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## 3-(4-Iodophenyl)pentanedinitrile

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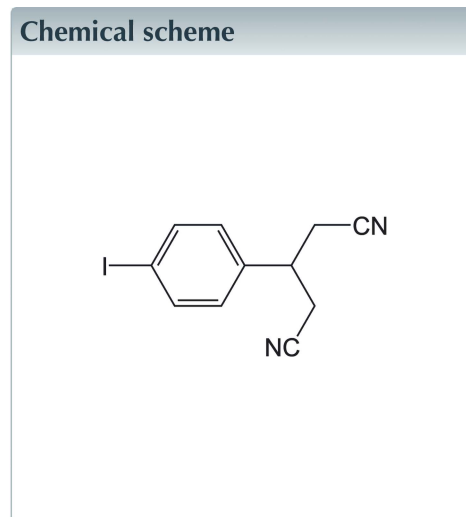
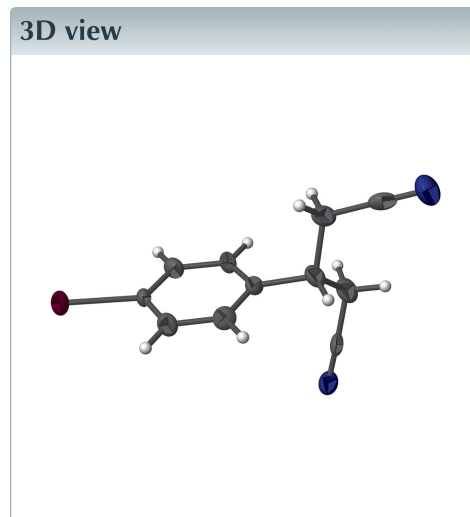
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Keywords: crystal structure; pentanedinitrile, glutaronitrile; hydrogen bond; halogen bond.

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Structural data: full structural data are available from [iucrdata.iucr.org](http://iucrdata.iucr.org)

In the title pentanedinitrile derivative,  $C_{11}H_9IN_2$ , the iodophenyl group is connected at the 3-position. The central propylene chain of the pentanedinitrile moiety contains one *gauche* conformation as a result of steric repulsion with the phenyl ring. Intermolecular close contacts in the crystal comprise a weak  $Csp^3-H \cdots N$  hydrogen bond and a  $C-I \cdots N$  halogen bond.

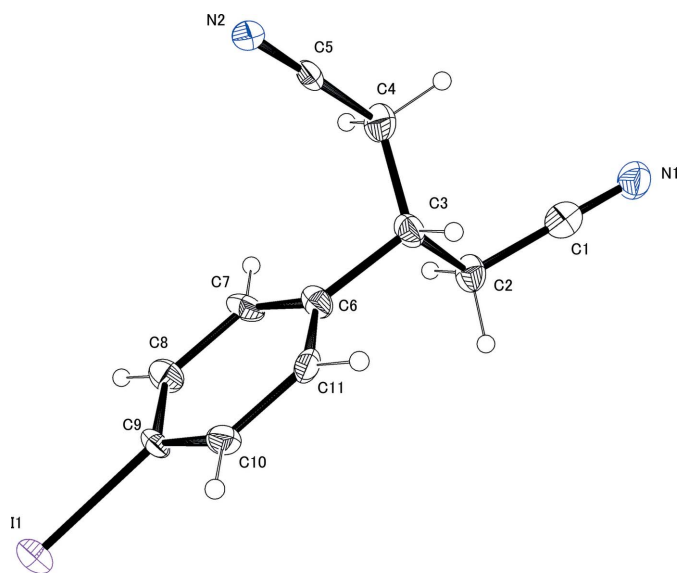


### Structure description

The title compound,  $C_{11}H_9IN_2$ , is a pentanedinitrile (glutaronitrile) derivative, in which the iodophenyl group is connected at the 3-position. Pentanedinitrile derivatives are used as precursors in the synthesis of 2,6-diaminopyridines, which are used as raw materials for insecticides (Kato *et al.*, 1989).

The title compound (Fig. 1) was obtained in a condensation reaction between 4-iodobenzaldehyde and cyanoacetic acid. The central propylene chain of the pentanedinitrile group contains one *gauche* conformation where the torsion angle of  $C1-C2-C3-C4$  is  $161.3(9)^\circ$ . Similarly, the conformations of related compounds (see *e.g.* Al-Arab *et al.*, 1988; Lorente *et al.*, 1995; Percino *et al.*, 2014) also do not show all-*anti* conformations. In the title compound, steric repulsion between the  $C1-N1$  cyano group and the phenyl ring is thought to result in the *gauche* conformation.

