

STRUCTURAL BIOLOGY COMMUNICATIONS

ISSN 2053-230X

Received 19 July 2024 Accepted 20 November 2024

Edited by Y. Chen, MAX IV Laboratory, Sweden

Keywords: negative staining; electron microscopy; single-particle analysis; structural biology; 3D reconstruction; sample preparation; methods development.

EMBD references: β -galactosidase, stained with uranyl formate, EMD-51069; stained with sodium phosphotungstate, EMD-51071; stained with ammonium molybdate, EMD-51072

Supporting information: this article has supporting information at journals.iucr.org/f



COPEN OPEN ACCESS

Revisiting sodium phosphotungstate and ammonium molybdate as nonradioactive negative-staining agents for single-particle analysis

Monika Gunkel,* Arthur Macha and Elmar Behrmann

University of Cologne, Institute of Biochemistry, Faculty of Mathematics and Natural Sciences, Zülpicher Strasse 47, 50674 Cologne, Germany. *Correspondence e-mail: monika.gunkel@uni-koeln.de

This study reports the successful replacement of uranyl-based stains by either sodium phosphotungstate or ammonium molybdate in negative-staining electron microscopy. Using apoferritin as a test specimen, it is demonstrated that in combination with a facile on-grid fixation step, both stains yield comparable images to uranyl formate. Subsequently, using β -galactosidase, it is shown that both stains can also successfully be employed for single-particle analysis, yielding virtually indistinguishable results from uranyl formate. As both replacement stains are nonradioactive, they are not subjected to the same handling restrictions as uranyl-based stains. Therefore they are not only cheaper to use, but also make decentralized sample-grid preparation, directly after purification, accessible to a broader range of scientists.

1. Introduction

In recent years cryo-electron microscopy (cryo-EM) has reached a level of maturity that allows it to routinely complement, and sometimes surpass, the capabilities of X-ray crystallography and NMR to uncover the structural basis of protein activity (Fukuda et al., 2023; Nakane et al., 2020; Yip et al., 2020). As with other structural biology tools, the success of cryo-EM relies on successful sample preparation. Often, this becomes the bottleneck in a structural biology project, requiring extensive and time-consuming optimization steps (Takizawa et al., 2017; Weissenberger et al., 2021). Negative staining is a well established technique to prepare a wide range of samples for electron microscopy (EM) characterization (Bradley, 1962; Brenner & Horne, 1959; De Carlo & Harris, 2011; Harris, 2015). It is a rapid, robust and costefficient approach, especially compared with cryo-EM. While associated with a range of drawbacks, particularly in terms of achievable resolution and preservation of the native structure, negative-stain EM has repeatedly been demonstrated to yield important insights (Fabre et al., 2017; Matsuike et al., 2023; Pramanick et al., 2021; Sasajima et al., 2022). For structural biology applications, negative-stain EM can provide information about general sample quality with regard to particle size, distribution, aggregation tendency, homogeneity and abundance of the protein of interest. Moreover, as only neglectable amounts of sample are needed, negative-stain EM does not require the dedicated production of sample material, but can instead be used with leftover material. Therefore, it lends itself as a diagnostic tool in cases where the interpretation of cryo-EM data sets has raised doubts about the quality of the input sample. Here, the enhanced image contrast due to embedding into a heavy-metal salt film allows challenges posed by the low particle contrast inherent to cryo-EM to be overcome. For research questions that do not require high-resolution structural details, such as the investigation of

conditions that allow the formation of protein complexes, determining protein dimensions and addressing binding sites, negative-staining EM has successfully been used as a standalone technique (Burgess *et al.*, 2004; Fabre *et al.*, 2017; Ha *et al.*, 2016; Francis *et al.*, 2024).

