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2-[[5-(4-Chlorophenoxymethyl)-4-phenyl-4*H*-1,2,4-triazol-3-yl]sulfonyl]-*N*-(4-nitrophenyl)acetamide

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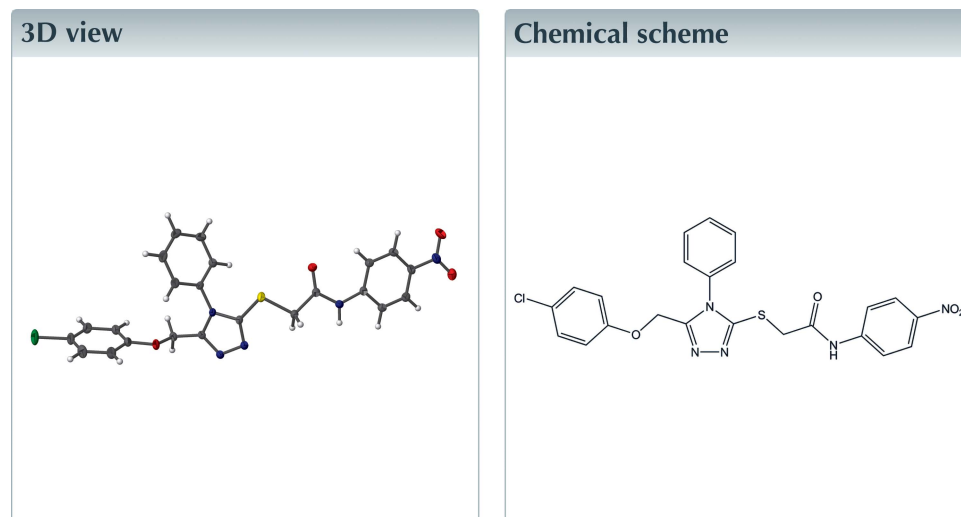
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Keywords: crystal structure; phenyltriazole; thiotriazole.

CCDC reference: 1468941

Structural data: full structural data are available from iucrdata.iucr.org

The title molecule, C₂₃H₁₈ClN₅O₄S, is in an 'extended' conformation. The central triazole ring makes dihedral angles of 58.14 (6), 86.29 (6) and 41.99 (6)°, respectively, with the adjacent phenyl, chlorophenyl and nitrophenyl rings. In the crystal, molecules are linked *via* a pair of N—H···N hydrogen bonds with an R₂²(16) ring motif, forming an inversion dimer. The dimers are connected by C—H···O hydrogen bonds into a tape structure along [011]. C—H···π interactions between the tapes are also observed.



Structure description

1,2,4-Triazole and its derivatives represent one of the most biologically active class of heterocyclic compounds. There are a large number of drugs containing a 1,2,4-triazole nucleus in their structures. These include Ribavirin (antiviral), Rizatriptan (anti-migraine), Estazolam and Alprazolam (anxiolytic), Letrozole and Anastrozole (breast cancer) (Godhani *et al.*, 2015). Others such as Itraconazole, Fluconazole and Posaconazole have been used for the treatment of fungal infection diseases (Godhani *et al.*, 2015). In view of the above facts, we report in this context the synthesis and crystal structure of the title compound (Fig. 1).

In the crystal, molecules form a tape structure through N—H···N and C—H···O hydrogen bonds (Table 1 and Fig. 2). C—H···π interactions between the tapes contribute to the stabilization of the molecular packing.

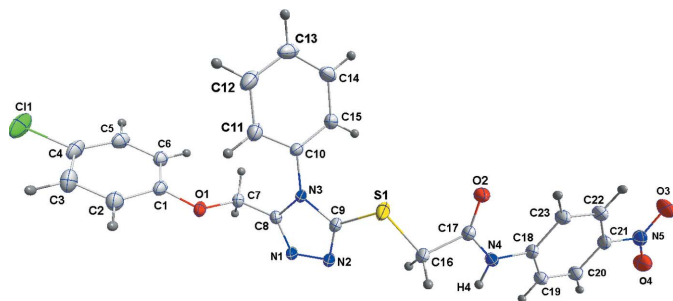


Figure 1
The molecular structure of the title compound, showing the atom-labeling scheme and 50% probability ellipsoids.

