

Tetragonal distortion of the multiferroic compound α -LiFe₅O₈ from first-principles calculations

Yaoshi Cheng and So-Hyun Park*

Applied Crystallography and Geomaterials, Department of Earth and Environmental Sciences, Ludwig-Maximilians-Universität München, Theresienstrasse 41C, Munich, Bavaria 80333, Germany. *Correspondence e-mail: sohyun.park@lmu.de

Received 13 April 2026

Accepted 25 June 2026

Edited by Th. Proffen, Oak Ridge National Laboratory, USA

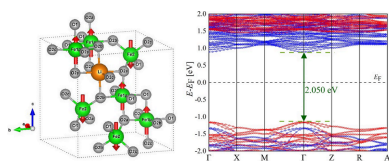
Keywords: α -LiFe₅O₈; ferrimagnetic multiferroics; band structure; first-principles calculations; density of states; Hubbard model; density functional theory; magnetic exchange interactions.

Supporting information: this article has supporting information at journals.iucr.org/j

The cubic space group $P4_32$ has been previously reported for the multiferroic lithium ferrite (α -LiFe₅O₈, denoted LFO). Instead, results from the present study using PBEsol+ U calculations with SSSP pseudopotentials suggest that the tetragonal space group $P4_32_12$ is true for the ground state of LFO. Spin-resolved projected density of states analysis shows strong $3d$ electron localization of Fe³⁺ in FeO₄ tetrahedra, mostly contributing to the high Hubbard energy U in the calculated tetragonal model. Its direct band gap of 2.050 eV is slightly lower than that of the cubic model (2.107 eV) in density of states calculations. The calculated ferrimagnetic ground state exhibits a collinear spin order along the crystallographic c axis in the tetragonal magnetic space group $P4_32_12'$, consistent with the group theoretical prediction. Decreasing ferromagnetic interactions and increasing antiferromagnetic interactions in the c direction, simultaneously, explain the origin of the tetragonality $a/c > 1$ with $a = 8.3528$ (2) Å and $c = 8.3511$ (1) Å. Strong Fe–O–Fe superexchange interactions contribute to the tetragonal lattice distortion of LFO as well.

1. Introduction

The photovoltaic (PV) effect in magnetoelectric (ME) multiferroics (MFs) has been increasingly investigated by finding spintronics showing helical magnetic spin order-induced electric polarization (Young *et al.*, 2013; Xiao *et al.*, 2023; Xu *et al.*, 2021). The bulk PV effect in ME-MFs occurs with photon-excited electrons jumping from the valence band to the conduction band under inherent electric fields in their polar electronic structures due to the magnetic spin order-induced breaking of space inversion (Baltz & Kraut, 1981). Thus, in contrast to p-n-junction semiconductor PVs, the resulting photovoltage in ME-based PVs is barely associated with the band-gap energy (Rangel *et al.*, 2017; Jalaja & Dutta, 2015). In ME-MFs, the coexistence of ferroelectricity and magnetic spin order allows adjustment of their PV performance by applying external magnetic and electric fields. Furthermore, the MF-PV effect can be significantly enhanced by mechanical and chemical stress using thin films and doping (Vopson, 2015; Dharmadhikari & Grannemann, 1982; Ji *et al.*, 2010; Peng *et al.*, 2017; Lee & Son, 2024). Hence, MFs can be photoactive over a wide solar spectrum. In particular, thin films of ME-MFs such as BaTiO₃ (Koch *et al.*, 1975), BiFeO₃ (Yang *et al.*, 2009) and Bi₂FeCrO₆ (Mahapatra *et al.*, 2024) have been given attention for PV applications due to their strong response to photons. Their relatively low band gaps of about 2.04–3.21 eV are advantageous compared with those of pure ferroelectric PVs (Nechache *et al.*, 2015; Jalaja & Dutta, 2015). For the bulk PV effect, the photocurrent contributes to the shift current induced by electric polarization, as well as to the ballistic



current induced by momentum asymmetry of photon-excited carriers (Burger *et al.*, 2019). Therefore, a low band-gap energy for MFs is desired for strong electron–electron interactions to overcome their photocurrent upper limit of milliamps per square centimetre in contrast to the limit of several thousand amps per square centimetre for semiconductor-based PVs (Jalaja & Dutta, 2015; Huang *et al.*, 2021; Nakamura *et al.*, 2024; Dhass *et al.*, 2020). MF-PVs with a low band-gap energy will be particularly effective for the infrared region from 1.24 meV to 1.7 eV, which is the major part of the solar energy spectrum (Jalaja & Dutta, 2015; Nechache *et al.*, 2015).

Among MFs, the ordered phase of lithium ferrite [α -LiFe₅O₈ (Liu *et al.*, 2019); LFO hereafter] is interesting for its low band-gap energy depending on preparation: 1.67 and 1.86 eV for nest-like crystallites with and without using a reduced graphene oxide substrate, respectively (Zhang & Zhang, 2016), 2.37 eV for a nanocrystalline film on a glass substrate (Babrekar & Jadhav, 2017), and 1.40–1.60 eV for mixed nanoparticles of α -LiFe₅O₈ and β -LiFe₅O₈ calcined between 973 and 673 K (Chireh & Naseri, 2019). On the other hand, it is desirable to introduce a high ME coupling effect into multiferroics, preferably at room temperature for PV applications. In this sense, LFO is interesting for its high ME coupling coefficient (α_{ME}) value of 2 mV Oe^{−1} cm^{−1} at 120 K; its α_{ME} value decreases with increasing temperature (0.2 mV Oe^{−1} cm^{−1} at 250 K), but the ME effect in LFO does exist up to room temperature (Liu *et al.*, 2019). The ferrimagnetic state of LFO is stable below its high Curie temperature of 907 K (Astafyev *et al.*, 2019), exhibiting a saturation magnetization of 2.5 μ_{B} per formula unit at room temperature (Pachauri *et al.*, 2015). This significant effective ME effect in LFO is attributed to the spin–orbital (SO) coupling of Fe³⁺ (3d⁵) residing in geometrically distorted octahedra (Mohapatra & Dobbidi, 2023). The resulting polyhedral distortion-induced electric polarization in a weak ferromagnetism (Mercier *et al.*, 1977) can be tuned by various external fields (Liu *et al.*, 2019).

LFO has been intensively investigated in both research and development of functional materials (Iliev *et al.*, 2011; De Picciotto & Thackeray, 1986; Gridnev *et al.*, 1997; Kinoshita *et al.*, 2016; Teixeira *et al.*, 2021; Soreto *et al.*, 2017; Kumawat *et al.*, 2021; Kim *et al.*, 2022; Sousa *et al.*, 2019; Moore *et al.*, 2024; Tan *et al.*, 2021). All these experimental and theoretical studies of LFO have been based on the atomic arrangements in the cubic space group $P4_332$ (Liu *et al.*, 2019). Its point group 432 is not a polar group and hence it disallows spontaneous electric polarization but allows the second-order ME effect. There are several unclear issues about its cubic space-group symmetry $P4_332$ and the consequent magnetic spin order in the cubic lattice (Kumawat *et al.*, 2021; Liu *et al.*, 2019). In particular, several strong magnetic Bragg reflections forbidden to its ferrimagnetic order in the cubic magnetic space group $P4_332$ were excluded in Rietveld refinements using high-resolution neutron powder diffraction (HRNPD) data (Kumawat *et al.*, 2021). By coincidence, the resulting collinear ferrimagnetic spin order is consistent with a theoretical magnetic spin order along the *c* axis predicted from *ab initio*

calculations, where only the cubic space group $P4_332$ was considered (Kim *et al.*, 2022). However, according to the group–subgroup relation between space group and magnetic space group in group theory, the collinear spin arrangement is best realized in the tetragonal magnetic space group $P4_32_12'$, the same as the ferrimagnetic structure of Al³⁺-doped LFO-type compounds in our recent HRNPD study (Inckemann *et al.*, 2025).