While there is a rich history of chemically diverse staining solutions (Scarff et al., 2018), current negative-staining EM for single-particle analysis (SPA) almost exclusively relies on either uranium acetate (UA; Epstein, 2001; Johnson et al., 1977) or uranium formate (UF; Finch, 1964; Leberman, 1965). Both uranium salts yield excellent contrast due to the high Z number of uranium, and both salts have fine grain sizes, allowing molecular details to be resolved (Ohi et al., 2004). Moreover, the counterions of uranium in these salts, acetate and formate, have the advantage of rapidly fixing biological samples during the staining process, with fixation reported to occur on the millisecond time range (Zhao & Craig, 2003). Still, the widespread use of uranium-based staining solutions not only produces potentially radioactive waste, but has additional drawbacks. Firstly, uranium hydroxide precipitates at neutral and basic pH values, and thus stains are commonly prepared as highly acidic solutions that potentially interfere with the biological sample of interest. Secondly, due to its classification as a radioactive substance, obtaining, handling and disposing of uranium salts can be challenging from a legal and administrative point of view, which often translates into high costs. These administrative burdens also complicate decentralized sample preparation, so that purified protein samples must often be brought to central EM laboratories before they can be stained, which can be an issue for unstable samples. Given these drawbacks, it is not surprising that alternative stains are sought after. However, these studies often focus on the staining and preservation of subcellular structures and not on SPA applications (Benmeradi et al., 2015; Inaga et al., 2007; Nakakoshi et al., 2011; Hosogi et al., 2015; Kuipers & Giepmans, 2020; Scarff et al., 2018).

Here, we report the successful application of two stains that were used for SPA before the advent of UF and UA, namely sodium phosphotungstate (SPT; Brenner & Horne, 1959) and ammonium molybdate (AMo; Bohonek, 1974; Mannella & Frank, 1984). We show that by including a facile on-grid fixation step both SPT and AMo can be used to visualize the 440 kDa capsid assembly of apoferritin. Furthermore, we demonstrate that β -galactosidase, a common test specimen in the field of cryo-EM, can be faithfully reconstructed using both stains.

2. Materials and methods

2.1. Protein expression and purification

Mus musculus apoferritin heavy chain (pET-24a-mmFTH1) was expressed in *Escherichia coli* BL21(DE3) competent cells. *E. coli* BL21(DE3) cells transformed with pET-24a-mmFTH1 were cultured in LB medium (10 g L⁻¹ tryptone, 5 g L⁻¹ yeast extract, 10 g L⁻¹ NaCl) supplemented with 50 μ g mL⁻¹ kanamycin. After reaching an OD₆₀₀ of 0.5, isopropyl β -D-1-

thiogalactopyranoside (IPTG) was added to a final concentration of 1 mM to induce expression. The cells were cultured for a further 4 h at 37°C and then collected by centrifugation at 3000 rev min $^{-1}$ for 15–20 min at 4°C. The cell pellets were resuspended in lysis buffer (20 mM HEPES-NaOH pH 7.5, $300 \text{ m}M \text{ NaCl}, 1 \text{ m}M \text{ MgSO}_4$) supplemented with 100 mg mL^{-1} lysozyme, followed by sonication (power 200 W, amplitude 60%, C 100%, sonotrode diameter 7 mm, 4 s work/8 s pause, 30 cycles) on ice. After clarifying by centrifugation at 70 000g for 30 min at 4°C, the supernatant was heated for 10 min to 70°C and then centrifuged at 70 000g for 30 min at 4°C. 8.4 g ammonium sulfate was added to the second supernatant to a final concentration of 52.5%. The solution was centrifuged at 50 000g for 20 min at 4°C. The supernatant was then discarded and the pellet was resuspended in a final volume of 2 mL sample buffer (20 mM HEPES-NaOH pH 7.5, 300 mM NaCl). The sample was further purified by size-exclusion chromatography on a HiLoad 16/600 Superdex 200 pg gel-filtration column using sample buffer. Fractions containing apoferritin were snap-frozen in liquid nitrogen and stored at -80° C until further use.

E. coli β -galactosidase was purchased from Sigma–Aldrich (catalogue No. G5635-3KU) as lyophilized protein. 2.5 g of the lyophilized protein was dissolved in 250 µL sample buffer (25 m*M* Tris–HCl pH 8.0, 50 m*M* NaCl, 0.5 m*M* TCEP) and aggregates were removed by centrifugation. The sample was further purified by size-exclusion chromatography on a Superdex 200 Increase 10/300 GL column using sample buffer. Fractions containing β -galactosidase were snap-frozen in liquid nitrogen and stored at -80° C until further use.