Synthesis and crystallization

A suspension of 5-(4-chlorophenoxy)methyl-4-phenyl-1,2,4-triazoline-3-thione (10 mmol), chloro-*N*-(*p*-nitrophenyl)-acetamide (10 mol) and anhydrous K_2CO_3 (2.0 g) in dry acetone (50 ml) was heated under reflux with stirring for 3 h. The hot reaction mixture was filtered to remove K_2CO_3 and the clear filtrate was evaporated until dryness. The solid residue was crystallized from ethanol to give the title compound. Yield: 93%; m.p. 207°C. IR (cm^{-1}): 3300 (NH), 1660 (C=O). 1H NMR ($CDCl_3$, p.p.m.): δ 9.2 (*s*, 1H, NH), 6.7–8.0 (*m*, 13H, ArH), 4.9 (*s*, 2H, OCH_2), 4.0 (*s*, 2H, SCH_2).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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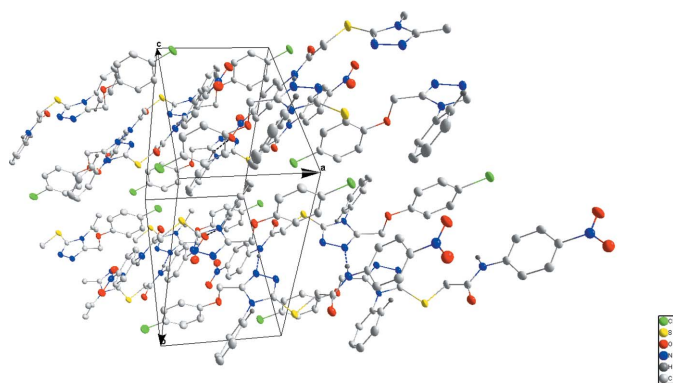


Figure 2
A packing diagram of the title compound, showing the tapes formed by N–H...N and C–H...O hydrogen bonds (blue and black dotted lines, respectively).

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$, $Cg2$ and $Cg3$ are the centroids of the 1,2,4-triazole (N1/N2/C9/N3/C8), benzene (C1–C6) and phenyl (C10–C15) rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N4-H4\cdots N1^i$	0.91	2.08	2.9934 (17)	176
$C15-H15\cdots O4^{ii}$	0.95	2.44	3.2324 (19)	140
$C5-H5\cdots Cg1^{iii}$	0.95	2.97	3.757 (2)	141
$C16-H16A\cdots Cg3^{iv}$	0.99	2.82	3.6284 (18)	139
$C22-H22\cdots Cg2^v$	0.95	2.69	3.5883 (16)	158

Symmetry codes: (i) $-x+2, -y+1, -z+2$; (ii) $-x+2, -y, -z+1$; (iii) $x-1, y, z$; (iv) $x+1, y, z$; (v) $x+1, y-1, z-1$.

Table 2

Experimental details.

Crystal data	
Chemical formula	$C_{23}H_{18}ClN_5O_4S$
M_r	495.93
Crystal system, space group	Triclinic, $P\bar{1}$
Temperature (K)	150
a, b, c (\AA)	8.1828 (2), 12.1541 (3), 12.3654 (3)
α, β, γ ($^\circ$)	114.474 (1), 93.681 (1), 96.949 (1)
V (\AA^3)	1102.14 (5)
Z	2
Radiation type	Cu $K\alpha$
μ (mm^{-1})	2.79
Crystal size (mm)	$0.21 \times 0.18 \times 0.03$
Data collection	
Diffractometer	Bruker D8 VENTURE PHOTON 100 CMOS
Absorption correction	Multi-scan (SADABS; Bruker, 2016)
T_{min}, T_{max}	0.80, 0.93
No. of measured, independent and observed [$I > 2\sigma(I)$] reflections	8470, 4084, 3709
R_{int}	0.023
$(\sin \theta/\lambda)_{max}$ (\AA^{-1})	0.618
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.034, 0.089, 1.04
No. of reflections	4084
No. of parameters	307
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{max}, \Delta\rho_{min}$ ($e \text{\AA}^{-3}$)	0.17, -0.35

Computer programs: APEX3 (Bruker, 2016), SAINT (Bruker, 2016), SHELXT (Sheldrick, 2015b), SHELXL2014 (Sheldrick, 2015a), DIAMOND (Brandenburg & Putz, 2012).

