

4,5-Dihydrocyclopenta[*b*]thiophen-6-one

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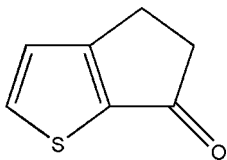
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The title compound crystallizes with two unique but closely similar molecules, 1 and 2, in the asymmetric unit. Both molecules are planar with r.m.s. deviations of only 0.0193 and 0.0107 Å for the planes through the nine non-hydrogen atoms of molecules 1 and 2 respectively. In the crystal structure, $\pi-\pi$ and C–H $\cdots\pi$ contacts stack the molecules into columns in an inverse fashion along the *b* axis. An extensive series of C–H \cdots O hydrogen bonds links the columns generating an extended network structure.

Keywords: thiophenes; hydrogen bonding

Chemical context

The title compound is a key intermediate in the synthesis of many conjugated thiophene vinylene oligomers for low band-gap and fluorescent applications with applications in non-linear optics and LED devices. (Blanchard *et al.*, 1997, 1998*a,b*, 2006). This is because the ethylene bridge in the resulting vinylene system flattens the ring and increases the conjugation between the neighbouring thiophene groups. This technique can also be used to control the bandgap in the corresponding polymers (Roncali *et al.*, 1994).



Structural commentary

The asymmetric unit of the triclinic unit cell of the title compound contains two unique molecules, 1 and 2, Fig. 1. These are closely similar and overlay almost perfectly with an r.m.s. deviation of only 0.024 Å (Macrae *et al.*, 2008). The molecules are approximately planar with r.m.s. deviations of 0.0193 and 0.0107 Å from the best fit planes through the nine non-hydrogen atoms in each of the two molecules. This is

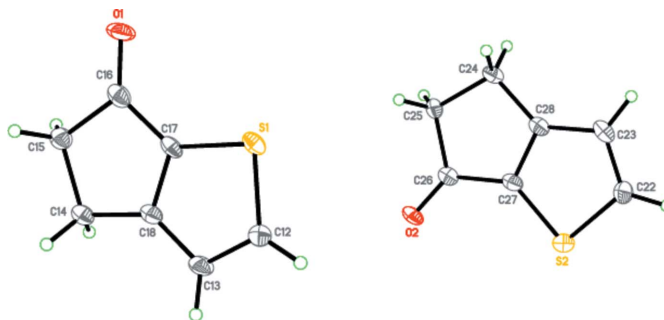


Figure 1

The asymmetric unit with ellipsoids drawn at the 50% probability level.

further illustrated by the fact that the thiophene and cyclopentadienone ring planes are inclined at 2.40 (13)° in molecule 1 and 0.64 (13)° in molecule 2. Bond distances in the molecule are normal (Allen *et al.*, 1987) and similar to those observed in related molecules such as *trans*-4,5-diphenyl-4,5-dihydro-6*H*-cyclopenta[*b*]thiophen-6-one (Chang *et al.*, 2004) or ethyl 5-hydroxy-6-oxo-4-phenyl-5,6-dihydro-4*H*-cyclopenta[*b*]thiophene-5-carboxylate (McAdam *et al.*, 2011).