In the crystal, intermolecular halogen bonds are formed between inversion-related molecules to give a dimeric structure (Fig. 2), where the distance of  $C-I \cdots N^i$  [symmetry code: (i)  $-x + 1, -y, -z + 1$ ] is  $3.369(7)$  Å and the  $C-I \cdots N^i$  angle is  $166.2(2)^\circ$ . Here, the intermolecular distance is shorter by 4.6% than the sum of the van der Waals radii of the nitrogen and iodine atoms, such that this halogen bond is classified as a weak interaction. Pairs of weak intermolecular  $Csp^3-H \cdots N^i$  hydrogen bonds also form between inversion-related dimers (Fig. 2, Table 1).

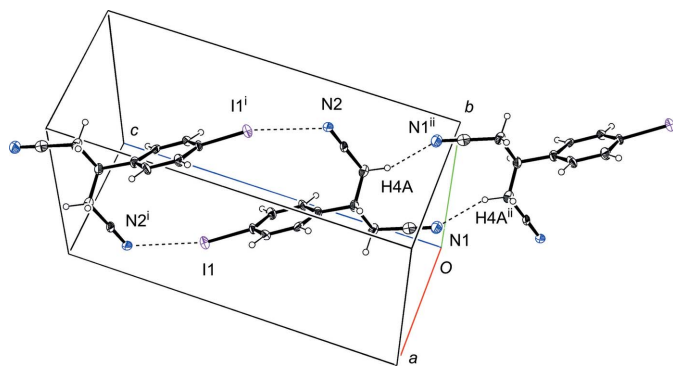


**Figure 1**  
The molecular structure of the title compound with displacement ellipsoids drawn at the 50% probability level, with H atoms shown as small spheres.

### Synthesis and crystallization

A solution of 4-iodobenzaldehyde (1.0 g, 4.3 mmol) and cyanoacetic acid (0.73 g, 8.6 mmol) in acetonitrile (30 ml) was refluxed overnight with piperidine (0.22 ml) as a basic catalyst. The solution was condensed using a rotary evaporator, and the residual yellowish oil was dissolved in chloroform (90 ml). The solution was washed with  $\text{HCl}_{\text{aq}}$  (1.0 M) and water. It was dried over  $\text{Na}_2\text{SO}_4$  and condensed under reduced pressure. The residue was purified by recrystallization from a hexane-ethanol ( $v:v = 7:1$ ) solution to give 3-(4-iodophenyl)pentanedinitrile in a yield of 41% (0.52 g).

$^1\text{H NMR}$  (400 MHz):  $\delta$  7.73 (*d*,  $J = 8.4$  Hz, 2H), 7.19 (*d*,  $J = 8.4$  Hz, 2H), 3.39 (*m*, 1H), 2.91 (*d*,  $J = 7.4$  Hz, 4H)



**Figure 2**  
A view of the intermolecular interactions in the title compound [symmetry codes: (i)  $-x + 1, -y, -z + 1$ ; (ii)  $-x, -y, -z$ ].

**Table 1**  
Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C4-H4B \cdots N1^i$	0.99	2.61	3.387 (12)	135

Symmetry code: (i)  $-x, -y, -z$ .

**Table 2**  
Experimental details.

Crystal data	
Chemical formula	$\text{C}_{11}\text{H}_9\text{IN}_2$
$M_r$	296.11
Crystal system, space group	Monoclinic, $P2_1/c$
Temperature (K)	93
$a, b, c$ ( $\text{\AA}$ )	6.508 (4), 10.524 (6), 16.047 (10)
$\beta$ ( $^\circ$ )	90.507 (10)
$V$ ( $\text{\AA}^3$ )	1099.0 (11)
$Z$	4
Radiation type	Mo $K\alpha$
$\mu$ ( $\text{mm}^{-1}$ )	2.88
Crystal size (mm)	$0.15 \times 0.10 \times 0.02$
Data collection	
Diffractometer	Rigaku Saturn724+
Absorption correction	Numerical (NUMABS; Rigaku, 1999)
$T_{\text{min}}, T_{\text{max}}$	0.744, 0.944
No. of measured, independent and observed [ $F^2 > 2.0\sigma(F^2)$ ] reflections	7204, 1920, 1618
$R_{\text{int}}$	0.065
$(\sin \theta/\lambda)_{\text{max}}$ ( $\text{\AA}^{-1}$ )	0.595
Refinement	
$R[F^2 > 2\sigma(F^2)], wR(F^2), S$	0.068, 0.095, 1.21
No. of reflections	1920
No. of parameters	127
H-atom treatment	H-atom parameters constrained
$\Delta\rho_{\text{max}}, \Delta\rho_{\text{min}}$ ( $\text{e \AA}^{-3}$ )	1.02, $-0.83$

Computer programs: *CrystalClear* (Rigaku, 2008), *SIR92* (Altomare *et al.*, 1994), *SHELXL2013* (Sheldrick, 2015), *ORTEP-3 for Windows* (Farrugia, 2012) and *CrystalStructure* (Rigaku, 2014).