Tulane Crystallography Laboratory are gratefully acknowledged.

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full crystallographic data

IUCrData (2016). **1**, x160452 [doi:10.1107/S2414314616004521]

2-[[5-(4-Chlorophenoxy)methyl]-4-phenyl-4*H*-1,2,4-triazol-3-yl]sulfanyl]-*N*-(4-nitrophenyl)acetamide

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2-[[5-(4-Chlorophenoxy)methyl]-4-phenyl-4*H*-1,2,4-triazol-3-yl]sulfanyl]-*N*-(4-nitrophenyl)acetamide

Crystal data

$C_{23}H_{18}ClN_5O_4S$
 $M_r = 495.93$
 Triclinic, $P\bar{1}$
 $a = 8.1828$ (2) Å
 $b = 12.1541$ (3) Å
 $c = 12.3654$ (3) Å
 $\alpha = 114.474$ (1)°
 $\beta = 93.681$ (1)°
 $\gamma = 96.949$ (1)°
 $V = 1102.14$ (5) Å³

$Z = 2$
 $F(000) = 512$
 $D_x = 1.494$ Mg m⁻³
 Cu $K\alpha$ radiation, $\lambda = 1.54178$ Å
 Cell parameters from 6886 reflections
 $\theta = 4.0$ – 72.3 °
 $\mu = 2.79$ mm⁻¹
 $T = 150$ K
 Plate, colourless
 $0.21 \times 0.18 \times 0.03$ mm

Data collection

Bruker D8 VENTURE PHOTON 100 CMOS
 diffractometer
 Radiation source: INCOATEC $I\mu$ S micro-focus
 source
 Mirror monochromator
 Detector resolution: 10.4167 pixels mm⁻¹
 ω scans
 Absorption correction: multi-scan
 (*SADABS*; Bruker, 2016)

$T_{\min} = 0.80$, $T_{\max} = 0.93$
 8470 measured reflections
 4084 independent reflections
 3709 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.023$
 $\theta_{\max} = 72.3$ °, $\theta_{\min} = 4.0$ °
 $h = -9 \rightarrow 10$
 $k = -14 \rightarrow 13$
 $l = -15 \rightarrow 15$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.089$
 $S = 1.04$
 4084 reflections
 307 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: mixed
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.048P)^2 + 0.4011P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.17$ e Å⁻³
 $\Delta\rho_{\min} = -0.35$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	−0.09452 (6)	1.08937 (5)	1.17426 (5)	0.04449 (14)
S1	0.85499 (5)	0.52367 (4)	0.73580 (3)	0.02790 (12)
O1	0.45177 (13)	0.80389 (9)	1.06669 (9)	0.0230 (2)
O2	0.96933 (13)	0.31268 (10)	0.56493 (9)	0.0260 (2)
O3	1.30812 (16)	−0.22278 (10)	0.30629 (10)	0.0333 (3)
O4	1.43386 (15)	−0.21478 (10)	0.46972 (11)	0.0335 (3)
N1	0.70644 (15)	0.61703 (11)	1.04995 (11)	0.0215 (3)
N2	0.82617 (15)	0.57622 (11)	0.97273 (11)	0.0219 (3)
N3	0.60856 (14)	0.59829 (11)	0.87198 (10)	0.0184 (2)
N4	1.14142 (15)	0.27311 (11)	0.69377 (11)	0.0215 (3)
H4	1.1865	0.3031	0.7717	0.026*
N5	1.34798 (16)	−0.17258 (12)	0.41598 (12)	0.0253 (3)
C1	0.31563 (19)	0.86299 (14)	1.08991 (13)	0.0218 (3)
C2	0.3516 (2)	0.98921 (15)	1.13020 (15)	0.0289 (3)
H2	0.4630	1.0278	1.1399	0.035*
C3	0.2256 (2)	1.05892 (16)	1.15629 (16)	0.0335 (4)
H3	0.2500	1.1454	1.1848	0.040*
C4	0.0633 (2)	1.00125 (16)	1.14047 (15)	0.0300 (4)
C5	0.0266 (2)	0.87596 (16)	1.09983 (14)	0.0281 (3)
H5	−0.0854	0.8376	1.0887	0.034*
C6	0.15317 (19)	0.80536 (14)	1.07504 (13)	0.0241 (3)
H6	0.1287	0.7191	1.0484	0.029*
C7	0.42651 (18)	0.67518 (13)	1.03645 (13)	0.0208 (3)
H7A	0.3285	0.6324	0.9753	0.025*
H7B	0.4086	0.6607	1.1083	0.025*
C8	0.57961 (18)	0.63017 (12)	0.98852 (12)	0.0189 (3)
C9	0.76419 (17)	0.56569 (13)	0.86746 (13)	0.0197 (3)
C10	0.49571 (17)	0.58946 (13)	0.77336 (12)	0.0188 (3)
C11	0.4338 (2)	0.69254 (14)	0.77869 (14)	0.0255 (3)
H11	0.4648	0.7689	0.8467	0.031*
C12	0.3251 (2)	0.68181 (18)	0.68222 (15)	0.0344 (4)
H12	0.2811	0.7514	0.6844	0.041*
C13	0.2809 (2)	0.57079 (19)	0.58354 (15)	0.0354 (4)
H13	0.2066	0.5643	0.5181	0.043*
C14	0.3442 (2)	0.46872 (17)	0.57942 (14)	0.0299 (4)

