

2-[(5-Amino-1,3,4-thiadiazol-2-yl)sulfanyl]-*N*-(4-chlorophenyl)acetamide

S. Madan Kumar,^a D. J. Madhu Kumar,^b K. S. Harish,^c Prasad D. Jagadeesha,^b K. Byrappa^d and M. M. M. Abdoh^{e*}

^aPURSE Lab, Mangalagangothri, Mangalore University, Mangaluru 574 199, India, ^bDepartment of Post-Graduate and Research In Chemistry, Mangalagangothri, Mangalore University, India, ^cDepartment of Physics, Sheshadripuram PU college, Mysore 570 017, India, ^dDepartment of Material Science, Mangalore University, Mangaluru 574 199, India, and ^eDepartment of Physics, Faculty of Science, An Najah National University, Nablus, West Bank, Palestinian Territories. *Correspondence e-mail: muneer@najah.edu

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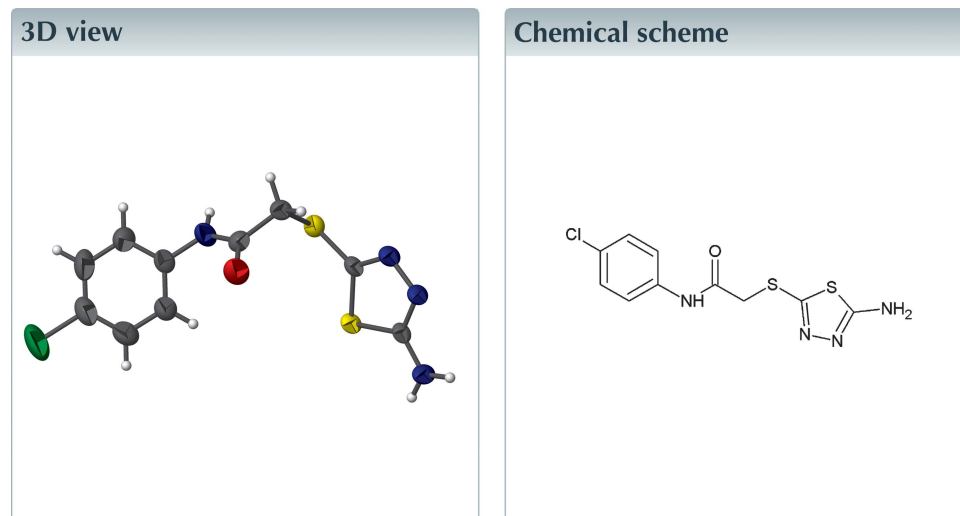
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Keywords: crystal structure; 1,3,4-thiadiazole; intermolecular hydrogen bonds.

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Structural data: full structural data are available from iucrdata.iucr.org

In the title compound, C₁₀H₉ClN₄OS₂, the dihedral angle between the planes of the chlorophenyl and thiadiazole groups is 32.93 (16)°. The molecules are connected through intermolecular N—H···N and N—H···O hydrogen bonds. An N—H···N hydrogen bond forms R₂²(8) ring motifs.



Structure description

As part of our research on synthesis and crystal structure determination of imidazo-[2,1-*b*][1,3,4]thiadiazole derivatives, we report here the crystal and molecular structure of 2-[(5-amino-1,3,4-thiadiazol-2-yl)sulfanyl]-*N*-(4-chlorophenyl)acetamide (Fig. 1).

The dihedral angle between the chlorophenyl (C5–C10) and thiadiazol (C1/C2/N1/N2/S1) rings is 32.93 (16)°. In the crystal (Fig. 2), the molecules are connected through N—H···N and N—H···O hydrogen bonds (Table 1), the N3—H3A···N2 hydrogen bonds forming R₂²(8) ring motifs. Overall, these interactions generate a two-dimensional network parallel to (100).

Synthesis and crystallization

An equimolar ratio of 2-[(5-amino-1,3,4-thiadiazol-2-yl)thio]-*N*-(4-chlorophenyl)acetamide (0.005 mol) and ethyl chloroacetate (0.005 mol) in glacial acetic acid (20 mL) was heated under reflux for 17 h. The reaction mixture was poured into ice cold water. The precipitated solid was filtered, dried and recrystallized from ethanol.

