routine starts with a few comment lines to facilitate the understanding of the program structure and help if modifications are required.

Availability: Copies of the EDAUTO-CAL source code are available from the author on 9-track magnetic tape (format 1600 or 6250 b.p.i. ASCII or EBCDIC code) or by direct transmission through the network.

Keywords: Energy dispersive, detector calibration, detector resolution, non-linear least-squares fit.

References

- Aitken, D. W. (1968). *IEEE Trans. Nucl. Sci.* **15**, 10–46.
- Amersham International plc, UK (1989). Product code AMC.2084. Catalogue of Instrument Calibration Sources, No. 44. Amersham International plc, Aylesbury, Bucks HP20 2TP, England.
- Bearden, J. A. (1967). NSRDS. Natl Bur. Stand. (US) Spec. Publ. No. 14.
- Buras, B., Olsen, J. S., Lindegaard Andersen, A., Gerward, L. & Selsmark, B. (1974). J. Appl. Cryst. 7, 296–297.
- Fukamachi, T., Togawa, S. & Hosoya, S. (1973). J. Appl. Cryst. 6, 297–298.
- Häusermann, D. (1987). New Techniques in X-ray Diffraction. PhD thesis, Univ. of London, England.
- Helmer, R. G., Heath, R. L., Putnam, M. & Gibson, D. H. (1967). *Nucl. Instrum. Methods*, **57**, 46–57.
- National Physical Laboratory, UK (1986), Private communication.
- Pohlers, A. (1988). *Exp. Tech. Phys.* 36, 97–103.

Crystallographers

J. Appl. Cryst. (1990). 23, 223

This section is intended to be a series of short paragraphs dealing with the activities of crystallographers, such as their changes of position, promotions, assumption of significant new duties, honours, etc. Items for inclusion, subject to the approval of the Editoral Board, should be sent to the Executive Secretary of the International Union of Crystallography (J. N. King, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England).

Professor Robin John Hawes Clark, Professor of Chemistry at University College London, has been elected a Fellow of the Royal Society. Distinguished for his application of vibrational spectroscopy, particularly resonance Raman spectroscopy, to the determination of the structures of inorganic compounds. His work on the linear-chain complexes of palladium and platinum has been particularly notable. Dr Nicholas Charles Handy, Reader in Quantum Chemistry in the University of Cambridge, has been elected a Fellow of the Royal Society. Distinguished for his development of very precise methods for the calculation of the structures and spectroscopic properties of molecules. Computer programs which he developed are now extensively used by the chemical community.

Dr Louise Napier Johnson, Lecturer in Molecular Biophysics in the University of Oxford, has been elected a Fellow of the Royal Society. Distinguished for her contributions to protein crystallography, in particular for her studies of the mechanism of action of the large complex enzyme glycogen phosphorylase b and its allosteric regulation.

Professor Harold Walker Kroto, Professor of Chemistry in the University of Sussex, has been elected a Fellow of the Royal Society. Distinguished for his spectroscopic studies of the structure of unstable molecules, notably leading to the discovery of cyanopolyacetylenes in space. Recent studies of carbon vapour led him to identify new structures, notably the 'carbon football' – C₆₀.

John Osbourne, former Director of the South London Science Centre, has been awarded the Bragg Medal and Prize of the British Institute of Physics for his contributions to the teaching of physics in schools.

New Commercial Products

Announcements of new commercial products are published by the Journal of Applied Crystallography free of charge. The descriptions, up to 300 words or the equivalent if a figure is included, should give the price and the manufacturer's full address. Full or partial inclusion is subject to the Editor's approval and to the space available. All correspondence should be sent to the Editor. Professor M. Schlenker, Editor Journal of Applied Crystallography, Laboratoire Louis Néel du CNRS, BP 166, F-38042 Grenoble CEDEX, France.

noble CLUEX, France. The International Union of Crystallography can assume no responsibility for the accuracy of the claims made. A copy of the version sent to the printer is sent to the company concerned.

J. Appl. Cryst. (1990). 23, 223-224

ChemDBS-3D Links Molecular Modelling and 3D Database systems

Chemical Design is pleased to announce a new member of the Chem-X suite of programs for computer-aided molecular modelling. The new module, **ChemDBS-3D**, is designed to help research chemists indentify potentially active compounds in 3D databases of chemical structures, and is fully integrated with the rest of the Chem-X system.