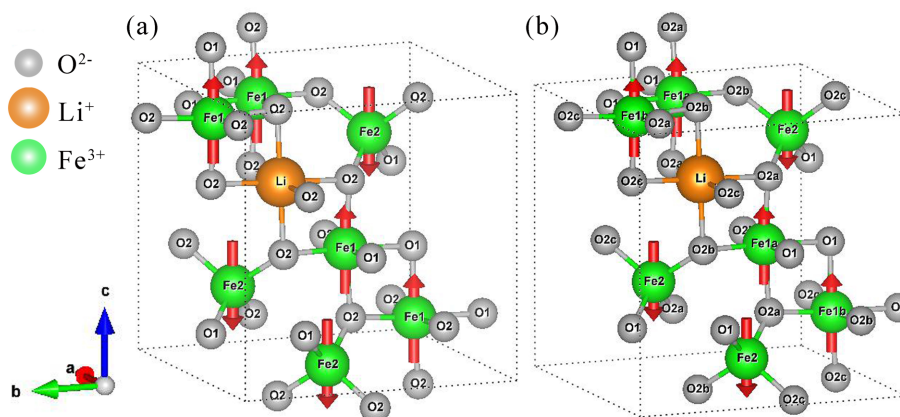
To date, there are no experimental band-gap energy data for LFO samples without microstructural effects. Results from previous density functional theory (DFT) calculations on LFO by different researchers are also contradictory to each other as they reported different band-gap energy values: Sousa *et al.* (2019) did not consider short-range interactions between localized Fe 3d electrons, giving rise to unreasonable valence band characteristics for LFO and an overestimated indirect band gap of 2.3 eV. Kim *et al.* (2022) calculated structural, electronic and magnetic properties of LFO in combination with the Hubbard model, resulting in a collinear ferrimagnetic spin configuration with a indirect band gap of 1.99 eV. However, in their DFT calculations, the Hubbard value *U* for the on-site Coulomb correction was determined merely by fitting the cubic lattice constant, rather than using the linear response method regarded as the standard way for obtaining *U* values (Moore *et al.*, 2024). For this reason, the underestimated interaction between Fe 3d electrons might be corrected further. According to the simulations done by Tan *et al.* (2021), the *U* value was set at 5.3 eV, where the valence band maximum was higher than the Fermi energy level. This could not be explained in terms of semiconductor behaviour. Thus, it is necessary to re-examine the true atomic and magnetic structures of LFO, which involves a symmetry lowering from cubic to tetragonal. The present study is an extensive re-investigation of the electronic structures of LFO in both cubic and tetragonal models using the DFT method with the Hubbard model. The simulated atomic and magnetic structures, as well as the electronic properties, agree with the tetragonal space group $P4_32_12$ being true for LFO, as described in the following.

2. Starting models

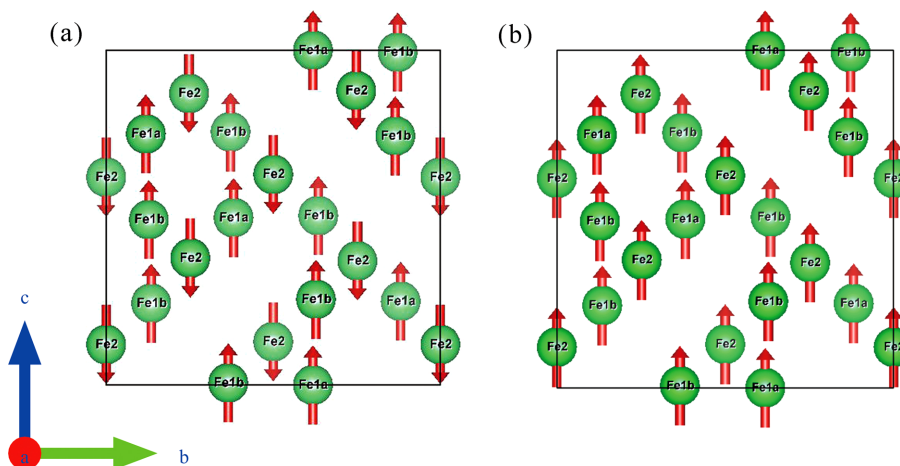
Following previous experimental and theoretical studies of LFO (Kim *et al.*, 2022; Kumawat *et al.*, 2021), we have taken a collinear ferrimagnetic spin configuration of Fe³⁺ in the tetragonal and octahedral sublattices in two starting models:

(i) The well known cubic model in $P4_332$. Li⁺ is located on an octahedral site (4b) while Fe³⁺ is found on both octahedral (12d) and tetrahedral (8c) sites. The respective Fe sites are labelled Fe1 and Fe2 [Fig. 1(a)].

(ii) The tetragonal model in $P4_32_12$, a direct subgroup of $P4_332$. Its magnetic subgroup $P4_32_12'$ is the only possible one to allow a collinear ferrimagnetic arrangement along *c* among the cubic ($P4_3'32'$ and $P4_332$) and tetragonal ($P4_32_12'$, $P4_3'2_12'$, $P4_3'2_12'$ and $P4_32_12$) magnetic subgroups of $P4_332$ [see the supporting information of Inckemann *et al.* (2025)]. This tetragonal distortion is accompanied by splitting on the Fe1

**Figure 1**

Initial models of α -LiFe₅O₈, (a) in $P4_332$ and (b) in $P4_3212$ with site splitting for DFT simulation. For both starting models, a collinear ferrimagnetic spin order is configured with two sublattices of Fe³⁺ in FeO₄ and FeO₆ polyhedra.

**Figure 2**

Two theoretical (a) ferrimagnetic and (b) ferromagnetic arrangements of Fe³⁺ in the starting models for the ground state of α -LiFe₅O₈ in $P4_3212$.

and O2 sites, *i.e.* the octahedral site for Fe (12d) and the site for O (24e) in the cubic model are split into Fe1a (8b) + Fe1b (4a) and O2a (8b) + O2b (8b) + O2c (8b), respectively, in the tetragonal model [Fig. 1(b)].

Using all the atomic parameters including magnetic moments, U values were fitted for all Fe³⁺ ions for both models. To prove the difference between the ferri- [Fig. 2(a)] and ferromagnetic [Fig. 2(b)] ground states, we calculated their total energies.

3. Computational methods

DFT, based on the electron density and exchange-correlation approximations, explains the behaviour of the electrons and atomic nuclei of the investigated system by solving the Kohn–Sham equation without any experimental data (Bagayoko, 2014). As a reliable first-principles calculation method, DFT has contributed greatly to predicting and explaining coupling effects in various types of multiferroic materials beyond their structural and electronic properties (Neaton *et al.*, 2005; Ederer & Spaldin, 2006). Combined with the first-order perturbation theory, the density functional perturbation

theory (DFPT) allows the simulation, with high computational efficiency, of crystal systems with a localized perturbation in one unit cell without requiring a supercell (Baroni *et al.*, 2001).

In the present study, for the multiferroic state of LFO, the magnetic spin of Fe³⁺ has been considered with the valence electronic configurations of Fe ($3p^6 3d^5 4s^2$), Li ($1s^2 2s^1$) and O ($2s^2 2p^4$). All Fe³⁺ species in LFO exhibit the high-spin state ($S = 5/2$). Strong correlations between the Fe 3d electrons were described by the Hubbard model. The DFPT+ U method was applied to correct the on-site Coulomb interaction U and spin polarization (Anisimov *et al.*, 1991), where U describes the potential energy of a 3d⁵ electron. In Hubbard-corrected DFPT calculations, the total energy of a spinel crystal system $E_{\text{DFPT}+U}$ is given by the sum of the energy without spins (E_{DFPT}) plus the energy of the on-site Coulomb correction term with a spin contribution (E_U),

$$E_{\text{DFPT}+U} = E_{\text{DFPT}} + E_U. \quad (1)$$

In order to determine U , DFPT calculations with the first-order linear-response theory in reciprocal space were self-consistently performed by the *HP* code (Timrov *et al.*, 2022).

The convergence threshold for the response functions was set to $1 \times 10^{-6} \text{ eV}^{-1}$.

With the corrected U values, DFT calculations were executed using the *Quantum ESPRESSO* software suite (Giannozzi *et al.*, 2009; Giannozzi *et al.*, 2017) with the standard solid-state pseudopotentials (SSSP) (Version 1.3.0; Prandini *et al.*, 2018). Using the Perdew–Burke–Ernzerhof (PBE) (Perdew *et al.*, 1996) method for the generalized gradient approximation, exchange–correlation functionals were implemented to optimize the lattice metric and atomic parameters, and subsequently to calculate band structures. The lattice parameters are strongly related to magnetic properties, such as magnetic exchange interactions. Hence, we applied the PBEsol method to DFT calculations to improve the accuracy of the lattice parameters (Perdew *et al.*, 2008). The kinetic energy cutoff for wavefunctions was tested and could be set to 100 Ry. According to the Monkhorst–Pack scheme, the k -point meshes in the Brillouin zone were sampled by $5 \times 5 \times 5$ grids for both cubic and tetragonal unit cells of LFO. The total energy convergence criterion of 1×10^{-6} Ry and the forces convergence threshold of 1×10^{-4} Ry per Bohr were parameterized. The charge density cutoff of 1100 Ry was set when using SSSP to describe details in orbital electronic structures near the iron nucleus (Prandini *et al.*, 2018).

To explain the origin of tetragonality for LFO, magnetic exchange interactions were calculated with the Green function method according to Liechtenstein–Katsnelson–Antropov–Gubanov (LKAG) (Liechtenstein *et al.*, 1987). For a classical Heisenberg model, the Hamiltonian is described as

$$\mathcal{H} = - \sum_{i \neq j} J_{ij} \mathbf{s}_i \cdot \mathbf{s}_j. \quad (2)$$

J_{ij} is an exchange interaction between two magnetic moments on sites i and j ; and $\mathbf{s}_i$ and $\mathbf{s}_j$ are the unit vectors pointing in the magnetic moment direction on sites i and j , respectively. The magnetic exchange parameters were calculated on the basis of the magnetic force theorem, while the change in electronic single-particle energies of the system was related to the perturbation in the magnetic spin subsystem. Accordingly, the exchange parameters J_{ij} could be described by the intersite Green functions. In DFT calculations, plane-wave-based methods were used to obtain electronic band structures with Bloch states. The resulting Bloch states wavefunctions were projected onto a localized basis (Wannier functions) using the *Wannier90* code (Pizzi *et al.*, 2020) for calculating J_{ij} , where Wannier functions construct the effective Hamiltonian and give the intersite Green functions. Finally, the magnetic exchange parameters in real space were obtained from DFT calculations using the *TB2J* package (He *et al.*, 2021).