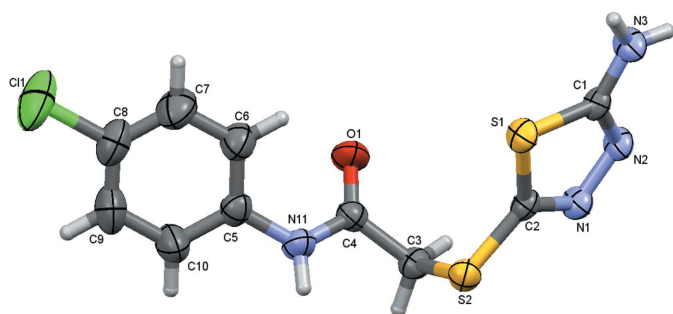


Figure 1
A view of the title molecule, showing the atom labelling. Displacement ellipsoids are drawn at the 50% probability level.

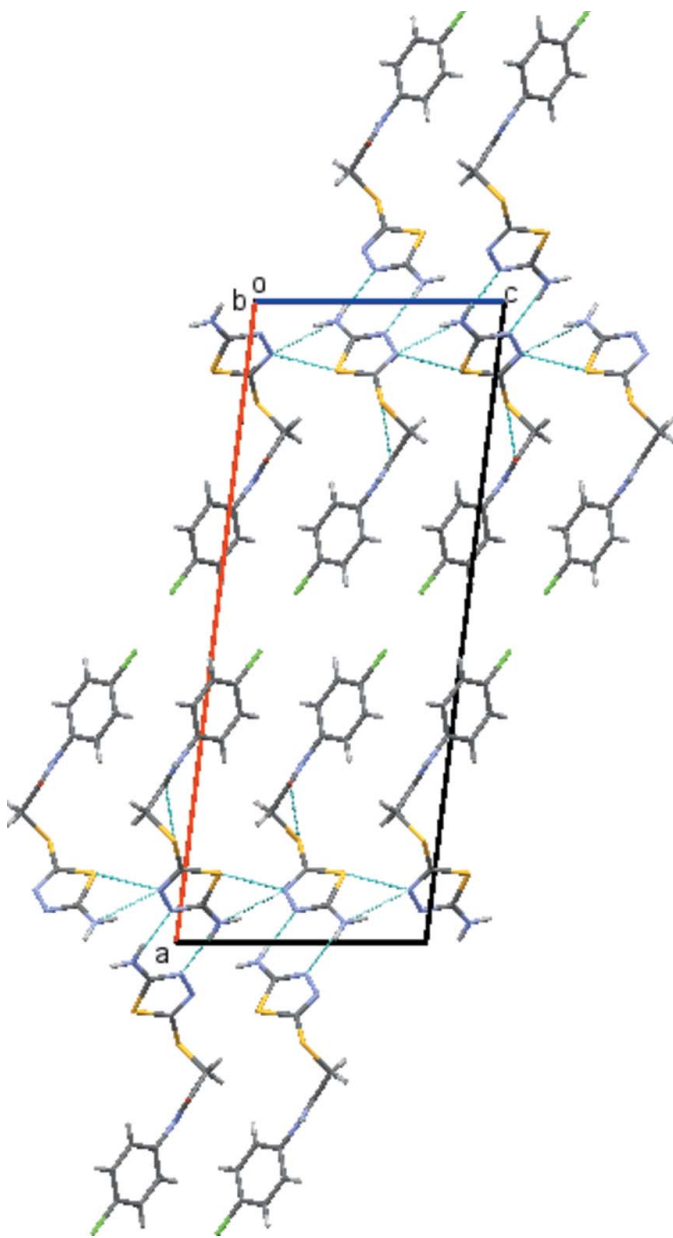


Figure 2
A view along the *b* axis of the crystal packing of the title compound. Hydrogen bonds are drawn as dashed lines.

Table 1
Hydrogen-bond geometry (Å, °).

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
N3—H3A···N2 ⁱ	0.86	2.12	2.978 (3)	172
N3—H3B···N1 ⁱⁱ	0.86	2.17	2.973 (3)	156
N3—H3B···N2 ⁱⁱ	0.86	2.58	3.319 (3)	144
N11—H11···O1 ⁱⁱⁱ	0.86	2.09	2.926 (3)	164

Symmetry codes: (i) $-x, -y + 2, -z + 1$; (ii) $x, -y + \frac{3}{2}, z - \frac{1}{2}$; (iii) $x, y - 1, z$.