2.2. Preparation and storage of staining solutions

UF was prepared by dissolving 2%(w/v) UF (Science Services, CAS No. 6984-95-1) in boiling double-distilled water (ddH₂O). The solution was stirring in darkness for 5 min. 6.4 µL 4 *M* NaOH per 1 mL UF solution was then added and stirred for another 5 min in darkness. The solution was filtered through a 0.22 µ*M* filter and 200 µL aliquots were shock-frozen in liquid nitrogen and stored at -80° C.

AMo was prepared by dissolving 1%(w/v) AMo (Sigma-Aldrich, CAS No. 13106-76-8) in ddH₂O. The pH was set to pH 7.0 with ammonium hydroxide. 200 µL aliquots were shock-frozen in liquid nitrogen and stored at -80° C.

SPT was prepared by dissolving 2%(w/v) SPT (Sigma-Aldrich, CAS No. 312696-30-3) in ddH₂O. The pH was set to pH 7.0 with sodium hydroxide. 200 µL aliquots were shock-frozen in liquid nitrogen and stored at -80° C.

For storage recommendations, see the supporting information and Supplementary Fig. S1.

2.3. Negative staining with UF

Continuous carbon grids (Quantifoil, Cu 200 mesh) were subjected to glow discharge using a Zepto Plasma Cleaner (Diener) for 30 s to clean the surface and render it hydrophilic. 3 μ l protein solution, at a concentration of 0.4 mg mL⁻¹ for apoferritin and 2.1 mg mL⁻¹ for β -galactosidase, was

incubated on the grid for approximately 1 min at room temperature; excess protein solution was then blotted away using filter paper, taking care to leave a thin liquid film on the grid. Using self-locking forceps (Dumoxel, N5), the grid was then washed by dipping it onto a droplet of protein buffer and blotting away the buffer using filter paper, taking care to leave a thin liquid film. This step was repeated three times. Next, the grid was briefly dipped onto a droplet of staining solution, followed by immediate blotting of excess liquid with filter paper. The grid was then placed onto fresh droplets of staining solution and incubated for 1 min. Excess stain was carefully removed with filter paper, taking care to leave a thin liquid film on the grid. The remaining stain solution was then rapidly dried using a gentle, indirect stream from a hair dryer without heating. The grids were stored in the dark in a dry atmosphere until further use.

2.4. Negative staining with AMo or SPT

Continuous carbon grids (Quantifoil, Cu 200 mesh) were subjected to glow discharge using a Zepto Plasma Cleaner (Diener) for 30 s to clean the surface and render it hydrophilic. $3 \,\mu\text{L}$ protein solution, at a concentration of 0.4 mg mL⁻¹ for apoferritin and 2.1 mg mL⁻¹ for β -galactosidase, was incubated on the grid for approximately 1 min at room temperature; excess protein solution was then blotted away using filter paper, taking care to leave a thin liquid film on the grid. Using self-locking forceps (Dumoxel, N5), the grid was then washed by dipping it onto a droplet of protein buffer and blotting away the buffer using filter paper, taking care to leave a thin liquid film. This step was repeated three times. Next, the grid was briefly dipped onto a droplet of fixation solution [0.15%(w/v) glutaraldehyde in ddH₂O, Sigma–Aldrich, CAS No. 111-30-8] followed by immediate blotting of excess liquid with filter paper. The grid was then placed onto a fresh droplet of fixation solution and incubated for 5 min. Excess fixation solution was carefully removed with filter paper. Next, the grid was briefly dipped onto a droplet of staining solution, followed by immediate blotting of excess liquid with filter paper. The grid was then placed onto fresh droplets of staining solution and incubated for 1 min. Excess stain was carefully removed with filter paper, taking care to leave a thin liquid film on the grid. The remaining stain solution was then rapidly dried using a gentle, indirect air stream from a hair dryer without heating. Grids were stored in the dark in a dry atmosphere until further use.