### Refinement

Crystal data, data collection and structure refinement details are summarized in Table 2.

### References

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## full crystallographic data

*IUCrData* (2019). 4, x190282 [https://doi.org/10.1107/S2414314619002827]

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*Crystal data*

$C_{11}H_9IN_2$

$M_r = 296.11$

Monoclinic,  $P2_1/c$

$a = 6.508$  (4) Å

$b = 10.524$  (6) Å

$c = 16.047$  (10) Å

$\beta = 90.507$  (10)°

$V = 1099.0$  (11) Å<sup>3</sup>

$Z = 4$

$F(000) = 568.00$

$D_x = 1.790$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71075$  Å

Cell parameters from 2574 reflections

$\theta = 2.5$ – $25.0$ °

$\mu = 2.88$  mm<sup>-1</sup>

$T = 93$  K

Prism, colorless

$0.15 \times 0.10 \times 0.02$  mm

*Data collection*

Rigaku Saturn724+  
diffractometer

Detector resolution: 7.111 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: numerical  
(NUMABS; Rigaku, 1999)

$T_{\min} = 0.744$ ,  $T_{\max} = 0.944$

7204 measured reflections

1920 independent reflections

1618 reflections with  $F^2 > 2.0\sigma(F^2)$

$R_{\text{int}} = 0.065$

$\theta_{\max} = 25.0$ °,  $\theta_{\min} = 3.1$ °

$h = -7 \rightarrow 6$

$k = -12 \rightarrow 12$

$l = -18 \rightarrow 19$

*Refinement*

Refinement on  $F^2$

$R[F^2 > 2\sigma(F^2)] = 0.068$

$wR(F^2) = 0.095$

$S = 1.21$

1920 reflections

127 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.P)^2 + 7.5421P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 1.02$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.83$  e Å<sup>-3</sup>

*Special details*

**Geometry.** ENTER SPECIAL DETAILS OF THE MOLECULAR GEOMETRY

**Refinement.** Refinement was performed using all reflections. The weighted R-factor (wR) and goodness of fit (S) are based on  $F^2$ . R-factor (gt) are based on F. The threshold expression of  $F^2 > 2.0 \sigma(F^2)$  is used only for calculating R-factor (gt).

The C-bound H atoms were placed at ideal positions and were refined as riding on their parent C atoms.  $U_{\text{iso}}(\text{H})$  values of the H atoms were set at  $1.2U_{\text{eq}}(\text{parent atom for C})$ .

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
I1	0.77853 (9)	0.16532 (6)	0.51640 (4)	0.0351 (2)
N1	0.2457 (12)	0.0590 (8)	-0.0356 (5)	0.045 (2)
N2	0.0921 (11)	-0.2452 (7)	0.2869 (4)	0.0362 (19)
C1	0.2949 (14)	0.0783 (8)	0.0303 (6)	0.035 (2)
C2	0.3609 (14)	0.1037 (8)	0.1178 (5)	0.035 (2)
C3	0.3499 (13)	-0.0126 (9)	0.1745 (5)	0.034 (2)
C4	0.1348 (14)	-0.0614 (9)	0.1804 (5)	0.041 (2)
C5	0.1166 (12)	-0.1689 (9)	0.2395 (5)	0.0301 (19)
C6	0.4440 (13)	0.0237 (8)	0.2588 (5)	0.030 (2)
C7	0.6281 (12)	-0.0277 (8)	0.2834 (5)	0.030 (2)
C8	0.7244 (13)	0.0096 (9)	0.3576 (5)	0.032 (2)
C9	0.6311 (12)	0.0997 (8)	0.4067 (5)	0.0248 (19)
C10	0.4428 (13)	0.1489 (9)	0.3852 (5)	0.035 (2)
C11	0.3499 (13)	0.1095 (8)	0.3110 (5)	0.034 (2)
H2A	0.2728	0.1714	0.1411	0.0425*
H2B	0.50393	0.13561	0.11769	0.0425*
H3	0.43663	-0.08104	0.14959	0.0402*
H4A	0.04378	0.00848	0.19859	0.0497*
H4B	0.08771	-0.08939	0.12454	0.0497*
H7	0.69157	-0.08991	0.24927	0.0363*
H8	0.85222	-0.02667	0.37399	0.0389*
H10	0.37682	0.20873	0.42025	0.0416*
H11	0.21879	0.14254	0.29604	0.0412*