4. Results and discussion

4.1. Assessing the ground-state structures and on-site Hubbard parameters

The Hubbard U values were calculated on the tetrahedrally and octahedrally coordinated sites for Fe^{3+} in $P4_332$ and

$P4_32_12$. The calculated U values between 5.0308 and 5.1668 eV from both SSSP+PBE (Table A.I in the supporting information) and SSSP+PBEsol (Table 1) methods agree with those typical for Fe^{3+} in various iron compounds (Moore *et al.*, 2024). Overall, as demonstrated in Fig. 3, the U values from the two methods are close to each other after the fourth iteration. The U values of Fe^{3+} in the tetragonal lattice in $P4_32_12$ are larger than those in the cubic lattice in $P4_332$, and this is especially significant on the tetrahedral site Fe2. Furthermore, this on-site Coulomb repulsion on Fe2 ($U = 5.1668 \text{ eV}$) is stronger than those on the octahedral sites Fe1a ($U = 5.0724 \text{ eV}$) and Fe1b ($U = 5.0839 \text{ eV}$). This implies more localized Fe 3d electrons in FeO_4 tetrahedra, particularly in the tetragonally distorted lattice (Table 1).

From DFT calculations with the U values from both SSSP+PBE (Table A.I in the supporting information) and SSSP+PBEsol (Table 1) methods, the magnitudes of the total energy E for the cubic and tetragonal models were computed as a function of the unit-cell volume V , as shown in Fig. 4. The ground-state unit-cell volume (V_0) and its total energy (E_0) were estimated by least-squares fitting of the E – V curve in the third-order Birch–Murnaghan equation of state (EOS) (Birch, 1947),

$$E(V) = E_0 + \frac{9V_0B_0}{16} \left\{ \left[\left(\frac{V_0}{V} \right)^{2/3} - 1 \right]^3 B'_0 + \left[\left(\frac{V_0}{V} \right)^{2/3} - 1 \right]^2 \times \left[6 - 4 \left(\frac{V_0}{V} \right)^{2/3} \right] \right\}, \quad (3)$$

where B_0 is the ground-state bulk modulus and B'_0 is the derivative of the bulk modulus against pressure. After four

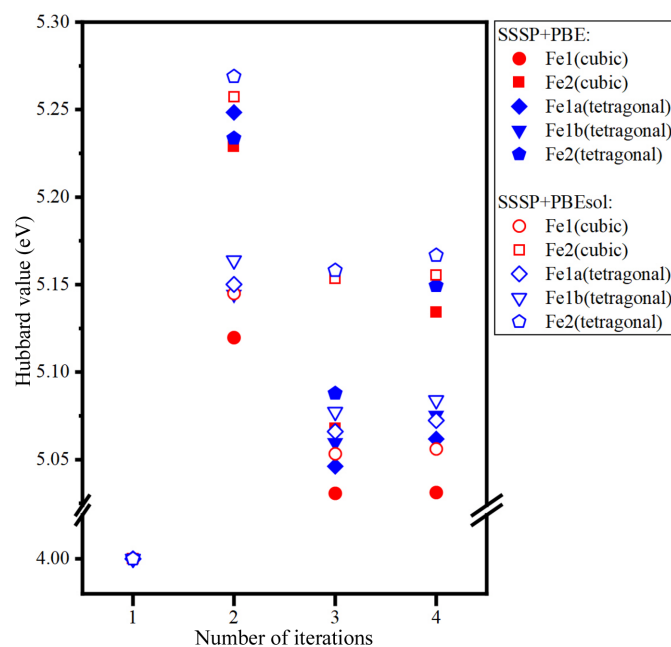


Figure 3
Hubbard values (U) calculated in this study for the cubic and tetragonal models of LFO in each iteration step.

Table 1
Results from DFT calculations for the ground state of LFO in both cubic and tetragonal models by the SSSP+PBEsol method.

	Crystal system			
	Cubic			Tetragonal
Space group	$P4_332$			$P4_32_12$
Unit-cell parameters ($\text{\AA}$)	$a = 8.3550$ (1)			$a = 8.3528$ (2), $c = 8.3511$ (1)
V_0 ($\text{\AA}^3$)	583.2293 (3)			582.6501 (3)
	Site (Wyckoff)	Coordinates		
	Fe1 (12d)	x	y	z
		0.1250	0.3644 (1)	0.8856 (1)
	Fe2 (8c)	0.9947 (1)	0.9947 (1)	0.9947 (1)
	Li (4b)	0.8750	0.3750	0.1250
	O1 (8c)	0.3897 (1)	0.3897 (1)	0.3897 (1)
	O2 (24e)	0.1212 (1)	0.1261 (1)	0.3817 (1)
	Site (Wyckoff)	Coordinates		
	Fe1a (4a)	x	y	z
		0.6173 (3)	0.6173 (3)	0
	Fe1b (8b)	0.8750 (1)	0.3674 (1)	0.0076 (2)
	Fe2 (8b)	0.2523 (1)	0.0023 (1)	0.6227 (2)
	Li (4a)	0.1250	0.1250	0
	O1 (8b)	0.1349 (1)	0.3849 (1)	0.5099 (1)
	O2a (8b)	0.8682 (1)	0.1269 (1)	0.5084 (1)
	O2b (8b)	0.1334 (1)	0.1183 (1)	0.2519 (1)
	O2c (8b)	0.8769 (1)	0.3835 (1)	0.2433 (1)
U (eV)	Fe1	5.0561		
	Fe2	5.1556		
	Fe1a	5.0724		
	Fe1b	5.0839		
Total energy (Ry)		−7911.0052		
B_0 (GPa)		168.30 ± 0.04		
B'_0 (d B_0 /d P)		4.256 ± 0.005		
	Fe1a	5.0724		
	Fe1b	5.0839		
	Fe2	5.1668		
		−7911.0057		
		166.20 ± 0.06		
		4.221 ± 0.003		

iteration steps of the geometry optimization, the value for V_0 tends toward convergence, particularly by applying SSSP pseudopotentials with PBEsol exchange-correlation functionals (Fig. 4). The obtained V_0 values of 583.2293 (3) and 582.6501 (3) $\text{\AA}^3$ for the cubic and tetragonal models, respectively, are comparable to experimental results, e.g. $V_{\text{exp}} = 578.01 \text{ \AA}^3$ at room temperature for the cubic model (Liu *et al.*,

2019), while for the tetragonal model (our own work, to be published elsewhere), we obtained $V_{\text{exp}} = 578.19$ (2) $\text{\AA}^3$ at room temperature using X-ray single-crystal diffraction and $V_{\text{exp}} = 576.49$ (1) $\text{\AA}^3$ at 3 K by Rietveld analysis with HRNPD data. There is an extremely small difference of 0.1% in the V_0 values between the tetragonal and cubic models, where the former with $a > c$ is relatively close to V_{exp} values. In comparison, the V_0 values from the SSSP+PBE method are much further from these experimental observations, as clearly seen in Fig. 4. The current study shows that the PBEsol exchange-correlation functional is suitable for calculating the ground-state unit-cell volume V_0 . Thus, we consider exclusively the results from the SSSP+PBEsol method hereafter. For the same reason, we employed the U values obtained using the SSSP+PBEsol method in the subsequent calculations of electronic band structures and magnetic interaction parameters (Sections 4.2 and 4.3).

The atomic positions were relaxed by geometry optimization iterations using the PBEsol method. The optimized structure parameters, bulk moduli and total energies are listed in Table 1. The tetragonal model of LFO shows an insignificantly lower total energy than the cubic model, i.e. a difference of 5×10^{-4} Ry, but with a high accuracy of 10^{-9} Ry. On the other hand, our calculations of the respective total energies −7911.0057 and −7910.4706 Ry for the respective ferri- and ferromagnetic orders in the tetragonal lattice (Fig. 2) point to the ferrimagnetic ground state in the tetragonal structure of LFO. As mentioned above, this ground-state ferrimagnetic spin order cannot be realized in the cubic magnetic subgroups of $P4_332$; the collinear ferrimagnetic spin arrangements along the crystallographic c axis are allowed exclusively in the tetragonal magnetic space group $P4_32_12'$ (Inckemann *et al.*, 2025). Hence, a reasonable conclusion can be drawn that LFO undergoes a slight tetragonal lattice distortion. This is highly possible due to the magnetic origin,

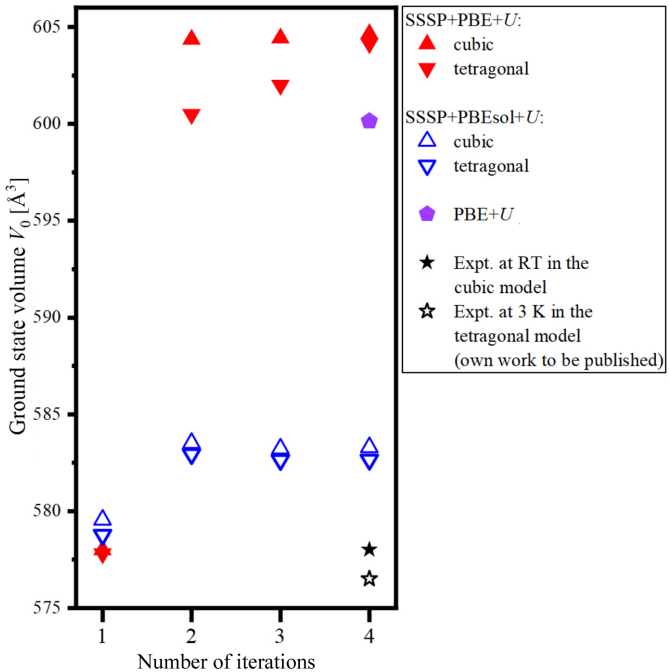


Figure 4
 V_0 values (triangles) calculated in this study for the cubic and tetragonal models of LFO in each iteration step. The volumes of LFO in the cubic space group reported by other experimental [filled star: Liu *et al.* (2019); open star: our work] and theoretical [pentagon: Tan *et al.* (2021)] studies are given for comparison.

