(3)	–100.30 (13)	0.5845 (14)	3.88 (14)	47 (2)	86.86 (8)	3.95 (6)	27.98 (5)

The quinolone skeleton of the NFX molecules remains essentially planar in all structures. The dihedral angle between the heterocyclic and benzene rings ranges from 1.67 (6) to 3.95 (6) $^{\circ}$, confirming the planarity of the fused bicyclic system (Table 1). The torsion angle C1–N1–C11–C12, ranging from –99.62 (13) to –104.23 (11) $^{\circ}$, indicates a nearly perpendicular orientation of the ethyl group with respect to the quinolone skeleton. The second independent NFX molecule in **(2)** adopts the opposite orientation, with the torsion angle of 105.88 (13) $^{\circ}$, showing the conformational flexibility of the ethyl group (Table 1).

The piperazine ring adopts a chair conformation, with the puckering amplitude (Q) lying within a narrow range of 0.5705 (12) – 0.5845 (14) $\text{\AA}$. The θ values are close to 180 $^{\circ}$ in structures **(1)** and **(2)**; however, in structure **(3)** the opposite chair conformation is observed with θ close to 0 $^{\circ}$ (Table 1). This difference is clearly illustrated by the superposition of the five independent NFX molecules in Fig. 2.

In all investigated crystal structures, the quinolone moiety occupies an equatorial position of the piperazine ring. The angle between the C7–N2 bond and the normal to the Cremer & Pople mean plane of the piperazine ring varies from 79.72 (8) to 88.23 (7) $^{\circ}$ (Cremer & Pople, 1975). Consequently, the dihedral angle between these molecular fragments ranges from 27.98 (5) to 44.70 (4) $^{\circ}$.

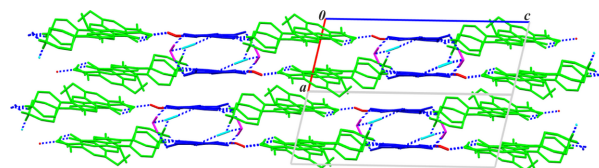
**Figure 2**

An overlay of five independent norfloxacin molecules present in the title molecular salts with the best fit for atoms N1, C3, C5 and C8. The color code is blue – molecule 1 in **(1)**; cyan – molecule 2 in **(1)**; green – molecule 1 in **(2)**, light green – molecule 2 in **(2)** and red – molecule in **(3)**. The root mean-square (r.m.s.) deviation between two independent NFX molecules is 0.484 $\text{\AA}$ for salt **(1)** and 0.142 $\text{\AA}$ for salt **(2)**, calculated for the best fit of 23 non-hydrogen atoms (Spek, 2020).

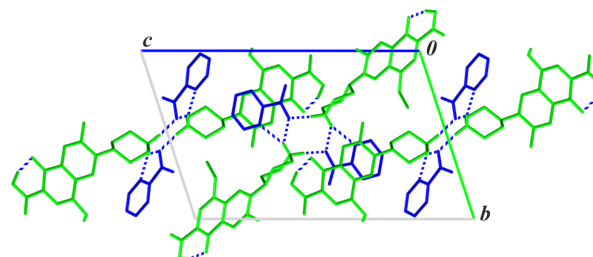
3. Supramolecular features

In the crystal structure of **(1)**, Py3,5DCA(–) and water molecules (O3W) form a hydrogen-bonded chain submotif with the sequence coformer-coformer-water. The remaining water molecules (O1W and O2W) connect two neighbouring chains, generating centrosymmetric channels. The first NFX molecule interacts with these chains through N–H $\cdots$ N_{coformer} and N–H $\cdots$ O_{coformer} hydrogen bonds (Table 2), whereas the second NFX molecule forms N–H $\cdots$ O hydrogen bonds with both the coformer and water molecules. Furthermore, the O3W–H2W3 $\cdots$ O5 hydrogen bond links adjacent columns into a supramolecular layer parallel to the (110) plane (Fig. 3).

In the crystal structure of **(2)**, each NFX(+) molecule interacts with Pyd3CA(–) coformer in the same manner. The terminal nitrogen atom of the piperazine ring forms three hydrogen bonds, including one N–H $\cdots$ O_{coformer} hydrogen bond and one bifurcated donor hydrogen bond involving oxygen and nitrogen atoms of the same coformer molecule as

**Figure 3**

Crystal packing of 2NFX(+)-2Py3,5DCA(–)-3H₂O **(1)** showing the di-periodic supramolecular layers. Norfloxacin cations are shown in green, Py3,5DCA anions in blue and water molecules in red, cyan and pink. (C)–H atoms not involved in hydrogen bonds have been omitted.

**Figure 4**

Crystal packing of NFX(+)-Pyd3CA(–) **(2)** showing the finite four-molecule supramolecular motifs. Norfloxacin cations are shown in green and Pyd3CA anions in blue. (C)–H atoms not involved in hydrogen bonds have been omitted.

Table 2

Hydrogen-bond geometry (Å, °) for (1).

<i>D</i> — <i>H</i> ... <i>A</i>	<i>D</i> — <i>H</i>	<i>H</i> ... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> — <i>H</i> ... <i>A</i>
O2—H2...O1	0.97 (2)	1.64 (2)	2.5554 (13)	156 (2)
O5—H5A...O4	0.98 (3)	1.54 (3)	2.4830 (14)	158 (2)
O9—H9...O11	0.95 (2)	1.69 (3)	2.6396 (13)	176 (2)
O13—H13...O3W	0.93 (2)	1.75 (2)	2.6342 (14)	157 (2)
N3—H3B...O7	0.94 (2)	1.66 (2)	2.5925 (14)	170 (2)
N3—H3A...N8 ⁱ	0.945 (19)	2.009 (19)	2.8825 (16)	153 (2)
N6—H6A...O2W	0.887 (18)	1.908 (19)	2.7646 (15)	162 (2)
N6—H6B...O12	0.94 (2)	1.81 (2)	2.6856 (14)	154 (2)
C5—H5...F1 ⁱⁱ	0.95	2.32	3.0193 (15)	130
C11—H11B...O6 ⁱⁱⁱ	0.99	2.35	3.2564 (16)	152
C12—H12B...F2 ⁱ	0.98	2.36	3.2576 (15)	151
C15—H15A...O11 ^{iv}	0.99	2.38	3.3290 (16)	161
C33—H33A...O2 ⁱⁱ	0.99	2.36	3.2282 (15)	145
O1W—H1W1...O8	0.95 (2)	1.76 (2)	2.7047 (2)	174 (2)
O1W—H2W1...O14 ^v	0.90 (2)	1.98 (2)	2.8834 (2)	173 (2)
O2W—H1W2...O1W	0.84 (2)	1.97 (2)	2.7974 (16)	167 (2)
O2W—H2W2...O11 ^v	0.88 (2)	1.95 (2)	2.8171 (14)	166 (2)
O3W—H1W3...N7 ^{vi}	0.87 (2)	1.96 (2)	2.7937 (15)	161 (2)
O3W—H2W3...O5 ^{vii}	0.83 (3)	2.41 (2)	3.0846 (14)	138 (2)

Symmetry codes: (i) $x+1, y-1, z$; (ii) $-x+1, -y, -z$; (iii) $-x+2, -y, -z$; (iv) $-x+1, -y+1, -z+1$; (v) $-x, -y+1, -z+1$; (vi) $x-1, y+1, z$; (vii) $x-1, y+1, z+1$.

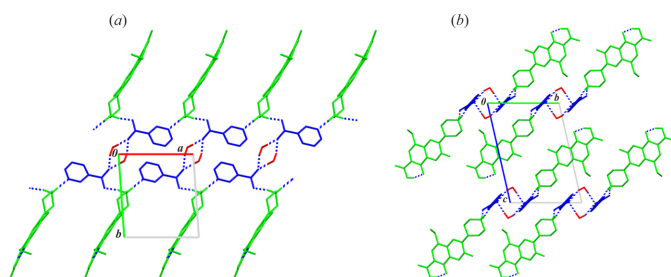
Table 3

Hydrogen-bond geometry (Å, °) for (2).

<i>D</i> — <i>H</i> ... <i>A</i>	<i>D</i> — <i>H</i>	<i>H</i> ... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> — <i>H</i> ... <i>A</i>
O2—H2...O1	0.946 (18)	1.610 (18)	2.5174 (10)	159 (2)
O5—H5A...O4	0.970 (19)	1.592 (18)	2.5208 (11)	159 (2)
N3—H3B...O8	0.987 (17)	1.685 (18)	2.6682 (12)	174 (2)
N6—H6B...O10	0.993 (17)	1.668 (17)	2.6591 (12)	176 (2)
N3—H3A...O8 ⁱ	0.921 (16)	2.254 (15)	2.8557 (13)	122 (1)
N3—H3A...N7 ⁱ	0.921 (16)	2.072 (15)	2.9552 (13)	160 (1)
N6—H6A...O10 ⁱⁱ	0.882 (15)	2.112 (14)	2.7721 (12)	131 (1)
N6—H6A...N9 ⁱⁱ	0.882 (15)	2.194 (15)	3.0060 (13)	153 (1)
C11—H11B...O9 ⁱⁱⁱ	0.99	2.46	3.2870 (13)	141
C11—H11A...F2 ⁱⁱⁱ	0.99	2.37	3.2083 (12)	142
C16—H16B...N8 ⁱ	0.99	2.50	3.4430 (14)	158
C36—H36B...N10 ⁱⁱ	0.99	2.49	3.4388 (14)	160
C41—H41...O6	0.95	2.45	3.1048 (13)	126
C51—H51...O3 ^{iv}	0.95	2.31	3.1186 (13)	143

Symmetry codes: (i) $-x, -y+1, -z+2$; (ii) $-x+2, -y+1, -z+1$; (iii) $x-1, y-1, z$; (iv) $x+1, y+1, z$.

acceptors (Table 3). These ionic pairs interact with each other forming finite supramolecular motifs composed of three fused hydrogen-bonded rings: $[R_1^2(5)R_4^2(8)R_1^2(5)]$ (Etter *et al.*, 1990; Bernstein *et al.*, 1995) (Fig. 4).

**Figure 5**

A part of the crystal structure of NFX(+).Prm5CA(-).H₂O (3) showing (a) a hydrogen-bonded chain motif and (b) the mono-periodic ribbons. Norfloxacin cations are shown in green, Prm5CA anions in blue and water molecules in red. (C)—H atoms not involved in hydrogen bonds have been omitted.

Table 4

Hydrogen-bond geometry (Å, °) for (3).

<i>D</i> — <i>H</i> ... <i>A</i>	<i>D</i> — <i>H</i>	<i>H</i> ... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> — <i>H</i> ... <i>A</i>
O2—H2...O1	0.96 (2)	1.60 (3)	2.5234 (13)	160 (2)
N3—H3A...O5	0.98 (2)	1.67 (2)	2.6378 (15)	170 (2)
N3—H3B...N5 ⁱ	0.94 (2)	1.93 (2)	2.8654 (16)	172 (2)
C1—H1...O3 ⁱⁱ	0.95	2.48	3.2305 (16)	136
C11—H11A...N4 ⁱⁱⁱ	0.99	2.58	3.3630 (17)	136
C13—H13A...F1	0.99	2.20	2.8752 (15)	124
O1W—H1W...O4	0.91 (2)	2.04 (2)	2.8863 (16)	155 (2)
O1W—H2W...O4 ^{iv}	0.94 (3)	2.01 (3)	2.9487 (17)	174 (2)

Symmetry codes: (i) $x+1, y, z$; (ii) $-x+1, -y+3, -z+1$; (iii) $-x+1, -y+1, -z+2$; (iv) $-x+2, -y, -z+2$.

Table 5Geometric parameters (Å, °) of aromatic π - π stacking interactions for norfloxacin salts (1)–(3).

$Cg1/Cg2$ are the centroids of the heterocyclic ring in NFX; $Cg3/Cg4$ are the centroids of the benzene ring in NFX; $Cg5/Cg6$ are the centroids of the aromatic ring in coformers (1)–(3). $CgI...CgJ$ – distance between ring centroids; α – dihedral angle between planes I and J ; CgI_{perp} and CgJ_{perp} (interplanar spacing) – perpendicular distance of CgI on ring J and CgJ on ring I , respectively; slippage – distance between CgI and perpendicular projection of CgJ on ring I .

Structure	Contact	$CgI...CgJ$	α	CgI_{perp}	CgJ_{perp}	Slippage
(1)	$Cg1...Cg1^{iii}$	3.4778 (7)	0.00 (6)	3.2329 (5)	3.2329 (5)	1.282
	$Cg1...Cg3^{iii}$	3.6013 (7)	1.67 (6)	3.2152 (5)	3.2500 (5)	1.552
	$Cg2...Cg2^{viii}$	3.8634 (7)	0.03 (6)	3.5678 (5)	3.5679 (5)	1.482
	$Cg6...Cg5^v$	3.7431 (7)	6.68 (6)	3.3319 (5)	3.4232 (5)	1.514
	$Cg6...Cg6^{ix}$	3.7718 (7)	0.00 (6)	3.5725 (5)	3.5724 (5)	1.210
	$Cg1...Cg6^{ii}$	3.6213 (6)	7.40 (5)	3.3032 (4)	3.4289 (5)	1.164
(2)	$Cg3...Cg6^{ii}$	3.6826 (7)	9.90 (5)	3.3350 (4)	3.4546 (5)	1.276
	$Cg4...Cg5^v$	3.7303 (7)	8.10 (5)	3.0801 (4)	3.3419 (5)	1.657
	$Cg5...Cg2^v$	3.5006 (6)	8.98 (5)	3.3215 (5)	3.4353 (4)	0.673
(3)	$Cg1...Cg1^v$	3.5480 (7)	0.00 (6)	3.4031 (5)	3.4031 (5)	1.003
	$Cg1...Cg3^v$	3.8093 (7)	3.95 (6)	3.3998 (5)	3.4876 (5)	1.532
	$Cg3...Cg3^{vi}$	3.7265 (7)	0.00 (6)	3.6818 (5)	3.6819 (5)	0.575

Symmetry codes: (1) (iii) $-x+2, -y, -z$; (viii) $-x+1, -y+1, -z$; (v) $-x, -y+1, -z+1$; (ix) $-x, -y+2, -z+1$; (2) (ii) $-x+2, -y+1, -z+1$; (v) $-x+1, -y+2, -z+2$; (3) (v) $-x+1, -y+2, -z+1$; (vi) $-x+2, -y+2, -z+1$.

In the crystal structure of (3), the NFX(+) molecules interact with Prm5CA(−) molecules through N—H...N_{coformer} and N—H...O_{coformer} hydrogen bonds, generating chains (Table 4, Fig. 5a). Centrosymmetrically related water molecules link two adjacent chains into a mono-periodic ribbon running along the [100] direction (Fig. 5b).

In all three crystal structures, (1)–(3), the hydrogen-bonded motifs described above are further connected by weak C—H...O and C—H...N interactions, resulting in tri-periodic supramolecular networks. These supramolecular architectures are further stabilized by aromatic π - π interactions (Table 5). However, each structure exhibits a different π -stacking arrangement. In (1), two types of molecular stacks are observed, consisting exclusively of either NFX or coformer molecules. In (3), only the NFX molecules form molecular stacks, in which the heterocyclic and benzene rings participate in π - π interactions. In contrast, in (2), aromatic π - π interactions occur only within isolated NFX-coformer pairs.