Table 2
Experimental details.

Crystal data	
Chemical formula	C ₁₀ H ₉ CIN ₄ OS ₂
<i>M_r</i>	300.78
Crystal system, space group	Monoclinic, <i>P</i> 2 ₁ / <i>c</i>
Temperature (K)	293
<i>a</i> , <i>b</i> , <i>c</i> (Å)	26.379 (5), 4.7704 (8), 10.1866 (18)
β (°)	96.996 (7)
<i>V</i> (Å ³)	1272.3 (4)
<i>Z</i>	4
Radiation type	Mo <i>K</i> α
μ (mm ⁻¹)	0.62
Crystal size (mm)	0.32 × 0.23 × 0.12
Data collection	
Diffractometer	Rigaku Saturn724+
Absorption correction	Multi-scan (<i>NUMABS</i> ; Rigaku 1999)
<i>T_{min}</i> , <i>T_{max}</i>	0.843, 0.928
No. of measured, independent and observed [<i>I</i> > 2 σ (<i>I</i>)] reflections	10559, 2881, 2019
<i>R_{int}</i>	0.066
(<i>sin</i> θ / λ) _{max} (Å ⁻¹)	0.649
Refinement	
<i>R</i> [<i>F</i> ² > 2 σ (<i>F</i> ²)], <i>wR</i> (<i>F</i> ²), <i>S</i>	0.051, 0.127, 1.06
No. of reflections	2881
No. of parameters	163
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\max}$, $\Delta\rho_{\min}$ (e Å ⁻³)	0.34, -0.34

Computer programs: *CrystalClear SM Expert* (Rigaku, 2011), *SHELXS97* (Sheldrick, 2008), *SHELXL2014* (Sheldrick, 2015), *Mercury* (Macrae *et al.*, 2008), *OLEX2* (Dolomanov *et al.*, 2009).

Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

Acknowledgements

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full crystallographic data

IUCrData (2016). **1**, x161139 [<https://doi.org/10.1107/S2414314616011391>]

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$C_{10}H_9ClN_4OS_2$

$M_r = 300.78$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 26.379$ (5) Å

$b = 4.7704$ (8) Å

$c = 10.1866$ (18) Å

$\beta = 96.996$ (7)°

$V = 1272.3$ (4) Å³

$Z = 4$

$F(000) = 616$

$D_x = 1.570$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71075$ Å

Cell parameters from 2881 reflections

$\theta = 3.1$ – 27.5 °

$\mu = 0.62$ mm⁻¹

$T = 293$ K

Block, brown

$0.32 \times 0.23 \times 0.12$ mm

Data collection

Rigaku Saturn724+

diffractometer

profile data from ω -scans

Absorption correction: multi-scan

(NUMABS; Rigaku 1999)

$T_{\min} = 0.843$, $T_{\max} = 0.928$

10559 measured reflections

2881 independent reflections

2019 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.066$

$\theta_{\max} = 27.5$ °, $\theta_{\min} = 3.1$ °

$h = -34 \rightarrow 34$

$k = -6 \rightarrow 5$

$l = -13 \rightarrow 13$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.051$

$wR(F^2) = 0.127$

$S = 1.06$

2881 reflections

163 parameters

0 restraints

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\Sigma^2(F_o^2) + (0.049P)^2 + 0.4848P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.34$ e Å⁻³

$\Delta\rho_{\min} = -0.34$ e Å⁻³

Special details

Geometry. Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell esds are taken into account in the estimation of distances, angles and torsion angles