H14	0.3138	0.3926	0.5109	0.036*
C15	0.45185 (19)	0.47706 (14)	0.67500 (13)	0.0230 (3)
H15	0.4946	0.4071	0.6730	0.028*
C16	1.02531 (19)	0.45839 (14)	0.77244 (13)	0.0244 (3)
H16A	1.1299	0.5171	0.7933	0.029*
H16B	1.0057	0.4410	0.8424	0.029*
C17	1.03937 (17)	0.34014 (13)	0.66470 (13)	0.0203 (3)
C18	1.18868 (18)	0.16229 (13)	0.61799 (13)	0.0202 (3)
C19	1.27723 (18)	0.10494 (14)	0.67439 (13)	0.0227 (3)
H19	1.3019	0.1421	0.7592	0.027*
C20	1.32914 (19)	-0.00431 (14)	0.60902 (13)	0.0237 (3)
H20	1.3893	-0.0429	0.6477	0.028*
C21	1.29176 (18)	-0.05704 (13)	0.48524 (13)	0.0222 (3)
C22	1.20524 (19)	-0.00199 (14)	0.42706 (13)	0.0232 (3)
H22	1.1816	-0.0396	0.3422	0.028*
C23	1.15320 (19)	0.10814 (14)	0.49296 (13)	0.0231 (3)
H23	1.0939	0.1466	0.4537	0.028*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0467 (3)	0.0525 (3)	0.0530 (3)	0.0368 (2)	0.0247 (2)	0.0304 (2)
S1	0.0279 (2)	0.0415 (2)	0.02153 (19)	0.02142 (17)	0.00882 (15)	0.01553 (16)
O1	0.0191 (5)	0.0182 (5)	0.0288 (5)	0.0078 (4)	0.0043 (4)	0.0056 (4)
O2	0.0258 (5)	0.0308 (6)	0.0187 (5)	0.0112 (5)	-0.0004 (4)	0.0067 (4)
O3	0.0445 (7)	0.0286 (6)	0.0223 (6)	0.0105 (5)	0.0096 (5)	0.0045 (5)
O4	0.0361 (6)	0.0279 (6)	0.0365 (6)	0.0152 (5)	0.0051 (5)	0.0111 (5)
N1	0.0220 (6)	0.0234 (6)	0.0170 (6)	0.0089 (5)	0.0021 (5)	0.0053 (5)
N2	0.0206 (6)	0.0253 (6)	0.0186 (6)	0.0098 (5)	0.0030 (5)	0.0063 (5)
N3	0.0177 (6)	0.0205 (6)	0.0169 (6)	0.0092 (5)	0.0023 (5)	0.0061 (4)
N4	0.0226 (6)	0.0235 (6)	0.0160 (6)	0.0089 (5)	0.0011 (5)	0.0049 (5)
N5	0.0260 (7)	0.0227 (6)	0.0256 (7)	0.0058 (5)	0.0083 (5)	0.0076 (5)
C1	0.0220 (7)	0.0243 (7)	0.0203 (7)	0.0107 (6)	0.0047 (6)	0.0085 (6)
C2	0.0275 (8)	0.0237 (8)	0.0349 (8)	0.0074 (7)	0.0060 (7)	0.0108 (6)
C3	0.0397 (9)	0.0246 (8)	0.0398 (9)	0.0144 (7)	0.0108 (8)	0.0138 (7)
C4	0.0320 (9)	0.0357 (9)	0.0312 (8)	0.0214 (7)	0.0122 (7)	0.0175 (7)
C5	0.0231 (7)	0.0354 (9)	0.0291 (8)	0.0102 (7)	0.0067 (6)	0.0151 (7)
C6	0.0238 (7)	0.0254 (7)	0.0235 (7)	0.0086 (6)	0.0055 (6)	0.0091 (6)
C7	0.0206 (7)	0.0190 (7)	0.0208 (7)	0.0064 (6)	0.0038 (6)	0.0055 (5)
C8	0.0205 (7)	0.0171 (6)	0.0173 (6)	0.0058 (6)	0.0025 (5)	0.0048 (5)
C9	0.0177 (7)	0.0201 (7)	0.0202 (7)	0.0077 (6)	0.0023 (5)	0.0064 (5)
C10	0.0164 (6)	0.0244 (7)	0.0174 (7)	0.0074 (6)	0.0021 (5)	0.0095 (5)
C11	0.0296 (8)	0.0262 (8)	0.0250 (7)	0.0129 (6)	0.0065 (6)	0.0125 (6)
C12	0.0390 (9)	0.0474 (10)	0.0313 (9)	0.0273 (8)	0.0115 (8)	0.0246 (8)
C13	0.0283 (8)	0.0634 (12)	0.0216 (8)	0.0206 (8)	0.0050 (7)	0.0211 (8)
C14	0.0237 (7)	0.0418 (9)	0.0187 (7)	0.0062 (7)	0.0020 (6)	0.0073 (6)
C15	0.0220 (7)	0.0252 (7)	0.0205 (7)	0.0077 (6)	0.0037 (6)	0.0074 (6)
C16	0.0206 (7)	0.0287 (8)	0.0201 (7)	0.0113 (6)	0.0014 (6)	0.0049 (6)