For a detailed, step-by-step guide, see the supporting information and Supplementary Fig. S2.

2.5. EM data collection

EM data were acquired using a Talos L120C (Thermo Fisher Scientific) electron microscope equipped with a LaB₆ emitter operated at 120 kV. Images were collected automatically using *EPU* (version 2.12.1.2782REL, Thermo Fisher Scientific) on Ceta16M CMOS detector with a calibrated pixel size of 1.86 Å per pixel. Defocus values were set to range from -0.3 to $-2.0 \,\mu\text{m}$.

2.6. Image processing, particle quantification and data analysis

Image processing was performed using *cryoSPARC* 4.4 (Punjani *et al.*, 2017). Random subsets of 25 micrographs were used for each staining condition to determine the optimal value for the amplitude contrast by comparing experimental and simulated CTF oscillations. Using this identified value (UF, 0.35; AMo, 0.15; SPT, 0.10), defocus and other CTF-related values were then calculated for the complete data sets. Only high-quality micrographs with low astigmatism and good agreement between experimental and calculated CTFs were further processed.

For apoferritin, a total of nine grids were prepared during three independent staining sessions, with each session using freshly prepared staining solution. During each session a single grid was prepared for each of the three stains, namely UF, SPT and AMo. From each grid, several micrographs were acquired. Of these, five micrographs were chosen at random, resulting in 15 micrographs for UF, SPT and AMo, respectively. On these micrographs particles were picked manually and classified as either 'dark core', 'bright core' or 'ambiguous' by hand. Based on this classification, the percentage of either 'dark core', 'bright core' or 'ambiguous' was calculated for each image independently to account for the different particle counts on each image. Finally, the mean of all particle numbers for a staining solution was calculated together with the standard deviation between the individual images.

For β -galactosidase, each data set was limited to 163 highquality micrographs. On these, putative particles were automatically picked based on the expected protein diameter of 180 Å, extracted and subjected to reference-free 2D classification. Representative 2D classes were then used for a template-based picking approach; particles were extracted again and subjected to reference-free 2D classification to exclude artefacts and subsequent 3D classification using C1 symmetry to identify high-quality particles. The particle population yielding a 3D volume showing four defined subunits was further refined using the homogeneous refinement strategy enforcing D2 symmetry. To account for the limiting grain size of the negative-stain salts and to minimize overfitting, we limited the alignment resolution for 3D classification to 15 Å and to 10 Å for the homogeneous refinement. The final maps were low-pass filtered to 10 Å for comparability. The atomic model of *E. coli* β -galactosidase (PDB entry 6x1q) was fitted into the electron-density maps using a rigid-body strategy as implemented in ChimeraX (version 1.8-rc2024.06.06; Meng et al., 2023).

For further details, see Supplementary Figs. S3, S4 and S5.

3. Results

3.1. SPT and AMo require on-grid fixation to faithfully stain apoferritin

As a first step to evaluate whether SPT and AMo are suitable replacements for the established uranyl-based stains, we investigated their ability to fixate and stain the iron-storage protein apoferritin (Hamaguchi *et al.*, 2019; Kayama *et al.*,

2021; Wu *et al.*, 2020). Ferritin is a spherical capsid, with an outer diameter of 12 nm and an inner cavity with a diameter of 7 nm, comprising 24 subunits in eukaryotes (Massover, 1993; Narayanan *et al.*, 2019). After staining with UF we observed mainly circular objects with a diameter of 12 nm that have an electron-dense, dark core with a diameter of approximately 7 nm (98 \pm 2% particles with a dark core; Fig. 1*a* and Supplementary Figs. S3, S4 and S6*a*).

Conversely, when using the same staining protocol but replacing UF with either SPT or AMo, we barely observed defined apoferritin capsids as the particles appeared fuzzy (Supplementary Fig. S7). Assuming disassembly of the capsids during the staining procedure, we included an on-grid fixation step prior to the staining step into our staining protocol (Fig. 2, supporting information and Supplementary Fig. S2).