Supramolecular features

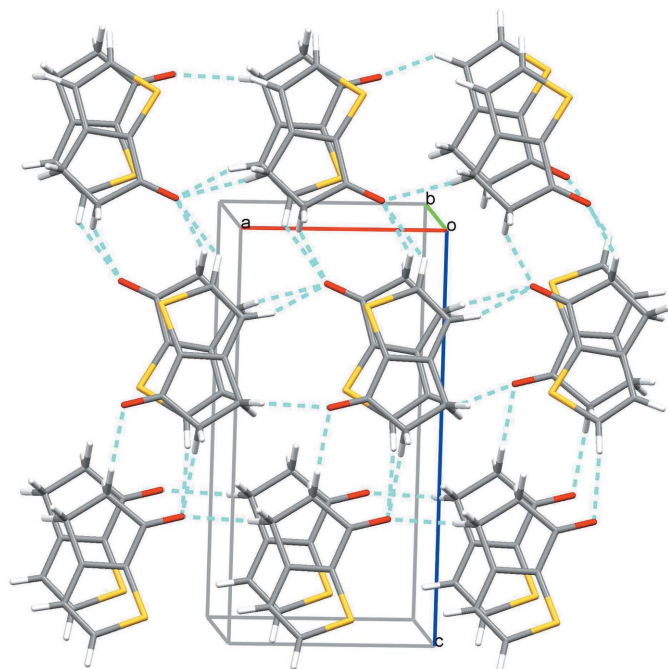
In the crystal structure, molecules 1 and 2 are linked by C15–H15*A* $\cdots\pi$ and C25–H25*B* $\cdots\pi$ contacts into dimers. $\pi-\pi$ contacts of 3.6542 (17) Å, between the centroids of the thio-

Table 1

Experimental details.

Crystal data	
Chemical formula	C ₇ H ₆ OS
<i>M_r</i>	138.18
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	92
<i>a</i> , <i>b</i> , <i>c</i> (Å)	6.6133 (9), 7.4894 (11), 13.3213 (16)
α , β , γ (°)	83.247 (8), 86.097 (7), 71.363 (8)
<i>V</i> (Å ³)	620.54 (14)
<i>Z</i>	4
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.095
Crystal size	0.55 × 0.28 × 0.02
Data collection	
Diffractometer	Bruker APEXII CCD area detector
Absorption correction	Multi-scan
<i>T_{min}</i> , <i>T_{max}</i>	0.617, 0.745
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	5251, 1522, 1264
<i>R_{int}</i>	0.040
($\sin \theta/\lambda$) _{max} (Å ⁻¹)	0.529
Refinement	
$R[F^2 > 2\sigma(F^2)]$, $wR(F^2)$, <i>S</i>	0.040, 0.119, 1.11
No. of reflections	1522
No. of parameters	163
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.30, -0.34

Computer programs used: APEX2, SADABS and SAINT (Bruker 2011), SHELXS97, SHELXL97 and SHELXTL (Sheldrick, 2008), TITAN (Hunter & Simpson, 1999), Mercury (Macrae *et al.*, 2008), enCIFer (Allen *et al.*, 2004), PLATON (Spek, 2009) and publCIF (Westrip 2010).


Figure 2

A general view of the crystal packing for the title compound along the *b* axis with hydrogen bonds drawn as dashed lines.

phene rings of molecules 1 and 2, link these dimers, forming columns along the *b* axis, Fig. 1*S*. An extensive series of C–H···O hydrogen bonds involving the carbonyl O atoms from both molecules join the columns into an extended network, Fig. 2. Interestingly the S atoms are not involved in any close intermolecular interactions.

Synthesis and crystallization

The starting material methyl 6-oxo-5,6-dihydro-4*H*-cyclopenta[*b*]thiophene-5-carboxylate was prepared by literature methods (Cai *et al.* 2002; More & Finney, 2002; Yang, 2009) and 124 mg, 0.63 mmol was dissolved in KOH (0.5 *M*) solution (H₂O/EtOH, *v/v* = 1) and refluxed for 4 h. The solution was cooled to room temperature, extracted with EtOAc (3 × 20 ml), then the combined organic layers were washed with water (2 × 50 ml) and dried over MgSO₄. The solvent was evaporated *in vacuo* to yield transparent gold crystals (52.3 mg, 0.38 mmol) 60% yield. ¹H NMR (δ p.p.m., CDCl₃, 300 Hz): 7.896–7.884 (1*H*, *d*, *J* = 3.6 Hz, C₄H₂S), 7.051–7.039 (1*H*, *d*, *J* = 3.6 Hz, C₄H₂S), 3.049–2.974 (4*H*, *m*, CH₂CH₂).