Using Chem-X, chemists can not only build complex structures quickly and easily, but also calculate a wide range of molecular properties and identify structure-activity relationships. This information may be used to generate templates called pharmacophores which define the spatial arrangement of functional groups and properties common to active molecules of a particular class. Once a pharmacophore has been established, it can be compared with candidate molecules to test whether they are potentially active members of this class.

The most effective way of doing this is to compare the pharmacophore structure with the 3D structures of compounds stored in a database. 3D databases can store data specific to individual atoms, such as partial charges or coordinates, which makes it possible to search for pharmacophores defined by both geometric and electronic criteria.

ChemDBS-3D offers rapid automatic screening of a 3D database for compounds that can adopt the required 3D arrangement of key atoms, as defined by the pharmacophore, in any lowenergy (stable) conformation. To save disk space, only one set of coordinates is stored for each molecule - the lowenergy conformational space used in screening is generated automatically. The exceptionally efficient screening method enables searches to be performed very rapidly: preliminary tests suggest that a database containing 5500 compounds can be screened in 0.16 s of CPU time on a VAX 8600.

Only compounds which pass the initial screening test (ideally less than 10% of the molecules in the database) are considered in subsequent searches. The chemist may wish to select only compounds containing a particular substructure (a 2D atom connectivity search) and/or with a certain range of values in one or more predefined property fields. This reduces the size of the set of accepted molecules still further.

Preliminary results show that searching can be performed on any combination of 100 fields at approximately 150 compounds per second.

The low-energy conformations of the accepted compounds which actually match the given pharmacophore are then generated automatically.

The results of each phase of the search (a set of molecules or conformations) are stored in a database and can be listed or displayed. 'Hit lists' can be combined using logical operations (AND, OR, NOT *etc*), and may also be imported from or exported to other database systems. Chem-X can be used to produce 2D or 3D plots of various properties for a set of candidate structures, and to analyse the results statistically.

ChemDBS-3D should be available in April 1990.

Chemical Design Ltd, Unit 12, 7 West Way, Oxford OX2 0JB, England

Book Reviews

Works intended for notice in this column should be sent direct to the Book-Review Editor (R. O. Gould, Department of Chemistry, University of Edinburgh, West Mains Road, Edinburgh EH9 3JJ, Scotland). As far as practicable books will be reviewed in a country different from that of publication.

J. Appl. Cryst. (1990). 23, 224

The physics of quasicrystals. Edited by *Paul J. Steinhardt* and *Stellan Ostlund.* Pp. xvi + 767. World Scientific Publishing, Singapore, 1987. (Distributed by J. Wiley.) Price £76.95 or US\$ 111.55 (cloth), £34.20 (paper). ISBN 0971502275.

This 767 page book on quasicrystals consists of about ninety pages of original text, about thirty pages of references and a large number of reprints, some of which were, according to the flyleaf of the book, reproduced without permission. While there is some value in the collection, most of the papers involved are very accessible. Thus it is doubtful that any other than a very select few will wish to have this book.

Fivefold symmetry should not be possible for a true crystal lattice, yet Shechtman, Blech, Gratias & Cahn [*Phys. Rev. Lett.* (1984). **53**, No. 20, 1951–1953] did a study of rapidly cooled alloys of Al with 10–14 at.% Mn, Fe or Cr, which display icosahedral symmetry with the sharp peaks one would expect of a crystal. Fortunately there is a model which can explain this surprising result: the interaction of two incommensurate periods, based on the concept of Penrose tilings [*Bull. Inst. Math. Applic.* (1978). **10**, No. 7/8, 266–271]. This book first considers the geometric properties of quasilattices (3D quasiperiodic tilings), then the applicable group theory, the diffraction properties and structural models. The most useful reprints are in the fifth chapter, where competing models are considered for icosahedral symmetry. Multiple twinning, while not generally a full explanation of experimental results in this area, must be considered for each new case.

The second part of the book reviews physical properties of quasicrystals including Landau theory, stability, defects, electronic and magnetic properties. The book ends with an Appendix of the classic papers on Penrose tilings. If basic teaching materials are needed, we would recommend reproducing this Appendix and the first 33 pages of the book, which define the essential terminology.

HERBERT J. BERNSTEIN University Computer Center New York University 715 Broadway New York NY 10003 USA