This additional fixation step resulted in a marked improvement in the particle quality, revealing circular objects with a diameter of 12 nm. While 98 \pm 2% of UF-stained particles have a dark core, most SPT- and AMo-stained particles do not show a dark core (SPT staining, 4.3 \pm 6.4% particles with a dark core; AMo staining, 9.4 \pm 5.1% particles

with a dark core; Figs. 1b and 1c and Supplementary Figs. S6b and S6c).

3.2. SPT and AMo can be used to assess the sample quality of β -galactosidase

As a proof of concept that SPT and AMo are also suitable for SPA applications, we relied on the well characterized protein β -galactosidase, which is a 464 kDa homotetramer with D2 symmetry (Jacobson *et al.*, 1994) that is commonly used both in biochemical assays and as a resolution standard for cryo-EM. Individual β -galactosidase molecules could be readily distinguished from the background for all three tested stains, although the particles did appear slightly fainter for SPT and AMo staining compared with the established UF stain (Fig. 3).

Nevertheless, all three stains allowed a rapid assessment of the sample quality. On the micrographs, β -galactosidase tetramers could be visually differentiated from smaller objects, such as incomplete assemblies or contaminants, and from stain-dense clusters of particles that are commonly associated



Figure 1

UF-stained apoferritin particles feature an electron-dense core, while SPT- or AMo-stained particles do not. Representative raw micrographs of apoferritin stained with (a) UF, (b) SPT or (c) AMo. Scale bars are 100 nm. The insets show the boxed area at $2.5 \times$ magnification. For micrographs of SPT and AMo staining without prior on-grid fixation, see Supplementary Fig. S7. For a quantification of particles according to their core area, see Supplementary Fig. S6.



Figure 2

Brief overview of the negative-staining workflow. Starting from glow-discharged sample-carrier grids with a continuous support film (a), protein in buffer solution is incubated on the grid surface (b). Grids are then washed with sample buffer (c). For AMo and SPT negative staining an on-grid fixation step using 0.15% glutaraldehyde solution is employed to fix the sample (d) before staining using a droplet of staining solution (e). Afterwards, the stain is rapidly dried to create an amorphous film around the sample (f). For a detailed step-by-step guide, see Supplementary Fig. S2.

with (micro-)aggregates (Chari *et al.*, 2015; Fig. 3, white arrows). We noted that the observed particle density is lower for SPT and AMo compared with UF, although identical protein concentrations and absorption times were used.

3.3. SPT and AMo can be used to reconstruct the 3D structure of β -galactosidase

We next processed data from all three staining conditions to evaluate the capability of SPT and AMo, compared with UF, to preserve the native structure of β -galactosidase (Supplementary Figs. S3, S4 and S5). In line with our observation of the raw micrographs, we note that CTF correction for both SPT and AMo worked best assuming a lower amplitude contrast than for UF (in our hands 0.10 for SPT, 0.15 for AMo and 0.35 for UF). Despite this difference, all three staining conditions yielded convincing 2D class averages that clearly showed the boundaries of the individual subunits. As expected for negative staining, secondary-structure elements were not visible in these 2D class averages. Ultimately, all three stains yielded comparable 3D density maps at resolutions around 10 Å that allowed rigid-body fitting of the known molecular model of the β -galactosidase tetramer (Merk et al., 2020; Fig. 4).

