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Structure and magnetic properties of W-type hexaferrites

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W-type hexaferrites (WHFs) (Sr Me_2 Fe₁₆O₂₇, Me = Mg, Co, Ni and Zn) are hard magnetic materials with high potential for permanent magnet applications owing to their large crystalline anisotropy and high cation tunability. However, little is known with regards to their complex structural and magnetic characteristics. Here, the substitution of metals (Me = Mg, Co, Ni and Zn) in WHFs is described and their crystal and magnetic structures investigated. From joined refinements of X-ray and neutron powder diffraction data, the atomic positions of the Me atoms were extracted along with the magnetic dipolar moment of the individual sites. The four types of WHFs exhibit ferrimagnetic ordering. For Mg, Ni and Zn the magnetic moments are found to be ordered colinearly and with the magnetic easy axis along the crystallographic c axis. In SrCo₂Fe₁₆O₂₇, however, the spontaneous magnetization changes from uniaxial to planar, with the moments aligning in the crystallographic ab plane. Macromagnetic properties were measured using a vibration sample magnetometer. The measured saturation magnetization (M_s) between the different samples follows the same trend as the calculated $M_{\rm s}$ extracted from the refined magnetic moments of the neutron powder diffraction data. Given the correlation between the calculated $M_{\rm s}$ and the refined substitution degree of the different Me in specific crystallographic sites, the agreement between the measured and calculated M_s values consolidates the robustness of the structural and magnetic Rietveld model.

1. Introduction

Permanent magnetic materials are a keystone in society today (Jacobs, 1969). Magnets have enabled the information age by providing data storage (Coey, 2001) and allowed generation of electricity through induction (Lewis & Jiménez-Villacorta, 2013). Given their importance in the modern world, there is a big incentive to improve magnets and the most common figure of merit for magnetic materials is the energy product (BH_{max}) (McCallum et al., 2014), i.e. the largest rectangle in the second quadrant under the BH curve in the hysteresis loop (O'Handley, 2000). Nd₂Fe₁₄B magnets have been the best performing magnets since they were discovered in 1982 and have continuously improved through laborious developments (Brown et al., 2002). Tightened export quotas of rare-earth elements (REE) in late 2009 (Bomgardner, 2015; Wübbeke, 2013), along with health concerns working with REE ores (International Atomic Energy Agency, 2011; Pagano et al. 2015) have prompted research of REE-free magnets. A widespread REE-free permanent magnet is the M-type hexaferrite (MHF), e.g. BaFe₁₂O₁₉ or SrFe₁₂O₁₉ (Pullar, 2012). Although not as powerful as Nd₂Fe₁₄B magnets, ferrites are the most abundantly used permanent magnetic material throughout the world as a result of their low cost, high stability and adequate performance in many applications (Pullar, 2012; Lewis & Jiménez-Villacorta, 2013). Recent work has focused on optimizing the performance of MHFs through bottom-up nanostructuring and morphology control (Eikeland *et al.*, 2018, 2017; Saura-Muzquiz *et al.*, 2016, 2018). However, there is still a huge gap in performance between REE magnets and REE-free magnets (Coey, 2012), encouraging further research into REE-free magnets. Although the MHF is the best known and widely commercially used ferrite, several other hexaferrite structures exist of which an overview can be found in the review by Pullar (2012). One of the promising candidates is the W-Type hexaferrite (WHF): $A^{2+}Me^{2+}Fe_{16}^{3+}O_{27}$, (A =alkaline earth metal, *e.g.* Sr or Ba, Me = Mg, Mn, Fe, Co, Ni, Cu, Zn). These hexaferrites hold great potential as future permanent magnetic materials owing to the wide tuneability of their magnetic properties.

In the common descriptions of hexagonal ferrites (Pullar, 2012; Smit & Wijn, 1959), R denotes a layer including an A ion, whereas S describes a layer with a spinel-like structure, and an asterisk (*) next to R or S denotes layers that are rotated 180° around the crystallographic *c* axis. According to this notation, the MHF can be described as RSR*S*, whereas WHFs can be described by introducing an additional S layer into the MHF structure, resulting in the structure RSSR*S*S*. Fig. 1 shows the WHF structure along with the obtained magnetic structures from neutron powder diffraction investigations obtained in this work. The introduction of the additional spinel layer in the WHF structure gives rise to a $6g_{oct}$ site which lies on the SS boundary. In the most conventional spin-arrangement (Gorter, 1957), the ratio of sites with parallel/antiparallel spin alignment per unit cell is maintained. WHF $(12\uparrow/6\downarrow)$ compared with MHF $(8\uparrow/4\downarrow)$ has a 50% net increase in parallel magnetic sites whereas the unit cell volume only increases by roughly 42-44% depending on Me substitution. Introduction of the additional S layer necessitates two Me^{2+} ions in the structure, as in spinel-ferrites, which enable