Table 2
Average polyhedral degrees of distortion in two different theoretical models for LFO.

Model	Polyhedron	Site	Volume (Å ³)	Distortion (%)	Electric dipole moment (e Å)			Modulus (e Å)
					<i>x</i>	<i>y</i>	<i>z</i>	
Cubic	LiO ₆	Li1	12.434 (1)	0.2 (1)	0	0	0	0
	FeO ₆	Fe1	10.873 (1)	0.5 (1)	−0.8572 (1)	0	0.8572 (1)	1.21 (1)
	FeO ₄	Fe2	3.527 (1)	0.3 (1)	−0.0157 (1)	0.0157 (1)	−0.0157 (1)	0.03 (1)
Tetragonal	LiO ₆	Li1	12.415 (1)	0.4 (1)	1.1409 (1)	1.1409 (1)	0	1.61 (1)
	FeO ₆	Fe1 _a	11.355 (1)	1.8 (1)	−0.2121 (1)	−0.2121 (1)	0	0.30 (1)
	FeO ₆	Fe1 _b	11.188 (1)	3.3 (1)	−1.1360 (1)	0.6582 (1)	0.6013 (1)	1.44 (1)
	FeO ₄	Fe2	3.344 (1)	0.9 (1)	0.0100 (1)	−0.4243 (1)	0.1303 (1)	0.44 (1)

i.e. the small tetragonality [$a = 8.3528$ (2) Å, $c = 8.3511$ (1) Å] may arise from weak antiferromagnetic interactions between the tetragonal and octahedral sublattices of Fe³⁺, which compete with the ferromagnetic interactions within each sublattice (Fig. 2). This issue is addressed quantitatively in Section 4.3. The bulk modulus of 166.20 GPa calculated by the PBEsol method for the tetragonal model is consistent with the experimental value reported by Massoudi *et al.* (2020) as well. Overall, the current study suggests the ground-state structure of LFO is tetragonal in $P4_32_12$. This finding was applied in the subsequent calculations of band structures in Section 4.2 and magnetic interactions in Section 4.3.

Comparing the local structures of LiO₆, FeO₆ and FeO₄ polyhedra (Fig. 5), their volumes, geometric degrees of distortion and distortion-induced electric dipole moments were evaluated using the best-fitted idealized polyhedron method (Song & Liu, 2019; Song *et al.*, 2023) (Table 2). Their degrees of polyhedral distortion and distortion-induced electric dipole moments (Fig. 5) show noticeable differences between the two models within calculation uncertainties (Table 2). In the tetragonal model, the small volume of FeO₄ with obviously short Fe–O bonding distances (Tables A.I and A.II) reflects strong interactions between Fe on Fe2 with four ligand oxygens. The electric dipole moment modulus of FeO₄ is negligibly small compared with those of highly distorted FeO₆, particularly on Fe1_b. The shrinkage in FeO₄ tetrahedra in the tetragonal model is consistent with the higher U value on Fe2 compared with that in the cubic lattice. This hints at the

presence of competing interactions, such as hybridization between O 2*p* and Fe 3*d* orbitals in a rigid and small FeO₄ unit in the tetragonal ground state of LFO (see Section 4.2). The significantly different degrees of distortion between Fe1_aO₆ and Fe1_bO₆ [Fig. 5(*b*)] are a strong sign for lattice distortion towards a tetragonal lattice. Otherwise, in the cubic structure model, there may not be any differences between these two octahedra.

4.2. Electronic band structures and density of states

The electronic band structures were calculated using the optimized structural and Hubbard parameters from self-consistent DFT calculations in both cubic and tetragonal models. The spin-up and spin-down states were calculated along the high-symmetry paths in Brillouin zones: $R(0.5, 0.5, 0.5) \rightarrow \Gamma(0, 0, 0) \rightarrow X(0.5, 0, 0) \rightarrow M(0.5, 0.5, 0) \rightarrow \Gamma(0, 0, 0)$ in $P4_332$ and $\Gamma(0, 0, 0) \rightarrow X(0, 0.5, 0) \rightarrow M(0.5, 0.5, 0) \rightarrow \Gamma(0, 0, 0) \rightarrow Z(0, 0, 0.5) \rightarrow R(0, 0.5, 0.5) \rightarrow A(0.5, 0.5, 0.5)$ in $P4_32_12$. The spin-up state in the octahedral sublattice (denoted the *B* sublattice) was set to $|\uparrow\rangle$, while the spin-down state in the tetrahedral sublattice (denoted the *A* sublattice) was set to $|\downarrow\rangle$, and then the electronic band gaps were calculated. The reverse configurations were also calculated (Fig. 6).

Regarding the *B* sublattice, their band structures for two reverse $|\uparrow\rangle$ [Figs. 6(*a*) and 6(*c*)] and $|\downarrow\rangle$ [Figs. 6(*b*) and 6(*d*)] states show direct band gaps. Our results do not agree with an indirect band-gap energy calculated for the spin-up state of Fe³⁺ in the octahedral *B* sublattice in a previous DFT study

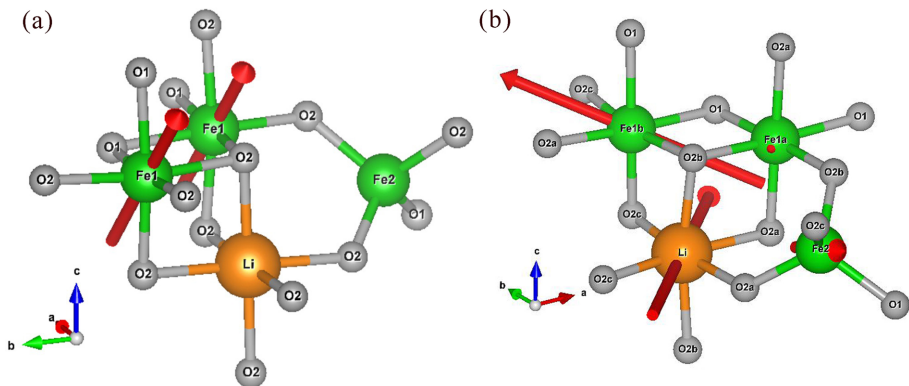


Figure 5
Polyhedral distortion-induced electric dipole moments (Table 2) are indicated by bold red arrows, (*a*) in the cubic model in $P4_332$ and (*b*) in the tetragonal model in $P4_32_12$. Both models were simulated by the PBEsol method.

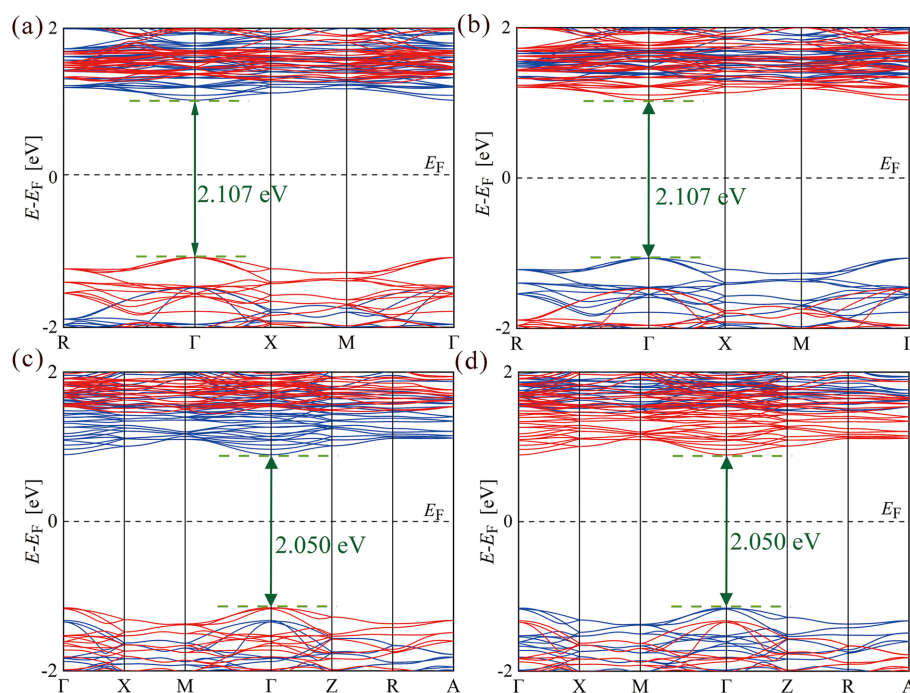


Figure 6

Electronic band structures of LFO along the high-symmetry paths: (a) $|\uparrow\rangle$ on Fe1 and $|\downarrow\rangle$ on Fe2 in the cubic model, (b) $|\uparrow\rangle$ on Fe2 and $|\downarrow\rangle$ on Fe1 in the cubic model, (c) $|\uparrow\rangle$ on Fe1a and Fe1b and $|\downarrow\rangle$ on Fe2 in the tetragonal model, and (d) $|\uparrow\rangle$ on Fe2 and $|\downarrow\rangle$ on Fe1a and Fe1b in the tetragonal model. The dashed lines at 0 eV indicate the Fermi energy. Spin-up and spin-down electrons are shown in red and blue, respectively.