4. Hirshfeld surface analysis

Hirshfeld surfaces and fingerprint plots (Spackman & McKinnon, 2002; Spackman & Jayatilaka, 2009) were gener-

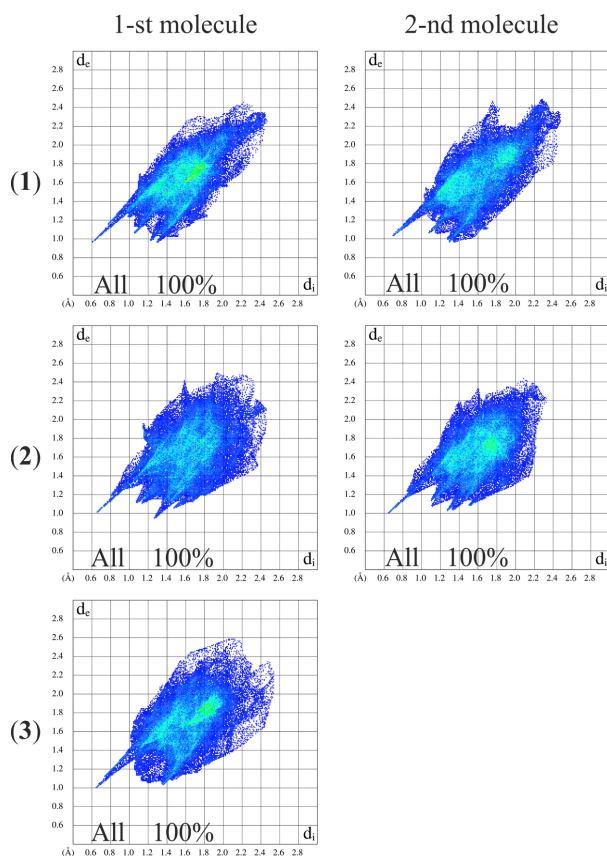


Figure 6

Two-dimensional fingerprint plots for the Hirshfeld surfaces of five independent norfloxacin molecules in molecular salts **(1)**–**(3)**. The d_i and d_e values are the closest internal and external distances (in Å) from given points on the Hirshfeld surface.

ated with *CrystalExplorer* (Spackman *et al.*, 2021) to visualize and quantify intermolecular interactions in three norfloxacin salts (Fig. 6). As shown in the fingerprint breakdown diagram (Fig. 7), the Hirshfeld surfaces of the NFX molecules are dominated by H...H and O...H/H...O contacts. The contribution of F...H/H...F contacts is similar in all three salts (6.8–9.7%) indicating a comparable involvement of the fluorine atom in the crystal packing. The relatively high contribution of C...C contacts in salts **(1)** (8.7–9.9%) and **(3)** (8.3%) suggests more pronounced π - π -stacking interactions, whereas these

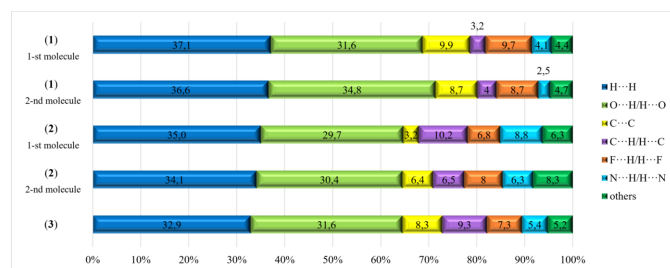


Figure 7

Diagram of the percentage contributions of different close contacts to the Hirshfeld surfaces of five independent norfloxacin molecules in molecular salts **(1)**–**(3)**.

contacts are considerably less significant for one of the molecules in salt **(2)** (3.2%). An opposite trend is observed for N...H/H...N and C...H/H...C contacts, which reach their highest contributions to the Hirshfeld surface of the first independent NFX molecule in salt **(2)**, accounting for 8.8% and 10.2%, respectively. Each of the other contact types contributes less than 6.6% in multicomponent forms **(1)**–**(3)**.

5. Database survey

A search of the Cambridge Structural Database (CSD version 6.01, February 2026; Groom *et al.*, 2016) revealed 16 crystal structures of multicomponent forms of norfloxacin containing benzoic acid derivatives as coformers (private communications have been omitted). Nine are molecular salts with substituted benzoate anions (BOLVUX01; Li & Jiang, 2025; LOQLAG; LOQLOU; LOQLUA; Xu *et al.*, 2014; ORUYEG; Reddy *et al.*, 2011; WOXRUA01; WOXSAH01; Liang *et al.*, 2024; INOMOR; de Paula Costa *et al.*, 2026; LUVCOY01; Wang *et al.*, 2025), whereas the remaining seven contain carboxybenzoate anions (HIGSAT; HIGSIB; HIGSUN; HIGTAU; UPAYOA01; Huang *et al.*, 2013; UPAYOA; Yan *et al.*, 2011; VOLMUI01; Sun *et al.*, 2024). Most of these crystal forms are hydrated, comprising nine monohydrates and two dihydrates.

A separate CSD search for multicomponent forms of norfloxacin with *N*-heteroaromatic coformers yielded only three entries: a cocrystal with pyridine-3-carboxylic acid (GADQEL; Ferreira *et al.*, 2020), a tetrahydrate molecular salt with a pyrazine-2-carboxylate (UKIXOF; de Paula Costa *et al.*, 2025) and a chloroform-solvated cocrystal with isonicotinamide, in which norfloxacin occurs in the zwitterionic form (VETVUM; Basavoju *et al.*, 2006).

6. Crystallization

Norfloxacin used in this study was purchased from TCI (Tokyo Chemical Industry Co., Ltd., Japan). Pyridine-3,5-dicarboxylic acid, pyridazine-3-carboxylic acid and pyrimidine-5-carboxylic acid were obtained from Fluorochem (Hadfield, United Kingdom). Ethanol (analytical grade; Chempur, Piekary Śląskie, Poland) was used as received.

Equimolar amounts of norfloxacin and the appropriate coformer were ground together and dissolved in ethanol. The resulting mixtures were heated under stirring until complete dissolution, filtered and left to evaporate slowly at ambient temperature. Single crystals suitable for X-ray diffraction were obtained from the filtrates.

7. Refinement

Crystal data, data collection and structure refinement details are summarized in Table 6. The H atoms bonded to C atoms were included in the refinement using riding models with constrained distances set to 0.95 Å (aromatic), 0.98 Å (methyl), 0.99 Å (methylene), and with $U_{\text{iso}}(\text{H}) = kU_{\text{eq}}(\text{C})$, where $k=1.5$ for the methyl group and 1.2 for all other H atoms

Table 6
Experimental details.

	(1)	(2)	(3)
Crystal data			
Chemical formula	$2\text{C}_{16}\text{H}_{19}\text{FN}_3\text{O}_3^+ \cdot 2\text{C}_7\text{H}_4\text{NO}_4^- \cdot 3\text{H}_2\text{O}$	$\text{C}_{16}\text{H}_{19}\text{FN}_3\text{O}_3^+ \cdot \text{C}_5\text{H}_3\text{N}_2\text{O}_2^-$	$\text{C}_{16}\text{H}_{19}\text{FN}_3\text{O}_3^+ \cdot \text{C}_5\text{H}_3\text{N}_2\text{O}_2^- \cdot \text{H}_2\text{O}$
M_r	1026.95	443.43	461.45
Crystal system, space group	Triclinic, $P\bar{1}$	Triclinic, $P\bar{1}$	Triclinic, $P\bar{1}$
Temperature (K)	100	100	100
a, b, c (Å)	9.7162 (2), 12.5261 (2), 19.3460 (4)	9.0971 (1), 12.9566 (1), 19.2412 (2)	8.3657 (1), 9.5115 (1), 13.6068 (2)
α, β, γ (°)	96.772 (2), 99.257 (1), 99.128 (1)	104.666 (1), 95.813 (1), 110.485 (1)	76.490 (1), 82.771 (1), 84.520 (1)
V (Å ³)	2269.44 (8)	2009.65 (4)	1041.92 (2)
Z	2	4	2
Radiation type	Cu $K\alpha$	Cu $K\alpha$	Cu $K\alpha$
μ (mm ⁻¹)	1.03	0.95	0.98
Crystal size (mm)	0.18 × 0.12 × 0.04	0.25 × 0.19 × 0.05	0.22 × 0.17 × 0.04
Data collection			
Diffraction	Rigaku, XtaLAB Synergy, Dual-flex, HyPix	Rigaku, XtaLAB Synergy, Dual-flex, HyPix	Rigaku, XtaLAB Synergy, Dual-flex, HyPix
Absorption correction	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2025)	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2025)	Gaussian (<i>CrysAlis PRO</i> ; Rigaku OD, 2025)
$T_{\min}, T_{\max}$	0.586, 1.000	0.626, 1.000	0.660, 1.000
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	28003, 9032, 8052	45406, 8113, 7634	31025, 4184, 3922
R_{int}	0.031	0.019	0.023
$(\sin \theta/\lambda)_{\text{max}}$ (Å ⁻¹)	0.633	0.633	0.632
Refinement			
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.036, 0.104, 1.04	0.031, 0.084, 1.05	0.037, 0.097, 1.07
No. of reflections	9032	8113	4184
No. of parameters	716	603	319
No. of restraints	2	0	0
H-atom treatment	H atoms treated by a mixture of independent and constrained refinement	H atoms treated by a mixture of independent and constrained refinement	H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ (e Å ⁻³)	0.71, -0.27	0.25, -0.21	0.31, -0.22

Computer programs: *CrysAlis PRO* (Rigaku OD, 2025), *SHELXT* (Sheldrick, 2015a), *SHELXL2025/1* (Sheldrick, 2015b), *Mercury* (Macrae *et al.*, 2020) and *publCIF* (Westrip, 2010).

bonded to C atoms. The hydrogen atoms bonded to heteroatoms (N and O) involved in hydrogen bonds were located in difference Fourier maps and refined isotropically.

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Structural characterization of three norfloxacin salts with N-heterocyclic mono- and dicarboxylic acids

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Computing details

4-(3-Carboxy-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinolin-7-yl)piperazin-1-ium 5-carboxy-pyridine-3-carboxylate trihydrate (1)

Crystal data

$2\text{C}_{16}\text{H}_{19}\text{FN}_3\text{O}_3^+ \cdot 2\text{C}_7\text{H}_4\text{NO}_4^- \cdot 3(\text{H}_2\text{O})$

$M_r = 1026.95$

Triclinic, $P\bar{1}$

$a = 9.7162(2) \text{ \AA}$

$b = 12.5261(2) \text{ \AA}$

$c = 19.3460(4) \text{ \AA}$

$\alpha = 96.772(2)^\circ$

$\beta = 99.257(1)^\circ$

$\gamma = 99.128(1)^\circ$

$V = 2269.44(8) \text{ \AA}^3$

$Z = 2$

$F(000) = 1076$

$D_x = 1.503 \text{ Mg m}^{-3}$

Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$

Cell parameters from 16961 reflections

$\theta = 3.6\text{--}76.5^\circ$

$\mu = 1.03 \text{ mm}^{-1}$

$T = 100 \text{ K}$

Plate, colourless

$0.18 \times 0.12 \times 0.04 \text{ mm}$

Data collection

Rigaku, XtaLAB Synergy, Dualflex, HyPix diffractometer

Radiation source: micro-focus sealed X-ray tube, PhotonJet (Cu) X-ray Source

Mirror monochromator

Detector resolution: $10.0000 \text{ pixels mm}^{-1}$

ω scans

Absorption correction: gaussian (CrysAlisPro; Rigaku OD, 2025)

$T_{\min} = 0.586$, $T_{\max} = 1.000$

28003 measured reflections

9032 independent reflections

8052 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.031$

$\theta_{\max} = 77.5^\circ$, $\theta_{\min} = 3.6^\circ$

$h = -11 \rightarrow 12$

$k = -14 \rightarrow 15$

$l = -24 \rightarrow 17$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.036$

$wR(F^2) = 0.104$

$S = 1.04$

9032 reflections

716 parameters

2 restraints

Primary atom site location: structure-invariant direct methods

Hydrogen site location: mixed

H atoms treated by a mixture of independent and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0565P)^2 + 0.7761P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.71 \text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.27 \text{ e \AA}^{-3}$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.57815 (8)	0.05966 (6)	0.06233 (4)	0.02211 (17)
O1	0.80626 (9)	−0.15482 (7)	−0.10697 (5)	0.01858 (18)
O2	0.98509 (10)	−0.25471 (7)	−0.15846 (5)	0.01963 (19)
O3	1.17176 (10)	−0.28635 (8)	−0.08633 (5)	0.0232 (2)
N1	1.06397 (11)	−0.11602 (8)	0.08782 (5)	0.0164 (2)
N2	0.72793 (11)	0.07193 (8)	0.19531 (5)	0.0169 (2)
N3	0.63609 (12)	0.10385 (9)	0.32831 (6)	0.0193 (2)
C1	1.09586 (13)	−0.17183 (10)	0.03083 (7)	0.0169 (2)
H1	1.180717	−0.201089	0.036536	0.020*
C2	1.01246 (13)	−0.18906 (9)	−0.03545 (6)	0.0166 (2)
C3	0.88432 (13)	−0.14612 (9)	−0.04702 (6)	0.0159 (2)
C4	0.84887 (13)	−0.08926 (9)	0.01590 (6)	0.0157 (2)
C5	0.72535 (13)	−0.04342 (10)	0.01017 (7)	0.0169 (2)
H5	0.665012	−0.049800	−0.034490	0.020*
C6	0.69282 (13)	0.01007 (10)	0.06889 (7)	0.0173 (2)
C7	0.77587 (13)	0.02058 (10)	0.13735 (6)	0.0163 (2)
C8	0.90041 (13)	−0.02048 (10)	0.14252 (6)	0.0166 (2)
H8	0.961326	−0.011885	0.187208	0.020*
C9	0.93817 (13)	−0.07492 (9)	0.08235 (6)	0.0159 (2)
C10	1.06412 (13)	−0.24769 (10)	−0.09447 (7)	0.0175 (2)
C11	1.16011 (14)	−0.10326 (11)	0.15723 (7)	0.0197 (3)
H11A	1.169190	−0.028044	0.182104	0.024*
H11B	1.255448	−0.114408	0.149545	0.024*
C12	1.10463 (16)	−0.18459 (11)	0.20290 (7)	0.0250 (3)
H12A	1.165672	−0.170393	0.249749	0.038*
H12B	1.104607	−0.258996	0.180550	0.038*
H12C	1.007747	−0.176947	0.207913	0.038*
C13	0.59338 (14)	0.00870 (10)	0.20668 (7)	0.0210 (3)
H13A	0.610827	−0.060841	0.222896	0.025*
H13B	0.524681	−0.009059	0.161439	0.025*
C14	0.53187 (14)	0.07334 (11)	0.26124 (7)	0.0202 (3)
H14A	0.507355	0.140163	0.243492	0.024*
H14B	0.444055	0.028938	0.269709	0.024*
C15	0.77156 (14)	0.16682 (10)	0.31686 (7)	0.0204 (3)
H15A	0.840660	0.183929	0.361977	0.025*
H15B	0.754883	0.236656	0.300860	0.025*
C16	0.83151 (14)	0.10108 (10)	0.26164 (7)	0.0188 (3)
H16A	0.919851	0.144628	0.253081	0.023*
H16B	0.854825	0.033789	0.279166	0.023*

F2	0.07596 (8)	0.60893 (6)	0.07893 (4)	0.02097 (16)
O4	0.28416 (11)	0.41374 (8)	−0.10935 (5)	0.0275 (2)
O5	0.43829 (12)	0.30850 (9)	−0.16961 (5)	0.0314 (2)
O6	0.59023 (11)	0.22416 (9)	−0.10910 (6)	0.0312 (2)
N4	0.46059 (11)	0.33020 (8)	0.07957 (6)	0.0174 (2)
N5	0.17230 (11)	0.54148 (8)	0.20568 (5)	0.0163 (2)
N6	0.06828 (11)	0.56020 (9)	0.33654 (6)	0.0177 (2)
C21	0.50266 (13)	0.29701 (10)	0.01929 (7)	0.0193 (3)
H21	0.574332	0.253360	0.021219	0.023*
C22	0.44701 (14)	0.32307 (10)	−0.04555 (7)	0.0194 (3)
C23	0.34107 (14)	0.38969 (10)	−0.05079 (7)	0.0195 (3)
C24	0.30098 (13)	0.42960 (10)	0.01522 (7)	0.0173 (2)
C25	0.20360 (14)	0.50162 (10)	0.01617 (7)	0.0183 (2)
H25	0.165573	0.526262	−0.026217	0.022*
C26	0.16448 (13)	0.53560 (10)	0.07835 (7)	0.0169 (2)
C27	0.21476 (13)	0.50077 (10)	0.14325 (6)	0.0160 (2)
C28	0.31211 (13)	0.43118 (10)	0.14203 (7)	0.0171 (2)
H28	0.348283	0.405932	0.184508	0.020*
C29	0.35846 (13)	0.39715 (10)	0.07921 (7)	0.0165 (2)
C30	0.49918 (14)	0.28049 (11)	−0.10966 (7)	0.0234 (3)
C31	0.52326 (14)	0.29572 (10)	0.14671 (7)	0.0201 (3)
H31A	0.572282	0.234058	0.135600	0.024*
H31B	0.446194	0.269378	0.171749	0.024*
C32	0.62818 (15)	0.38788 (11)	0.19540 (7)	0.0238 (3)
H32A	0.656093	0.364436	0.241380	0.036*
H32B	0.583635	0.452369	0.201985	0.036*
H32C	0.712311	0.406142	0.174152	0.036*
C33	0.01886 (13)	0.51107 (10)	0.20646 (7)	0.0183 (2)
H33A	−0.004918	0.431826	0.208581	0.022*
H33B	−0.036645	0.525995	0.162206	0.022*
C34	−0.02090 (13)	0.57483 (10)	0.26940 (7)	0.0184 (2)
H34A	−0.007775	0.653382	0.264429	0.022*
H34B	−0.122097	0.549385	0.270788	0.022*
C35	0.22129 (14)	0.59142 (10)	0.33424 (7)	0.0190 (3)
H35A	0.278875	0.578969	0.378650	0.023*
H35B	0.243530	0.670137	0.330308	0.023*
C36	0.25860 (13)	0.52464 (10)	0.27164 (6)	0.0179 (2)
H36A	0.360404	0.546922	0.270211	0.021*
H36B	0.241189	0.446177	0.276780	0.021*
O7	0.52493 (11)	0.19182 (8)	0.42897 (5)	0.0269 (2)
O8	0.47013 (12)	0.33271 (8)	0.37645 (5)	0.0287 (2)
O9	0.19434 (11)	0.57629 (8)	0.50908 (5)	0.0230 (2)
O10	0.17782 (10)	0.55155 (8)	0.62052 (5)	0.0234 (2)
N7	0.42844 (12)	0.31409 (9)	0.61737 (6)	0.0198 (2)
C40	0.46609 (13)	0.28029 (10)	0.55558 (7)	0.0189 (2)
H40	0.519316	0.222923	0.554081	0.023*
C41	0.43164 (13)	0.32440 (10)	0.49402 (7)	0.0180 (2)
C42	0.35243 (13)	0.40754 (10)	0.49574 (7)	0.0181 (2)