4. Discussion

The majority of structures submitted to the Electron Microscopy Data Bank (EMDB) originate from uranyl-based stains. While this number is potentially skewed by publication bias, since researchers new to the field are more likely to employ well established and documented stains, there are objective reasons for choosing uranyl-based stains such as a strong beam-scattering effect (Nakakoshi *et al.*, 2011), a fine grain size (Haschemeyer & Myers, 1972) and the strong fixating properties of formate and acetate (Zhao & Craig, 2003). However, uranium is a radioactive material and it requires a low pH to prevent crystallization of the staining solution (Cao et al., 2011). In contrast, SPT and AMo are based on the nonradioactive elements tungsten and molybdenum, which also show adequate solubility in solutions at neutral pH values, with SPT having been reported to be stable in the pH range 6-9 (Scarff et al., 2018) and AMo in the pH range 5-7 (Bohonek, 1974). As a proof of concept we tested both SPT and AMo stains with two different proteins, apoferritin and β -galactosidase, and compared the results with the established UF stain. When working with apoferritin, we observed that the lack of an organic counterion with protein-fixating properties required us to include a simple on-grid fixation step into our staining procedure (Fig. 2, supporting information and Supplementary Fig. S2) in order to stabilize both proteins (Figs. 1 and 3 and Supplementary Fig. S7). The on-grid fixation vielded a homogeneous particle distribution on the grids, while in our hands using in-solution fixation instead (Harris, 1999) resulted in aggregation of the test proteins (data not shown). Presumably, the absorption of the protein onto the carbon film minimizes the risk of intermolecular cross-links.

While apoferritin stained by UF appeared as particles with a dark core, implying that the central cavity of the capsid-like structure was filled with staining solution, the majority of particles stained by either SPT or AMo did not (Figs. 1*a*, 1*b* and 1*c*, Supplementary Fig. S6). This difference in staining has been observed before (Harris, 1982). Possibly, the different behaviour is due to the pH difference of the staining solutions. It has been reported that low-pH solutions can cause opening of the pores of apoferritin, putatively allowing uranium ions to penetrate the capsid (Mollazadeh *et al.*, 2022).

While SPT and AMo have been reported to have approximately twice the grain size of UF (Harris *et al.*, 2006; Hosogi *et al.*, 2015), we do not observe any resolution-limiting effect compared with UF when working with the tetrameric protein β -galactosidase (Supplementary Figs. S3, S4 and S5), and the resulting 3D density maps are virtually identical (Fig. 4). However, this implies that when working with SPT and AMo



Figure 3

SPT or AMo staining allows assessment of the quality of β -galactosidase preparations similar to UF staining. Representative raw micrographs of β -galactosidase stained with (a) UF, (b) SPT or (c) AMo are shown. Scale bars are 100 nm. The insets show the boxed area at 2.5× magnification. Instances of stain accumulation, commonly associated with protein aggregates, are marked by arrows.

methods communications



Figure 4

UF, SPT and AMo all allow faithful 3D reconstructions of β -galactosidase to be obtained by single-particle analysis. (*a*) Selected 2D class averages for β -galactosidase obtained from grids prepared by either UF, SPT or AMo negative staining. (*b*) Surface representation of 3D density maps calculated from these data sets, filtered to 10 Å for comparability. The atomic model of β -galactosidase (PDB entry 6x1q, coloured by individual subunits) was rigid-body fitted to each map. The scale bar is 25 Å. For the complete 3D reconstruction workflow of each condition, see Supplementary Figs. S3, S4 and S5.

stains 3D reconstructions should be limited to about 20 Å resolution to account for the worse sampling rate compared with UF.

Moreover, the apparently lower contrast of SPT and AMo (Figs. 1 and 3), which is likely to be due to the lower atomic numbers of tungsten and molybdenum compared with uranium, did not negatively affect particle picking, 2D classification or 3D reconstruction. The correct amplitude contrast for processing SPT and AMo data was easy to determine by systematically varying this value for a small subset of micrographs during the CTF determination step.

In combination with the easy-to-integrate on-grid fixation step, SPT and AMo offer adequate contrast and structure preservation, establishing themselves as viable alternatives to uranyl salts for sample preparation for SPA applications. Both SPT and AMo can be utilized at higher pH ranges, permitting the imaging of proteins at neutral pH, in contrast to UF which has limited solubility at non-acidic pH values. Additionally, as nonradioactive alternatives, SPT and AMo can be used in any laboratory without the administrative complexities associated with the purchase and disposal of radioactive materials. This advantage enables researchers to negatively stain their proteins directly after preparation within their own laboratory, eliminating the need to transport samples to an external, possibly remote, EM facility.