(Kim *et al.*, 2022). The DFT calculations performed in the present study clearly show direct band gaps for these two different ferrimagnetic spin configurations. Their identical magnitudes can be understood by the intrinsic symmetry of the spin-polarized Hamiltonian. These two magnetic states, which are the reverse of each other, belong to the same energetic and electronic solution for a collinear spin alignment in the DFT calculations, and are related by a global spin-flip symmetry. This results in a physically equivalent charge-density distribution. Therefore, in the subsequent calculations, we focused on the one case with $|\uparrow\rangle$ in the *B* sublattice and $|\downarrow\rangle$ in the *A* sublattice. The direct band-gap energy of 2.050 eV for the tetragonal model in $P4_32_1$ is slightly narrower than the value of 2.107 eV for the cubic model in $P4_33_2$. These band gaps calculated for the microstructure-free LFO crystalline structure are larger than the band-gap energy measured on nest-like crystallites of LFO (about 1.9 eV; Zhang & Zhang, 2016) and the direct band gap probed with LFO nanoparticles (1.6 eV; Chireh & Naseri, 2019; Li *et al.*, 2020).

The spin-resolved projected density of states (PDOS) on each atomic site in the cubic and tetragonal models for LFO are graphically summarized in Fig. 7. The total DOS indicates the following:

(i) There is a strong hybridization between O 2*p* and Fe 3*d* orbitals in the valence bands. This hybridization is profound for the tetragonal model.

(ii) The conduction bands predominantly contribute to Fe 3*d* states. In particular, the conduction-band minimum is determined by the edge of the Fe 3*d* orbitals near the Fermi level. The spin-up electrons mainly originate from tetrahedral

Fe sites, whereas the spin-down electrons are associated with octahedral Fe sites. A noticeable contribution of O 2*p* to these spin states is also seen in the total DOS.

(iii) The PDOS profiles on sites Fe1a and Fe1b in the tetragonal model differ from each other. This accords with the symmetry splitting with respect to the Fe1 site in the cubic model. The total DOS in the tetragonal model shows that the conduction-band minimum is predominantly contributed to by Fe 3*d* on the octahedral sites Fe1a and Fe1b. The minimum energy level of Fe 3*d* on Fe2 is much further from the Fermi level. This is highly possible due to a higher localization of electrons. In contrast, the Fe 3*d* orbitals on both Fe1 and Fe2 contribute to the conduction-band minimum in the cubic model. A higher localization of Fe 3*d* electrons in the tetragonal structure of LFO is associated with its *U* parameters being slightly larger than those in the cubic model (Table 1).

The PDOS profiles of Fe 3*d* on tetrahedral and octahedral sites are compared in Fig. 8, where the tetragonal model reveals a redistribution and extension of the unoccupied spin-up states, rather than a simple energy shift. Moreover, the PDOS profile of Fe 3*d* states on Fe2 exhibits a noticeably broader energy distribution in the tetragonal model compared with that in the cubic one. This suggests an increase in the bandwidth due to the electronic delocalization enhanced on the tetrahedral site Fe2 [Fig. 8(a)]. This contradicts the *U* values for a higher localization of Fe 3*d* electrons in the tetragonal model, as mentioned above. This conflict could be explained by *p*–*d* hybridization effects competing with on-site Coulomb interactions. The stability of LFO in the tetragonal structure might be enhanced by hybridization towards strong

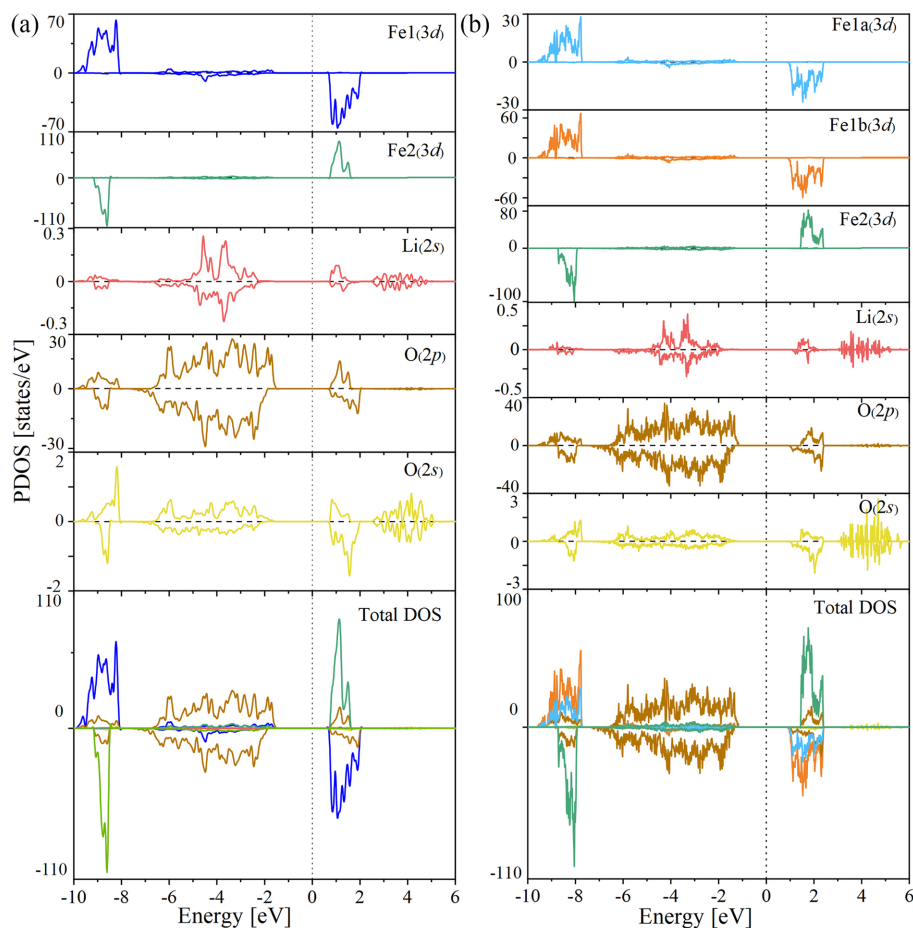


Figure 7

PDOS on all atomic sites of LFO, (a) in $P4_32$ and (b) in $P4_32_{12}$. The spin-up electron states are above 0 eV in the PDOS, while the spin-down electron states are below 0 eV. The Fermi energy levels are indicated with vertical dotted lines.

covalent interactions between Fe 3d electrons on Fe2 and O 2p electrons in the spin-up ferrimagnetic state.

4.3. Magnetic properties

The calculated cubic and tetragonal LFO models commonly show a total magnetization value of $2.5 \mu_B$ per formula unit,

consistent with experimentally determined values (Boyras *et al.*, 2011; Kumawat *et al.*, 2021; Liu *et al.*, 2019). The calculated magnetic moments of Fe^{3+} on the octahedral site Fe1 are larger than those on the tetrahedral site Fe2 in the cubic model (Table 3). Similarly, the magnetic moments of Fe^{3+} on Fe1a ($4.156 \mu_B$) and Fe1b ($4.149 \mu_B$) are larger than that of the

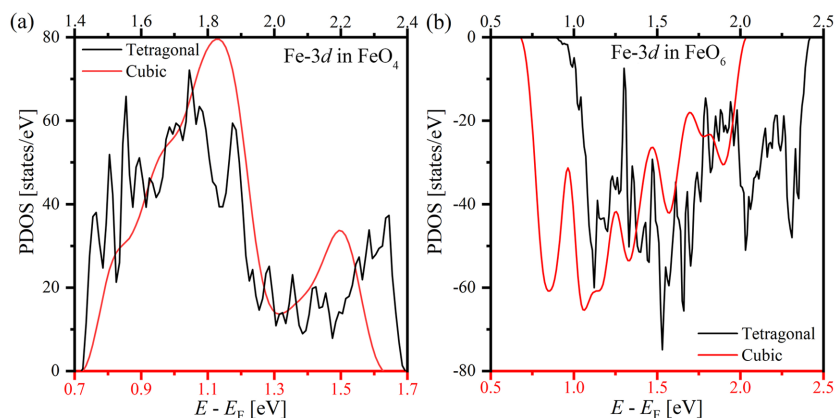


Figure 8

PDOS of the cubic model (red lines) against the $E - E_F$ scale at the bottom, compared with PDOS of the tetragonal model (black lines) against the $E - E_F$ scale at the top: (a) Fe 3d spin-up electrons on tetrahedral sites and (b) Fe 3d spin-down electrons on octahedral sites.