H42	0.327874	0.440131	0.454631	0.022*
C43	0.30968 (13)	0.44226 (10)	0.55855 (7)	0.0178 (2)
C44	0.35107 (14)	0.39327 (10)	0.61772 (7)	0.0191 (3)
H44	0.322627	0.417573	0.660740	0.023*
C45	0.47909 (14)	0.28018 (10)	0.42722 (7)	0.0214 (3)
C46	0.22116 (13)	0.52903 (10)	0.56630 (7)	0.0183 (2)
O11	0.02791 (10)	0.72051 (7)	0.52760 (5)	0.02092 (19)
O12	0.01164 (11)	0.73611 (8)	0.41277 (5)	0.0239 (2)
O13	−0.34905 (11)	1.07996 (8)	0.60863 (5)	0.0251 (2)
O14	−0.21681 (11)	0.98202 (8)	0.67231 (5)	0.0238 (2)
N8	−0.24408 (12)	0.96650 (9)	0.42223 (6)	0.0205 (2)
C50	−0.16020 (14)	0.89128 (10)	0.42121 (7)	0.0187 (2)
H50	−0.137927	0.864803	0.377035	0.022*
C51	−0.10355 (13)	0.84945 (9)	0.48116 (6)	0.0163 (2)
C52	−0.13637 (13)	0.88881 (10)	0.54549 (7)	0.0175 (2)
H52	−0.099803	0.862807	0.587608	0.021*
C53	−0.22352 (13)	0.96686 (10)	0.54757 (7)	0.0174 (2)
C54	−0.27473 (14)	1.00335 (10)	0.48487 (7)	0.0198 (3)
H54	−0.334015	1.056701	0.486515	0.024*
C55	−0.01369 (13)	0.76158 (10)	0.47301 (6)	0.0168 (2)
C56	−0.26179 (14)	1.00938 (10)	0.61586 (7)	0.0190 (3)
O1W	0.24155 (13)	0.22831 (9)	0.27960 (6)	0.0337 (2)
O2W	0.07059 (12)	0.34373 (8)	0.35168 (6)	0.0290 (2)
O3W	−0.44895 (12)	1.18574 (8)	0.70984 (6)	0.0258 (2)
H2	0.906 (2)	−0.2189 (18)	−0.1516 (11)	0.049 (6)*
H3A	0.6603 (19)	0.0414 (15)	0.3468 (9)	0.030 (4)*
H3B	0.597 (2)	0.1427 (16)	0.3624 (11)	0.040 (5)*
H5A	0.368 (3)	0.352 (2)	−0.1569 (12)	0.060 (7)*
H6A	0.0518 (18)	0.4917 (15)	0.3449 (9)	0.026 (4)*
H6B	0.045 (2)	0.6073 (16)	0.3728 (10)	0.035 (5)*
H9	0.136 (2)	0.6294 (19)	0.5180 (12)	0.056 (6)*
H13	−0.368 (2)	1.1088 (19)	0.6518 (12)	0.055 (6)*
H1W1	0.325 (2)	0.266 (2)	0.3110 (12)	0.073 (8)*
H2W1	0.229 (3)	0.1601 (15)	0.2913 (13)	0.067 (7)*
H1W2	0.126 (2)	0.3056 (18)	0.3362 (11)	0.048 (6)*
H2W2	0.052 (2)	0.3183 (17)	0.3905 (12)	0.044 (5)*
H1W3	−0.494 (2)	1.2315 (18)	0.6894 (11)	0.044 (6)*
H2W3	−0.435 (2)	1.2139 (19)	0.7522 (13)	0.052 (6)*

Atomic displacement parameters ($\text{\AA}^2$)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0179 (4)	0.0264 (4)	0.0218 (4)	0.0107 (3)	0.0017 (3)	−0.0036 (3)
O1	0.0192 (4)	0.0188 (4)	0.0171 (4)	0.0045 (3)	0.0029 (3)	−0.0005 (3)
O2	0.0228 (5)	0.0186 (4)	0.0178 (4)	0.0060 (4)	0.0047 (4)	−0.0005 (3)
O3	0.0229 (5)	0.0251 (5)	0.0234 (5)	0.0107 (4)	0.0063 (4)	−0.0007 (4)
N1	0.0177 (5)	0.0161 (5)	0.0163 (5)	0.0050 (4)	0.0042 (4)	0.0015 (4)
N2	0.0159 (5)	0.0170 (5)	0.0174 (5)	0.0019 (4)	0.0053 (4)	−0.0010 (4)

N3	0.0253 (6)	0.0187 (5)	0.0178 (5)	0.0100 (4)	0.0087 (4)	0.0043 (4)
C1	0.0187 (6)	0.0133 (5)	0.0207 (6)	0.0044 (4)	0.0075 (5)	0.0029 (5)
C2	0.0190 (6)	0.0128 (5)	0.0192 (6)	0.0032 (5)	0.0066 (5)	0.0020 (5)
C3	0.0188 (6)	0.0115 (5)	0.0171 (6)	0.0003 (4)	0.0048 (5)	0.0018 (4)
C4	0.0159 (6)	0.0126 (5)	0.0190 (6)	0.0017 (4)	0.0054 (5)	0.0018 (4)
C5	0.0160 (6)	0.0158 (6)	0.0180 (6)	0.0020 (5)	0.0025 (5)	0.0006 (5)
C6	0.0143 (6)	0.0160 (6)	0.0219 (6)	0.0049 (5)	0.0039 (5)	0.0009 (5)
C7	0.0190 (6)	0.0127 (5)	0.0173 (6)	0.0017 (4)	0.0064 (5)	0.0004 (4)
C8	0.0180 (6)	0.0149 (5)	0.0170 (6)	0.0025 (5)	0.0039 (5)	0.0022 (4)
C9	0.0171 (6)	0.0117 (5)	0.0199 (6)	0.0026 (4)	0.0061 (5)	0.0031 (4)
C10	0.0190 (6)	0.0137 (5)	0.0198 (6)	0.0019 (5)	0.0058 (5)	0.0014 (5)
C11	0.0181 (6)	0.0219 (6)	0.0188 (6)	0.0081 (5)	0.0009 (5)	−0.0014 (5)
C12	0.0367 (8)	0.0189 (6)	0.0192 (6)	0.0095 (6)	0.0015 (5)	0.0008 (5)
C13	0.0206 (6)	0.0194 (6)	0.0225 (6)	0.0002 (5)	0.0088 (5)	−0.0007 (5)
C14	0.0195 (6)	0.0217 (6)	0.0215 (6)	0.0057 (5)	0.0079 (5)	0.0033 (5)
C15	0.0232 (6)	0.0188 (6)	0.0193 (6)	0.0047 (5)	0.0049 (5)	0.0000 (5)
C16	0.0193 (6)	0.0196 (6)	0.0174 (6)	0.0047 (5)	0.0036 (5)	0.0005 (5)
F2	0.0241 (4)	0.0205 (4)	0.0217 (4)	0.0124 (3)	0.0059 (3)	0.0029 (3)
O4	0.0371 (6)	0.0321 (5)	0.0179 (5)	0.0153 (4)	0.0084 (4)	0.0050 (4)
O5	0.0381 (6)	0.0367 (6)	0.0231 (5)	0.0131 (5)	0.0136 (4)	−0.0002 (4)
O6	0.0249 (5)	0.0316 (5)	0.0360 (6)	0.0085 (4)	0.0098 (4)	−0.0099 (4)
N4	0.0176 (5)	0.0144 (5)	0.0206 (5)	0.0043 (4)	0.0044 (4)	0.0011 (4)
N5	0.0162 (5)	0.0165 (5)	0.0169 (5)	0.0040 (4)	0.0049 (4)	0.0013 (4)
N6	0.0207 (5)	0.0156 (5)	0.0183 (5)	0.0038 (4)	0.0076 (4)	0.0019 (4)
C21	0.0171 (6)	0.0148 (6)	0.0258 (6)	0.0027 (5)	0.0069 (5)	−0.0018 (5)
C22	0.0198 (6)	0.0156 (6)	0.0226 (6)	0.0015 (5)	0.0077 (5)	−0.0016 (5)
C23	0.0214 (6)	0.0161 (6)	0.0209 (6)	0.0013 (5)	0.0067 (5)	0.0011 (5)
C24	0.0190 (6)	0.0140 (5)	0.0187 (6)	0.0016 (5)	0.0053 (5)	0.0008 (5)
C25	0.0199 (6)	0.0166 (6)	0.0184 (6)	0.0040 (5)	0.0035 (5)	0.0025 (5)
C26	0.0169 (6)	0.0132 (5)	0.0213 (6)	0.0052 (5)	0.0039 (5)	0.0015 (5)
C27	0.0162 (6)	0.0132 (5)	0.0181 (6)	0.0006 (4)	0.0050 (5)	0.0001 (4)
C28	0.0184 (6)	0.0151 (5)	0.0181 (6)	0.0029 (5)	0.0046 (5)	0.0027 (5)
C29	0.0160 (6)	0.0117 (5)	0.0217 (6)	0.0020 (4)	0.0043 (5)	0.0009 (4)
C30	0.0215 (7)	0.0202 (6)	0.0270 (7)	0.0002 (5)	0.0093 (5)	−0.0048 (5)
C31	0.0218 (6)	0.0176 (6)	0.0230 (6)	0.0090 (5)	0.0039 (5)	0.0038 (5)
C32	0.0240 (7)	0.0230 (6)	0.0237 (6)	0.0072 (5)	0.0012 (5)	0.0007 (5)
C33	0.0165 (6)	0.0182 (6)	0.0198 (6)	0.0026 (5)	0.0055 (5)	−0.0006 (5)
C34	0.0188 (6)	0.0180 (6)	0.0197 (6)	0.0059 (5)	0.0056 (5)	0.0020 (5)
C35	0.0188 (6)	0.0189 (6)	0.0183 (6)	0.0013 (5)	0.0043 (5)	0.0002 (5)
C36	0.0181 (6)	0.0183 (6)	0.0180 (6)	0.0046 (5)	0.0046 (5)	0.0023 (5)
O7	0.0351 (6)	0.0244 (5)	0.0257 (5)	0.0125 (4)	0.0137 (4)	0.0014 (4)
O8	0.0378 (6)	0.0256 (5)	0.0249 (5)	0.0073 (4)	0.0106 (4)	0.0041 (4)
O9	0.0297 (5)	0.0211 (4)	0.0225 (5)	0.0126 (4)	0.0087 (4)	0.0048 (4)
O10	0.0271 (5)	0.0228 (5)	0.0254 (5)	0.0103 (4)	0.0125 (4)	0.0049 (4)
N7	0.0208 (5)	0.0179 (5)	0.0214 (5)	0.0048 (4)	0.0049 (4)	0.0017 (4)
C40	0.0179 (6)	0.0152 (6)	0.0235 (6)	0.0032 (5)	0.0054 (5)	−0.0002 (5)
C41	0.0175 (6)	0.0147 (5)	0.0209 (6)	0.0002 (5)	0.0053 (5)	−0.0004 (5)
C42	0.0183 (6)	0.0146 (5)	0.0206 (6)	0.0013 (5)	0.0040 (5)	0.0016 (5)

C43	0.0161 (6)	0.0141 (5)	0.0226 (6)	0.0014 (5)	0.0042 (5)	0.0011 (5)
C44	0.0195 (6)	0.0172 (6)	0.0207 (6)	0.0032 (5)	0.0062 (5)	0.0000 (5)
C45	0.0243 (7)	0.0170 (6)	0.0231 (6)	0.0025 (5)	0.0083 (5)	0.0003 (5)
C46	0.0175 (6)	0.0152 (6)	0.0225 (6)	0.0025 (5)	0.0052 (5)	0.0018 (5)
O11	0.0265 (5)	0.0207 (4)	0.0188 (4)	0.0116 (4)	0.0059 (4)	0.0039 (3)
O12	0.0342 (5)	0.0228 (5)	0.0183 (4)	0.0133 (4)	0.0089 (4)	0.0011 (4)
O13	0.0329 (5)	0.0253 (5)	0.0232 (5)	0.0175 (4)	0.0107 (4)	0.0038 (4)
O14	0.0321 (5)	0.0223 (5)	0.0192 (4)	0.0106 (4)	0.0070 (4)	0.0016 (4)
N8	0.0246 (6)	0.0183 (5)	0.0204 (5)	0.0077 (4)	0.0055 (4)	0.0037 (4)
C50	0.0215 (6)	0.0161 (6)	0.0194 (6)	0.0046 (5)	0.0059 (5)	0.0015 (5)
C51	0.0175 (6)	0.0117 (5)	0.0194 (6)	0.0017 (5)	0.0045 (5)	0.0008 (5)
C52	0.0192 (6)	0.0143 (5)	0.0188 (6)	0.0026 (5)	0.0033 (5)	0.0020 (5)
C53	0.0192 (6)	0.0136 (5)	0.0198 (6)	0.0034 (5)	0.0054 (5)	0.0005 (5)
C54	0.0220 (6)	0.0163 (6)	0.0232 (6)	0.0065 (5)	0.0068 (5)	0.0027 (5)
C55	0.0168 (6)	0.0137 (5)	0.0194 (6)	0.0025 (5)	0.0037 (5)	0.0003 (4)
C56	0.0215 (6)	0.0145 (5)	0.0216 (6)	0.0040 (5)	0.0061 (5)	0.0013 (5)
O1W	0.0412 (6)	0.0317 (6)	0.0294 (5)	0.0156 (5)	0.0021 (5)	0.0028 (4)
O2W	0.0377 (6)	0.0224 (5)	0.0349 (6)	0.0117 (4)	0.0194 (5)	0.0099 (4)
O3W	0.0363 (6)	0.0235 (5)	0.0232 (5)	0.0143 (4)	0.0122 (4)	0.0050 (4)

Geometric parameters (Å, °)