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.45400 (4)	0.6782 (3)	0.30886 (13)	0.1062 (6)
S1	0.10780 (3)	0.59735 (16)	0.36162 (7)	0.0389 (2)
S2	0.15876 (3)	0.17391 (16)	0.55874 (7)	0.0409 (2)
O1	0.24709 (8)	0.7463 (4)	0.6213 (2)	0.0543 (8)
N1	0.08143 (8)	0.5253 (5)	0.5913 (2)	0.0386 (8)
N2	0.04937 (8)	0.7301 (5)	0.5344 (2)	0.0372 (7)
N3	0.03064 (8)	0.9734 (6)	0.3363 (2)	0.0422 (8)
N11	0.27991 (9)	0.3229 (5)	0.5821 (3)	0.0476 (9)
C1	0.05795 (10)	0.7883 (6)	0.4124 (2)	0.0331 (8)
C2	0.11406 (10)	0.4409 (6)	0.5160 (2)	0.0348 (9)
C3	0.20296 (10)	0.3442 (6)	0.6846 (3)	0.0394 (9)
C4	0.24508 (10)	0.4915 (6)	0.6270 (3)	0.0382 (9)
C5	0.32186 (11)	0.4125 (6)	0.5165 (3)	0.0435 (10)
C6	0.31545 (12)	0.6126 (7)	0.4193 (3)	0.0554 (11)
C7	0.35637 (14)	0.6942 (9)	0.3561 (4)	0.0704 (14)
C8	0.40275 (13)	0.5755 (9)	0.3893 (4)	0.0638 (14)
C9	0.40974 (13)	0.3734 (9)	0.4832 (4)	0.0719 (15)
C10	0.36869 (13)	0.2904 (8)	0.5480 (4)	0.0654 (16)
H3A	0.00600	1.06190	0.36610	0.0510*
H3B	0.03760	1.00480	0.25730	0.0510*
H3C	0.18440	0.47870	0.73200	0.0470*
H3D	0.21750	0.20480	0.74750	0.0470*
H6	0.28350	0.69330	0.39610	0.0670*
H7	0.35210	0.83100	0.29060	0.0840*
H9	0.44160	0.29080	0.50410	0.0860*
H10	0.37310	0.15200	0.61270	0.0790*
H11	0.27660	0.14560	0.59390	0.0570*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0674 (7)	0.1549 (13)	0.1040 (9)	-0.0259 (7)	0.0419 (6)	0.0066 (8)
S1	0.0432 (4)	0.0475 (5)	0.0289 (3)	0.0042 (3)	0.0162 (3)	-0.0013 (3)
S2	0.0409 (4)	0.0356 (4)	0.0466 (4)	-0.0003 (3)	0.0076 (3)	-0.0044 (3)
O1	0.0574 (13)	0.0294 (12)	0.0784 (17)	0.0021 (10)	0.0176 (12)	-0.0010 (11)
N1	0.0427 (13)	0.0462 (15)	0.0290 (11)	0.0023 (11)	0.0127 (10)	0.0006 (11)
N2	0.0389 (12)	0.0488 (15)	0.0259 (11)	0.0054 (10)	0.0127 (9)	0.0016 (10)
N3	0.0479 (14)	0.0551 (17)	0.0260 (11)	0.0120 (12)	0.0140 (10)	0.0046 (11)
N11	0.0482 (15)	0.0326 (14)	0.0654 (17)	-0.0006 (11)	0.0206 (12)	0.0012 (13)
C1	0.0339 (13)	0.0420 (17)	0.0251 (12)	-0.0035 (12)	0.0105 (10)	-0.0049 (12)
C2	0.0353 (13)	0.0394 (17)	0.0315 (14)	-0.0064 (12)	0.0109 (11)	-0.0019 (12)
C3	0.0434 (15)	0.0429 (18)	0.0328 (14)	0.0029 (13)	0.0086 (12)	0.0016 (13)
C4	0.0378 (14)	0.0358 (17)	0.0399 (15)	0.0025 (12)	0.0000 (12)	-0.0009 (13)
C5	0.0424 (16)	0.0399 (18)	0.0497 (18)	-0.0041 (13)	0.0116 (13)	-0.0058 (15)
C6	0.0480 (18)	0.065 (2)	0.054 (2)	-0.0009 (16)	0.0099 (15)	0.0080 (18)

C7	0.066 (2)	0.087 (3)	0.060 (2)	-0.006 (2)	0.0153 (18)	0.020 (2)
C8	0.051 (2)	0.082 (3)	0.061 (2)	-0.0152 (19)	0.0178 (17)	-0.006 (2)
C9	0.0408 (18)	0.091 (3)	0.086 (3)	0.0049 (19)	0.0159 (18)	-0.003 (2)
C10	0.050 (2)	0.072 (3)	0.077 (3)	0.0151 (18)	0.0187 (18)	0.019 (2)

Geometric parameters (Å, °)