C17	0.0164 (6)	0.0237 (7)	0.0198 (7)	0.0055 (6)	0.0037 (5)	0.0076 (6)
C18	0.0174 (6)	0.0219 (7)	0.0199 (7)	0.0059 (6)	0.0037 (5)	0.0066 (6)
C19	0.0228 (7)	0.0260 (7)	0.0176 (7)	0.0063 (6)	0.0020 (6)	0.0072 (6)
C20	0.0234 (7)	0.0258 (7)	0.0237 (7)	0.0082 (6)	0.0029 (6)	0.0113 (6)
C21	0.0213 (7)	0.0210 (7)	0.0222 (7)	0.0058 (6)	0.0068 (6)	0.0061 (6)
C22	0.0243 (7)	0.0254 (7)	0.0172 (7)	0.0048 (6)	0.0028 (6)	0.0062 (6)
C23	0.0241 (7)	0.0261 (8)	0.0189 (7)	0.0081 (6)	0.0019 (6)	0.0084 (6)

Geometric parameters (Å, °)

C11—C4	1.7387 (16)	C7—C8	1.487 (2)
S1—C9	1.7383 (15)	C7—H7A	0.9900
S1—C16	1.8061 (15)	C7—H7B	0.9900
O1—C1	1.3796 (18)	C10—C11	1.386 (2)
O1—C7	1.4343 (18)	C10—C15	1.387 (2)
O2—C17	1.2181 (18)	C11—C12	1.392 (2)
O3—N5	1.2360 (17)	C11—H11	0.9500
O4—N5	1.2281 (18)	C12—C13	1.379 (3)
N1—C8	1.3111 (19)	C12—H12	0.9500
N1—N2	1.4010 (18)	C13—C14	1.384 (3)
N2—C9	1.3151 (18)	C13—H13	0.9500
N3—C8	1.3734 (19)	C14—C15	1.389 (2)
N3—C9	1.3765 (18)	C14—H14	0.9500
N3—C10	1.4403 (17)	C15—H15	0.9500
N4—C17	1.3634 (19)	C16—C17	1.525 (2)
N4—C18	1.4031 (18)	C16—H16A	0.9900
N4—H4	0.9098	C16—H16B	0.9900
N5—C21	1.4586 (19)	C18—C19	1.399 (2)
C1—C2	1.388 (2)	C18—C23	1.400 (2)
C1—C6	1.389 (2)	C19—C20	1.374 (2)
C2—C3	1.383 (2)	C19—H19	0.9500
C2—H2	0.9500	C20—C21	1.388 (2)
C3—C4	1.386 (3)	C20—H20	0.9500
C3—H3	0.9500	C21—C22	1.384 (2)
C4—C5	1.377 (2)	C22—C23	1.384 (2)
C5—C6	1.393 (2)	C22—H22	0.9500
C5—H5	0.9500	C23—H23	0.9500
C6—H6	0.9500		
C9—S1—C16	100.00 (7)	C11—C10—N3	119.63 (13)
C1—O1—C7	117.68 (12)	C15—C10—N3	118.71 (13)
C8—N1—N2	107.77 (12)	C10—C11—C12	118.59 (15)
C9—N2—N1	106.62 (11)	C10—C11—H11	120.7
C8—N3—C9	104.71 (11)	C12—C11—H11	120.7
C8—N3—C10	127.83 (12)	C13—C12—C11	120.44 (15)
C9—N3—C10	127.22 (12)	C13—C12—H12	119.8
C17—N4—C18	128.21 (12)	C11—C12—H12	119.8
C17—N4—H4	117.2	C12—C13—C14	120.28 (15)

C18—N4—H4	114.6	C12—C13—H13	119.9
O4—N5—O3	123.20 (13)	C14—C13—H13	119.9