Table 3

Magnetic moments of Fe^{3+} along the c axis in the LFO structure models in $P4_332$ and $P4_32_12$, based on Bloch states.

Space group	Atomic site	Magnetic moment (μ_B)
$P4_332$	Fe1	4.156
	Fe2	−4.065
$P4_32_12$	Fe1a	4.156
	Fe1b	4.149
	Fe2	−4.044

tetrahedral site Fe2 (4.044 μ_B), corresponding to the high-spin state expected for LFO.

We calculated the long-range magnetic interactions in LFO, resulting in negligibly small values near zero up to the fifth order. Hence, in this study, only the nearest magnetic interactions were considered. As shown in Fig. 9, there are four different nearest magnetic exchange interactions, *i.e.* the face-diagonal interaction J_{AB} between Fe (A sublattice) and Fe (B sublattice), the space-diagonal term J'_{AB} , and J_{BB} and J'_{BB} which are the corresponding face- and space-diagonal terms within the B sublattice. Their magnitude ratios affect the degree and fashion of tetragonality in spinels (Menyuk *et al.*, 1962). Therefore, to clarify the origin of the tetragonal lattice distortion $a/c > 1$ feasible in the ground state of LFO (Table 1), these magnetic exchange interaction parameters were calculated using SSSP pseudopotentials plus PBEsol. The resulting magnetic exchange parameters are highly reliable, as the band structure calculated using Wannier functions matches well that using Bloch states (Fig. 10).

To construct the effective spin Hamiltonian, magnetic exchange interactions were first evaluated using the magnetic moments projected onto Fe alone. In this case, the resulting exchange parameters correspond to effective Fe–Fe interactions, where O is not treated as an independent magnetic degree of freedom; the magnetic interaction contribution of O,

Table 4

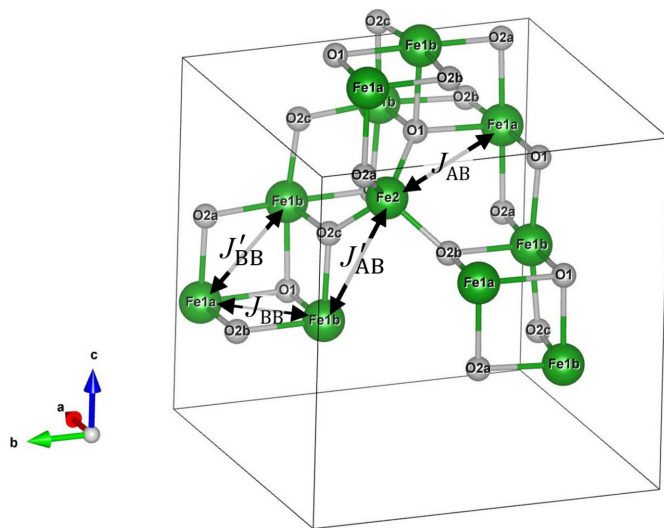
Effective magnetic exchange interaction parameters (meV) and magnetic moments (μ_B) in the tetragonal LFO model, calculated using Wannier functions.

Effective magnetic exchange interaction		Atomic site	Magnetic moment projected on Fe	Magnetic moment projected on Fe and O
J_{BB}	5.3791	Fe1a	4.3300	4.2455
J'_{BB}	5.1697	Fe1b	4.4070	4.2381
J_{AB}	−1.4478	Fe2	−4.9928	−4.1476
J'_{AB}	−1.8574	O1	−	0.1344
		O2a	−	0.0642
		O2b	−	0.0519
		O2c	−	0.0363

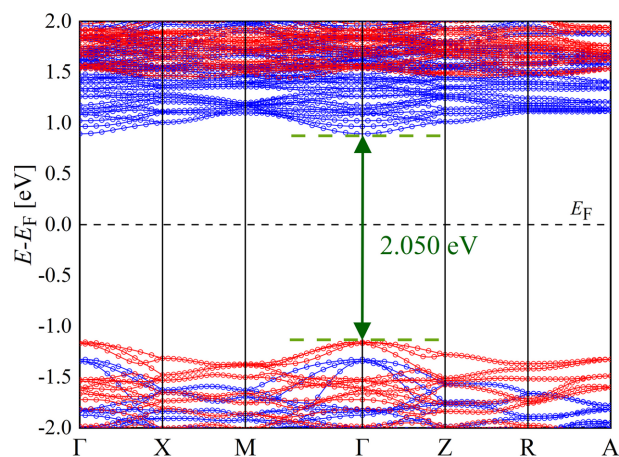
arising from Fe–O–Fe superexchange processes, is implicitly included in the effective Fe–Fe exchange parameters. The total Hamiltonian (H_{total}) is expressed by the static Hamiltonian H_{static} with the effective interaction J_{eff} ,

$$H_{\text{total}} = H_{\text{static}} - J_{\text{eff}}P, \quad (4)$$

where P is the spin exchange operator. This formalism allows us to derive a minimal Heisenberg model for the Fe sublattices. In another case, to prove the hybridization, particularly in Fe_2O_4 as mentioned in Section 4.2, the magnetic moments were projected onto both Fe and O, with the spin polarization of ligand oxygens explicitly taken into account. As a result, a part of the magnetic coupling, which was incorporated into the effective Fe–Fe interaction in the previous case, is redistributed into Fe–O–Fe superexchange terms. The contribution of Fe–O–Fe superexchange interactions to J_{eff} corresponds to p – d hybridization. The magnetic moments for both projection cases are listed in Table 4, along with the four nearest magnetic interactions calculated using Wannier functions. The magnitude differences on the octahedral sites Fe1a and Fe1b between the two projection cases are similar to each other. In contrast, the magnetic moment on

**Figure 9**

Nearest magnetic interaction paths of Fe–Fe within an interatomic distance of 3.4 Å in the calculated tetragonal model of LFO (see the main text).

**Figure 10**

Electronic band structures calculated by Wannier functions (circles) for the tetragonal model of LFO compared with those in Bloch states (lines; see also Fig. 6). Spin-up and spin-down electrons are presented in red and blue, respectively.

Fe2 projected onto Fe ($-4.9928 \mu_B$) is much larger than that from the projection onto both Fe and O ($-4.1476 \mu_B$). Furthermore, the latter is very similar to the magnetic moment magnitude on Fe2 calculated in Bloch states ($-4.044 \mu_B$). This large discrepancy (Δ) of $0.8542 \mu_B$ on Fe2 between the two projection cases points to strong Fe–O–Fe superexchange interactions in Fe₂O₄. The superexchange effect in Fe1aO₆ and Fe1bO₆ might be smaller, as indicated by the respective low Δ values of $0.0845 \mu_B$ and $0.1689 \mu_B$.

The suggested strong Fe–O–Fe superexchange interactions are consistent with the pronounced *p*–*d* hybridization observed in the PDOS of the Fe 3*d* and O 2*p* states (Fig. 7). The Goodenough–Kanamori–Anderson (GKA) rules (Anderson, 1959; Kanamori, 1957; Kanamori, 1959) state that the sign and strength of superexchange interactions are influenced by $\angle(\text{Fe–O–Fe})$ angles and the orbital occupancy. The superexchanges Fe1a($\uparrow$)–O($\uparrow$)–Fe1b($\uparrow$) and Fe1b($\uparrow$)–O($\uparrow$)–Fe1b($\uparrow$) occur over $\angle(\text{Fe–O–Fe}) \simeq 90^\circ$ (Table A.III). According to the GKA rules, the bonding geometry of $\angle(\text{Fe–O–Fe}) \simeq 90^\circ$ favours ferromagnetic interactions due to the orthogonality of the involved orbitals. In contrast, the Fe1b($\uparrow$)–O($\uparrow$)–Fe2($\downarrow$) and Fe1a($\uparrow$)–O($\uparrow$)–Fe2($\downarrow$) pathways exhibit $\angle(\text{Fe–O–Fe}) \gg 90^\circ$. This favours antiferromagnetic coupling due to maximal overlap of the half-filled Fe 3*d* orbitals with the O 2*p* orbitals. The inclusion of O 2*p* states accounts for the strong Fe 3*d*–O 2*p*–Fe 3*d* hybridization on Fe₂O₄, where superexchange interactions are significant.