Acknowledgements

We acknowledge access to the cryo-EM infrastructure of StruBiTEM (Cologne, funded by DFG Grant INST 216/949-1 FUGG) and to the computing infrastructure of CHEOPS (Cologne, funded by DFG Grant INST 216/512/1 FUGG). We

methods communications

thank Alexandra Schneider and Jennifer Lange for the preparation of apoferritin and β -galactosidase. We thank Sandra Jordan for contributions at an early stage of the project. We thank Dr Kristina Barragán Sanz for the kind donation of the pET-24a-mmFTH1 plasmid. Author contributions were as follows. MG conceived the study; MG and AM prepared samples; MG performed electron microscopy; MG analyzed data; MG performed 3D reconstruction; MG and EB validated data; MG prepared figures; MG and EB wrote the manuscript with input from all authors. Open access funding enabled and organized by Projekt DEAL.

Conflict of interest

The authors declare no competing interests.

Funding information

This work was funded by the Ministry of Culture and Science of the State of North Rhine-Westphalia (iHEAD to EB).

References

- Benmeradi, N., Payre, B. & Goodman, S. L. (2015). *Microsc. Micro*anal. 21, 721–722.
- Bohonek, J. (1974). Colloid Polym. Sci. 252, 417-418.
- Bradley, D. E. (1962). J. Gen. Microbiol. 29, 503-516.
- Brenner, S. & Horne, R. W. (1959). *Biochim. Biophys. Acta*, **34**, 103–110.
- Burgess, S. A., Walker, M. L., Sakakibara, H., Oiwa, K. & Knight, P. J. (2004). J. Struct. Biol. 146, 205–216.
- Cao, B., Xu, H. & Mao, C. (2011). Microsc. Res. Tech. 74, 627-635.
- Chari, A., Haselbach, D., Kirves, J.-M., Ohmer, J., Paknia, E., Fischer, N., Ganichkin, O., Möller, V., Frye, J. J., Petzold, G., Jarvis, M., Tietzel, M., Grimm, C., Peters, J.-M., Schulman, B. A., Tittmann, K., Markl, J., Fischer, U. & Stark, H. (2015). *Nat. Methods*, **12**, 859–865. De Carlo, S. & Harris, J. R. (2011). *Micron*, **42**, 117–131.
- Epstein, M. A. (2001). J. Infect. **43**, 111–115.
- Fabre, L., Bao, H., Innes, J., Duong, F. & Rouiller, I. (2017). J. Biol.
- *Chem.* **292**, 5457–5464. Finch, J. (1964). *J. Mol. Biol.* **8**, 872–874.
- Francis, S. M., Pattar Kadavan, S. & Natesh, R. (2024). Arch. Microbiol. 206, 230.
- Fukuda, Y., Stapleton, K. & Kato, T. (2023). Microscopy, 72, 135-143.
- Ha, J. Y., Chou, H.-T., Ungar, D., Yip, C. K., Walz, T. & Hughson, F. M. (2016). *Nat. Struct. Mol. Biol.* **23**, 758–760.
- Hamaguchi, T., Maki-Yonekura, S. M., Naitow, H., Matsuura, Y., Ishikawa, T. & Yonekura, K. (2019). J. Struct. Biol. 207, 40-48.
- Harris, J. R. (1982). Micron, 13, 169–184.
- Harris, J. R. (1999). Methods Mol. Biol. 117, 13-30.
- Harris, J. R. (2015). Arch. Biochem. Biophys. 581, 3-18.
- Harris, J. R., Bhella, D. & Adrian, M. (2006). Wiley Analytical Science Magazine. https://analyticalscience.wiley.com/do/10.1002/micro.559/.