Figure 1

(a) Unit cell and seven distinct Me sites for $SrMe_2Fe_{16}O_{27}$ (Me = Mg, Co, Ni, Zn), the green spheres are Sr and the colored polyhedra show the Me sites and their relative coordination. Oxygen atoms at the corners of polyhedra are not shown. (b) Uniaxial magnetic ordering of $SrMg_2Fe_{16}O_{27}$, $SrNi_2Fe_{16}O_{27}$ and $SrZn_2Fe_{16}O_{27}$. (c) Planar magnetic ordering of $SrCo_2Fe_{16}O_{27}$. The brackets in the right part of the figure indicates the common hexaferrite layers. Figure prepared using VESTA (Momma & Izumi, 2011).

the direct substitution of magnetic or nonmagnetic Me^{2+} ions without the need for additional substitutions to maintain the charge balance (Tokunaga *et al.*, 2010; Wang *et al.*, 2012). This enhances the tuneability of WHF in comparison with MHF (Gorter, 1950; Andersen *et al.*, 2018). By partially substituting nonmagnetic species into specific sites, which align opposite to the net magnetization of the unit cell, *i.e.* the tetrahedrally coordinated 4*e* and 4*f* sites in the S layer and the octahedral 4*f* site in the R layer, the magnetization of the compound can be potentially increased with minimal reduction of the superexchange coupling in and between the layers (Lilot *et al.*, 1982; Ram & Joubert, 1991*b*).

Ease of synthesis has caused research to primarily focus on Ba-containing WHFs (Collomb et al., 1986a,c, 1988; Lotgering et al., 1980b; Ahmed et al., 2009; Cao et al., 2018), but considering Ba health concerns (Choudhury & Cary, 2001), further research towards Sr-containing WHFs is highly interesting as they are nontoxic (Kirrane et al., 2006). Some efforts have already gone into Sr-containing WHFs: Ram & Joubert (1991a) reached a saturation magnetization (M_s) of 91 A m² kg⁻¹ at room temperature by Li/Zn substitution (SrZnLi_{0.5}Fe_{16.5}O₂₇). This value is about 25% larger than achieved for MHFs. Toyota (1997) also reported a BH_{max} of 42 kJ m⁻³ for SrFe₁₈O₂₇, which is significantly higher than for pure $SrFe_{12}O_{19}$, normally limited to about 30 kJ m⁻³ (Eclipse Magnetics Ltd, 2019; Eikeland et al., 2017). Furthermore, WHFs have also revealed themselves as promising materials for other applications, e.g. multiferroics (Song et al., 2014; Kimura, 2012) or microwave absorbance (Ahmad et al., 2012). As Sr-containing hexaferrites in general do not always display the same cation distribution as Ba-containing hexaferrites (Albanese et al., 1973; Sizov, 1968), it is highly relevant and interesting to investigate both the cation distribution and subsequent magnetic ordering of the Sr-containing hexaferrites. Aside from the cation distribution, the crystallite size also has a huge impact on the macroscopic magnetic properties of hexaferrites (Gjørup et al., 2018) and is an important part of optimizing their performances.

This work involved rigorous investigation of $SrMe^{2+}Fe_{16}^{3+}O_{27}$ (where Me = Mg, Co, Ni and Zn) synthesized by high-temperature sol-gel synthesis. Neutron powder diffraction (NPD) data along with X-ray powder diffraction (PXRD) data extracted using both synchrotron and laboratory sources were analyzed with combined Rietveld refinements. The data obtained were used to determine the magnetic structure and the occupation of transition metal atoms in the hexaferrite structures. The crystallographic structural and magnetic data is compared with macroscopic magnetic measurements.

2. Experimental

2.1. Synthesis

 $SrMe_2Fe_{16}O_{27}$ particles were synthesized utilizing a sol-gel autocombustion method due to the high chemical homogeneity achieved in these syntheses (Hench & West, 1990).

The method contains three steps: first $Sr(NO_3)_2$, $Me(NO_3)_2 \cdot 6H_2O$ and $Fe(NO_3)_3 \cdot 9H_2O$ (all Sigma-Aldrich technical grade with purity >98%) were dissolved in demineralized water in stoichiometric molar ratios $[Sr^{2+}]:[Me^{2+}]:[Fe^{3+}] = 1:2:16$. Citric Acid was dissolved and added in equal ratio to the nitrates, $[C_8H_8O_7] = 2[Sr^{2+}] +$ $4[Me^{2+}] + 48[Fe^{3+}]$, under constant stirring. The solution was neutralized with NH₄OH and dried overnight in a convection oven at 100°C until a gel was formed. In the second step, the gel was fired in a preheated furnace at 350°C for 30 min until the autocombustion had finished and subsequently cooled to room temperature in air. Finally, the resulting powder was crushed and fired in a furnace at 1200°C (SrMg₂Fe₁₆O₂₇ and SrZn₂Fe₁₆O₂₇) or 1300°C (SrNi₂Fe₁₆O₂₇ and SrCo₂Fe₁₆O₂₇) according to the following heating scheme:

$$\mathrm{RT} \xrightarrow{600^{\mathrm{o}}\mathrm{C}\,\mathrm{h}^{-1}} 1000^{\mathrm{o}}\mathrm{C} \xrightarrow{100^{\mathrm{o}}\mathrm{C}\,\mathrm{h}^{-1}} 1200/1300^{\mathrm{o}}\mathrm{C},$$

with a holding time of 2 h before cooling to room temperature.