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
I1	0.0355 (3)	0.0376 (3)	0.0322 (3)	0.0030 (3)	-0.0112 (2)	-0.0077 (3)
N1	0.041 (5)	0.064 (6)	0.030 (5)	0.002 (4)	0.000 (4)	0.002 (4)
N2	0.040 (5)	0.040 (5)	0.029 (4)	-0.017 (4)	0.000 (3)	0.001 (4)
C1	0.036 (5)	0.020 (5)	0.047 (6)	0.009 (4)	0.008 (5)	0.005 (4)
C2	0.046 (6)	0.032 (5)	0.028 (5)	0.000 (4)	0.003 (4)	0.006 (4)
C3	0.039 (5)	0.039 (5)	0.023 (4)	-0.001 (4)	-0.005 (4)	0.006 (4)
C4	0.047 (6)	0.051 (6)	0.025 (5)	-0.006 (5)	-0.004 (4)	0.007 (4)
C5	0.037 (5)	0.033 (5)	0.020 (4)	-0.011 (5)	-0.007 (4)	-0.007 (4)
C6	0.044 (6)	0.025 (5)	0.021 (4)	-0.004 (4)	-0.007 (4)	0.002 (4)
C7	0.032 (5)	0.035 (5)	0.024 (5)	-0.002 (4)	0.011 (4)	-0.001 (4)
C8	0.029 (5)	0.038 (5)	0.030 (5)	-0.004 (4)	-0.005 (4)	0.003 (4)
C9	0.031 (5)	0.022 (4)	0.020 (4)	-0.001 (4)	-0.009 (4)	-0.002 (4)
C10	0.035 (5)	0.032 (5)	0.037 (5)	0.003 (4)	-0.013 (4)	0.002 (4)
C11	0.029 (5)	0.035 (5)	0.039 (5)	-0.008 (4)	-0.020 (4)	0.005 (4)

*Geometric parameters (Å, °)*

I1—C9	2.113 (8)	C9—C10	1.371 (11)
N1—C1	1.120 (12)	C10—C11	1.393 (12)
N2—C5	1.118 (11)	C2—H2A	0.990
C1—C2	1.489 (13)	C2—H2B	0.990
C2—C3	1.526 (12)	C3—H3	1.000
C3—C4	1.495 (12)	C4—H4A	0.990
C3—C6	1.529 (11)	C4—H4B	0.990
C4—C5	1.482 (13)	C7—H7	0.950
C6—C7	1.369 (12)	C8—H8	0.950
C6—C11	1.379 (12)	C10—H10	0.950
C7—C8	1.398 (11)	C11—H11	0.950
C8—C9	1.378 (12)		
N1—C1—C2	179.8 (9)	C3—C2—H2A	108.805
C1—C2—C3	113.8 (7)	C3—C2—H2B	108.800
C2—C3—C4	111.2 (7)	H2A—C2—H2B	107.673
C2—C3—C6	107.9 (7)	C2—C3—H3	108.068
C4—C3—C6	113.4 (7)	C4—C3—H3	108.070
C3—C4—C5	112.6 (7)	C6—C3—H3	108.068
N2—C5—C4	175.0 (9)	C3—C4—H4A	109.087
C3—C6—C7	119.9 (7)	C3—C4—H4B	109.085
C3—C6—C11	121.6 (7)	C5—C4—H4A	109.082
C7—C6—C11	118.5 (7)	C5—C4—H4B	109.080
C6—C7—C8	121.3 (8)	H4A—C4—H4B	107.844
C7—C8—C9	119.0 (8)	C6—C7—H7	119.373
I1—C9—C8	120.1 (6)	C8—C7—H7	119.368
I1—C9—C10	119.0 (6)	C7—C8—H8	120.514
C8—C9—C10	120.9 (7)	C9—C8—H8	120.523
C9—C10—C11	118.9 (8)	C9—C10—H10	120.539
C6—C11—C10	121.4 (8)	C11—C10—H10	120.550
C1—C2—H2A	108.800	C6—C11—H11	119.287
C1—C2—H2B	108.802	C10—C11—H11	119.285
C1—C2—C3—C4	-61.3 (9)	C3—C6—C11—C10	-175.6 (7)
C1—C2—C3—C6	173.8 (6)	C7—C6—C11—C10	3.2 (12)
C2—C3—C4—C5	-176.1 (6)	C11—C6—C7—C8	-2.9 (12)
C2—C3—C6—C7	-110.6 (8)	C6—C7—C8—C9	0.1 (12)
C2—C3—C6—C11	68.2 (9)	C7—C8—C9—I1	-177.0 (6)
C4—C3—C6—C7	125.8 (8)	C7—C8—C9—C10	2.6 (12)
C4—C3—C6—C11	-55.4 (10)	I1—C9—C10—C11	177.3 (5)
C6—C3—C4—C5	-54.3 (9)	C8—C9—C10—C11	-2.3 (12)
C3—C6—C7—C8	175.9 (7)	C9—C10—C11—C6	-0.7 (12)

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*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C4—H4B···N1 <sup>i</sup>	0.99	2.61	3.387 (12)	135

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Symmetry code: (i)  $-x, -y, -z$ .