F1—C6	1.3549 (14)	C23—C24	1.4505 (17)
O1—C3	1.2622 (15)	C24—C29	1.4064 (17)
O2—C10	1.3331 (16)	C24—C25	1.4078 (18)
O2—H2	0.97 (2)	C25—C26	1.3604 (18)
O3—C10	1.2160 (16)	C25—H25	0.9500
N1—C1	1.3404 (16)	C26—C27	1.4159 (17)
N1—C9	1.3933 (16)	C27—C28	1.3853 (18)
N1—C11	1.4832 (16)	C28—C29	1.4058 (17)
N2—C7	1.4074 (15)	C28—H28	0.9500
N2—C16	1.4664 (16)	C31—C32	1.5233 (18)
N2—C13	1.4798 (16)	C31—H31A	0.9900
N3—C14	1.4808 (17)	C31—H31B	0.9900
N3—C15	1.4864 (17)	C32—H32A	0.9800
N3—H3A	0.945 (19)	C32—H32B	0.9800
N3—H3B	0.94 (2)	C32—H32C	0.9800
C1—C2	1.3756 (18)	C33—C34	1.5132 (16)
C1—H1	0.9500	C33—H33A	0.9900
C2—C3	1.4285 (18)	C33—H33B	0.9900
C2—C10	1.4828 (17)	C34—H34A	0.9900
C3—C4	1.4547 (17)	C34—H34B	0.9900
C4—C5	1.4047 (18)	C35—C36	1.5153 (17)
C4—C9	1.4053 (17)	C35—H35A	0.9900
C5—C6	1.3574 (17)	C35—H35B	0.9900
C5—H5	0.9500	C36—H36A	0.9900
C6—C7	1.4157 (17)	C36—H36B	0.9900
C7—C8	1.3818 (18)	O7—C45	1.2589 (17)

C8—C9	1.4086 (17)	O8—C45	1.2442 (17)
C8—H8	0.9500	O9—C46	1.3245 (16)
C11—C12	1.5167 (19)	O9—H9	0.95 (2)
C11—H11A	0.9900	O10—C46	1.2113 (16)
C11—H11B	0.9900	N7—C44	1.3362 (17)
C12—H12A	0.9800	N7—C40	1.3463 (17)
C12—H12B	0.9800	C40—C41	1.3861 (18)
C12—H12C	0.9800	C40—H40	0.9500
C13—C14	1.5105 (17)	C41—C42	1.3900 (18)
C13—H13A	0.9900	C41—C45	1.5143 (17)
C13—H13B	0.9900	C42—C43	1.3904 (18)
C14—H14A	0.9900	C42—H42	0.9500
C14—H14B	0.9900	C43—C44	1.3945 (18)
C15—C16	1.5178 (17)	C43—C46	1.4978 (17)
C15—H15A	0.9900	C44—H44	0.9500
C15—H15B	0.9900	O11—C55	1.2649 (15)
C16—H16A	0.9900	O12—C55	1.2451 (15)
C16—H16B	0.9900	O13—C56	1.3225 (16)
F2—C26	1.3546 (14)	O13—H13	0.93 (2)
O4—C23	1.2696 (16)	O14—C56	1.2180 (16)
O5—C30	1.3265 (18)	N8—C50	1.3405 (17)
O5—H5A	0.98 (3)	N8—C54	1.3409 (17)
O6—C30	1.2156 (18)	C50—C51	1.3972 (17)
N4—C21	1.3390 (17)	C50—H50	0.9500
N4—C29	1.3969 (16)	C51—C52	1.3880 (17)
N4—C31	1.4852 (16)	C51—C55	1.5188 (17)
N5—C27	1.4037 (15)	C52—C53	1.3924 (18)
N5—C36	1.4669 (16)	C52—H52	0.9500
N5—C33	1.4821 (16)	C53—C54	1.3916 (18)
N6—C35	1.4854 (16)	C53—C56	1.4907 (17)
N6—C34	1.4886 (16)	C54—H54	0.9500
N6—H6A	0.887 (18)	O1W—H1W1	0.947 (17)
N6—H6B	0.94 (2)	O1W—H2W1	0.904 (16)
C21—C22	1.3808 (19)	O2W—H1W2	0.84 (2)
C21—H21	0.9500	O2W—H2W2	0.88 (2)
C22—C23	1.4230 (18)	O3W—H1W3	0.87 (2)
C22—C30	1.4872 (18)	O3W—H2W3	0.83 (3)
C10—O2—H2	106.3 (12)	C26—C25—C24	119.37 (12)
C1—N1—C9	120.08 (11)	C26—C25—H25	120.3
C1—N1—C11	119.43 (10)	C24—C25—H25	120.3
C9—N1—C11	120.42 (10)	F2—C26—C25	118.56 (11)
C7—N2—C16	115.79 (10)	F2—C26—C27	117.99 (11)
C7—N2—C13	111.44 (10)	C25—C26—C27	123.42 (11)
C16—N2—C13	110.95 (10)	C28—C27—N5	122.92 (11)
C14—N3—C15	111.06 (10)	C28—C27—C26	116.80 (11)
C14—N3—H3A	111.4 (11)	N5—C27—C26	120.18 (11)
C15—N3—H3A	105.5 (11)	C27—C28—C29	121.34 (11)

C14—N3—H3B	109.8 (12)	C27—C28—H28	119.3
C15—N3—H3B	111.8 (12)	C29—C28—H28	119.3
H3A—N3—H3B	107.2 (16)	N4—C29—C28	120.50 (11)
N1—C1—C2	123.61 (12)	N4—C29—C24	119.47 (11)
N1—C1—H1	118.2	C28—C29—C24	120.03 (11)
C2—C1—H1	118.2	O6—C30—O5	120.97 (12)
C1—C2—C3	120.46 (11)	O6—C30—C22	124.20 (13)
C1—C2—C10	117.87 (11)	O5—C30—C22	114.83 (12)
C3—C2—C10	121.59 (11)	N4—C31—C32	112.61 (10)
O1—C3—C2	123.68 (11)	N4—C31—H31A	109.1
O1—C3—C4	121.19 (11)	C32—C31—H31A	109.1
C2—C3—C4	115.13 (11)	N4—C31—H31B	109.1
C5—C4—C9	118.65 (11)	C32—C31—H31B	109.1
C5—C4—C3	119.63 (11)	H31A—C31—H31B	107.8
C9—C4—C3	121.67 (11)	C31—C32—H32A	109.5
C6—C5—C4	119.54 (11)	C31—C32—H32B	109.5
C6—C5—H5	120.2	H32A—C32—H32B	109.5
C4—C5—H5	120.2	C31—C32—H32C	109.5
F1—C6—C5	119.21 (11)	H32A—C32—H32C	109.5
F1—C6—C7	117.41 (11)	H32B—C32—H32C	109.5
C5—C6—C7	123.34 (12)	N5—C33—C34	110.95 (10)
C8—C7—N2	124.24 (11)	N5—C33—H33A	109.4
C8—C7—C6	117.05 (11)	C34—C33—H33A	109.4
N2—C7—C6	118.70 (11)	N5—C33—H33B	109.4
C7—C8—C9	120.82 (11)	C34—C33—H33B	109.4
C7—C8—H8	119.6	H33A—C33—H33B	108.0
C9—C8—H8	119.6	N6—C34—C33	110.65 (10)
N1—C9—C4	118.97 (11)	N6—C34—H34A	109.5
N1—C9—C8	120.60 (11)	C33—C34—H34A	109.5
C4—C9—C8	120.43 (11)	N6—C34—H34B	109.5
O3—C10—O2	121.26 (11)	C33—C34—H34B	109.5
O3—C10—C2	123.26 (12)	H34A—C34—H34B	108.1
O2—C10—C2	115.48 (11)	N6—C35—C36	110.39 (10)
N1—C11—C12	111.19 (11)	N6—C35—H35A	109.6
N1—C11—H11A	109.4	C36—C35—H35A	109.6
C12—C11—H11A	109.4	N6—C35—H35B	109.6
N1—C11—H11B	109.4	C36—C35—H35B	109.6
C12—C11—H11B	109.4	H35A—C35—H35B	108.1
H11A—C11—H11B	108.0	N5—C36—C35	109.91 (10)
C11—C12—H12A	109.5	N5—C36—H36A	109.7
C11—C12—H12B	109.5	C35—C36—H36A	109.7
H12A—C12—H12B	109.5	N5—C36—H36B	109.7
C11—C12—H12C	109.5	C35—C36—H36B	109.7
H12A—C12—H12C	109.5	H36A—C36—H36B	108.2
H12B—C12—H12C	109.5	C46—O9—H9	109.0 (13)
N2—C13—C14	110.58 (10)	C44—N7—C40	116.97 (11)
N2—C13—H13A	109.5	N7—C40—C41	123.76 (12)
C14—C13—H13A	109.5	N7—C40—H40	118.1

N2—C13—H13B	109.5	C41—C40—H40	118.1
C14—C13—H13B	109.5	C40—C41—C42	118.34 (11)
H13A—C13—H13B	108.1	C40—C41—C45	119.51 (11)
N3—C14—C13	109.91 (11)	C42—C41—C45	122.15 (11)
N3—C14—H14A	109.7	C41—C42—C43	118.98 (12)
C13—C14—H14A	109.7	C41—C42—H42	120.5
N3—C14—H14B	109.7	C43—C42—H42	120.5
C13—C14—H14B	109.7	C42—C43—C44	118.20 (12)
H14A—C14—H14B	108.2	C42—C43—C46	123.77 (11)
N3—C15—C16	110.17 (10)	C44—C43—C46	118.03 (11)
N3—C15—H15A	109.6	N7—C44—C43	123.74 (12)
C16—C15—H15A	109.6	N7—C44—H44	118.1
N3—C15—H15B	109.6	C43—C44—H44	118.1
C16—C15—H15B	109.6	O8—C45—O7	126.06 (12)
H15A—C15—H15B	108.1	O8—C45—C41	118.52 (12)
N2—C16—C15	110.09 (10)	O7—C45—C41	115.42 (11)
N2—C16—H16A	109.6	O10—C46—O9	124.15 (12)
C15—C16—H16A	109.6	O10—C46—C43	121.68 (12)
N2—C16—H16B	109.6	O9—C46—C43	114.16 (11)
C15—C16—H16B	109.6	C56—O13—H13	112.3 (14)
H16A—C16—H16B	108.2	C50—N8—C54	117.50 (11)
C30—O5—H5A	106.1 (14)	N8—C50—C51	123.98 (12)
C21—N4—C29	120.05 (11)	N8—C50—H50	118.0
C21—N4—C31	119.69 (11)	C51—C50—H50	118.0
C29—N4—C31	120.25 (10)	C52—C51—C50	117.61 (11)
C27—N5—C36	115.34 (10)	C52—C51—C55	123.16 (11)
C27—N5—C33	114.99 (10)	C50—C51—C55	119.19 (11)
C36—N5—C33	111.25 (9)	C51—C52—C53	119.22 (11)
C35—N6—C34	110.73 (9)	C51—C52—H52	120.4
C35—N6—H6A	108.6 (11)	C53—C52—H52	120.4
C34—N6—H6A	111.7 (11)	C54—C53—C52	118.81 (11)
C35—N6—H6B	109.2 (11)	C54—C53—C56	120.95 (11)
C34—N6—H6B	106.4 (11)	C52—C53—C56	120.24 (11)
H6A—N6—H6B	110.2 (16)	N8—C54—C53	122.89 (12)
N4—C21—C22	123.25 (12)	N8—C54—H54	118.6
N4—C21—H21	118.4	C53—C54—H54	118.6
C22—C21—H21	118.4	O12—C55—O11	125.90 (12)
C21—C22—C23	120.32 (11)	O12—C55—C51	116.70 (11)
C21—C22—C30	119.10 (12)	O11—C55—C51	117.40 (11)
C23—C22—C30	120.57 (12)	O14—C56—O13	123.72 (12)
O4—C23—C22	122.72 (12)	O14—C56—C53	123.35 (12)
O4—C23—C24	121.21 (12)	O13—C56—C53	112.93 (11)
C22—C23—C24	116.07 (11)	H1W1—O1W—H2W1	104 (2)
C29—C24—C25	118.92 (11)	H1W2—O2W—H2W2	105.7 (19)
C29—C24—C23	120.71 (12)	H1W3—O3W—H2W3	101 (2)
C25—C24—C23	120.37 (11)		
C9—N1—C1—C2	2.19 (18)	C36—N5—C27—C26	−163.87 (11)

C11—N1—C1—C2	179.19 (11)	C33—N5—C27—C26	64.60 (14)
N1—C1—C2—C3	0.47 (18)	F2—C26—C27—C28	−176.02 (10)
N1—C1—C2—C10	177.09 (11)	C25—C26—C27—C28	2.07 (18)
C1—C2—C3—O1	176.79 (11)	F2—C26—C27—N5	0.35 (17)
C10—C2—C3—O1	0.30 (18)	C25—C26—C27—N5	178.44 (11)
C1—C2—C3—C4	−2.58 (16)	N5—C27—C28—C29	−176.19 (11)
C10—C2—C3—C4	−179.07 (10)	C26—C27—C28—C29	0.07 (17)
O1—C3—C4—C5	0.25 (17)	C21—N4—C29—C28	179.09 (11)
C2—C3—C4—C5	179.64 (10)	C31—N4—C29—C28	−1.05 (17)
O1—C3—C4—C9	−177.16 (11)	C21—N4—C29—C24	−0.29 (17)
C2—C3—C4—C9	2.23 (16)	C31—N4—C29—C24	179.57 (11)
C9—C4—C5—C6	−2.20 (17)	C27—C28—C29—N4	177.62 (11)
C3—C4—C5—C6	−179.68 (11)	C27—C28—C29—C24	−3.00 (18)
C4—C5—C6—F1	176.11 (10)	C25—C24—C29—N4	−176.75 (11)
C4—C5—C6—C7	−1.52 (19)	C23—C24—C29—N4	3.26 (17)
C16—N2—C7—C8	12.22 (17)	C25—C24—C29—C28	3.87 (17)
C13—N2—C7—C8	−115.84 (13)	C23—C24—C29—C28	−176.12 (11)
C16—N2—C7—C6	−167.46 (11)	C21—C22—C30—O6	−0.8 (2)
C13—N2—C7—C6	64.49 (14)	C23—C22—C30—O6	179.25 (12)
F1—C6—C7—C8	−173.54 (10)	C21—C22—C30—O5	179.12 (11)
C5—C6—C7—C8	4.14 (18)	C23—C22—C30—O5	−0.81 (18)
F1—C6—C7—N2	6.16 (16)	C21—N4—C31—C32	105.85 (13)
C5—C6—C7—N2	−176.16 (11)	C29—N4—C31—C32	−74.01 (14)
N2—C7—C8—C9	177.31 (10)	C27—N5—C33—C34	−169.65 (10)
C6—C7—C8—C9	−3.01 (17)	C36—N5—C33—C34	56.90 (13)
C1—N1—C9—C4	−2.47 (17)	C35—N6—C34—C33	55.74 (13)
C11—N1—C9—C4	−179.44 (10)	N5—C33—C34—N6	−55.09 (13)
C1—N1—C9—C8	177.09 (10)	C34—N6—C35—C36	−57.47 (13)
C11—N1—C9—C8	0.13 (17)	C27—N5—C36—C35	168.53 (10)
C5—C4—C9—N1	−177.22 (10)	C33—N5—C36—C35	−58.19 (12)
C3—C4—C9—N1	0.21 (17)	N6—C35—C36—N5	58.46 (13)
C5—C4—C9—C8	3.22 (17)	C44—N7—C40—C41	1.38 (18)
C3—C4—C9—C8	−179.35 (10)	N7—C40—C41—C42	−0.70 (19)
C7—C8—C9—N1	179.90 (11)	N7—C40—C41—C45	−179.89 (11)
C7—C8—C9—C4	−0.55 (17)	C40—C41—C42—C43	−0.70 (18)
C1—C2—C10—O3	4.32 (18)	C45—C41—C42—C43	178.47 (11)
C3—C2—C10—O3	−179.10 (11)	C41—C42—C43—C44	1.32 (18)
C1—C2—C10—O2	−174.93 (10)	C41—C42—C43—C46	−178.43 (11)
C3—C2—C10—O2	1.65 (16)	C40—N7—C44—C43	−0.70 (18)
C1—N1—C11—C12	−99.65 (13)	C42—C43—C44—N7	−0.63 (19)
C9—N1—C11—C12	77.34 (14)	C46—C43—C44—N7	179.13 (11)
C7—N2—C13—C14	−171.06 (11)	C40—C41—C45—O8	−167.22 (12)
C16—N2—C13—C14	58.32 (14)	C42—C41—C45—O8	13.62 (19)
C15—N3—C14—C13	57.01 (13)	C40—C41—C45—O7	12.66 (18)
N2—C13—C14—N3	−57.07 (14)	C42—C41—C45—O7	−166.50 (12)
C14—N3—C15—C16	−57.21 (13)	C42—C43—C46—O10	174.72 (12)
C7—N2—C16—C15	173.61 (10)	C44—C43—C46—O10	−5.02 (18)
C13—N2—C16—C15	−58.08 (13)	C42—C43—C46—O9	−4.47 (18)