O4—N5—C21	118.18 (13)	C13—C14—C15	120.28 (15)
O3—N5—C21	118.62 (13)	C13—C14—H14	119.9
O1—C1—C2	114.63 (14)	C15—C14—H14	119.9
O1—C1—C6	124.88 (14)	C10—C15—C14	118.74 (15)
C2—C1—C6	120.50 (14)	C10—C15—H15	120.6
C3—C2—C1	120.19 (16)	C14—C15—H15	120.6
C3—C2—H2	119.9	C17—C16—S1	109.03 (10)
C1—C2—H2	119.9	C17—C16—H16A	109.9
C2—C3—C4	119.29 (16)	S1—C16—H16A	109.9
C2—C3—H3	120.4	C17—C16—H16B	109.9
C4—C3—H3	120.4	S1—C16—H16B	109.9
C5—C4—C3	120.85 (15)	H16A—C16—H16B	108.3
C5—C4—C11	120.03 (14)	O2—C17—N4	125.33 (13)
C3—C4—C11	119.12 (13)	O2—C17—C16	122.69 (13)
C4—C5—C6	120.16 (15)	N4—C17—C16	111.96 (12)
C4—C5—H5	119.9	C19—C18—C23	119.61 (13)
C6—C5—H5	119.9	C19—C18—N4	116.02 (13)
C1—C6—C5	119.00 (15)	C23—C18—N4	124.37 (13)
C1—C6—H6	120.5	C20—C19—C18	121.02 (13)
C5—C6—H6	120.5	C20—C19—H19	119.5
O1—C7—C8	106.20 (12)	C18—C19—H19	119.5
O1—C7—H7A	110.5	C19—C20—C21	118.54 (14)
C8—C7—H7A	110.5	C19—C20—H20	120.7
O1—C7—H7B	110.5	C21—C20—H20	120.7
C8—C7—H7B	110.5	C22—C21—C20	121.67 (14)
H7A—C7—H7B	108.7	C22—C21—N5	119.88 (13)
N1—C8—N3	110.23 (13)	C20—C21—N5	118.45 (13)
N1—C8—C7	125.83 (13)	C21—C22—C23	119.76 (14)
N3—C8—C7	123.93 (12)	C21—C22—H22	120.1
N2—C9—N3	110.67 (13)	C23—C22—H22	120.1
N2—C9—S1	128.89 (11)	C22—C23—C18	119.40 (13)
N3—C9—S1	120.39 (10)	C22—C23—H23	120.3
C11—C10—C15	121.66 (13)	C18—C23—H23	120.3
C8—N1—N2—C9	-0.55 (16)	C8—N3—C10—C15	-118.65 (16)
C7—O1—C1—C2	173.83 (13)	C9—N3—C10—C15	54.9 (2)
C7—O1—C1—C6	-6.6 (2)	C15—C10—C11—C12	-0.1 (2)
O1—C1—C2—C3	179.81 (15)	N3—C10—C11—C12	179.91 (14)
C6—C1—C2—C3	0.2 (2)	C10—C11—C12—C13	-0.2 (3)
C1—C2—C3—C4	-0.7 (3)	C11—C12—C13—C14	0.0 (3)
C2—C3—C4—C5	0.3 (3)	C12—C13—C14—C15	0.5 (3)
C2—C3—C4—C11	179.61 (13)	C11—C10—C15—C14	0.6 (2)
C3—C4—C5—C6	0.6 (3)	N3—C10—C15—C14	-179.44 (13)
C11—C4—C5—C6	-178.67 (12)	C13—C14—C15—C10	-0.7 (2)
O1—C1—C6—C5	-178.85 (14)	C9—S1—C16—C17	136.78 (11)
C2—C1—C6—C5	0.7 (2)	C18—N4—C17—O2	0.8 (3)