C11—C8	1.735 (4)	N3—H3A	0.8600
S1—C1	1.730 (3)	C5—C10	1.368 (5)
S1—C2	1.730 (2)	C5—C6	1.371 (4)
S2—C2	1.754 (3)	C6—C7	1.379 (5)
S2—C3	1.816 (3)	C7—C8	1.353 (5)
O1—C4	1.218 (3)	C8—C9	1.355 (6)
N1—N2	1.373 (3)	C9—C10	1.393 (5)
N1—C2	1.285 (3)	N11—H11	0.8600
N2—C1	1.320 (3)	C3—H3C	0.9700
N3—C1	1.328 (4)	C3—H3D	0.9700
N11—C4	1.343 (4)	C6—H6	0.9300
N11—C5	1.426 (4)	C7—H7	0.9300
C3—C4	1.494 (4)	C9—H9	0.9300
N3—H3B	0.8600	C10—H10	0.9300
C1—S1—C2	86.98 (12)	C6—C7—C8	120.0 (4)
C2—S2—C3	101.82 (13)	C11—C8—C9	119.2 (3)
N2—N1—C2	113.2 (2)	C11—C8—C7	119.8 (3)
N1—N2—C1	112.3 (2)	C7—C8—C9	121.0 (4)
C4—N11—C5	125.6 (2)	C8—C9—C10	119.4 (3)
S1—C1—N2	113.3 (2)	C5—C10—C9	120.0 (3)
S1—C1—N3	123.62 (17)	C4—N11—H11	117.00
N2—C1—N3	123.1 (2)	C5—N11—H11	117.00
S1—C2—S2	121.71 (14)	S2—C3—H3C	109.00
S1—C2—N1	114.2 (2)	S2—C3—H3D	109.00
S2—C2—N1	123.92 (18)	C4—C3—H3C	109.00
S2—C3—C4	112.2 (2)	C4—C3—H3D	109.00
H3A—N3—H3B	120.00	H3C—C3—H3D	108.00
C1—N3—H3B	120.00	C5—C6—H6	120.00
C1—N3—H3A	120.00	C7—C6—H6	120.00
N11—C4—C3	115.1 (2)	C6—C7—H7	120.00
O1—C4—C3	121.8 (3)	C8—C7—H7	120.00
O1—C4—N11	123.1 (3)	C8—C9—H9	120.00
N11—C5—C6	120.8 (3)	C10—C9—H9	120.00
N11—C5—C10	119.5 (3)	C5—C10—H10	120.00
C6—C5—C10	119.6 (3)	C9—C10—H10	120.00
C5—C6—C7	120.0 (3)		
C2—S1—C1—N2	0.4 (2)	C4—N11—C5—C6	-44.4 (5)
C2—S1—C1—N3	-179.1 (3)	C4—N11—C5—C10	137.8 (3)
C1—S1—C2—S2	176.98 (19)	S2—C3—C4—O1	107.9 (3)

C1—S1—C2—N1	1.0 (2)	S2—C3—C4—N11	-71.8 (3)
C3—S2—C2—S1	112.82 (18)	N11—C5—C6—C7	-179.3 (3)
C3—S2—C2—N1	-71.6 (3)	C10—C5—C6—C7	-1.5 (5)
C2—S2—C3—C4	-89.6 (2)	N11—C5—C10—C9	179.1 (3)
C2—N1—N2—C1	2.3 (3)	C6—C5—C10—C9	1.2 (5)
N2—N1—C2—S1	-2.1 (3)	C5—C6—C7—C8	0.5 (6)
N2—N1—C2—S2	-177.95 (19)	C6—C7—C8—C11	179.7 (3)
N1—N2—C1—S1	-1.5 (3)	C6—C7—C8—C9	0.9 (6)
N1—N2—C1—N3	177.9 (2)	C11—C8—C9—C10	180.0 (3)
C5—N11—C4—O1	-3.4 (5)	C7—C8—C9—C10	-1.1 (6)
C5—N11—C4—C3	176.3 (3)	C8—C9—C10—C5	0.1 (6)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
N3—H3 <i>A</i> ...N2 ⁱ	0.86	2.12	2.978 (3)	172
N3—H3 <i>B</i> ...N1 ⁱⁱ	0.86	2.17	2.973 (3)	156
N3—H3 <i>B</i> ...N2 ⁱⁱ	0.86	2.58	3.319 (3)	144
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