2.2. Powder diffraction

Four diffraction data sets were measured for all of the prepared WHF samples and were collected using the following instruments.

X-rays: (i) Synchrotron radiation at the MS beamline at SLS, PSI, Switzerland (Willmott *et al.*, 2013) with a chosen wavelength of $\lambda = 0.778$ Å. The samples were packed in 0.3 mm capillaries and the data were collected using Debye–Scherrer geometry and a Mythen-II detector in the angular range $2\theta = 2-121^{\circ}$. (ii) An in-house Rigaku SmartLab diffractometer equipped with a Co $K\alpha$ source ($K\alpha_1 = 1.789$ and $K\alpha_2 = 1.793$ Å). Data were collected in parallel-beam geometry in reflection mode in the angular range $2\theta = 18-90^{\circ}$ using a D/teX Ultra detector. The powder samples were packed on zero-background single-crystal silicon sample holders.

Neutrons: HRPT at SINQ, PSI, Switzerland (Fischer *et al.*, 2000) with chosen wavelengths (iii) $\lambda = 1.89$ Å and (iv) $\lambda = 2.45$ Å. The powders were packed in vanadium cans ($\phi = 6$ mm) and data were collected in Debye–Scherrer geometry with a large position-sensitive ³He detector having an angular range of $2\theta = 5-165^{\circ}$. During the measurement, the detector was shifted by 0.05° to enhance the number of data points across the diffraction peaks.

The four diffraction patterns were collected on the basis of their individual strengths, which improve the robustness of the analysis. The individual strengths attributed to each method are as follows. *X-rays*: (i) MS-SLS: high peak resolution and large coverage of *Q*-Space; (ii) Rigaku SmartLab: precise wavelengths ($K\alpha_1$ and $K\alpha_2$) for accurate *d* spacings and readily available for purity control, slight contrast effect due to Fe resonance $f'_{Fe}(\lambda)$. *Neutrons*: (iii) HRPT (1.89 Å): largest *Q*space coverage of neutrons; (iv) HRPT (2.45 Å): highest resolution at low *Q* of neutron powder diffraction data. All Rietveld refinements were carried out using the *FullProf Suite* software package (Rodríguez-Carvajal, 1993), where four combined refinements of four sets of diffraction patterns were performed. Fig. 2 shows the four different patterns with accompanying Rietveld refinements and difference plots for $SrCo_2Fe_{16}O_{27}$. The refinements of the other WHFs are shown in Figs. S1–S3 of the supporting information.

2.2.1. Technical refinement details. All four samples belong to the space group $P6_3/mmc$, unit-cell parameters and isotropic thermal parameters are given in Table S4 of the supporting information. The peak profile was described with the Thomson-Cox-Hastings pseudo-Voigt profile with axial divergence asymmetry (Thompson *et al.*, 1987), where Y and X are the only profile parameters refined relating to size and strain, respectively. Instrumental broadening effects were described by refinement of standards at the MS beamline [Si NIST 640D) with a Rigaku SmartLab diffractometer (LaB₆ NIST 660B)] and at HRPT (Na₂Al₁₂Ca₃F₁₄); all standards were measured under the same conditions as the samples. Figs. S10 and S11 show the pseudo-Voigt FWHM of the different instruments as a function of Q and 2θ . Peak-broadening resulting from the samples was constrained to be equal between the different patterns with Y constrained with respect to the difference in wavelength (X has no wavelength)dependency). For all patterns of a specific sample, the unit-cell parameters and atomic positions within the unit cells were constrained to be identical, as they should not vary across the different measurements carried out under equal conditions. A total of three isotropic thermal vibration parameters were used, describing Sr, Fe/Me and O. Fourier cosine series were used to model the neutron background in HRPT patterns, whereas Chebychev polynomials were used for the X-ray patterns. The refined magnetic moments between Fe³⁺, Co²⁺ and Ni²⁺ were constrained to maintain the ratio between the number of unpaired 3d electrons of the individual elements, assuming the orbit moment to be quenched, e.g. $\mu_{\rm B}({\rm Fe}^{3+}):\mu_{\rm B}({\rm Co}^{2+}) = 5:3$. The difference in neutron scattering length of Fe (9.45 fm) compared with Co (2.49 fm), Ni (10.6 fm) and Zn (5.68 fm) made it possible to determine the site occupancies of the neighboring Me elements in the refinements. The initial predictions for the Me occupancies were found by refining the structures with only Fe and looking at their individual thermal parameters. The occupancies were subsequently refined by introducing Me atoms into the sites