C4—C5—C6—C1	-1.1 (2)	C18—N4—C17—C16	-177.53 (14)
C1—O1—C7—C8	166.87 (12)	S1—C16—C17—O2	14.43 (19)
N2—N1—C8—N3	0.69 (16)	S1—C16—C17—N4	-167.17 (11)
N2—N1—C8—C7	-178.20 (13)	C17—N4—C18—C19	-171.74 (14)
C9—N3—C8—N1	-0.55 (16)	C17—N4—C18—C23	8.4 (2)
C10—N3—C8—N1	174.12 (13)	C23—C18—C19—C20	-0.4 (2)
C9—N3—C8—C7	178.37 (13)	N4—C18—C19—C20	179.69 (14)
C10—N3—C8—C7	-7.0 (2)	C18—C19—C20—C21	0.0 (2)
O1—C7—C8—N1	95.45 (17)	C19—C20—C21—C22	0.4 (2)
O1—C7—C8—N3	-83.30 (16)	C19—C20—C21—N5	179.58 (14)
N1—N2—C9—N3	0.21 (16)	O4—N5—C21—C22	175.23 (14)
N1—N2—C9—S1	177.76 (11)	O3—N5—C21—C22	-4.3 (2)
C8—N3—C9—N2	0.19 (16)	O4—N5—C21—C20	-3.9 (2)
C10—N3—C9—N2	-174.52 (13)	O3—N5—C21—C20	176.57 (14)
C8—N3—C9—S1	-177.60 (10)	C20—C21—C22—C23	-0.3 (2)
C10—N3—C9—S1	7.7 (2)	N5—C21—C22—C23	-179.48 (14)
C16—S1—C9—N2	15.55 (16)	C21—C22—C23—C18	-0.1 (2)
C16—S1—C9—N3	-167.10 (12)	C19—C18—C23—C22	0.5 (2)
C8—N3—C10—C11	61.4 (2)	N4—C18—C23—C22	-179.62 (14)
C9—N3—C10—C11	-125.12 (16)		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1, Cg2 and Cg3 are the centroids of the 1,2,4-triazole (N1/N2/C9/N3/C8), benzene (C1–C6) and phenyl (C10–C15) rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N4—H4 \cdots N1 ⁱ	0.91	2.08	2.9934 (17)	176
C15—H15 \cdots O4 ⁱⁱ	0.95	2.44	3.2324 (19)	140
C5—H5 \cdots Cg1 ⁱⁱⁱ	0.95	2.97	3.757 (2)	141
C16—H16A \cdots Cg3 ^{iv}	0.99	2.82	3.6284 (18)	139
C22—H22 \cdots Cg2 ^v	0.95	2.69	3.5883 (16)	158

Symmetry codes: (i) $-x+2, -y+1, -z+2$; (ii) $-x+2, -y, -z+1$; (iii) $x-1, y, z$; (iv) $x+1, y, z$; (v) $x+1, y-1, z-1$.