N3—C15—C16—N2	57.27 (13)	C44—C43—C46—O9	175.79 (11)
C29—N4—C21—C22	−1.89 (18)	C54—N8—C50—C51	0.05 (19)
C31—N4—C21—C22	178.25 (11)	N8—C50—C51—C52	−0.16 (19)
N4—C21—C22—C23	1.00 (19)	N8—C50—C51—C55	178.00 (11)
N4—C21—C22—C30	−178.93 (11)	C50—C51—C52—C53	0.24 (18)
C21—C22—C23—O4	−178.25 (12)	C55—C51—C52—C53	−177.83 (11)
C30—C22—C23—O4	1.68 (19)	C51—C52—C53—C54	−0.23 (18)
C21—C22—C23—C24	1.91 (17)	C51—C52—C53—C56	179.23 (11)
C30—C22—C23—C24	−178.17 (11)	C50—N8—C54—C53	−0.03 (19)
O4—C23—C24—C29	176.17 (12)	C52—C53—C54—N8	0.12 (19)
C22—C23—C24—C29	−3.98 (17)	C56—C53—C54—N8	−179.34 (11)
O4—C23—C24—C25	−3.82 (19)	C52—C51—C55—O12	−177.75 (11)
C22—C23—C24—C25	176.02 (11)	C50—C51—C55—O12	4.20 (17)
C29—C24—C25—C26	−1.83 (18)	C52—C51—C55—O11	2.64 (17)
C23—C24—C25—C26	178.17 (11)	C50—C51—C55—O11	−175.40 (11)
C24—C25—C26—F2	176.90 (10)	C54—C53—C56—O14	−178.09 (12)
C24—C25—C26—C27	−1.18 (19)	C52—C53—C56—O14	2.46 (19)
C36—N5—C27—C28	12.27 (16)	C54—C53—C56—O13	1.69 (17)
C33—N5—C27—C28	−119.26 (13)	C52—C53—C56—O13	−177.76 (11)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
O2—H2...O1	0.97 (2)	1.64 (2)	2.5554 (13)	156 (2)
O5—H5 <i>A</i> ...O4	0.98 (3)	1.54 (3)	2.4830 (14)	158 (2)
O9—H9...O11	0.95 (2)	1.69 (3)	2.6396 (13)	176 (2)
O13—H13...O3 <i>W</i>	0.93 (2)	1.75 (2)	2.6342 (14)	157 (2)
N3—H3 <i>B</i> ...O7	0.94 (2)	1.66 (2)	2.5925 (14)	170 (2)
N3—H3 <i>A</i> ...N8 ⁱ	0.945 (19)	2.009 (19)	2.8825 (16)	153 (2)
N6—H6 <i>A</i> ...O2 <i>W</i>	0.887 (18)	1.908 (19)	2.7646 (15)	162 (2)
N6—H6 <i>B</i> ...O12	0.94 (2)	1.81 (2)	2.6856 (14)	154 (2)
C5—H5...F1 ⁱⁱ	0.95	2.32	3.0193 (15)	130
C11—H11 <i>B</i> ...O6 ⁱⁱⁱ	0.99	2.35	3.2564 (16)	152
C12—H12 <i>B</i> ...F2 ⁱ	0.98	2.36	3.2576 (15)	151
C15—H15 <i>A</i> ...O11 ^{iv}	0.99	2.38	3.3290 (16)	161
C33—H33 <i>A</i> ...O2 ⁱⁱ	0.99	2.36	3.2282 (15)	145
O1 <i>W</i> —H1 <i>W</i> 1...O8	0.95 (2)	1.76 (2)	2.7047 (2)	174 (2)
O1 <i>W</i> —H2 <i>W</i> 1...O14 ^v	0.90 (2)	1.98 (2)	2.8834 (2)	173 (2)
O2 <i>W</i> —H1 <i>W</i> 2...O1 <i>W</i>	0.84 (2)	1.97 (2)	2.7974 (16)	167 (2)
O2 <i>W</i> —H2 <i>W</i> 2...O11 ^v	0.88 (2)	1.95 (2)	2.8171 (14)	166 (2)
O3 <i>W</i> —H1 <i>W</i> 3...N7 ^{vi}	0.87 (2)	1.96 (2)	2.7937 (15)	161 (2)
O3 <i>W</i> —H2 <i>W</i> 3...O5 ^{vii}	0.83 (3)	2.41 (2)	3.0846 (14)	138 (2)

Symmetry codes: (i) $x+1, y-1, z$; (ii) $-x+1, -y, -z$; (iii) $-x+2, -y, -z$; (iv) $-x+1, -y+1, -z+1$; (v) $-x, -y+1, -z+1$; (vi) $x-1, y+1, z$; (vii) $x-1, y+1, z+1$.

4-(3-Carboxy-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinolin-7-yl)piperazin-1-ium pyridazine-3-carboxylate (2)

Crystal data $C_{16}H_{19}FN_3O_3^+ \cdot C_5H_3N_2O_2^-$ $M_r = 443.43$ Triclinic, $P\bar{1}$ $a = 9.0971(1) \text{ \AA}$ $b = 12.9566(1) \text{ \AA}$ $c = 19.2412(2) \text{ \AA}$ $\alpha = 104.666(1)^\circ$ $\beta = 95.813(1)^\circ$ $\gamma = 110.485(1)^\circ$ $V = 2009.65(4) \text{ \AA}^3$ $Z = 4$ $F(000) = 928$ $D_x = 1.466 \text{ Mg m}^{-3}$ Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$

Cell parameters from 34780 reflections

 $\theta = 3.8\text{--}77.3^\circ$ $\mu = 0.95 \text{ mm}^{-1}$ $T = 100 \text{ K}$

Plate, colourless

 $0.25 \times 0.19 \times 0.05 \text{ mm}$ *Data collection*Rigaku, XtaLAB Synergy, Dualflex, HyPix
diffractometerRadiation source: micro-focus sealed X-ray
tube, PhotonJet (Cu) X-ray Source

Mirror monochromator

Detector resolution: $10.0000 \text{ pixels mm}^{-1}$ ω scansAbsorption correction: gaussian
(CrysAlisPro; Rigaku OD, 2025) $T_{\min} = 0.626$, $T_{\max} = 1.000$

45406 measured reflections

8113 independent reflections

7634 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.019$ $\theta_{\max} = 77.4^\circ$, $\theta_{\min} = 3.8^\circ$ $h = -10 \rightarrow 11$ $k = -15 \rightarrow 16$ $l = -24 \rightarrow 23$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.031$ $wR(F^2) = 0.084$ $S = 1.05$

8113 reflections

603 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Hydrogen site location: mixed

H atoms treated by a mixture of independent
and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.0428P)^2 + 0.6879P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.25 \text{ e \AA}^{-3}$ $\Delta\rho_{\min} = -0.21 \text{ e \AA}^{-3}$ *Special details*

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.36929 (8)	0.56898 (5)	0.71179 (3)	0.02401 (14)
O1	0.51080 (9)	0.36291 (6)	0.48658 (4)	0.02014 (16)
O2	0.53287 (9)	0.20568 (7)	0.38541 (4)	0.02135 (16)
O3	0.47224 (9)	0.02945 (6)	0.39598 (4)	0.02155 (16)
N1	0.29377 (10)	0.11660 (7)	0.58316 (5)	0.01631 (17)
N2	0.25725 (10)	0.41334 (7)	0.78774 (5)	0.01732 (17)
N3	0.12634 (11)	0.44497 (8)	0.91723 (5)	0.0232 (2)

C1	0.34766 (12)	0.08884 (9)	0.52180 (5)	0.01644 (19)
H1	0.333827	0.010486	0.500777	0.020*
C2	0.42220 (12)	0.16825 (9)	0.48760 (5)	0.0164 (2)
C3	0.44332 (11)	0.28707 (9)	0.51632 (6)	0.0165 (2)
C4	0.38257 (12)	0.31573 (9)	0.58243 (5)	0.01630 (19)
C5	0.39718 (12)	0.43021 (9)	0.61530 (6)	0.0178 (2)
H5	0.440187	0.488056	0.592529	0.021*
C6	0.34937 (12)	0.45726 (8)	0.67973 (6)	0.0181 (2)
C7	0.28665 (12)	0.37591 (9)	0.71729 (5)	0.0168 (2)
C8	0.26614 (12)	0.26245 (9)	0.68323 (5)	0.0170 (2)
H8	0.219683	0.204727	0.705623	0.020*
C9	0.31273 (11)	0.23123 (8)	0.61624 (5)	0.01576 (19)
C10	0.47810 (12)	0.12736 (9)	0.41999 (5)	0.0172 (2)
C11	0.20953 (13)	0.02375 (9)	0.61409 (6)	0.0189 (2)
H11A	0.251163	0.050424	0.667828	0.023*
H11B	0.232630	−0.045391	0.592531	0.023*
C12	0.02891 (13)	−0.00969 (9)	0.59900 (6)	0.0242 (2)
H12A	−0.019484	−0.059436	0.628219	0.036*
H12B	−0.016158	−0.051622	0.546671	0.036*
H12C	0.005979	0.060435	0.612374	0.036*
C13	0.12390 (12)	0.45322 (9)	0.79015 (6)	0.0201 (2)
H13A	0.020357	0.385714	0.770054	0.024*
H13B	0.132536	0.504891	0.759316	0.024*
C14	0.12832 (13)	0.51772 (9)	0.86843 (6)	0.0233 (2)
H14A	0.226588	0.589760	0.886579	0.028*
H14B	0.034393	0.539528	0.869611	0.028*
C15	0.25414 (13)	0.39901 (9)	0.91187 (6)	0.0225 (2)
H15A	0.242311	0.345488	0.941576	0.027*
H15B	0.360354	0.463788	0.932013	0.027*
C16	0.24551 (13)	0.33513 (9)	0.83249 (6)	0.0194 (2)
H16A	0.334427	0.307915	0.829811	0.023*
H16B	0.142862	0.266632	0.813290	0.023*
F2	1.19558 (7)	1.00676 (5)	0.77694 (3)	0.02172 (14)
O4	1.03759 (9)	1.19786 (6)	0.99992 (4)	0.02192 (16)
O5	0.89197 (9)	1.23291 (6)	1.10250 (4)	0.02316 (16)
O6	0.66496 (9)	1.09502 (6)	1.10469 (4)	0.02197 (16)
N4	0.68014 (10)	0.87336 (7)	0.90886 (5)	0.01628 (17)
N5	0.95393 (10)	0.79482 (7)	0.70781 (5)	0.01697 (17)
N6	0.96497 (11)	0.62052 (8)	0.58612 (5)	0.01960 (18)
C21	0.67623 (12)	0.94555 (9)	0.97149 (5)	0.0169 (2)
H21	0.589271	0.919297	0.994759	0.020*
C22	0.79145 (12)	1.05599 (8)	1.00415 (5)	0.0165 (2)
C23	0.92738 (12)	1.09775 (9)	0.97242 (5)	0.0169 (2)
C24	0.93343 (12)	1.01687 (8)	0.90620 (5)	0.0163 (2)
C25	1.06560 (12)	1.04859 (9)	0.87253 (5)	0.0171 (2)
H25	1.151927	1.121916	0.893741	0.020*
C26	1.06891 (12)	0.97381 (9)	0.80964 (6)	0.0171 (2)
C27	0.94327 (12)	0.86402 (9)	0.77466 (5)	0.01634 (19)

C28	0.81318 (12)	0.83293 (8)	0.80764 (5)	0.0168 (2)
H28	0.726120	0.760302	0.785211	0.020*
C29	0.80791 (12)	0.90726 (8)	0.87388 (5)	0.01571 (19)
C30	0.77508 (12)	1.12812 (9)	1.07417 (6)	0.0180 (2)
C31	0.55239 (12)	0.75514 (9)	0.87923 (6)	0.0191 (2)
H31A	0.514984	0.737813	0.825853	0.023*
H31B	0.460126	0.752590	0.902969	0.023*
C32	0.60900 (13)	0.66285 (9)	0.89173 (6)	0.0234 (2)
H32A	0.524286	0.586020	0.866974	0.035*
H32B	0.633711	0.674000	0.944539	0.035*
H32C	0.705539	0.668964	0.871761	0.035*
C33	1.08165 (12)	0.74969 (9)	0.71392 (6)	0.0193 (2)
H33A	1.047393	0.684652	0.734735	0.023*
H33B	1.180674	0.811459	0.747120	0.023*
C34	1.11428 (12)	0.70845 (9)	0.63836 (6)	0.0206 (2)
H34A	1.158144	0.775297	0.619813	0.025*
H34B	1.195547	0.674305	0.641710	0.025*
C35	0.82914 (13)	0.65866 (9)	0.58539 (6)	0.0195 (2)
H35A	0.730063	0.592981	0.555266	0.023*
H35B	0.850732	0.721307	0.562431	0.023*
C36	0.80354 (12)	0.70219 (9)	0.66243 (6)	0.0184 (2)
H36A	0.717377	0.732118	0.660625	0.022*
H36B	0.770777	0.637893	0.683949	0.022*
O7	0.40275 (12)	0.71962 (9)	1.02567 (5)	0.0453 (3)
O8	0.18454 (10)	0.59661 (7)	1.04995 (5)	0.0324 (2)
N7	0.16228 (11)	0.77033 (8)	1.15200 (5)	0.01979 (18)
N8	0.15170 (11)	0.85107 (8)	1.20891 (5)	0.02063 (18)
C40	0.29378 (13)	0.79270 (9)	1.12391 (6)	0.0210 (2)
C41	0.42495 (13)	0.89960 (10)	1.15102 (6)	0.0243 (2)
H41	0.516498	0.914854	1.129385	0.029*
C42	0.41628 (13)	0.98111 (9)	1.20973 (6)	0.0247 (2)
H42	0.502470	1.054462	1.231155	0.030*
C43	0.27586 (13)	0.95225 (9)	1.23692 (6)	0.0227 (2)
H43	0.268519	1.008091	1.277777	0.027*
C44	0.29505 (14)	0.69461 (10)	1.06044 (6)	0.0274 (2)
O9	1.15796 (12)	0.79246 (7)	0.47867 (5)	0.0333 (2)
O10	1.03207 (10)	0.59978 (7)	0.45318 (4)	0.02688 (18)
N9	1.17088 (11)	0.55982 (8)	0.33985 (5)	0.01952 (18)
N10	1.22934 (11)	0.53331 (8)	0.27920 (5)	0.02252 (19)
C50	1.19622 (12)	0.66957 (9)	0.37419 (6)	0.0184 (2)
C51	1.28665 (13)	0.76160 (9)	0.35105 (6)	0.0207 (2)
H51	1.304587	0.839401	0.376648	0.025*
C52	1.34837 (13)	0.73528 (10)	0.29006 (6)	0.0222 (2)
H52	1.412129	0.794308	0.272192	0.027*
C53	1.31435 (13)	0.61876 (10)	0.25507 (6)	0.0225 (2)
H53	1.353835	0.599347	0.211819	0.027*
C54	1.12238 (13)	0.69055 (9)	0.44173 (6)	0.0214 (2)
H2	0.536 (2)	0.2757 (16)	0.4172 (10)	0.052 (5)*

H3A	0.0276 (18)	0.3844 (13)	0.9055 (8)	0.032 (4)*
H3B	0.144 (2)	0.4966 (15)	0.9675 (10)	0.047 (4)*
H5A	0.966 (2)	1.2374 (15)	1.0694 (10)	0.053 (5)*
H6A	0.9365 (17)	0.5543 (13)	0.5961 (8)	0.027 (3)*
H6B	0.9885 (19)	0.6087 (14)	0.5361 (9)	0.043 (4)*