Four diffraction patterns and corresponding Rietveld refinements and difference spectra for SrCo₂Fe₁₆O₂₇. Inset is data from SLS-MS at PSI for $Q = 6.6-14 \text{ Å}^{-1}$.

where thermal parameters mostly deviated from the only Fe refinement, while constraining thermal parameters to be equal between all metal sites. The total site occupation is fixed to have all sites fully occupied.

2.3. Magnetic measurements

The magnetic hysteresis loops were measured using a physical properties measuring system (PPMS) from Quantum Design equipped with a vibrating sample magnetometer

(VSM). Powder samples of 20–45 mg masses were packed in cylindrical powder capsules (height 2.7–3.0 mm, diameter 3.5 mm), and measurements were performed at 300 K with an applied field of ± 3 T. The graphical (near) infinite slope method was used to correct the data for demagnetization, whereas the approach to saturation was used to extract $M_{\rm s}$ (O'Handley, 2000). Fig. 7 shows hysteresis loops for all four WHFs.

Curie temperatures ($T_{\rm C}$) were measured with thermogravimetry using an STA 449 F3 Jupiter from Netzsch. Pellets of 60–120 mg were measured with attached external magnets by ramping the temperature at a rate of 10 °C min⁻¹ in an Ar flow of 50 ml min⁻¹. The weight in % of the original mass is plotted against temperature in Fig. S4 and the extracted Curie temperatures are given in Fig. 4.

3. Results and discussion

3.1. Structural model

The results of the combined refinement of all diffraction patterns are shown for $SrCo_2Fe_{16}O_{27}$ in Fig. 2, while the refined model and patterns for $SrMe_2Fe_{16}O_{27}$ (*Me* = Mg, Ni and Zn) are given in Figs. S1–S3.

3.2. Purity and occupation

Refinements showed almost phase-pure samples, with small impurities of $CoFe_2O_4$ [5.27 (5) wt%], MgFe_2O_4 [8.24 (5) wt%], NiFe_2O_4 [7.08 (6) wt%] and ZnFe_2O_4 [5.28 (4) wt%] which were quantifiable by the high-quality data from SLS-MS. The SrMg_2Fe_{16}O_{27} sample additionally contained a small impurity of X-type hexaferrite Sr₂Mg₂Fe₂₈O₄₆ [8.77 (5) wt%]. A few undescribed peaks in this dataset stem from a small impurity that remains to be determined but is assumed to be very small based on the peak intensities. A selection of the impurity peaks are highlighted in Figs. S12–S16. The resulting structure of the WHF, with seven distinct *Me*–O polyhedra that make up the structure, is shown in Fig. 1(*a*)

The refined occupation fractions (%) of the full site occupancies are given in Fig. 3. The sites occupied by *Mes* are as follows: in all samples the bipyramidal 2*d* site is displaced from the center and described by a half-occupied 4*f* site in the Rlayer $[4f_{\text{\frac{5}{bi}(R)}}]$, this is exclusively occupied by Fe. Mg predo-

Structure	Site	Mg (%)	Co (%)	Ni (%)	Zn (%)
	$4f_{_{1/2}\text{bi}(R)}$	0	0	0	0
	$4f_{oct(S)}$	9.9(3)	8(2)	11(2)	2.8(9)
Î 👫 👗	6g _{oct(S-S)}	45.8(2)	33(1)	31(1)	0
	$4e_{tet(S)}$	3.5(3)	14(1)	12(2)	55.1(9)
	$4f_{tet(S)}$	2.4(3)	11(2)	5(2)	10.5(8)
	$4f_{oct(R)}$	0	2(2)	15(2)	0
	12k _{oct(R-S)}	0	8(1)	6(1)	3.5(6)
Sr <i>Me</i> ₂ Fe ₁₆ O ₂₇	Total <i>Me</i>	1.69(1)	2.2(1)	2.1(1)	1.9(1)

Figure 3

Overview of the occupation of Me_s given as a percentage of the full site occupancies and the total amount of Me for one formula unit $SrMe_2Fe_{16}O_{27}$.

minantly occupies octahedral S sites $[4f_{oct(S)} \text{ and } 6g_{oct(S-S)}]$ and, to a minor degree, tetrahedral S sites $[4e_{tet(S)} \text{ and } 4f_{tet(S)}]$. Zn predominantly occupies $4e_{tet(S)}$ and $4f_{tet(S)}$ with a minor occupation of $4f_{oct(S)}$ and $12k_{oct(R-S)}$. Co and Ni are spread across all sites, but the predominantly occupied site for both is $6g_{oct(S-S)}$. For SrNi₂Fe₁₆O₂₇, the data does not support unambiguous determination of *Me* occupation, given the poor scattering length contrast between Ni and Fe for both neutrons and X-rays. Comparisons between the occupancies obtained in this work and in previously reported studies (Collomb *et al.*, 1986*c*,*a*; Graetsch *et al.*, 1986) of (Ba/ Sr) Me_2 Fe₁₆O₂₇ (Me = Mg, Zn and Co) are given in Tables S1– S3.