When applying the Yafet–Kittel model for spinels, the magnetic energy for a tetragonal distortion can be characterized by three dimensionless parameters, which describe the relative exchange interaction strength and the magnetic anisotropy (Menyuk *et al.*, 1962). These parameters are defined as

$$u \equiv \frac{2(J_{BB} + J'_{BB})S_B}{(2J_{AB} + J'_{AB})S_A}, \quad (5)$$

$$v \equiv \frac{J_{BB}}{J_{BB} + J'_{BB}}, \quad (6)$$

$$w \equiv \frac{J'_{AB}}{2J_{AB} + J'_{AB}}. \quad (7)$$

The spin vectors $S_B = 5/2$ and $S_A = -5/2$ correspond to the respective high-spin states of Fe³⁺ in the *B* and *A* sublattices. Using the calculated parameters ($J_{BB} = 5.3791$ meV, $J'_{BB} = 5.1697$ meV, $J_{AB} = -1.4478$ meV and $J'_{AB} = -1.8574$ meV), the magnitudes of *u*, *v* and *w* were calculated to determine the tetragonality of LFO as follows:

$$u = \frac{2(5.3791 + 5.1697)(5/2)}{[2(-1.4478) + (-1.8574)](-5/2)} = \frac{2(10.5488)}{(4.7530)} \simeq 4.44 > 1, \quad (8)$$

$$v = \frac{5.3791}{5.3791 + 5.1697} \simeq 0.51 > \frac{1}{2}, \quad (9)$$

$$w = \frac{-1.8574}{2(-1.4478) + (-1.8574)} \simeq 0.39 < \frac{1}{2}. \quad (10)$$

The parameter $u > 1$ characterizes the relative strength of ferromagnetic interactions in the *B* sublattice in comparison with the antiferromagnetic interactions between the *A* and *B* sublattices. The condition $v > 1/2$ indicates that ferromagnetic interactions within the *B* sublattice are stronger on the *ab* plane than those parallel to the *c* direction; $w < 1/2$ implies that antiferromagnetic interactions between the *A* and *B* sublattices along the *c* axis are stronger than those in the *ab* plane. These parameters explain how a total energy lowering is realized by the tetragonal distortion $a > c$ in LFO, induced by decreasing ferromagnetic interactions and increasing antiferromagnetic interactions along *c*, simultaneously. This explains the feasibility of the tetragonal structure model for LFO with an *a/c* ratio slightly larger than 1 (Table 1). In summary, the tiny tetragonality of LFO has a magnetic origin from strong ferromagnetic interactions in the *B* sublattice competing with the anisotropy of antiferromagnetic interactions, including the Fe–O–Fe superexchange.

5. Conclusions

There has been a need to find the true structure of α -LiFe₅O₈ (denoted LFO) compatible with a collinear ferrimagnetic spin arrangement on the crystallographic *c* axis. Thus, the aim of this study was to determine the true ground-state symmetry of LFO to re-examine and re-interpret its structural and physical properties. In a comprehensive first-principles investigation on the atomic, electronic and magnetic structures of LFO, we have dealt with two structural models in the cubic space group *P*4₃32, which has been known to be true for LFO to date, and in its tetragonal subgroup *P*4₃2₁2. Results from SSSP+PBEsol+*U* calculations suggest that the tetragonal lattice is more favourable than the ideal cubic one within the calculation accuracy.

Our magnetic structure simulations show that LFO inherits the ferrimagnetic structure in the tetragonal magnetic space group *P*4₃2'₁2', which is consistent with the refined model from our recent analysis with HRNPD (to be published). This ferrimagnetic structure model comprises a spin-down tetrahedral diamond lattice (*A* sublattice) and a spin-up octahedral hyper-Kagome lattice (*B* sublattice). The calculated nearest-neighbour effective magnetic exchange interactions indicate that ferromagnetic interactions dominate over antiferromagnetic interactions in this ferrimagnetic arrangement. The new and insightful findings in this relevant prototype for MF-PVs are highlighted as follows:

- (i) there is a tiny tetragonal lattice distortion $a/c > 1$ with $a = 8.3528$ (2) Å and $c = 8.3511$ (1) Å;
- (ii) this faint tetragonality has a magnetic origin;
- (iii) the Fe–O–Fe magnetic superexchange interaction is consistent with Fe 3*d*–O 2*p* hybridization, particularly strong in Fe₂O₄ tetrahedra;

(iv) the direct band gap of 2.050 eV for microstructure-free LFO in the tetragonal model is smaller than that in the cubic model.

As a relevant aspect, the extremely small tetragonality of LFO (and its variants) might be considered for developing LFO-based MF-PVs. Slight shifts in atomic positions caused by magnetic interactions in (multi)ferroic structures could be verified with greater accuracy at sub-picometre resolution using new resonant X-ray diffraction techniques (Richter *et al.*, 2018), *e.g.* YMn_2O_5 (Weigel *et al.*, 2024). On the other hand, to realize LFO-type PVs in industrial applications, further experimental and theoretical work is necessary to determine their dynamic magnetic charges (Ye & Vanderbilt, 2014) and the shift-current response under external fields (Jalaja & Dutta, 2015; Gong *et al.*, 2024; Burger *et al.*, 2019).

Last but not least, the present study could serve as a basis for enhancing PV effectiveness for LFO-type MFs. For instance, Al-doped LFO solid-solution compounds that exhibit a relatively strong tetragonal distortion (Inckemann *et al.*, 2025) show significantly lower band-gap energies. Details will be reported elsewhere.

Acknowledgements

The authors acknowledge the Leibniz Supercomputing Center at Garching near Munich, Germany. Open access funding enabled and organized by Projekt DEAL.

Funding information

Yaoshi Cheng thanks the China Scholarship Council for financial support under grant No. 202204890001.