Atomic displacement parameters ($\text{\AA}^2$)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0344 (3)	0.0128 (3)	0.0232 (3)	0.0077 (3)	0.0105 (3)	0.0033 (2)
O1	0.0236 (4)	0.0173 (3)	0.0216 (4)	0.0066 (3)	0.0102 (3)	0.0094 (3)
O2	0.0249 (4)	0.0210 (4)	0.0202 (4)	0.0080 (3)	0.0107 (3)	0.0088 (3)
O3	0.0261 (4)	0.0187 (4)	0.0194 (4)	0.0086 (3)	0.0077 (3)	0.0042 (3)
N1	0.0191 (4)	0.0135 (4)	0.0162 (4)	0.0051 (3)	0.0054 (3)	0.0055 (3)
N2	0.0187 (4)	0.0169 (4)	0.0160 (4)	0.0063 (3)	0.0060 (3)	0.0043 (3)
N3	0.0228 (5)	0.0192 (4)	0.0206 (5)	0.0017 (4)	0.0105 (4)	0.0013 (4)
C1	0.0178 (4)	0.0152 (5)	0.0159 (5)	0.0064 (4)	0.0036 (4)	0.0040 (4)
C2	0.0154 (4)	0.0172 (5)	0.0162 (5)	0.0057 (4)	0.0036 (4)	0.0053 (4)
C3	0.0143 (4)	0.0166 (5)	0.0183 (5)	0.0048 (4)	0.0030 (4)	0.0069 (4)
C4	0.0152 (4)	0.0157 (5)	0.0168 (5)	0.0046 (4)	0.0029 (4)	0.0051 (4)
C5	0.0183 (5)	0.0152 (5)	0.0188 (5)	0.0040 (4)	0.0045 (4)	0.0067 (4)
C6	0.0196 (5)	0.0119 (4)	0.0200 (5)	0.0050 (4)	0.0037 (4)	0.0023 (4)
C7	0.0150 (4)	0.0176 (5)	0.0159 (5)	0.0050 (4)	0.0035 (4)	0.0043 (4)
C8	0.0167 (4)	0.0159 (5)	0.0172 (5)	0.0042 (4)	0.0046 (4)	0.0062 (4)
C9	0.0150 (4)	0.0148 (5)	0.0165 (5)	0.0051 (4)	0.0028 (4)	0.0044 (4)
C10	0.0147 (4)	0.0187 (5)	0.0165 (5)	0.0046 (4)	0.0032 (4)	0.0053 (4)
C11	0.0258 (5)	0.0139 (5)	0.0186 (5)	0.0068 (4)	0.0089 (4)	0.0074 (4)
C12	0.0248 (5)	0.0192 (5)	0.0263 (6)	0.0032 (4)	0.0084 (4)	0.0098 (4)
C13	0.0192 (5)	0.0183 (5)	0.0220 (5)	0.0068 (4)	0.0060 (4)	0.0047 (4)
C14	0.0235 (5)	0.0173 (5)	0.0258 (6)	0.0055 (4)	0.0107 (4)	0.0027 (4)
C15	0.0244 (5)	0.0214 (5)	0.0175 (5)	0.0042 (4)	0.0064 (4)	0.0050 (4)
C16	0.0224 (5)	0.0178 (5)	0.0174 (5)	0.0064 (4)	0.0068 (4)	0.0054 (4)
F2	0.0194 (3)	0.0228 (3)	0.0206 (3)	0.0032 (2)	0.0096 (2)	0.0078 (2)
O4	0.0220 (4)	0.0165 (3)	0.0194 (4)	0.0007 (3)	0.0041 (3)	0.0024 (3)
O5	0.0262 (4)	0.0173 (4)	0.0194 (4)	0.0038 (3)	0.0053 (3)	0.0012 (3)
O6	0.0220 (4)	0.0228 (4)	0.0195 (4)	0.0078 (3)	0.0072 (3)	0.0042 (3)
N4	0.0152 (4)	0.0149 (4)	0.0160 (4)	0.0033 (3)	0.0041 (3)	0.0037 (3)
N5	0.0167 (4)	0.0173 (4)	0.0164 (4)	0.0065 (3)	0.0048 (3)	0.0042 (3)
N6	0.0237 (4)	0.0157 (4)	0.0204 (4)	0.0074 (4)	0.0094 (4)	0.0057 (3)
C21	0.0156 (4)	0.0190 (5)	0.0161 (5)	0.0063 (4)	0.0041 (4)	0.0058 (4)
C22	0.0175 (5)	0.0165 (5)	0.0147 (5)	0.0061 (4)	0.0028 (4)	0.0045 (4)
C23	0.0173 (5)	0.0160 (5)	0.0163 (5)	0.0048 (4)	0.0013 (4)	0.0062 (4)
C24	0.0171 (5)	0.0159 (5)	0.0159 (5)	0.0054 (4)	0.0030 (4)	0.0067 (4)
C25	0.0167 (5)	0.0151 (5)	0.0170 (5)	0.0025 (4)	0.0022 (4)	0.0067 (4)
C26	0.0159 (4)	0.0191 (5)	0.0183 (5)	0.0058 (4)	0.0065 (4)	0.0097 (4)
C27	0.0185 (5)	0.0168 (5)	0.0150 (5)	0.0074 (4)	0.0036 (4)	0.0064 (4)
C28	0.0172 (5)	0.0144 (4)	0.0174 (5)	0.0045 (4)	0.0031 (4)	0.0051 (4)
C29	0.0153 (4)	0.0161 (5)	0.0164 (5)	0.0055 (4)	0.0039 (4)	0.0069 (4)

C30	0.0192 (5)	0.0178 (5)	0.0168 (5)	0.0077 (4)	0.0021 (4)	0.0050 (4)
C31	0.0151 (4)	0.0162 (5)	0.0193 (5)	0.0003 (4)	0.0047 (4)	0.0021 (4)
C32	0.0223 (5)	0.0166 (5)	0.0257 (5)	0.0011 (4)	0.0050 (4)	0.0066 (4)
C33	0.0178 (5)	0.0204 (5)	0.0207 (5)	0.0080 (4)	0.0055 (4)	0.0067 (4)
C34	0.0195 (5)	0.0192 (5)	0.0237 (5)	0.0069 (4)	0.0092 (4)	0.0068 (4)
C35	0.0220 (5)	0.0179 (5)	0.0177 (5)	0.0076 (4)	0.0046 (4)	0.0041 (4)
C36	0.0173 (5)	0.0179 (5)	0.0181 (5)	0.0059 (4)	0.0045 (4)	0.0036 (4)
O7	0.0377 (5)	0.0459 (6)	0.0320 (5)	−0.0018 (4)	0.0214 (4)	−0.0013 (4)
O8	0.0347 (5)	0.0240 (4)	0.0257 (4)	0.0009 (4)	0.0143 (4)	−0.0015 (3)
N7	0.0217 (4)	0.0183 (4)	0.0162 (4)	0.0044 (3)	0.0035 (3)	0.0052 (3)
N8	0.0234 (4)	0.0189 (4)	0.0183 (4)	0.0078 (4)	0.0022 (3)	0.0051 (3)
C40	0.0217 (5)	0.0216 (5)	0.0156 (5)	0.0029 (4)	0.0035 (4)	0.0071 (4)
C41	0.0211 (5)	0.0253 (6)	0.0218 (5)	0.0015 (4)	0.0031 (4)	0.0108 (4)
C42	0.0246 (5)	0.0176 (5)	0.0243 (5)	0.0011 (4)	−0.0031 (4)	0.0072 (4)
C43	0.0270 (5)	0.0182 (5)	0.0205 (5)	0.0086 (4)	−0.0010 (4)	0.0046 (4)
C44	0.0255 (5)	0.0296 (6)	0.0181 (5)	0.0023 (5)	0.0076 (4)	0.0032 (4)
O9	0.0503 (5)	0.0188 (4)	0.0266 (4)	0.0105 (4)	0.0141 (4)	0.0018 (3)
O10	0.0360 (4)	0.0202 (4)	0.0244 (4)	0.0085 (3)	0.0155 (3)	0.0068 (3)
N9	0.0220 (4)	0.0180 (4)	0.0184 (4)	0.0069 (3)	0.0064 (3)	0.0060 (3)
N10	0.0263 (5)	0.0222 (5)	0.0216 (4)	0.0106 (4)	0.0094 (4)	0.0076 (4)
C50	0.0191 (5)	0.0171 (5)	0.0171 (5)	0.0056 (4)	0.0015 (4)	0.0051 (4)
C51	0.0219 (5)	0.0166 (5)	0.0208 (5)	0.0049 (4)	0.0010 (4)	0.0062 (4)
C52	0.0190 (5)	0.0234 (5)	0.0245 (5)	0.0047 (4)	0.0035 (4)	0.0132 (4)
C53	0.0221 (5)	0.0276 (6)	0.0211 (5)	0.0107 (4)	0.0077 (4)	0.0106 (4)
C54	0.0262 (5)	0.0190 (5)	0.0182 (5)	0.0082 (4)	0.0052 (4)	0.0053 (4)

Geometric parameters (Å, °)

F1—C6	1.3604 (11)	N6—C34	1.4878 (14)
O1—C3	1.2663 (12)	N6—H6A	0.882 (15)
O2—C10	1.3325 (12)	N6—H6B	0.993 (17)
O2—H2	0.946 (18)	C21—C22	1.3794 (14)
O3—C10	1.2151 (13)	C21—H21	0.9500
N1—C1	1.3377 (13)	C22—C23	1.4306 (14)
N1—C9	1.4002 (13)	C22—C30	1.4869 (14)
N1—C11	1.4826 (12)	C23—C24	1.4502 (14)
N2—C7	1.4055 (13)	C24—C25	1.4074 (14)
N2—C16	1.4713 (13)	C24—C29	1.4076 (14)
N2—C13	1.4760 (13)	C25—C26	1.3558 (15)
N3—C15	1.4815 (15)	C25—H25	0.9500
N3—C14	1.4866 (15)	C26—C27	1.4157 (14)
N3—H3A	0.921 (16)	C27—C28	1.3833 (14)
N3—H3B	0.987 (17)	C28—C29	1.4063 (14)
C1—C2	1.3769 (14)	C28—H28	0.9500
C1—H1	0.9500	C31—C32	1.5180 (15)
C2—C3	1.4329 (14)	C31—H31A	0.9900
C2—C10	1.4852 (14)	C31—H31B	0.9900
C3—C4	1.4519 (14)	C32—H32A	0.9800

C4—C9	1.4081 (14)	C32—H32B	0.9800
C4—C5	1.4099 (14)	C32—H32C	0.9800
C5—C6	1.3582 (14)	C33—C34	1.5141 (14)
C5—H5	0.9500	C33—H33A	0.9900
C6—C7	1.4167 (14)	C33—H33B	0.9900
C7—C8	1.3859 (14)	C34—H34A	0.9900
C8—C9	1.4040 (14)	C34—H34B	0.9900
C8—H8	0.9500	C35—C36	1.5193 (14)
C11—C12	1.5209 (15)	C35—H35A	0.9900
C11—H11A	0.9900	C35—H35B	0.9900
C11—H11B	0.9900	C36—H36A	0.9900
C12—H12A	0.9800	C36—H36B	0.9900
C12—H12B	0.9800	O7—C44	1.2380 (14)
C12—H12C	0.9800	O8—C44	1.2647 (14)
C13—C14	1.5152 (14)	N7—C40	1.3322 (14)
C13—H13A	0.9900	N7—N8	1.3471 (13)
C13—H13B	0.9900	N8—C43	1.3294 (14)
C14—H14A	0.9900	C40—C41	1.4021 (15)
C14—H14B	0.9900	C40—C44	1.5277 (16)
C15—C16	1.5211 (14)	C41—C42	1.3675 (17)
C15—H15A	0.9900	C41—H41	0.9500
C15—H15B	0.9900	C42—C43	1.3947 (17)
C16—H16A	0.9900	C42—H42	0.9500
C16—H16B	0.9900	C43—H43	0.9500
F2—C26	1.3581 (11)	O9—C54	1.2379 (13)
O4—C23	1.2659 (13)	O10—C54	1.2642 (13)
O5—C30	1.3299 (13)	N9—C50	1.3326 (13)
O5—H5A	0.970 (19)	N9—N10	1.3466 (12)
O6—C30	1.2159 (13)	N10—C53	1.3316 (14)
N4—C21	1.3377 (13)	C50—C51	1.3977 (14)
N4—C29	1.3956 (13)	C50—C54	1.5295 (14)
N4—C31	1.4830 (12)	C51—C52	1.3688 (16)
N5—C27	1.4036 (13)	C51—H51	0.9500
N5—C36	1.4672 (13)	C52—C53	1.3944 (16)
N5—C33	1.4793 (13)	C52—H52	0.9500
N6—C35	1.4828 (13)	C53—H53	0.9500
C10—O2—H2	105.0 (11)	C21—C22—C30	118.84 (9)
C1—N1—C9	120.23 (8)	C23—C22—C30	120.99 (9)
C1—N1—C11	119.03 (8)	O4—C23—C22	123.18 (9)
C9—N1—C11	120.71 (8)	O4—C23—C24	121.17 (9)
C7—N2—C16	116.04 (8)	C22—C23—C24	115.64 (9)
C7—N2—C13	115.01 (8)	C25—C24—C29	118.97 (9)
C16—N2—C13	110.47 (8)	C25—C24—C23	119.77 (9)
C15—N3—C14	112.12 (8)	C29—C24—C23	121.26 (9)
C15—N3—H3A	109.7 (9)	C26—C25—C24	119.57 (9)
C14—N3—H3A	109.6 (9)	C26—C25—H25	120.2
C15—N3—H3B	109.7 (10)	C24—C25—H25	120.2

C14—N3—H3B	105.5 (10)	C25—C26—F2	118.85 (9)
H3A—N3—H3B	110.2 (13)	C25—C26—C27	123.04 (9)
N1—C1—C2	123.24 (9)	F2—C26—C27	118.07 (9)
N1—C1—H1	118.4	C28—C27—N5	123.62 (9)
C2—C1—H1	118.4	C28—C27—C26	117.39 (9)
C1—C2—C3	120.57 (9)	N5—C27—C26	118.94 (9)
C1—C2—C10	118.39 (9)	C27—C28—C29	120.97 (9)
C3—C2—C10	121.03 (9)	C27—C28—H28	119.5
O1—C3—C2	122.70 (9)	C29—C28—H28	119.5
O1—C3—C4	121.68 (9)	N4—C29—C28	120.83 (9)
C2—C3—C4	115.62 (9)	N4—C29—C24	119.15 (9)
C9—C4—C5	118.69 (9)	C28—C29—C24	120.01 (9)
C9—C4—C3	120.98 (9)	O6—C30—O5	121.39 (9)
C5—C4—C3	120.29 (9)	O6—C30—C22	123.50 (9)
C6—C5—C4	119.57 (9)	O5—C30—C22	115.11 (9)
C6—C5—H5	120.2	N4—C31—C32	112.58 (8)
C4—C5—H5	120.2	N4—C31—H31A	109.1
C5—C6—F1	118.68 (9)	C32—C31—H31A	109.1
C5—C6—C7	123.23 (9)	N4—C31—H31B	109.1
F1—C6—C7	118.02 (9)	C32—C31—H31B	109.1
C8—C7—N2	123.12 (9)	H31A—C31—H31B	107.8
C8—C7—C6	116.95 (9)	C31—C32—H32A	109.5
N2—C7—C6	119.82 (9)	C31—C32—H32B	109.5
C7—C8—C9	121.23 (9)	H32A—C32—H32B	109.5
C7—C8—H8	119.4	C31—C32—H32C	109.5
C9—C8—H8	119.4	H32A—C32—H32C	109.5
N1—C9—C8	120.53 (9)	H32B—C32—H32C	109.5
N1—C9—C4	119.27 (9)	N5—C33—C34	109.10 (8)
C8—C9—C4	120.18 (9)	N5—C33—H33A	109.9
O3—C10—O2	121.33 (9)	C34—C33—H33A	109.9
O3—C10—C2	123.60 (9)	N5—C33—H33B	109.9
O2—C10—C2	115.05 (9)	C34—C33—H33B	109.9
N1—C11—C12	112.23 (8)	H33A—C33—H33B	108.3
N1—C11—H11A	109.2	N6—C34—C33	110.96 (8)
C12—C11—H11A	109.2	N6—C34—H34A	109.4
N1—C11—H11B	109.2	C33—C34—H34A	109.4
C12—C11—H11B	109.2	N6—C34—H34B	109.4
H11A—C11—H11B	107.9	C33—C34—H34B	109.4
C11—C12—H12A	109.5	H34A—C34—H34B	108.0
C11—C12—H12B	109.5	N6—C35—C36	111.66 (8)
H12A—C12—H12B	109.5	N6—C35—H35A	109.3
C11—C12—H12C	109.5	C36—C35—H35A	109.3
H12A—C12—H12C	109.5	N6—C35—H35B	109.3
H12B—C12—H12C	109.5	C36—C35—H35B	109.3
N2—C13—C14	110.37 (9)	H35A—C35—H35B	107.9
N2—C13—H13A	109.6	N5—C36—C35	108.86 (8)
C14—C13—H13A	109.6	N5—C36—H36A	109.9
N2—C13—H13B	109.6	C35—C36—H36A	109.9