3.3. Magnetic structure

From the NPD data, it is also possible to determine the magnetic structure as the spin of the neutrons interacts with the atomic magnetic dipole moment. Gorter (1957) proposed ferrimagnetic ordering in WHFs with the $4e_{tet(S)}$, $4f_{tet(S)}$ and $4f_{oct(R)}$ sites ordering antiparallel to the remaining sites and this has been demonstrated for BaCo₂Fe₁₆O₂₇ using neutron scattering (Collomb *et al.*, 1986*b*). The magnetic and crystallographic unit cells are coinciding and three possibilities for the orientation of the magnetic moments have been considered for the obtained data: (i) uniaxial ordering where the magnetic moments are ordered along the crystallographic *c* axis, (ii) planar ordering where the magnetic ordering is in the *ab* plane and (iii) conical ordering where the moments have an angle with respect to the *c* axis.

The propagation vector is zero ($\mathbf{k} = 0$) as no additional magnetic peaks can be seen aside from those coinciding with the crystallographic Bragg peaks. Consequently, a collinear magnetic model is imposed with the same unit cell size as the crystallographic unit cell. For Mg, Ni and Zn there is no magnetic contribution to the (00*l*) reflections, indicating that the magnetic moment is aligned along the *c* axis. As the magnetic scattering is given by $\hat{\mathbf{q}} \times (\mathbf{s}_j \times \hat{\mathbf{q}}) \equiv \mathbf{s}_{j,\perp}$ (Lefmann, 2017; Marshall & Lovesey, 1971), showing that the spin component on site j (\mathbf{s}_j) is only visible if it is perpendicular to the scattering vector $\hat{\mathbf{q}}$. Refinement of the neutron diffraction data confirms that $SrMe_2Fe_{16}O_{27}$ (Me = Mg, Ni and Zn) can be satisfactorily described by a ferrimagnetic model, having the magnetic moments aligned along the *c* axis. The parallel and antiparallel alignment of magnetic moments on the different sites follows what is described by Gorter (1957) and as a result the magnetic structure belongs to the magnetic space group (Shubnikov group) $P6_3/mm'c'$. The magnetic contribution to the diffraction signal is highlighted by a red line for $SrZn_2Fe_{16}O_{27}$ and shown in Fig. 5, where the inset highlights the different magnetic signals for the respective WHFs. The refined magnetic moments from NPD were summed over all sites in the unit cell and divided by the mass of the unit cells as an estimation of M_s . Fig. 4 shows the refined moments for each crystallographic site and the M_s calculated from $M_{s,NPD}$, as well as the M_s extracted from the macroscopic magnetic measurement ($M_{s,VSM}$) for comparison. In Figs. S5–S7, the refined magnetic contribution for all WHFs is shown in greater detail. A table of the refined positions of all seven Me sites

Magnetic Structure	Site	${ m SrMg_2Fe_{16}O_{27}}\ (\mu_{ m B})$	${ m SrCo_2Fe_{16}O_{27}}\ (\mu_{ m B})$	${ m SrNi_2Fe_{16}O_{27}}\ (\mu_{ m B})$	$SrZn_2Fe_{16}O_{27}$ (μ_B)
	4f	3.6(4)	2.4(1)	3.7(5)	3.5(6)
	$4f_{oct(S)}$	3.6(3)	3.2(1)	3.8(4)	3.4(5)
	$6g_{oct(S-S)}$	2.0(6)	2.9(1)	3.1(5)	2.4(5)
	$4e_{tet(S)}$	-4.5(4)	-4.2(1)	-5.0(5)	-1.6(1.3)
	$4f_{tet(S)}$	-4.2(3)	-3.3(1)	-4.3(3)	-3.5(6)
	$4f_{oct(R)}$	-4.7(3)	-3.8(1)	-4.6(4)	-4.4(4)
	12k _{oct(R-S)}	3.5(2)	3.5(1)	3.5(3)	2.9(3)
M _{s,NPD} (Am ² /kg)		41(3)	59(3)	50(5)	57(5)
M _{s.VSM} (Am²/kg)		60.3(1)	78.5(1)	67.7(1)	76.2(1)
Wt% Corrected M _s (Am ² /kg)		63.7(2)	78.4(2)	69.0(3)	80.4(3)
T_C (°C)		435(2)	470(3)	500(2)	345(3)