- Haschemeyer, R. H. & Myers, R. J. (1972). Principles and Techniques of Electron Microscopy, Vol. 2, edited by A. F. Hayat, pp. 101–147. New York: Van Norstrand Reinhold.
- Hosogi, N., Nishioka, H. & Nakakoshi, M. (2015). Microscopy (Tokyo), **64**, 429–435.
- Inaga, S., Katsumoto, T., Tanaka, K., Kameie, T., Nakane, H. & Naguro, T. (2007). Arch. Histol. Cytol. 70, 43–49.
- Jacobson, R. H., Zhang, X.-J., DuBose, R. F. & Matthews, B. W. (1994). Nature, 369, 761–766.
- Johnson, M. C., Bozzola, J. J., Shechmeister, I. L. & Shklair, I. L. (1977). J. Bacteriol. 129, 351–357.
- Kayama, Y., Burton-Smith, R. N., Song, C., Terahara, N., Kato, T. & Murata, K. (2021). Sci. Rep. 11, 8395.
- Kuipers, J. & Giepmans, B. N. G. (2020). *Histochem. Cell Biol.* **153**, 271–277.
- Leberman, R. (1965). J. Mol. Biol. 13, 606.
- Mannella, C. A. & Frank, J. (1984). Ultramicroscopy, 13, 93-102.
- Massover, W. H. (1993). Ultramicroscopy, 52, 383-387.
- Matsuike, D., Tahara, Y., Nonaka, T., Wu, H., Hamaguchi, T., Kudo, H., Hayashi, Y., Arai, M. & Miyata, M. (2023). *J. Bacteriol.* **205**, e00340-22.
- Meng, E. C., Goddard, T. D., Pettersen, E. F., Couch, G. S., Pearson, Z. J., Morris, J. H. & Ferrin, T. E. (2023). *Protein Sci.* 32, e4792.
- Merk, A., Fukumura, T., Zhu, X., Darling, J. E., Grisshammer, R., Ognjenovic, J. & Subramaniam, S. (2020). *IUCrJ*, **7**, 639–643.
- Mollazadeh, S., Yazdimamaghani, M., Yazdian-Robati, R. & Pirhadi, S. (2022). *Comput. Biol. Med.* **141**, 105158.
- Nakakoshi, M., Nishioka, H. & Katayama, E. (2011). J. Electron Microsc. 60, 401–407.
- Nakane, T., Kotecha, A., Sente, A., McMullan, G., Masiulis, S., Brown, P. M. G. E., Grigoras, I. T., Malinauskaite, L., Malinauskas, T., Miehling, J., Uchański, T., Yu, L., Karia, D., Pechnikova, E. V., de Jong, E., Keizer, J., Bischoff, M., McCormack, J., Tiemeijer, P., Hardwick, S. W., Chirgadze, D. Y., Murshudov, G., Aricescu, A. R. & Scheres, S. H. W. (2020). *Nature*, **587**, 152–156.
- Narayanan, S., Shahbazian-Yassar, R. & Shokuhfar, T. (2019). J. Phys. D Appl. Phys. 52, 453001.
- Ohi, M., Li, Y., Cheng, Y. & Walz, T. (2004). Biol. Proced. Online, 6, 23–34.
- Pramanick, I., Sengupta, N., Mishra, S., Pandey, S., Girish, N., Das, A. & Dutta, S. (2021). *Structure*, **29**, 834–845.
- Punjani, A., Rubinstein, J. L., Fleet, D. J. & Brubaker, M. A. (2017). *Nat. Methods*, 14, 290–296.
- Sasajima, Y., Kato, T., Miyata, T., Kawamoto, A., Namba, K. & Miyata, M. (2022). Front. Microbiol. 13, 1004601.
- Scarff, C. A., Fuller, M. J. G., Thompson, R. F. & Iadanza, M. G. (2018). J. Vis. Exp., e57199.
- Takizawa, Y., Binshtein, E., Erwin, A. L., Pyburn, T. M., Mittendorf, K. F. & Ohi, M. D. (2017). Protein Sci. 26, 69–81.
- Weissenberger, G., Henderikx, R. J. M. & Peters, P. J. (2021). Nat. Methods, 18, 463–471.
- Wu, M., Lander, G. C. & Herzik, M. A. Jr (2020). J. Struct. Biol. X, 4, 100020.
- Yip, K. M., Fischer, N., Paknia, E., Chari, A. & Stark, H. (2020). *Nature*, 587, 157–161.
- Zhao, F. J. & Craig, R. (2003). J. Struct. Biol. 141, 43-52.