C14—C13—H13B	109.6	N5—C36—H36B	109.9
H13A—C13—H13B	108.1	C35—C36—H36B	109.9
N3—C14—C13	110.92 (9)	H36A—C36—H36B	108.3
N3—C14—H14A	109.5	C40—N7—N8	120.36 (9)
C13—C14—H14A	109.5	C43—N8—N7	118.60 (9)
N3—C14—H14B	109.5	N7—C40—C41	122.35 (10)
C13—C14—H14B	109.5	N7—C40—C44	116.27 (9)
H14A—C14—H14B	108.0	C41—C40—C44	121.38 (10)
N3—C15—C16	110.98 (9)	C42—C41—C40	117.56 (10)
N3—C15—H15A	109.4	C42—C41—H41	121.2
C16—C15—H15A	109.4	C40—C41—H41	121.2
N3—C15—H15B	109.4	C41—C42—C43	117.34 (10)
C16—C15—H15B	109.4	C41—C42—H42	121.3
H15A—C15—H15B	108.0	C43—C42—H42	121.3
N2—C16—C15	109.52 (8)	N8—C43—C42	123.75 (10)
N2—C16—H16A	109.8	N8—C43—H43	118.1
C15—C16—H16A	109.8	C42—C43—H43	118.1
N2—C16—H16B	109.8	O7—C44—O8	127.75 (11)
C15—C16—H16B	109.8	O7—C44—C40	116.98 (10)
H16A—C16—H16B	108.2	O8—C44—C40	115.27 (10)
C30—O5—H5A	105.7 (11)	C50—N9—N10	120.27 (9)
C21—N4—C29	120.07 (8)	C53—N10—N9	118.77 (9)
C21—N4—C31	119.43 (8)	N9—C50—C51	122.53 (10)
C29—N4—C31	120.42 (8)	N9—C50—C54	116.33 (9)
C27—N5—C36	116.51 (8)	C51—C50—C54	121.14 (9)
C27—N5—C33	114.37 (8)	C52—C51—C50	117.41 (10)
C36—N5—C33	110.41 (8)	C52—C51—H51	121.3
C35—N6—C34	113.11 (8)	C50—C51—H51	121.3
C35—N6—H6A	109.5 (9)	C51—C52—C53	117.65 (10)
C34—N6—H6A	110.6 (9)	C51—C52—H52	121.2
C35—N6—H6B	107.0 (9)	C53—C52—H52	121.2
C34—N6—H6B	107.2 (10)	N10—C53—C52	123.33 (10)
H6A—N6—H6B	109.3 (13)	N10—C53—H53	118.3
N4—C21—C22	123.64 (9)	C52—C53—H53	118.3
N4—C21—H21	118.2	O9—C54—O10	127.80 (10)
C22—C21—H21	118.2	O9—C54—C50	117.26 (9)
C21—C22—C23	120.11 (9)	O10—C54—C50	114.93 (9)
C9—N1—C1—C2	−0.45 (15)	C29—C24—C25—C26	0.55 (14)
C11—N1—C1—C2	177.66 (9)	C23—C24—C25—C26	−178.39 (9)
N1—C1—C2—C3	−1.17 (15)	C24—C25—C26—F2	178.53 (8)
N1—C1—C2—C10	179.55 (9)	C24—C25—C26—C27	0.93 (15)
C1—C2—C3—O1	−179.84 (9)	C36—N5—C27—C28	16.86 (14)
C10—C2—C3—O1	−0.57 (15)	C33—N5—C27—C28	−113.97 (11)
C1—C2—C3—C4	0.37 (14)	C36—N5—C27—C26	−160.74 (9)
C10—C2—C3—C4	179.64 (9)	C33—N5—C27—C26	68.44 (12)
O1—C3—C4—C9	−177.82 (9)	C25—C26—C27—C28	−0.77 (15)
C2—C3—C4—C9	1.98 (14)	F2—C26—C27—C28	−178.38 (8)

O1—C3—C4—C5	0.19 (15)	C25—C26—C27—N5	176.98 (9)
C2—C3—C4—C5	179.98 (9)	F2—C26—C27—N5	−0.63 (14)
C9—C4—C5—C6	2.37 (15)	N5—C27—C28—C29	−178.52 (9)
C3—C4—C5—C6	−175.69 (9)	C26—C27—C28—C29	−0.89 (14)
C4—C5—C6—F1	177.88 (9)	C21—N4—C29—C28	−179.37 (9)
C4—C5—C6—C7	1.05 (16)	C31—N4—C29—C28	3.90 (14)
C16—N2—C7—C8	16.02 (14)	C21—N4—C29—C24	1.38 (14)
C13—N2—C7—C8	−115.08 (11)	C31—N4—C29—C24	−175.35 (9)
C16—N2—C7—C6	−160.06 (9)	C27—C28—C29—N4	−176.89 (9)
C13—N2—C7—C6	68.84 (12)	C27—C28—C29—C24	2.35 (15)
C5—C6—C7—C8	−3.70 (15)	C25—C24—C29—N4	177.10 (9)
F1—C6—C7—C8	179.45 (9)	C23—C24—C29—N4	−3.98 (14)
C5—C6—C7—N2	172.62 (9)	C25—C24—C29—C28	−2.15 (14)
F1—C6—C7—N2	−4.24 (14)	C23—C24—C29—C28	176.77 (9)
N2—C7—C8—C9	−173.25 (9)	C21—C22—C30—O6	0.10 (15)
C6—C7—C8—C9	2.93 (15)	C23—C22—C30—O6	−177.26 (9)
C1—N1—C9—C8	−175.75 (9)	C21—C22—C30—O5	179.27 (9)
C11—N1—C9—C8	6.16 (14)	C23—C22—C30—O5	1.91 (14)
C1—N1—C9—C4	2.81 (14)	C21—N4—C31—C32	−104.23 (11)
C11—N1—C9—C4	−175.27 (9)	C29—N4—C31—C32	72.52 (12)
C7—C8—C9—N1	178.89 (9)	C27—N5—C33—C34	−163.21 (8)
C7—C8—C9—C4	0.33 (15)	C36—N5—C33—C34	63.05 (10)
C5—C4—C9—N1	178.38 (9)	C35—N6—C34—C33	51.15 (11)
C3—C4—C9—N1	−3.58 (14)	N5—C33—C34—N6	−56.11 (11)
C5—C4—C9—C8	−3.05 (14)	C34—N6—C35—C36	−50.97 (11)
C3—C4—C9—C8	174.99 (9)	C27—N5—C36—C35	165.01 (8)
C1—C2—C10—O3	−5.51 (15)	C33—N5—C36—C35	−62.34 (10)
C3—C2—C10—O3	175.21 (9)	N6—C35—C36—N5	55.62 (11)
C1—C2—C10—O2	172.92 (9)	C40—N7—N8—C43	−1.04 (14)
C3—C2—C10—O2	−6.36 (14)	N8—N7—C40—C41	−0.61 (16)
C1—N1—C11—C12	−104.20 (11)	N8—N7—C40—C44	178.88 (9)
C9—N1—C11—C12	73.91 (12)	N7—C40—C41—C42	1.87 (16)
C7—N2—C13—C14	−166.31 (8)	C44—C40—C41—C42	−177.60 (10)
C16—N2—C13—C14	59.97 (11)	C40—C41—C42—C43	−1.44 (16)
C15—N3—C14—C13	53.04 (12)	N7—N8—C43—C42	1.44 (15)
N2—C13—C14—N3	−55.53 (11)	C41—C42—C43—N8	−0.16 (16)
C14—N3—C15—C16	−53.94 (11)	N7—C40—C44—O7	168.04 (11)
C7—N2—C16—C15	166.44 (8)	C41—C40—C44—O7	−12.46 (17)
C13—N2—C16—C15	−60.36 (11)	N7—C40—C44—O8	−12.10 (16)
N3—C15—C16—N2	57.09 (11)	C41—C40—C44—O8	167.41 (11)
C29—N4—C21—C22	1.70 (15)	C50—N9—N10—C53	0.93 (15)
C31—N4—C21—C22	178.46 (9)	N10—N9—C50—C51	−1.85 (15)
N4—C21—C22—C23	−2.14 (15)	N10—N9—C50—C54	178.92 (9)
N4—C21—C22—C30	−179.52 (9)	N9—C50—C51—C52	0.85 (16)
C21—C22—C23—O4	−179.58 (9)	C54—C50—C51—C52	−179.96 (9)
C30—C22—C23—O4	−2.26 (15)	C50—C51—C52—C53	0.96 (15)
C21—C22—C23—C24	−0.46 (14)	N9—N10—C53—C52	0.96 (16)
C30—C22—C23—C24	176.87 (8)	C51—C52—C53—N10	−1.90 (16)

O4—C23—C24—C25	1.50 (15)	N9—C50—C54—O9	172.86 (10)
C22—C23—C24—C25	−177.65 (9)	C51—C50—C54—O9	−6.38 (15)
O4—C23—C24—C29	−177.42 (9)	N9—C50—C54—O10	−6.61 (14)
C22—C23—C24—C29	3.44 (14)	C51—C50—C54—O10	174.15 (10)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O2—H2 $\cdots$ O1	0.946 (18)	1.610 (18)	2.5174 (10)	159 (2)
O5—H5A $\cdots$ O4	0.970 (19)	1.592 (18)	2.5208 (11)	159 (2)
N3—H3B $\cdots$ O8	0.987 (17)	1.685 (18)	2.6682 (12)	174 (2)
N6—H6B $\cdots$ O10	0.993 (17)	1.668 (17)	2.6591 (12)	176 (2)
N3—H3A $\cdots$ O8 ⁱ	0.921 (16)	2.254 (15)	2.8557 (13)	122 (1)
N3—H3A $\cdots$ N7 ⁱ	0.921 (16)	2.072 (15)	2.9552 (13)	160 (1)
N6—H6A $\cdots$ O10 ⁱⁱ	0.882 (15)	2.112 (14)	2.7721 (12)	131 (1)
N6—H6A $\cdots$ N9 ⁱⁱ	0.882 (15)	2.194 (15)	3.0060 (13)	153 (1)
C11—H11B $\cdots$ O9 ⁱⁱⁱ	0.99	2.46	3.2870 (13)	141
C11—H11A $\cdots$ F2 ⁱⁱⁱ	0.99	2.37	3.2083 (12)	142
C16—H16B $\cdots$ N8 ⁱ	0.99	2.50	3.4430 (14)	158
C36—H36B $\cdots$ N10 ⁱⁱ	0.99	2.49	3.4388 (14)	160
C41—H41 $\cdots$ O6	0.95	2.45	3.1048 (13)	126
C51—H51 $\cdots$ O3 ^{iv}	0.95	2.31	3.1186 (13)	143

Symmetry codes: (i) $-x, -y+1, -z+2$; (ii) $-x+2, -y+1, -z+1$; (iii) $x-1, y-1, z$; (iv) $x+1, y+1, z$.

4-(3-Carboxy-1-ethyl-6-fluoro-4-oxo-1,4-dihydroquinolin-7-yl)piperazin-1-ium pyrimidine-5-carboxylate monohydrate (3)

Crystal data

$\text{C}_{16}\text{H}_{19}\text{FN}_3\text{O}_3^+ \cdot \text{C}_5\text{H}_3\text{N}_2\text{O}_2^- \cdot \text{H}_2\text{O}$

$M_r = 461.45$

Triclinic, $P\bar{1}$

$a = 8.3657$ (1) Å

$b = 9.5115$ (1) Å

$c = 13.6068$ (2) Å

$\alpha = 76.490$ (1)°

$\beta = 82.771$ (1)°

$\gamma = 84.520$ (1)°

$V = 1041.92$ (2) Å³

$Z = 2$

$F(000) = 484$

$D_x = 1.471$ Mg m^{−3}

Cu $K\alpha$ radiation, $\lambda = 1.54184$ Å

Cell parameters from 22879 reflections

$\theta = 3.3\text{--}76.5^\circ$

$\mu = 0.98$ mm^{−1}

$T = 100$ K

Plate, colourless

$0.22 \times 0.17 \times 0.04$ mm

Data collection

Rigaku, XtaLAB Synergy, Dualflex, HyPix diffractometer

Radiation source: micro-focus sealed X-ray tube, PhotonJet (Cu) X-ray Source

Mirror monochromator

Detector resolution: 10.0000 pixels mm^{−1}

ω scans

Absorption correction: gaussian (CrysAlisPro; Rigaku OD, 2025)

$T_{\min} = 0.660$, $T_{\max} = 1.000$

31025 measured reflections

4184 independent reflections

3922 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.023$

$\theta_{\max} = 77.0^\circ$, $\theta_{\min} = 3.4^\circ$

$h = -10 \rightarrow 10$

$k = -11 \rightarrow 12$

$l = -17 \rightarrow 16$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.097$ $S = 1.07$

4184 reflections

319 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Hydrogen site location: mixed

H atoms treated by a mixture of independent
and constrained refinement $w = 1/[\sigma^2(F_o^2) + (0.043P)^2 + 0.5856P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.31 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.22 \text{ e } \text{\AA}^{-3}$ *Special details*

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ($\text{\AA}^2$)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	1.00744 (10)	0.70135 (9)	0.46524 (6)	0.0275 (2)
O1	0.65784 (11)	1.12025 (10)	0.28850 (7)	0.0207 (2)
O2	0.49367 (12)	1.35975 (11)	0.25190 (7)	0.0245 (2)
O3	0.43857 (14)	1.48539 (11)	0.37189 (8)	0.0320 (3)
N1	0.61538 (13)	1.14104 (11)	0.58840 (8)	0.0167 (2)
N2	0.96501 (13)	0.69701 (11)	0.67052 (8)	0.0171 (2)
N3	1.06829 (14)	0.46490 (12)	0.83098 (9)	0.0204 (2)
C1	0.55431 (15)	1.24892 (13)	0.51833 (10)	0.0183 (3)
H1	0.498024	1.329758	0.540074	0.022*
C2	0.56856 (15)	1.24917 (13)	0.41660 (10)	0.0176 (3)
C3	0.65135 (15)	1.12979 (13)	0.38021 (9)	0.0164 (2)
C4	0.72927 (14)	1.01952 (13)	0.45551 (9)	0.0154 (2)
C5	0.82828 (15)	0.90495 (13)	0.42572 (10)	0.0184 (3)
H5	0.843295	0.898475	0.356497	0.022*
C6	0.90209 (15)	0.80378 (13)	0.49700 (10)	0.0186 (3)
C7	0.88033 (15)	0.80219 (13)	0.60212 (9)	0.0161 (2)
C8	0.78058 (15)	0.91432 (13)	0.63112 (9)	0.0160 (2)
H8	0.760466	0.916324	0.701082	0.019*
C9	0.70881 (14)	1.02496 (13)	0.55868 (9)	0.0152 (2)
C10	0.49419 (16)	1.37499 (14)	0.34660 (10)	0.0214 (3)
C11	0.58814 (16)	1.15048 (14)	0.69656 (10)	0.0195 (3)
H11A	0.562787	1.054175	0.739384	0.023*
H11B	0.493945	1.219228	0.705946	0.023*
C12	0.73378 (17)	1.20006 (15)	0.73108 (10)	0.0231 (3)
H12A	0.711638	1.203265	0.803005	0.035*
H12B	0.756881	1.296925	0.690519	0.035*
H12C	0.827264	1.132123	0.722081	0.035*
C13	0.94337 (16)	0.54377 (13)	0.67222 (10)	0.0202 (3)
H13A	0.947010	0.531123	0.601772	0.024*