Figure 4

Average magnetic moments of the seven Me sites, the derived M_s , the M_s from the macroscopic hysteresis measurement and the Curie temperature from thermogravimetry. Correction of M_s is with respect to impurity phases and is discussed in Section S8



Figure 5

NPD data for $SrZn_2Fe_{16}O_{27}$ measured at SINQ-HRPT $\lambda = 2.45$ Å. The contribution from the refined magnetic scattering is shown in red. The inset shows a comparison between the four different samples in the range Q = 1.0-3.5 Å⁻¹ with the two dominant magnetic peaks indicated at 006 and 101 (see Fig. S9 for a magnified version). $SrCo_2Fe_{16}O_{27}$ differs from the other three WHFs by having planar magnetic ordering instead of uniaxial ordering.

along with the seven O sites is given in Table S5.

While the PXRD is similar for the four different structures, there is a clear difference in the NPD in Fig. 5, the reason being that the magnetic structure of $SrCo_2Fe_{16}O_{27}$ differs from the other WHFs. The strong peak at 1.15 Å⁻¹ is the (006) peak. This peak dictates a component of the magnetic moment in the crystallographic *ab* plane, as magnetic scattering is perpendicular to the direction of the magnetic moment. Previous studies (Samaras *et al.*, 1989; Collomb *et al.*, 1986; Graetsch *et al.*, 1984; Lotgering *et al.*, 1961; Paoluzi *et al.*, 1988; Asti *et al.*, 1978) of BaCo₂Fe₁₆O₂₇ have reported either a conical magnetic structure with an angle to the *c* axis of 69–71°, or planar magnetic ordering. To the best of our knowledge, the most accepted magnetic structure is the conical ordering reported by Samaras *et al.* (1989); however, the

structure is known to be complex and changes as a function of both temperature and exact Co substitution, as described by Yamzin et al. (1966). To investigate the direction of the magnetic moment in SrCo₂Fe₁₆O₂₇, the refinements were carried out with the orientation of the atomic magnetic dipolar moment varying in angle with respect to the c axis in steps of 5° from uniaxial (0° from the *c* axis) to planar (90° from the c axis). The magnetic ordering conical to, or in the *ab* plane defined by the magnetic space groups belonging under the crystallographic space group $P6_3/$ mmc have either 0 moment or are antiparallel within each Wyckoff site based on the magnetic space groups (Litvin, 2013). As a result, the symmetry restrictions on the magnetic moments were reduced, so each Wyckoff site must be equal and pointing in the same direction, but is not dictated by any magnetic group of P63/mmc. In Section S5 of the supporting information, further discussion on the magnetic symmetry is given. A graph of the resulting R factors is shown in Fig. 6 while the decreases in R factors are small between conical and planar ordering, the lowest value is for 90° and there is no evidence to suggest that the ordering should be conical rather than planar. SrCo₂Fe₁₆O₂₇ must belong to a lower symmetry space group than $P6_3/$ *mmc*, which remains to be determined.

Refinements were carried out with two options for peak broadening: (i) peak broadening of the magnetic contribution is constrained to be equal to that of the peak broadening of the nuclear contribution and (ii) the peak



Resulting R factors from the refinements where the only difference is the constrained angle for the magnetic ordering with respect to the c axis. The R factors shown are for the refined model of the data measured at SINQ-HRPT at PSI.

broadening for magnetic and nuclear contribution are independent. If the Lorentzian peak broadening attributed to size (Y) is refined independently, the refined broadening of the magnetic diffraction signal is an order of magnitude larger than the refined broadening of the nuclear contribution, along with a small decrease in R_{wp} . A comparison of the peak width parameter Y along with the resulting R_{wp} is given in Table S7. The peak width originating from the crystal structure is very sharp suggesting the crystallite sizes to be out of resolution, on the other hand, the magnetic scattering has a size corresponding to ~ 100 nm, which can be explained by the crystallites breaking up into smaller magnetic domains. The underlying property responsible for the difference in magnetic ordering between $SrMe_2Fe_{16}O_{27}$ (Me = Mg, Ni, Zn) compared with SrCo₂Fe₁₆O₂₇ is a strong coupling of the unquenched orbital momentum present in Co and the trigonal axis of the crystal field, which has been described previously for cobaltsubstituted magnetite (Slonczewski, 1961, 1958) and BaCo₂Fe₁₆O₂₇ (Bickford, 1962). Samaras et al. (1989) have investigated the magnetic ordering for BaCo₂Fe₁₆O₂₇ with an occupation of 70% Co^{2+} in the $6g_{oct}$ site, which exhibits the trigonal axis around it. They concluded that there is a tilt of the magnetic moment between 69 and 71°, *i.e.* conical ordering in contrast to the planar ordering (90°) observed here. A small difference in the total Co content $[1.87 (7) \text{ Co}^{2+} \text{ per formula}]$ unit (f.u.) versus 2.2 (1) Co²⁺ per f.u.] could possibly explain this discrepancy as the amount of Co has previously been shown to alter the tilt of the magnetic moments (Yamzin et al., 1966).