References

- Anderson, P. W. (1959). *Phys. Rev.* **115**, 2–13.
- Anisimov, V. I., Zaanen, J. & Andersen, O. K. (1991). *Phys. Rev. B* **44**, 943–954.
- Astafyev, A., Lysenko, E. & Surzhikov, A. (2019). *J. Therm. Anal. Calorim.* **136**, 441–445.
- Babrekar, M. & Jadhav, K. (2017). *Int. Res. J. Sci. Eng. Spec.* **1**, 73–76.
- Bagayoko, D. (2014). *AIP Adv.* **4**, 127104.
- Baroni, S., de Gironcoli, S., Dal Corso, A. & Giannozzi, P. (2001). *Rev. Mod. Phys.* **73**, 515–562.
- Birch, F. (1947). *Phys. Rev.* **71**, 809–824.
- Boyraz, C., Mazumdar, D., Iliev, M., Marinova, V., Ma, J., Srinivasan, G. & Gupta, A. (2011). *Appl. Phys. Lett.* **98**, 012507.
- Burger, A. M., Agarwal, R., Aprelev, A., Schrubba, E., Gutierrez-Perez, A., Fridkin, V. M. & Spanier, J. E. (2019). *Sci. Adv.* **5**, eaau5588.
- Chireh, M. & Naseri, M. (2019). *Adv. Powder Technol.* **30**, 952–960.
- de Picciotto, L. A. & Thackeray, M. M. (1986). *Mater. Res. Bull.* **21**, 583–592.
- Dharmadhikari, V. S. & Grannemann, W. W. (1982). *J. Appl. Phys.* **53**, 8988–8992.
- Dhass, A. D., Kumar, R. S., Lakshmi, P., Natarajan, E. & Arivarasan, A. (2020). *Mater. Today Proc.* **22**, 330–334.
- Ederer, C. & Spaldin, N. A. (2006). *Phys. Rev. B* **74**, 024102.
- Giannozzi, P., Andreussi, O., Brumme, T., Bunau, O., Buongiorno Nardelli, M., Calandra, M., Car, R., Cavazzoni, C., Ceresoli, D., Cococcioni, M., Colonna, N., Carnimeo, I., Dal Corso, A., de Gironcoli, S., Delugas, P., DiStasio, R. A., Ferretti, A., Floris, A., Fratesi, G., Fugallo, G., Gebauer, R., Gerstmann, U., Giustino, F., Gorni, T., Jia, J., Kawamura, M., Ko, H. Y., Kokalj, A., Küçükbenli, E., Lazzeri, M., Marsili, M., Marzari, N., Mauri, F., Nguyen, N. L., Nguyen, H. V., Otero-de-la-Roza, A., Paulatto, L., Poncé, S., Rocca, D., Sabatini, R., Santra, B., Schlipf, M., Seitsonen, A. P., Smogunov, A., Timrov, I., Thonhauser, T., Umari, P., Vast, N., Wu, X. & Baroni, S. (2017). *J. Phys. Condens. Matter* **29**, 465901.
- Giannozzi, P., Baroni, S., Bonini, N., Calandra, M., Car, R., Cavazzoni, C., Ceresoli, D., Chiarotti, G. L., Cococcioni, M., Dabo, I., Dal Corso, A., de Gironcoli, S., Fabris, S., Fratesi, G., Gebauer, R., Gerstmann, U., Gougoussis, C., Kokalj, A., Lazzeri, M., Martin-Samos, L., Marzari, N., Mauri, F., Mazzarello, R., Paolini, S., Pasquarello, A., Paulatto, L., Sbraccia, C., Scandolo, S., Sclauzero, G., Seitsonen, A. P., Smogunov, A., Umari, P. & Wentzcovitch, R. M. (2009). *J. Phys. Condens. Matter* **21**, 395502.
- Gong, Z., Xun, Y., Qian, Z., Chang, K., Qi, J. & Wang, H. (2024). *Phys. Rev. B* **110**, 094408.
- Gridnev, V. N., Krichevskov, B. B., Pavlov, V. V. & Pisarev, R. V. (1997). *JETP Lett.* **65**, 68–73.
- He, X., Helbig, N., Verstraete, M. J. & Bousquet, E. (2021). *Comput. Phys. Commun.* **264**, 107938.
- Huang, X., Lan, N., Chen, W., Yan, Y., Zeng, W. & Liu, S. (2021). *J. Mater. Sci.* **56**, 8334–8357.
- Iliev, M. N., Ivanov, V. G., Todorov, N. D., Marinova, V., Abrashev, M. V., Petrova, R., Wang, Y.-Q. & Litvinchuk, A. P. (2011). *Phys. Rev. B* **83**, 174111.
- Inckemann, S., Park, S.-H., Arauzo, A. & Avdeev, M. (2025). *J. Solid State Chem.* **347**, 125325.
- Jalaja, M. A. & Dutta, S. (2015). *Adv. Mater. Lett.* **6**, 568–584.
- Ji, W., Yao, K. & Liang, Y. C. (2010). *Adv. Mater.* **22**, 1763–1766.
- Kanamori, J. (1957). *Prog. Theor. Phys.* **17**, 197–222.
- Kanamori, J. (1959). *J. Phys. Chem. Solids* **10**, 87–98.
- Kim, S.-Y., Kim, K.-S., Jong, U.-G., Kang, C.-J., Ri, S.-C. & Yu, C.-J. (2022). *RSC Adv.* **12**, 15973–15979.
- Kinoshita, Y., Kida, N., Sotome, M., Miyamoto, T., Iguchi, Y., Onose, Y. & Okamoto, H. (2016). *ACS Photonics* **3**, 1170–1175.
- Koch, W. T. H., Munser, R., Ruppel, W. & Würfel, P. (1975). *Solid State Commun.* **17**, 847–850.
- Kumawat, K. K., Jain, A., Meena, S. S. & Yusuf, S. M. (2021). *J. Alloys Compd.* **865**, 158849.
- Lee, E. & Son, J. Y. (2024). *ACS Mater. Lett.* **6**, 4248–4254.
- Li, H., Wang, X., Zhou, P., Wu, H., Zhong, C., Dong, Z. & Liu, J. (2020). *J. Alloys Compd.* **821**, 153199.
- Lichtenstein, A. I., Katsnelson, M. I., Antropov, V. P. & Gubanov, V. A. (1987). *J. Magn. Magn. Mater.* **67**, 65–74.
- Liu, R., Pan, L., Peng, S., Qin, L., Bi, J., Wu, J., Wu, H. & Ye, Z.-G. (2019). *J. Mater. Chem. C* **7**, 1999–2004.
- Mahapatra, M., Pati, D. K., Sahu, B., Sahoo, P. K., Parida, R. K., Parida, B. N. & Padhee, R. (2024). *J. Mater. Sci. Mater. Electron.* **35**, 582.
- Massoudi, J., Bouekkeze, D., Bougoffa, A., Khirouni, K., Dhahri, E. & Bessais, L. (2020). *Adv. Powder Technol.* **31**, 4714–4730.
- Menyuk, N., Dwight, K., Lyons, D. & Kaplan, T. A. (1962). *Phys. Rev.* **127**, 1983–1996.
- Mercier, M., Velleaud, G. & Puvinel, J. (1977). *Phys. B+C* **86–88**, 1089–1090.
- Mohapatra, P. P. & Dobbidi, P. (2023). *Appl. Surf. Sci.* **619**, 156706.
- Moore, G. C., Horton, M. K., Linscott, E., Ganose, A. M., Siron, M., O'Regan, D. D. & Persson, K. A. (2024). *Phys. Rev. Mater.* **8**, 014409.
- Nakamura, M., Chan, Y.-H., Yasunami, T., Huang, Y.-S., Guo, G.-Y., Hu, Y., Ogawa, N., Chiew, Y., Yu, X., Morimoto, T., Nagaosa, N., Tokura, Y. & Kawasaki, M. (2024). *Nat. Commun.* **15**, 9672.
- Neaton, J. B., Ederer, C., Waghmare, U. V., Spaldin, N. A. & Rabe, K. M. (2005). *Phys. Rev. B* **71**, 014113.
- Nechache, R., Harnagea, C., Li, S., Cardenas, L., Huang, W., Chakrabarty, J. & Rosei, F. (2015). *Nat. Photonics* **9**, 61–67.

- Pachauri, N., Khodadadi, B., Althammer, M., Singh, A. V., Loukya, B., Datta, R., Iliev, M., Bezmaternykh, L., Gudim, I., Mewes, T. & Gupta, A. (2015). *J. Appl. Phys.* **117**, 233907.
- Peng, Y.-T., Chiou, S.-H., Hsiao, C.-H., Hao Ouyang, C. & Tu, C.-S. (2017). *Sci. Rep.* **7**, 45164.
- Perdew, J. P., Burke, K. & Ernzerhof, M. (1996). *Phys. Rev. Lett.* **77**, 3865–3868.
- Perdew, J. P., Ruzsinszky, A., Csonka, G. I., Vydrov, O. A., Scuseria, G. E., Constantin, L. A., Zhou, X. & Burke, K. (2008). *Phys. Rev. Lett.* **100**, 136406.
- Pizzi, G., Vitale, V., Arita, R., Blügel, S., Freimuth, F., Géranton, G., Gibertini, M., Gresch, D., Johnson, C., Koretsune, T., Ibañez-Azpiroz, J., Lee, H., Lihm, J. M., Marchand, D., Marrazzo, A., Mokrousov, Y., Mustafa, J. I., Nohara, Y., Nomura, Y., Paulatto, L., Poncé, S., Ponweiser, T., Qiao, J., Thöle, F., Tsirkin, S. S., Wierzbowska, M., Marzari, N., Vanderbilt, D., Souza, I., Mostofi, A. A. & Yates, J. R. (2020). *J. Phys. Condens. Matter* **32**, 165902.
- Prandini, G., Marrazzo, A., Castelli, I. E., Mounet, N. & Marzari, N. (2018). *npj Comput. Mater.* **4**, 72.
- Rangel, T., Fregoso, B. M., Mendoza, B. S., Morimoto, T., Moore, J. E. & Neaton, J. B. (2017). *Phys. Rev. Lett.* **119**, 067402.
- Richter, C., Zschornak, M., Novikov, D., Mehner, E., Nentwich, M., Hanzig, J., Gorfman, S. & Meyer, D. C. (2018). *Nat. Commun.* **9**, 178.
- Song, Z., Li, Z., Zhang, J., Chen, Z., Suescun, L. & Liu, Q. (2023). *J. Appl. Cryst.* **56**, 884–888.
- Song, Z. & Liu, Q. (2019). *Phys. Chem. Chem. Phys.* **21**, 2372–2377.
- Soreto, S., Graça, M., Valente, M. & Costa, L. (2017). *Magnetic Spinel – Synthesis, Properties and Applications*, pp. 31–50. InTech.
- Sousa, O. M., Araujo, R. S. & Freitas, S. M. (2019). *Comput. Theor. Chem.* **1159**, 27–30.
- Tan, S., Zhang, W., Jiao, F., Zhou, Y., Yang, L., Shi, W. & Wang, Z. (2021). *Cryst. Res. Technol.* **56**, 2100076.
- Teixeira, S. S., Graça, M. P. F., Lucas, J., Valente, M. A., Soares, P. I. P., Lança, M. C., Vieira, T., Silva, J. C., Borges, J. P., Jinga, L.-I., Socol, G., Mello Salgueiro, C., Nunes, J. & Costa, L. C. (2021). *Nanomaterials* **11**, 193.
- Timrov, I., Marzari, N. & Cococcioni, M. (2022). *Comput. Phys. Commun.* **279**, 108455.
- von Baltz, R. & Kraut, W. (1981). *Phys. Rev. B* **23**, 5590–5596.
- Vopson, M. M. (2015). *Crit. Rev. Solid State Mater. Sci.* **40**, 223–250.
- Weigel, T., Richter, C., Nentwich, M., Mehner, E., Garbe, V., Bouchenoire, L., Novikov, D., Meyer, D. C. & Zschornak, M. (2024). *Phys. Rev. B* **109**, 054101.
- Xiao, R.-C., Jin, Y. J. & Jiang, H. (2023). *APL Mater.* **11**, 070903.
- Xu, H., Wang, H., Zhou, J. & Li, J. (2021). *Nat. Commun.* **12**, 4330.
- Yang, S. Y., Martin, L. W., Byrnes, S. J., Conry, T. E., Basu, S. R., Paran, D., Reichertz, L., Ihlefeld, J., Adamo, C., Melville, A., Chu, Y., Yang, C., Musfeldt, J. L., Schlom, D. G., Ager, J. W. III & Ramesh, R. (2009). *Appl. Phys. Lett.* **95**, 062909.
- Ye, M. & Vanderbilt, D. (2014). *Phys. Rev. B* **89**, 064301.
- Young, S. M., Zheng, F. & Rappe, A. M. (2013). *Phys. Rev. Lett.* **110**, 057201.
- Zhang, D. & Zhang, L. (2016). *New J. Chem.* **40**, 7171–7180.