H13B	0.836598	0.516786	0.708405	0.024*
C14	1.07576 (16)	0.44603 (14)	0.72533 (10)	0.0223 (3)
H14A	1.061348	0.343682	0.726032	0.027*
H14B	1.182542	0.471168	0.688327	0.027*
C15	1.08134 (16)	0.61905 (14)	0.83281 (10)	0.0217 (3)
H15A	1.190013	0.648806	0.802018	0.026*
H15B	1.067965	0.629505	0.904143	0.026*
C16	0.95380 (16)	0.71720 (14)	0.77477 (10)	0.0194 (3)
H16A	0.845114	0.694564	0.809733	0.023*
H16B	0.969299	0.819575	0.773344	0.023*
O4	0.85628 (11)	0.18288 (11)	1.00834 (8)	0.0268 (2)
O5	0.78038 (12)	0.40830 (11)	0.92836 (8)	0.0273 (2)
N4	0.36819 (14)	0.09850 (13)	1.08628 (9)	0.0240 (3)
N5	0.30900 (13)	0.29866 (13)	0.95040 (9)	0.0229 (2)
C20	0.57755 (15)	0.24858 (14)	0.99701 (9)	0.0182 (3)
C21	0.52305 (16)	0.13136 (15)	1.07032 (10)	0.0209 (3)
H21	0.598467	0.071528	1.111149	0.025*
C22	0.26994 (16)	0.18287 (15)	1.02331 (10)	0.0225 (3)
H22	0.161053	0.157882	1.031252	0.027*
C23	0.46306 (16)	0.33280 (14)	0.93893 (10)	0.0218 (3)
H23	0.494853	0.416807	0.889780	0.026*
C24	0.75307 (16)	0.28187 (14)	0.97704 (10)	0.0199 (3)
O1W	0.92942 (15)	−0.08187 (14)	1.15420 (9)	0.0391 (3)
H2	0.552 (3)	1.268 (3)	0.2514 (17)	0.060 (7)*
H3A	0.965 (3)	0.432 (2)	0.8672 (16)	0.045 (5)*
H3B	1.151 (2)	0.407 (2)	0.8647 (14)	0.036 (5)*
H1W	0.887 (3)	0.007 (3)	1.1247 (17)	0.049 (6)*
H2W	1.003 (3)	−0.115 (3)	1.105 (2)	0.077 (8)*

Atomic displacement parameters ($\text{\AA}^2$)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0346 (5)	0.0214 (4)	0.0231 (4)	0.0141 (3)	−0.0008 (3)	−0.0059 (3)
O1	0.0254 (5)	0.0195 (5)	0.0166 (4)	0.0002 (4)	−0.0053 (4)	−0.0017 (3)
O2	0.0287 (5)	0.0205 (5)	0.0222 (5)	0.0021 (4)	−0.0094 (4)	0.0013 (4)
O3	0.0421 (6)	0.0182 (5)	0.0362 (6)	0.0111 (4)	−0.0183 (5)	−0.0050 (4)
N1	0.0187 (5)	0.0131 (5)	0.0180 (5)	0.0023 (4)	−0.0038 (4)	−0.0032 (4)
N2	0.0193 (5)	0.0117 (5)	0.0203 (5)	0.0027 (4)	−0.0062 (4)	−0.0027 (4)
N3	0.0166 (5)	0.0171 (5)	0.0246 (6)	0.0022 (4)	−0.0061 (4)	0.0017 (4)
C1	0.0180 (6)	0.0125 (6)	0.0243 (6)	0.0017 (5)	−0.0057 (5)	−0.0030 (5)
C2	0.0172 (6)	0.0137 (6)	0.0211 (6)	−0.0008 (5)	−0.0059 (5)	−0.0007 (5)
C3	0.0155 (6)	0.0141 (6)	0.0192 (6)	−0.0038 (5)	−0.0034 (5)	−0.0009 (5)
C4	0.0146 (6)	0.0134 (6)	0.0181 (6)	−0.0015 (5)	−0.0033 (5)	−0.0019 (5)
C5	0.0208 (6)	0.0165 (6)	0.0177 (6)	−0.0005 (5)	−0.0017 (5)	−0.0040 (5)
C6	0.0193 (6)	0.0135 (6)	0.0223 (6)	0.0037 (5)	−0.0002 (5)	−0.0055 (5)
C7	0.0165 (6)	0.0119 (6)	0.0193 (6)	−0.0014 (5)	−0.0040 (5)	−0.0014 (5)
C8	0.0178 (6)	0.0132 (6)	0.0170 (6)	0.0000 (5)	−0.0035 (5)	−0.0028 (5)
C9	0.0147 (6)	0.0116 (5)	0.0191 (6)	−0.0009 (5)	−0.0023 (5)	−0.0024 (4)

C10	0.0207 (6)	0.0163 (6)	0.0260 (7)	−0.0001 (5)	−0.0086 (5)	0.0002 (5)
C11	0.0219 (6)	0.0168 (6)	0.0188 (6)	0.0045 (5)	−0.0020 (5)	−0.0044 (5)
C12	0.0276 (7)	0.0190 (6)	0.0239 (7)	0.0009 (5)	−0.0062 (5)	−0.0063 (5)
C13	0.0251 (7)	0.0122 (6)	0.0237 (6)	0.0016 (5)	−0.0081 (5)	−0.0029 (5)
C14	0.0232 (7)	0.0157 (6)	0.0261 (7)	0.0046 (5)	−0.0051 (5)	−0.0019 (5)
C15	0.0215 (6)	0.0185 (6)	0.0247 (7)	−0.0012 (5)	−0.0089 (5)	−0.0007 (5)
C16	0.0228 (6)	0.0156 (6)	0.0197 (6)	0.0018 (5)	−0.0076 (5)	−0.0025 (5)
O4	0.0184 (5)	0.0263 (5)	0.0346 (5)	0.0034 (4)	−0.0064 (4)	−0.0045 (4)
O5	0.0220 (5)	0.0256 (5)	0.0310 (5)	−0.0045 (4)	−0.0044 (4)	0.0018 (4)
N4	0.0203 (6)	0.0260 (6)	0.0234 (6)	−0.0012 (5)	−0.0032 (4)	−0.0006 (5)
N5	0.0184 (5)	0.0243 (6)	0.0249 (6)	0.0022 (4)	−0.0062 (4)	−0.0027 (5)
C20	0.0185 (6)	0.0189 (6)	0.0176 (6)	0.0014 (5)	−0.0029 (5)	−0.0057 (5)
C21	0.0192 (6)	0.0226 (6)	0.0198 (6)	0.0018 (5)	−0.0050 (5)	−0.0022 (5)
C22	0.0177 (6)	0.0261 (7)	0.0233 (6)	−0.0003 (5)	−0.0028 (5)	−0.0050 (5)
C23	0.0202 (6)	0.0202 (6)	0.0235 (6)	0.0010 (5)	−0.0041 (5)	−0.0017 (5)
C24	0.0187 (6)	0.0223 (6)	0.0194 (6)	0.0003 (5)	−0.0042 (5)	−0.0053 (5)
O1W	0.0367 (6)	0.0383 (7)	0.0363 (6)	0.0112 (5)	−0.0060 (5)	−0.0009 (5)

Geometric parameters (Å, °)

F1—C6	1.3600 (14)	C11—H11A	0.9900
O1—C3	1.2663 (16)	C11—H11B	0.9900
O2—C10	1.3306 (17)	C12—H12A	0.9800
O2—H2	0.96 (2)	C12—H12B	0.9800
O3—C10	1.2136 (17)	C12—H12C	0.9800
N1—C1	1.3373 (16)	C13—C14	1.5179 (17)
N1—C9	1.3992 (16)	C13—H13A	0.9900
N1—C11	1.4830 (16)	C13—H13B	0.9900
N2—C7	1.4016 (15)	C14—H14A	0.9900
N2—C16	1.4656 (16)	C14—H14B	0.9900
N2—C13	1.4804 (15)	C15—C16	1.5202 (17)
N3—C14	1.4826 (18)	C15—H15A	0.9900
N3—C15	1.4868 (17)	C15—H15B	0.9900
N3—H3A	0.98 (2)	C16—H16A	0.9900
N3—H3B	0.94 (2)	C16—H16B	0.9900
C1—C2	1.3738 (18)	O4—C24	1.2503 (16)
C1—H1	0.9500	O5—C24	1.2566 (16)
C2—C3	1.4284 (18)	N4—C22	1.3355 (17)
C2—C10	1.4835 (17)	N4—C21	1.3409 (17)
C3—C4	1.4526 (17)	N5—C22	1.3344 (18)
C4—C9	1.4053 (17)	N5—C23	1.3404 (17)
C4—C5	1.4098 (18)	C20—C21	1.3817 (18)
C5—C6	1.3598 (18)	C20—C23	1.3899 (18)
C5—H5	0.9500	C20—C24	1.5093 (18)
C6—C7	1.4156 (18)	C21—H21	0.9500
C7—C8	1.3904 (17)	C22—H22	0.9500
C8—C9	1.4082 (17)	C23—H23	0.9500
C8—H8	0.9500	O1W—H1W	0.91 (2)

C11—C12	1.5115 (18)	O1W—H2W	0.94 (3)
C10—O2—H2	104.1 (14)	C11—C12—H12A	109.5
C1—N1—C9	119.98 (11)	C11—C12—H12B	109.5
C1—N1—C11	118.82 (10)	H12A—C12—H12B	109.5
C9—N1—C11	121.15 (10)	C11—C12—H12C	109.5
C7—N2—C16	116.27 (10)	H12A—C12—H12C	109.5
C7—N2—C13	116.59 (10)	H12B—C12—H12C	109.5
C16—N2—C13	109.80 (10)	N2—C13—C14	110.00 (10)
C14—N3—C15	111.23 (10)	N2—C13—H13A	109.7
C14—N3—H3A	108.0 (12)	C14—C13—H13A	109.7
C15—N3—H3A	109.6 (12)	N2—C13—H13B	109.7
C14—N3—H3B	110.5 (11)	C14—C13—H13B	109.7
C15—N3—H3B	109.7 (11)	H13A—C13—H13B	108.2
H3A—N3—H3B	107.7 (16)	N3—C14—C13	109.20 (11)
N1—C1—C2	123.55 (12)	N3—C14—H14A	109.8
N1—C1—H1	118.2	C13—C14—H14A	109.8
C2—C1—H1	118.2	N3—C14—H14B	109.8
C1—C2—C3	120.34 (11)	C13—C14—H14B	109.8
C1—C2—C10	118.33 (12)	H14A—C14—H14B	108.3
C3—C2—C10	121.32 (12)	N3—C15—C16	111.10 (10)
O1—C3—C2	122.66 (11)	N3—C15—H15A	109.4
O1—C3—C4	121.84 (11)	C16—C15—H15A	109.4
C2—C3—C4	115.51 (11)	N3—C15—H15B	109.4
C9—C4—C5	118.74 (11)	C16—C15—H15B	109.4
C9—C4—C3	121.10 (11)	H15A—C15—H15B	108.0
C5—C4—C3	120.16 (11)	N2—C16—C15	110.39 (11)
C6—C5—C4	119.37 (12)	N2—C16—H16A	109.6
C6—C5—H5	120.3	C15—C16—H16A	109.6
C4—C5—H5	120.3	N2—C16—H16B	109.6
C5—C6—F1	118.22 (11)	C15—C16—H16B	109.6
C5—C6—C7	123.74 (12)	H16A—C16—H16B	108.1
F1—C6—C7	117.99 (11)	C22—N4—C21	115.60 (11)
C8—C7—N2	123.02 (11)	C22—N5—C23	116.45 (11)
C8—C7—C6	116.51 (11)	C21—C20—C23	116.60 (12)
N2—C7—C6	120.34 (11)	C21—C20—C24	122.46 (11)
C7—C8—C9	121.21 (11)	C23—C20—C24	120.89 (12)
C7—C8—H8	119.4	N4—C21—C20	122.77 (12)
C9—C8—H8	119.4	N4—C21—H21	118.6
N1—C9—C4	119.12 (11)	C20—C21—H21	118.6
N1—C9—C8	120.58 (11)	N5—C22—N4	126.69 (12)
C4—C9—C8	120.30 (11)	N5—C22—H22	116.7
O3—C10—O2	121.35 (12)	N4—C22—H22	116.7
O3—C10—C2	123.52 (12)	N5—C23—C20	121.76 (12)
O2—C10—C2	115.12 (11)	N5—C23—H23	119.1
N1—C11—C12	112.01 (11)	C20—C23—H23	119.1
N1—C11—H11A	109.2	O4—C24—O5	126.42 (12)
C12—C11—H11A	109.2	O4—C24—C20	118.08 (12)

N1—C11—H11B	109.2	O5—C24—C20	115.49 (11)
C12—C11—H11B	109.2	H1W—O1W—H2W	108 (2)
H11A—C11—H11B	107.9		
C9—N1—C1—C2	4.53 (19)	C3—C4—C9—N1	−1.82 (17)
C11—N1—C1—C2	−178.12 (11)	C5—C4—C9—C8	−2.70 (18)
N1—C1—C2—C3	0.32 (19)	C3—C4—C9—C8	177.72 (11)
N1—C1—C2—C10	179.51 (11)	C7—C8—C9—N1	−176.73 (11)
C1—C2—C3—O1	174.88 (12)	C7—C8—C9—C4	3.75 (18)
C10—C2—C3—O1	−4.28 (19)	C1—C2—C10—O3	9.3 (2)
C1—C2—C3—C4	−5.52 (17)	C3—C2—C10—O3	−171.56 (13)
C10—C2—C3—C4	175.31 (11)	C1—C2—C10—O2	−171.44 (11)
O1—C3—C4—C9	−174.18 (11)	C3—C2—C10—O2	7.74 (18)
C2—C3—C4—C9	6.22 (17)	C1—N1—C11—C12	−100.30 (13)
O1—C3—C4—C5	6.24 (18)	C9—N1—C11—C12	77.01 (14)
C2—C3—C4—C5	−173.36 (11)	C7—N2—C13—C14	163.70 (11)
C9—C4—C5—C6	−0.51 (18)	C16—N2—C13—C14	−61.36 (13)
C3—C4—C5—C6	179.09 (11)	C15—N3—C14—C13	−56.66 (14)
C4—C5—C6—F1	−174.62 (11)	N2—C13—C14—N3	59.80 (14)
C4—C5—C6—C7	2.9 (2)	C14—N3—C15—C16	54.97 (14)
C16—N2—C7—C8	−4.76 (17)	C7—N2—C16—C15	−166.17 (10)
C13—N2—C7—C8	127.26 (13)	C13—N2—C16—C15	58.74 (13)
C16—N2—C7—C6	170.95 (11)	N3—C15—C16—N2	−55.66 (14)
C13—N2—C7—C6	−57.02 (16)	C22—N4—C21—C20	3.0 (2)
C5—C6—C7—C8	−1.85 (19)	C23—C20—C21—N4	−0.2 (2)
F1—C6—C7—C8	175.64 (11)	C24—C20—C21—N4	−177.82 (12)
C5—C6—C7—N2	−177.83 (12)	C23—N5—C22—N4	0.5 (2)
F1—C6—C7—N2	−0.35 (18)	C21—N4—C22—N5	−3.3 (2)
N2—C7—C8—C9	174.38 (11)	C22—N5—C23—C20	2.6 (2)
C6—C7—C8—C9	−1.49 (18)	C21—C20—C23—N5	−2.8 (2)
C1—N1—C9—C4	−3.67 (17)	C24—C20—C23—N5	174.92 (12)
C11—N1—C9—C4	179.05 (11)	C21—C20—C24—O4	18.43 (19)
C1—N1—C9—C8	176.80 (11)	C23—C20—C24—O4	−159.12 (13)
C11—N1—C9—C8	−0.48 (17)	C21—C20—C24—O5	−162.59 (13)
C5—C4—C9—N1	177.77 (11)	C23—C20—C24—O5	19.87 (18)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
O2—H2 $\cdots$ O1	0.96 (2)	1.60 (3)	2.5234 (13)	160 (2)
N3—H3A $\cdots$ O5	0.98 (2)	1.67 (2)	2.6378 (15)	170 (2)
N3—H3B $\cdots$ N5 ⁱ	0.94 (2)	1.93 (2)	2.8654 (16)	172 (2)
C1—H1 $\cdots$ O3 ⁱⁱ	0.95	2.48	3.2305 (16)	136
C11—H11A $\cdots$ N4 ⁱⁱⁱ	0.99	2.58	3.3630 (17)	136
C13—H13A $\cdots$ F1	0.99	2.20	2.8752 (15)	124

O1 <i>W</i> —H1 <i>W</i> ···O4	0.91 (2)	2.04 (2)	2.8863 (16)	155 (2)
O1 <i>W</i> —H2 <i>W</i> ···O4 ^{iv}	0.94 (3)	2.01 (3)	2.9487 (17)	174 (2)

Symmetry codes: (i) $x+1, y, z$; (ii) $-x+1, -y+3, -z+1$; (iii) $-x+1, -y+1, -z+2$; (iv) $-x+2, -y, -z+2$.