In comparison with the theoretical work on superexchange interactions by Lilot *et al.* (1982), the $T_{\rm C}$ of the four samples can be related to how important each individual site is in the exchange interaction: $4e_{\rm tet} > 4f_{\rm tet} > [12k_{\rm oct}, 4f_{\rm oct(R)}] > [4f_{\rm jbi}, 6g_{\rm oct}, 4f_{\rm oct(S)}]$. This is done by relating the cation substitution and magnetic moment on different crystallographic sites to the measured $T_{\rm C}$ of the samples. The two samples with diamagnetic substitutions have the lowest $T_{\rm C}$, leading to $T_{\rm C}$ values of ZnWHF < MgWHF < CoWHF < NiWHF. The low $T_{\rm C}$ seen for SrZn₂Fe₁₆O₂₇ can be understood by the diamagnetic Zn²⁺ cation occupying the two most influential sites for the super-exchange, $4e_{\rm tet}$ and $4f_{\rm tet}$. SrMg₂Fe₁₆O₂₇ is the second lowest

owing to the diamagnetic Mg²⁺. SrCo₂Fe₁₆O₂₇ has a slightly higher Co occupation in $4e_{tet}$ and $4f_{tet}$ sites and additionally, the magnitude of the moment is lower on these important antiparallel sites (see Fig. 4) when compared with SrNi₂Fe₁₆O₂₇, which has the highest T_C of these four samples. The Curie temperatures here are in agreement with previously reported values for Ba Me_2 Fe₁₆O₂₇, where Me = Co, Mg (Collomb *et al.*, 1986*a*) and Me = Ni, Zn, Cu (Besagni *et al.*, 1981; Licci *et al.*, 1981).

3.4. Magnetic measurements

The hysteresis curves (mass magnetization M versus apparent field H) of the four synthesized WHFs are given in Fig. 7. From the magnetic measurements it is seen that $SrCo_2Fe_{16}O_{27}$ exhibits the highest M_s of 78.5 A m² kg⁻¹, while none of the samples show any appreciable coercivity or energy product (between 5–35 kA m⁻¹ and 0.5–2.2 kJ m⁻³). This is due to the large crystallites splitting into multiple magnetic domains, which is consistent with the observed difference in broadening of the magnetic and nuclear diffraction signals. The result of multiple domains is a negligible coercivity, as domain wall motion allows continuous rotation of the magnetic moments rather than a collective flip as for single domain particles. Future reduction in the size of large particles may result in appreciable coercivities, as BaMe₂Fe₁₆O₂₇ has previously been demonstrated to have respectable anisotropy constants and coercivity (Lotgering et al., 1961, 1980a; Pullar, 2012). Inspecting the approach to magnetic saturation of the different samples in Fig. 7, it is clear SrCo₂Fe₁₆O₂₇ varies from the three other WHFs in the curvature of magnetization. Considering the planar ordering seen in the neutron powder diffraction data, the difference could be explained by an easier reorientation of the magnetic moment with the *ab* plane rather than forcing the moment away for the uniaxial easy axis in the systems.



Magnetic hysteresis for the four $SrMe_2Fe_{16}O_{27}$ (Me = Mg, Co, Ni, Zn) samples. The samples were measured using an applied field of ± 3 T with a PPMS-VSM.