Refinement

All H-atoms were refined using a riding model with d(C–H) = 0.95 Å, for aromatic and 0.99 Å, for CH₂ H atoms with *U*_{iso} = 1.2*U*_{eq} (C). Crystals were very weakly diffracting and data of reasonable

Table 2

Hydrogen-bond geometry (Å, °).

C*g*1 and C*g*4 are the centroids of the S1, C12, C13, C17, C18 and S2, C22, C23, C27, C28 thiophene rings, respectively.

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C14–H14A···O1 ⁱ	0.99	2.52	3.473 (4)	160
C15–H15B···O1 ⁱⁱ	0.99	2.59	3.489 (3)	151
C12–H12···O2 ⁱⁱⁱ	0.95	2.43	3.370 (4)	168
C13–H13···O2 ^{iv}	0.95	2.65	3.275 (4)	124
C23–H23···O1 ^v	0.95	2.62	3.425 (4)	143
C24–H24A···O2 ^{vi}	0.99	2.52	3.480 (4)	164
C25–H25A···O2 ^{vii}	0.99	2.71	3.704 (3)	180
C25–H25B···C <i>g</i> 1 ^{viii}	0.99	2.79	3.561 (3)	135
C15–H15A···C <i>g</i> 4 ⁱⁱⁱ	0.99	2.84	3.571 (3)	131

Symmetry codes: (i) *x* + 1, *y*, *z*; (ii) *−x* + 1, *−y* + 1, *−z* + 2; (iii) *x*, *y* + 1, *z*; (iv) *−x* + 2, *−y*, *−z* + 1; (v) *−x*, *−y* + 1, *−z* + 1; (vi) *x* − 1, *y*, *z*; (vii) *−x* + 1, *−y*, *−z* + 1; (viii) *x*, *y* − 1, *z*.

intensity could not be obtained beyond $\theta = 22^\circ$. This also contributes to the relatively poor data/parameter ratio observed for this refinement.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: zzyy05). Services for accessing these data are described at the back of the journal.

References

- Allen, F. H., Johnson, O., Shields, G. P., Smith, B. R. & Towler, M. (2004). *J. Appl. Cryst.* **37**, 335–338.
- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.
- Blanchard, P., Brisset, H., Riou, A., Hierle, R. & Roncali, J. (1998*a*). *J. Org. Chem.* **63**, 8310–8319.
- Blanchard, P., Brisset, H., Riou, A., Hierle, R. & Roncali, J. (1998*b*). *New J. Chem.* **22**, 547–549.
- Blanchard, P., Brisset, H., Riou, A., Illien, B., Hierle, R., Riou, A. & Roncali, J. (1997). *J. Org. Chem.* **62**, 2401–2408.
- Blanchard, P., Verlhac, P., Michaux, L., Frere, P. & Roncali, J. (2006). *Chem. Eur. J.* **12**, 1244–1255.
- Bruker (2011). *APEX2*, *SAINTE* and *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cai, J. X., Zhou, Z. H., Zhao, G. F. & Tang, C. C. (2002). *Org. Lett.* **26**, 4723–4725.
- Chang, K.-J., Rayabarapu, D. K. & Cheng, C.-H. (2004). *J. Org. Chem.* **69**, 4781–4787.
- Hunter, K. A. & Simpson, J. (1999). *TITAN2000*. University of Otago, New Zealand.
- Macrae, C. F., Bruno, I. J., Chisholm, J. A., Edgington, P. R., McCabe, P., Pidcock, E., Rodriguez-Monge, L., Taylor, R., van de Streek, J. & Wood, P. A. (2008). *J. Appl. Cryst.* **41**, 466–470.
- McAdam, C. J., Moratti, S. C., Simpson, J. & Shi, Z. (2011). *Acta Cryst.* **E67**, o2764–o2765.
- More, J. D. & Finney, N. S. (2002). *Org. Lett.* **17**, 3001–3003.
- Roncali, J., Thobie-Gautier, C., Elandaloussi, E.-H. & Frere, P. (1994). *J. Chem. Soc., Chem. Commun.* (1994), 2249–2250.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.
- Yang, B. (2009). World patent WO2009158380.