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The macroscopically measured $M_{s,VSM}$ of the four WHFs were compared with the calculated $M_{s,NPD}$ from the refined magnetic moments of the NPD data (see Fig. 4 for numerical values). Across the four samples, M_s is higher for the VSM data compared with the NPD data. Notably, numerical equivalence between the calculated and measured magnetizations will most likely differ, given the distinct macroscopic and atomic nature of the two probes, as well as possibly unaccounted effects in the NPD refinement. Nevertheless, the trend from the magnetic measurements (M_8 of SrCo₂Fe₁₆O₂₇ > $SrZn_2Fe_{16}O_{27} > SrNi_2Fe_{16}O_{27} > SrMg_2Fe_{16}O_{27}$) is consistent with the refined data. When accounting for the M_s of the impurity phases, $SrZn_2Fe_{16}O_{27}$ has a higher M_s compared with $SrCo_2Fe_{16}O_{27}$, but there are uncertainties of their $M_{s,NPD}$ overlap. A comparison of M_s accounting for impurities and M_s values from the literature is given in Section S8 and Table S10 of the supporting information. Refinements where all seven crystallographic sites were constrained to have the same magnetic moment give rise to an increased calculated M_s at the cost of a significantly worse fit of the NPD data. The resulting R_{mag} and R_{wp} values for the two options are given in Table S6; $\mu_{\rm B}$ and $M_{\rm s}$ are also given for the constrained option and compared with the individual and VSM values in Table S9. Looking at both the site occupation of the different Mes and the moment of the seven crystallographic sites, the underlying reason for the difference in M_s can be rationalized. The higher $M_{\rm s}$ derived from the NPD of SrCo₂Fe₁₆O₂₇ in comparison with that of SrZn₂Fe₁₆O₂₇ and SrNi₂Fe₁₆O₂₇ can be explained by comparing the magnitude of the magnetic moments on the $12k_{oct(R-S)}$ site. This is higher for SrCo₂Fe₁₆O₂₇ than for SrZn₂Fe₁₆O₂₇, while the moment on the antiparallel sites $4e_{\text{tet}(S)}$, $4f_{\text{tet}(S)}$ and $4f_{\text{oct}(R)}$ are lower for SrCo₂Fe₁₆O₂₇ than for SrNi₂Fe₁₆O₂₇. When comparing $SrNi_2Fe_{16}O_{27}$ with $SrZn_2Fe_{16}O_{27}$, the former has a higher moment on $12k_{oct(R-S)}$ and the antiparallel sites also have significantly higher magnetic moments, especially $4e_{tet(S)}$ due to the Zn²⁺ contents on this site. $SrMg_2Fe_{16}O_{27}$ exhibits the lowest M_s , which is clearly understood from the low moment on $6g_{oct(S-S)}$ resulting from the high Mg^{2+} occupation on this site.

Further enhancement of W-type hexaferrite magnetic properties will be sought by means of nanostructuring, moving both to the single-domain region and optimizing the texture in a compacted magnet and ternary doping, with the aim of increasing $M_{\rm s}$ further while retaining sufficient $H_{\rm c}$ for optimal $BH_{\rm max}$.

4. Conclusions

To improve the performance of magnetic materials and multiferroics, an in-depth understanding of their atomic structure is essential as it reveals the intrinsic magnetic properties of the compound. In W-type hexaferrites, the co-existence of divalent and trivalent metal ions results in a complex crystal structure allowing for a tunable magnetic structure. Based on combined refinement of neutron and X-ray powder diffraction data on sol-gel synthesized W-type hexaferrites (Sr Me_2 Fe₁₆O₂₇, Me = Mg, Co, Ni, Zn), we

extracted the magnetic structure, where Mg, Ni and Zn exhibit uniaxial magnetic ordering along the crystallographic c axis, whereas SrCo₂Fe₁₆O₂₇ is ordered magnetically in the *ab* plane.

The specific site occupation of the *Me* atoms was extracted from the joint refinements. Co and Ni have similar substituted structures with the majority of the transition metal sitting on the $6g_{oct}$ site, whereas Zn was mostly found on the $4e_{tet}$ site as opposed to Mg which it located on $6g_{oct}$. The atomic structure occupancies allow us to rationalize the Curie temperature where SrNi₂Fe₁₆O₂₇, $T_{C} = 500 (2)^{\circ}C > SrCo_2Fe_{16}O_{27}$, $T_{C} =$ $470 (3)^{\circ}C > SrMg_2Fe_{16}O_{27}$, $T_{C} = 435 (2)^{\circ}C > SrZn_2Fe_{16}O_{27}$, $T_{C} =$ $345 (3)^{\circ}C$. The affinity of the different cations for different crystallographic sites leads to an alteration of the magnetic moments resulting in variation in the strength of the superexchange interactions.

The $M_{s,VSM}$ and the $M_{s,NPD}$ calculated on the basis of the NPD reveal a similar trend where the M_s is decreasing in the following order for the corrected VSM data: $SrZn_2Fe_{16}O_{27}$, $M_{s,VSM} = 80.4$ (3) > $SrCo_2Fe_{16}O_{27}$, $M_{s,VSM} = 78.4$ (2) > $SrNi_2Fe_{16}O_{27}$, $M_{s,VSM} = 69.0$ (3) > $SrMg_2Fe_{16}O_{27}$, $M_{s,VSM} = 63.7$ (2) A m² kg⁻¹. The relative variation of M_s between the different WHFs can be understood by the refined occupation fractions and magnetic moments on each sample.

Overall, the results illustrate the strength of combined synchrotron and neutron powder diffraction and the possibilities for rationalizing macroscopic properties from crystal-lographic and magnetic atomic structures. In conclusion, it is possible to tune magnetic properties and magnetic ordering of W-type hexaferrites by substitution of different divalent metals within the structure. This provides a handle for optimizing the magnetic properties of the compound, with the most promising results in terms of permanent magnets for SrCo₂Fe₁₆O₂₇ and SrZn₂Fe₁₆O₂₇.

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